# Connecting via Winsock to STN

Welcome to STN International! Enter x:x

LOGINID:ssspta1204bxd

PASSWORD:

TERMINAL (ENTER 1, 2, 3, OR ?):2

```
Welcome to STN International
                 Web Page URLs for STN Seminar Schedule - N. America
NEWS
                 "Ask CAS" for self-help around the clock
NEWS
                 EXTEND option available in structure searching
NEWS
         May 12
                 Polymer links for the POLYLINK command completed in REGISTRY
         May 12
NEWS
                 New UPM (Update Code Maximum) field for more efficient patent
NEWS
         May 27
                 SDIs in CAplus
                 CAplus super roles and document types searchable in REGISTRY
         May 27
NEWS
                 Additional enzyme-catalyzed reactions added to CASREACT
      7
         Jun 28
NEWS
         Jun 28
                 ANTE, AQUALINE, BIOENG, CIVILENG, ENVIROENG, MECHENG,
NEWS
      8
                 and WATER from CSA now available on STN(R)
         Jul 12
                 BEILSTEIN enhanced with new display and select options,
NEWS
                 resulting in a closer connection to BABS
                 BEILSTEIN on STN workshop to be held August 24 in conjunction
NEWS 10
         Jul 30
                 with the 228th ACS National Meeting
                 IFIPAT/IFIUDB/IFICDB reloaded with new search and display
NEWS 11
         AUG 02
                 fields
                 CAplus and CA patent records enhanced with European and Japan
NEWS 12
         AUG 02
                 Patent Office Classifications
                 STN User Update to be held August 22 in conjunction with the
NEWS 13
         AUG 02
                 228th ACS National Meeting
                 The Analysis Edition of STN Express with Discover!
         AUG 02
NEWS 14
                 (Version 7.01 for Windows) now available
                 Pricing for the Save Answers for SciFinder Wizard within
NEWS 15
         AUG 04
                 STN Express with Discover! will change September 1, 2004
NEWS 16
         AUG 27
                 BIOCOMMERCE: Changes and enhancements to content coverage
NEWS 17
         AUG 27
                 BIOTECHABS/BIOTECHDS: Two new display fields added for legal
                 status data from INPADOC
                 INPADOC: New family current-awareness alert (SDI) available
NEWS 18
         SEP 01
                 New pricing for the Save Answers for SciFinder Wizard within
NEWS 19
         SEP 01
                 STN Express with Discover!
                 New display format, HITSTR, available in WPIDS/WPINDEX/WPIX
NEWS 20
         SEP 01
                 STN Patent Forum to be held October 13, 2004, in Iselin, NJ
NEWS 21
         SEP 14
              JULY 30 CURRENT WINDOWS VERSION IS V7.01, CURRENT
NEWS EXPRESS
              MACINTOSH VERSION IS V6.0c(ENG) AND V6.0Jc(JP),
              AND CURRENT DISCOVER FILE IS DATED 11 AUGUST 2004
              STN Operating Hours Plus Help Desk Availability
NEWS HOURS
NEWS INTER
              General Internet Information
              Welcome Banner and News Items
NEWS LOGIN
              Direct Dial and Telecommunication Network Access to STN
NEWS PHONE
              CAS World Wide Web Site (general information)
NEWS WWW
```

Enter NEWS followed by the item number or name to see news on that specific topic.

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FILE 'HOME' ENTERED AT 16:17:05 ON 14 SEP 2004

=> fil caplus
COST IN U.S. DOLLARS

SINCE FILE TOTAL ENTRY SESSION 0.21 0.21

FULL ESTIMATED COST

FILE 'CAPLUS' ENTERED AT 16:17:17 ON 14 SEP 2004 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2004 AMERICAN CHEMICAL SOCIETY (ACS)

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FILE COVERS 1907 - 14 Sep 2004 VOL 141 ISS 12 FILE LAST UPDATED: 13 Sep 2004 (20040913/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> s cataly?

L1 1207996 CATALY?

=> s reduct?

424200 REDUCT?

821303 REDN

44517 REDNS

849207 REDN

(REDN OR REDNS)

L2 1087400 REDUCT?

(REDUCT? OR REDN)

=> s hydrogenat?

L3 256764 HYDROGENAT?

=> s 12 or 13

L4 1299917 L2 OR L3

=> s l1 and l4

L5 224466 L1 AND L4

=> s Ni or nickel

573448 NI

3631 NIS

575622 NI

(NI OR NIS)

558300 NICKEL

189 NICKELS

558327 NICKEL

(NICKEL OR NICKELS)

L6 775627 NI OR NICKEL

=> s copper or cu

816969 COPPER

415 COPPERS

817032 COPPER

(COPPER OR COPPERS)

763176 CU

4661 CUS

765163 CU

(CU OR CUS)

L7 1091254 COPPER OR CU

=> s chromium or cr

333603 CHROMIUM

74 CHROMIUMS

333606 CHROMIUM

(CHROMIUM OR CHROMIUMS)

375850 CR

1549 CRS

376837 CR

(CR OR CRS)

L8 513944 CHROMIUM OR CR

75% OF LIMIT FOR TOTAL ANSWERS REACHED

=> s 15 and 16

L9 40533 L5 AND L6

=> s regen?

L10 162143 REGEN?

=> s 19 and 110

L11 1162 L9 AND L10

=> s 111 and 17

L12 239 L11 AND L7

=> fil reg

COST IN U.S. DOLLARS

SINCE FILE TOTAL

ENTRY SESSION

FULL ESTIMATED COST 21.72 21.93

FILE 'REGISTRY' ENTERED AT 16:21:50 ON 14 SEP 2004 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT.

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Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 13 SEP 2004 HIGHEST RN 744170-41-0 DICTIONARY FILE UPDATES: 13 SEP 2004 HIGHEST RN 744170-41-0

TSCA INFORMATION NOW CURRENT THROUGH MAY 21, 2004

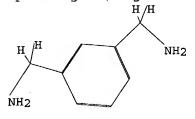
Please note that search-term pricing does apply when conducting SmartSELECT searches.

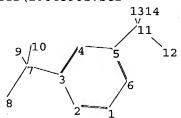
Crossover limits have been increased. See HELP CROSSOVER for details.

Experimental and calculated property data are now available. For more information enter HELP PROP at an arrow prompt in the file or refer to the file summary sheet on the web at: http://www.cas.org/ONLINE/DBSS/registryss.html

=>

Uploading C:\Program Files\Stnexp\Queries\10643981.str





chain nodes :

7 8 9 10 11 12 13 14

ring nodes : 1 2 3 4 5 6 chain bonds :

3-7 5-11 7-8 7-9 7-10 11-12 11-13 11-14

ring bonds :

1-2 1-6 2-3 3-4 4-5 5-6

exact/norm bonds :

7-8 11-12 exact bonds :

3-7 5-11 7-9 7-10 11-13 11-14

normalized bonds :

1-2 1-6 2-3 3-4 4-5 5-6

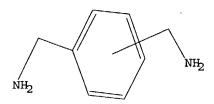
Match level :

1:Atom 2:Atom 3:Atom 4:Atom 5:Atom 6:Atom 7:CLASS 8:CLASS 9:CLASS 10:CLASS 11:CLASS 12:CLASS 13:CLASS 14:CLASS

#### L13 STRUCTURE UPLOADED

=> d query

L13 STR



Structure attributes must be viewed using STN Express query preparation.

=> s 113

SAMPLE SEARCH INITIATED 16:22:09 FILE 'REGISTRY'

Page 4

## SAMPLE SCREEN SEARCH COMPLETED - 182573 TO ITERATE

0.5% PROCESSED 1000 ITERATIONS

INCOMPLETE SEARCH (SYSTEM LIMIT EXCEEDED)

SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE \*\*INCOMPLETE\*\*

BATCH \*\*INCOMPLETE\*\*

PROJECTED ITERATIONS: EXCEEDS 1000000 PROJECTED ANSWERS: EXCEEDS 2841

L14 1 SEA SSS SAM L13

=>
Uploading C:\Program Files\Stnexp\Queries\10643981.str

H H NH2

1 ANSWERS

chain nodes :

7 8 9 10 11 12 13 14

ring nodes :
1 2 3 4 5 6
chain bonds :

3-7 5-11 7-8 7-9 7-10 11-12 11-13 11-14

ring bonds :

1-2 1-6 2-3 3-4 4-5 5-6

exact/norm bonds :

7-8 11-12

exact bonds :

3-7 5-11 7-9 7-10 11-13 11-14

normalized bonds :

1-2 1-6 2-3 3-4 4-5 5-6

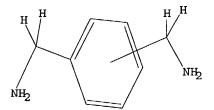
Match level :

1:Atom 2:Atom 3:Atom 4:Atom 5:Atom 6:Atom 7:CLASS 8:CLASS 9:CLASS 10:CLASS 11:CLASS 12:CLASS 13:CLASS 14:CLASS

L15 STRUCTURE UPLOADED

=> d query

L15 STR



Structure attributes must be viewed using STN Express query preparation.

=> s 115

SAMPLE SEARCH INITIATED 16:23:02 FILE 'REGISTRY' SAMPLE SCREEN SEARCH COMPLETED - 182573 TO ITERATE

0.5% PROCESSED 1000 ITERATIONS

1 ANSWERS

INCOMPLETE SEARCH (SYSTEM LIMIT EXCEEDED)

**SEARCH TIME: 00.00.01** 

FULL FILE PROJECTIONS: ONLINE \*\*INCOMPLETE\*\*

BATCH \*\*INCOMPLETE\*\*

PROJECTED ITERATIONS:

EXCEEDS 1000000

PROJECTED ANSWERS:

EXCEEDS 2841

L16

1 SEA SSS SAM L15

=>

Uploading C:\Program Files\Stnexp\Queries\10643981.str

9 10 4 5 12 8 2 1

chain nodes :

7 8 9 10 11 12 13 14

ring nodes:
1 2 3 4 5 6

chain bonds:

3-7 5-11 7-8 7-9 7-10 11-12 11-13 11-14

ring bonds :

1-2 1-6 2-3 3-4 4-5 5-6

exact/norm bonds :

7-8 11-12

exact bonds :

3-7 5-11 7-9 7-10 11-13 .11-14

normalized bonds :

1-2 1-6 2-3 3-4 4-5 5-6

Match level :

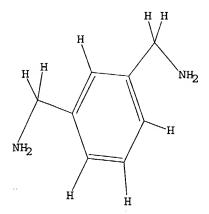
1:Atom 2:Atom 3:Atom 4:Atom 5:Atom 6:Atom 7:CLASS 8:CLASS 9:CLASS 10:CLASS

11:CLASS 12:CLASS 13:CLASS 14:CLASS

L17 STRUCTURE UPLOADED

=> d query

L17 STR



Structure attributes must be viewed using STN Express query preparation.

=> s 117 SAMPLE SEARCH INITIATED 16:24:04 FILE 'REGISTRY' SAMPLE SCREEN SEARCH COMPLETED - 15002 TO ITERATE

6.7% PROCESSED 1000 ITERATIONS
INCOMPLETE SEARCH (SYSTEM LIMIT EXCEEDED)

SEARCH TIME: 00.00.01

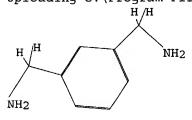
FULL FILE PROJECTIONS: ONLINE \*\*COMPLETE\*\*
BATCH \*\*COMPLETE\*\*

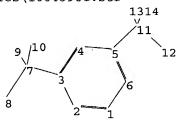
PROJECTED ITERATIONS: 292707 TO 307373 PROJECTED ANSWERS: 1486 TO 2714

L18 7 SEA SSS SAM L17

=>

Uploading C:\Program Files\Stnexp\Queries\10643981.str





7 ANSWERS

chain nodes :

7 8 9 10 11 12 13 14

ring nodes : 1 2 3 4 5 6 chain bonds :

3-7 5-11 7-8 7-9 7-10 11-12 11-13 11-14

ring bonds :

1-2 1-6 2-3 3-4 4-5 5-6

exact/norm bonds :

7-8 11-12

exact bonds :

3-7 5-11 7-9 7-10 11-13 11-14

normalized bonds :

1-2 1-6 2-3 3-4 4-5 5-6

Match level :

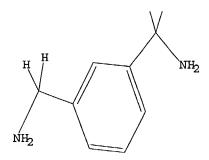
1:Atom 2:Atom 3:Atom 4:Atom 5:Atom 6:Atom 7:CLASS 8:CLASS 9:CLASS 10:CLASS 11:CLASS 12:CLASS 13:CLASS 14:CLASS

#### L19 STRUCTURE UPLOADED

STR

=> d query

L19



Structure attributes must be viewed using STN Express query preparation.

7 ANSWERS

=> s 119

SAMPLE SEARCH INITIATED 16:24:38 FILE 'REGISTRY' SAMPLE SCREEN SEARCH COMPLETED - 15002 TO ITERATE

6.7% PROCESSED 1000 ITERATIONS
INCOMPLETE SEARCH (SYSTEM LIMIT EXCEEDED)

SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE \*\*COMPLETE\*\*

BATCH \*\*COMPLETE\*\*

PROJECTED ITERATIONS: 292707 TO 307373 PROJECTED ANSWERS: 1486 TO 2714

L20 7 SEA SSS SAM L19

=> s 119 full

FULL SEARCH INITIATED 16:24:45 FILE 'REGISTRY'
FULL SCREEN SEARCH COMPLETED - 302263 TO ITERATE

100.0% PROCESSED 302263 ITERATIONS 1647 ANSWERS

SEARCH TIME: 00.00.02

L21 1647 SEA SSS FUL L19

=> fil caplus

COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION

FULL ESTIMATED COST 157.10 179.03

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FILE COVERS 1907 - 14 Sep 2004 VOL 141 ISS 12 FILE LAST UPDATED: 13 Sep 2004 (20040913/ED)
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This file contains CAS Registry Numbers for easy and accurate substance identification.

```
=> s 121
          3471 L21
L22
=> d his
     (FILE 'HOME' ENTERED AT 16:17:05 ON 14 SEP 2004)
     FILE 'CAPLUS' ENTERED AT 16:17:17 ON 14 SEP 2004
        1207996 S CATALY?
L1
        1087400 S REDUCT?
L2
L3
         256764 S HYDROGENAT?
L4
        1299917 S L2 OR L3
         224466 S L1 AND L4
L5
         775627 S NI OR NICKEL
L6
1.7
        1091254 S COPPER OR CU
         513944 S CHROMIUM OR CR
1.8
          40533 S L5 AND L6
1.9
         162143 S REGEN?
T.10
L11
           1162 S L9 AND L10
L12
            239 S L11 AND L7
     FILE 'REGISTRY' ENTERED AT 16:21:50 ON 14 SEP 2004
L13
                STRUCTURE UPLOADED
L14
              1 S L13
                STRUCTURE UPLOADED
L15
L16
              1 S L15
L17
                STRUCTURE UPLOADED
L18
              7 S L17
                STRUCTURE UPLOADED
L19
L20
              7 S L19
L21
           1647 S L19 FULL
     FILE 'CAPLUS' ENTERED AT 16:24:50 ON 14 SEP 2004
L22
           3471 S L21
=> s 122 and 15
            90 L22 AND L5
L23
=> s 123 and 16
L24
            40 L23 AND L6
```

=> s 124 and 110

=> d 125 1-5 abs ibib hitstr

L25 ANSWER 1 OF 5 CAPLUS COPYRIGHT 2004 ACS on STN
XYlylenediamines are prepared by hydrogenating dicyanobenzenes in
a liquid phase in the presence of a catalyst, where the
catalyst having its activity decreased during the course of the
hydrogenation of the dicyanobenzenes is contacted with a
hydrogen-containing gas at 200-500° while controlling a temperature rise
speed of the catalyst to \$40°/min, thus
regenerating the catalyst which can then be reused in
subsequent hydrogenations of the dicyanobenzenes.
ACCESSION NUMBER: 2004:162518 CAPLUS 140:219730 DOCUMENT NUMBER: 140:219730

Rydrogenation process and catalyst
for the production of xylylenediamines from
dicyanobenzenes and a regeneration process
for the catalyst
Amakawa, Razuhiko TITLE:

PATENT ASSIGNEE(S): SOURCE: Japan U.S. Pat. Appl. Publ., 7 pp. CODEN: USXXCO

DOCUMENT TYPE: Patent English FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

INVENTOR(S):

PATENT NO. KIND DATE APPLICATION NO. DATE US 2004039232 A1 20040226 US 2003-643981 20030820
JP 2004107327 A2 2004008 JP 2003-292139 20030812
EP 1394146 A1 20040303 EP 2003-18532 20030816
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LI, NL, SE, MC, PT,
IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK
PRIORITY APPLM. INFO:: JP 2002-245222 A 20020826 A1 A2 A1

нам-сна \_сн<sub>2</sub>-мн<sub>2</sub>

L25 ANSWER 3 OF 5 CAPLUS COPYRIGHT 2004 ACS on STN

Raney Ni ostalyst modified by chemical reduction
was used for liquid phase hydrogenation of isophthalonitrile (IPN)
to prepare m-phenylenediamine (m-NDA). The activity of Raney Ni
catalyst was improved greatly by introducing a and g element as
modifier, and yield of m-NDA was increased from 85.13 to 97.01. Service
lifetime of the catalyst was prolonged from 1-2 times to 5
times, and deactivated catalyst could regain original activity
after regeneration.

ACCESSION NUMBER: 2000:714722 CAPLUS
DOINDER NUMBER: 134:25802

134:268028

DOCUMENT NUMBER: TITLE:

134:268028
Study on catalyst for liquid phase
hydrogenation of isophthalonitrile to prepare
m-xylenediamine
Shen, Qin; Lu, Xiaoyuan; Hou, Minbo; Liu, Zhongneng
Shanghai Petrochemical Institute, Shanghai, 201208, AUTHOR(S): CORPORATE SOURCE:

SOURCE:

Peop. Rep. China Shanghai Huagong (2000), 25(17), 12-15 CODEN: SHAMEZ; ISSN: 1004-017X Shanghai Huagong Bianjibu

PUBLISHER: DOCUMENT TYPE: Journal

DOCUMENT TYPE: Journal
LANGUAGE: Chinese
IT 1477-55-Op, 1,3-Benzenedimethanamine
RL: SPN (Synthetic preparation); PREP (Preparation)
(cocatalyst study on Raney mickel-catalysed
hydrogenation of isophthalonitrile)
RN 1477-55-O CAPUS
CN 1,3-Benzenedimethanamine (9CI) (CA INDEX NAME)

H2N-CH2 \_CH2-NH2

L25 ANSWER 2 OF 5 CAPLUS COPYRIGHT 2004 ACS on STN

Modified Raney Ni catalysts were used for the preparation
of m-xylenediamine (m-XDA) from isophthalonictile (IFN) by
hydrogenation, and effects of solvent, temperature, pressure and amount catalyst were investigated. When mixture of MeOH and PhMe (resp. volume ratio 1:2) was used as solvent, conversion of IPN was nearly 100% and yield of m-XDA reached 97% at mixed solvent/IPN ratio 3:1, reaction yleid of m-Xun leading of the property of the Synthesis of m-xylenediamine by hydrogenation of isophthalonitrile with modified Raney Ni catalyst DOCUMENT NUMBER: TITLE: Shen, Qin; Liu, Zhong-Neng; Hou, Min-Bo; Lu, AUTHOR (S): Xiao-Yuan
CORPORATE SOURCE: Shanghai Research Institute of Petrochemical
Technology, SINOPEC, Shanghai, 201208, Peop. Rep.
China
SOURCE: Jingxi Huagong (2000), 17(9), 544-546, 551
CODEN: JIHUFJ: ISSN: 1003-5214
Jingxi Huagong Bianjibu
DOCUMENT TYPE: Journal
LANGUAGE: Chinese
IT 1477-55-0P, 1,3-Benzenedimethanamine
RL: IMF (Industrial manufacture): PREP (Preparation)
(preparation of m-xylenediamine by hydrogenation of
isophthalonitrile in presence of modified Raney Ni
catalyst) Xiao-Yuan CORPORATE SOURCE:

catalyst)
1477-55-0 CAPLUS
1,3-Benzenedimethanamine (9CI) (CA INDEX NAME)

нам-сна \_CH2-NH2

L25 ANSWER 4 OF 5 CAPLUS COPYRIGHT 2004 ACS on STN

AB M-Xylyenediamine was prepared by reducing the isophthalonitrile with
hydrogen in the presence of Raney nickel promoted with iron and
chromium. By utilizing this particular catalyst, it is not only
possible to provide a drop in reaction temperature and an increase in rate of consumed hydrogen which is substantially above that obtained with Raney nickel alone, but also to attain yield of the desired product more than 80%. The deactivated catalyst was recovered to its original reactivity by simple process of regeneration.

ACCESSION NUMBER: 1999:666051 CAPIUS

DOCUMENT NUMBER: 131:352075

FITTLE: Preparation of m-vvl. 100.

Preparation of m-xylylenediamine with Raney Ni comprising iron and chromium Lu, Zihong; Zhu, Zhiqing; Zhou, Congying Dep. Chem. Eng., ECUST, Shanghai, 200237, Peop. Rep. China Huadong Ligong Daxue Xuebao (1999), 25(3), 257-259 CODEN: HLIXEV: ISSN: 1006-3080 Huadong Ligong Daxue Xuebao Bianjibu Journal Chinese tenedimethanamine AUTHOR(S): CORPORATE SOURCE:

SOURCE:

PUBLISHER:

DOCUMENT TYPE:

DOCUMENT TYPE: OUR IN TYPE: OUR IN TABLE OF THE PROPERTY OF T

H2N-CH2 \_CH2--NH2

```
L25 ANSWER 5 OF 5 CAPLUS COPYRIGHT 2004 ACS on STN
AB A process for preparing a cyanoarylmethylamine by hydrogenating
only one of the two cyano groups of an aromatic dinitrile, which enables
the
        preparation of the amine in a high yield through the reaction of the
dinitrile
```

trile
at a high conversion by the use of a small amount of a catalyst
under the reaction conditions of low temperature and low pressure. In

this

process, at least one member selected from among activated Raney
catalysts prepared by treating Raney catalysts with
hydrogen in solvents and regenerated Raney catalysts
prepared by treating Raney catalysts with hydrogen in the presence
of alkalia in solvents is used as the catalyst in preparing a
cyanoarylmethylamine from an aromatic dinitrile. The amount of the
activated
Raney catalyst to be used in

vated
Raney catalyst to be used is preferably 0.1 to 10 weight%, still
preferably 0.5 to 5 weight% based on the aromatic dinitrile and that of

regenerated Raney catalyst to be used is preferably 0.1 to 50 weight%, still preferably 0.5 to 20 weight% based thereon. The

vated Raney catalyst is preferably one containing nickel and/or cobalt and prepared through activation in a solvent under a hydrogen

1998:543046 CAPUS
1998:19421
Process for the preparation of cyanoarylmethylamine
Miura, Motoo; Suyama, Yuseki; Kondo, Hideyuki;
Morikawa, Kouhei
Showa Denko K.K., Japan
PCT Int. Appl., 30 pp.
CODEN: PIXXO2
Patent
Japanese PATENT ASSIGNEE(S):

SOURCE: DOCUMENT TYPE: Japanese 2 LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PAT	ENT	NO.			KIN	D	DATE			APPL	1CAT	ION	NO.		D.	ATE	
						_									-		
WO	9833	767			A1		1998	0806		WO 1	998-	JP46	4		1	9980	204
	W:	AL.	AM.	AT.	AU.	AZ.	BA.	BB,	BG,	BR.	BY.	CA,	CH,	CN,	CU,	CZ,	DE
		DK.	EE.	ES.	FI.	GB,	GE.	GH,	HU,	ID,	IL,	ıs.	JP,	KE,	KG,	KR,	K2
																NZ,	
		PT.	RO,	RU,	SD,	SE,	SG,	SI,	SK,	SL,	TJ,	TM,	TR,	TT,	UA,	UG,	US
		UZ.	VN,	YU,	ZW,	AM,	AZ,	BY,	KG,	KZ,	MD,	RU,	TJ,	TM			
	RW:	GH,	GM.	KE,	LS,	MW,	SD,	SZ,	UG,	ZW,	AT,	BE,	CH,	DE,	DK,	ES,	FI
		FR.	GB.	GR.	IE.	IT,	LU,	MC,	NL,	PT,	SE,	BF,	BJ,	CF,	CG,	CI,	CM
		GA.	GN.	ML.	MR.	NE.	SN,	TD,	TG	-							
		- ^ ·					1000	000E		n., 1	000	E 770	α.			0000	204

4 E, L,  L25 ANSWER 5 OF 5 CAPLUS COPYRIGHT 2004 ACS on STN (CONTINUED)
US 6114277 A 20000905 US 1998-155450 19980930
NO 1998-4622 19981002
PRIORITY APPLM INFO:: NO 1997-JP270 W 19970204 WO 1998-JP464 A 19980204

OTHER SOURCE(S):

CASREACT 129:161421

17 1477-55-0P, m-Xylylenediamine
RL: BYP (Byproduct); PREP (Preparation)
(process for preparation of cyanoarylmethylamine)
RN 1477-55-0 CAPUS
CN 1,3-Benzenedimethanamine (9CI) (CA INDEX NAME)

CH2-NH2

REFERENCE COUNT: THIS THERE ARE 13 CITED REFERENCES AVAILABLE FOR RECORD. ALL CITATIONS AVAILABLE IN THE RE

## => d his

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(FILE 'HOME' ENTERED AT 16:17:05 ON 14 SEP 2004)
     FILE 'CAPLUS' ENTERED AT 16:17:17 ON 14 SEP 2004
        1207996 S CATALY?
L1
        1087400 S REDUCT?
L2
         256764 S HYDROGENAT?
L3
        1299917 S L2 OR L3
L4
L5
         224466 S L1 AND L4
         775627 S NI OR NICKEL
L6
        1091254 S COPPER OR CU
L7
         513944 S CHROMIUM OR CR
L8
          40533 S L5 AND L6
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L10
         162143 S REGEN?
           1162 S L9 AND L10
L11
L12
            239 S L11 AND L7
     FILE 'REGISTRY' ENTERED AT 16:21:50 ON 14 SEP 2004
L13
                STRUCTURE UPLOADED
L14
              1 S L13
                STRUCTURE UPLOADED
L15
              1 S L15
L16
                STRUCTURE UPLOADED
L17
L18
              7 S L17
                STRUCTURE UPLOADED
L19
L20
              7 S L19
L21
           1647 S L19 FULL
     FILE 'CAPLUS' ENTERED AT 16:24:50 ON 14 SEP 2004
L22
           3471 S L21
             90 S L22 AND L5
L23
L24
             40 S L23 AND L6
L25
              5 S L24 AND L10
=> s (catalyst regeneration)
        665986 CATALYST
        670508 CATALYSTS
        853598 CATALYST
                  (CATALYST OR CATALYSTS)
         94665 REGENERATION
           595 REGENERATIONS
         94949 REGENERATION
                  (REGENERATION OR REGENERATIONS)
L26
          4927 (CATALYST REGENERATION)
                  (CATALYST (W) REGENERATION)
=> s 126 and 14
          1176 L26 AND L4
L27
=> s 127 and 16
           258 L27 AND L6
L28
=> s temperature
        498531 TEMPERATURE
         72912 TEMPERATURES
        562083 TEMPERATURE
                  (TEMPERATURE OR TEMPERATURES)
       2718942 TEMP
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700192 TEMPS 3027176 TEMP

(TEMP OR TEMPS)

SYSTEM LIMITS EXCEEDED - SEARCH ENDED

The search profile you entered was too complex or gave too many answers. Simplify or subdivide the query and try again. If you have exceeded the answer limit, enter DELETE HISTORY at an arrow prompt (=>) to remove all previous answers sets and begin at L1. Use the SAVE command to store any important profiles or answer sets before using DELETE HISTORY.

=> d 128 1-258 abs ibib

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L28 ANSWER 1 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB A continuous and regenerative process for removing VOC from effluents and regenerating VOC loaded catalysts is provided. In particular, the
process
             employs two fixed bed reactors in series filled with adsorbent/catalyst. The VOC containing effluent is passed over the first reactor for
adsorbing VOC
from the effluent, and part of the treated gas is made up with an
appropriate amount of air or hot regenerator exhaust gas to form a
regeneration gas. The regeneration gas is passed over the second reactor
to regenerate the adsorbent/catalyst, which is loaded with VOC from the
previous cycle and to convert the adsorbed VOC into innocuous compds.
such
            as carbon dioxide and water. Just before the VOC breakthrough occurs at the first reactor, the effluent is switched to the second reactor while the regeneration gas is switched to the first reactor to start another operation cycle. The process is characterized by using fixed bed
            cors
of simple design, high VOC removal efficiency, requiring no or little
external heat, high and flexible treating capacity, low costs of
operation
and maintenance and most importantly, safe and robust to operate.

ACCESSION NUMBER: 2004:331560 CAPLUS

DOCUMENT NUMBER: 140:325891
                                                             14U:325891
Process for removing volatile organic compounds
Yan, Tsoung Y.; Chang, Jen-Ray
USA
U.S. Pat. Appl. Publ., 9 pp.
CODEN: USXXCO
Patent
English
 TITLE:
INVENTOR(S):
 PATENT ASSIGNEE(S):
 SOURCE:
 DOCUMENT TYPE:
                                                              English
 LANGUAGE:
 FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
                                                                                                                                                                      DATE
             PATENT NO.
                                                              KIND
                                                                            DATE
                                                                                                             APPLICATION NO.
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20040422

US 2004076568
PRIORITY APPLN. INFO.:

Al

its

activity decreased during the course of the hydrogenation of the dicyanobenzenes is contacted with a hydrogen-containing gas at 200-500° while controlling a temperature rise speed of the catalyst to \$40°/min, thus regenerating the catalyst which can then be reused in subsequent hydrogenations of the dicyanobenzenes.

ACCESSION NUMBER: 2004:162518 CAPLUS
DOCUMENT NUMBER: 140:219730

TITLE: Hydrogenation ANSWER 3 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Xylylenediamines are prepared by hydrogenating dicyanobenzenes in a liquid phase in the presence of a catalyst, where the catalyst having 100:21930 CAPUS
Hydrogenation process and catalyst for the
production of xylylenediamines from dicyanobenzenes
and a regeneration process for the catalyst
Amakawa, Kazuhiko INVENTOR (S): PATENT ASSIGNEE (S): Japan U.S. Pat. Appl. Publ., 7 pp. CODEN: USXXCO SOURCE: DOCUMENT TYPE: English 1 FAMILY ACC. NUM. COUNT: PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE US 2004039232 A1 20040226 US 2003-643981 20030820
JP 2004107327 A2 20040408 JP 2003-292139 20030812
EP 1394146 A1 20040303 EP 2003-18532 20030816
R: AT, BE, CH, DE, DK, ES, FR. GB, GR, IT, LI, LI, NL, SE, MC, PT,
IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK
PRIORITY APPLM. INFO:: JP 2002-245222 A 20020826

CASREACT 140:219730

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L28 ANSWER 4 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Compns. for reduction of gas-phase reduced nitrogen species and NOx, generated from partial or incomplete combustion during fluid catalytic cracking, are comprised of: (1) 5.0 weight of a non-zeolite acidic metal oxide, (2) 20.3 weight% an alkali metal oxide or an alkaline earth metal oxide, (3) 20.1 weight% of a metal oxide or an alkaline earth metal oxide, (3) 20.1 weight% of a metal coxponent, preferably Rh of Ir.

Freferably, the compns. are used as sep. additives particles circulated along with the circulating FCC catalyst inventory. Overall Nox emissions are decreased as the product combustion gas stream is passed from the cracking catalyst regenerator to a CO boiler, in which the CO is oxidized to CO2 and a decreased amount of the reduced nitrogen species is oxidized to NOX.

ACCESSION NUMBER: 2003:961164 CAPLUS
DOCUMENT NUMBER: 100:6918

Nitrogen oxide emission reduction from petroleum cracking units using communities.
                                                                                     2003:961164 CAPLUS
140:6918
Nitrogen oxide emission reduction from
petroleum cracking units using compositions
                                                                                     exhaust gas catalysts and oxygen storage components Yaluris, George: Rudesill, John Allen W.R. Grace & Co.-Conn., USA U.S., 15 pp. CODEN: USXXAM Patent English 1
   containing
   INVENTOR(S):
PATENT ASSIGNEE(S):
SOURCE:
   DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                           DATE
                   PATENT NO.
                                                                                       KIND
                                                                                                                                                      APPLICATION NO.
REFERENCE COUNT:
                                                                                                         THERE ARE 25 CITED REFERENCES AVAILABLE FOR
                                                                                     25
                                                                                                          RECORD. ALL CITATIONS AVAILABLE IN THE RE
   FORMAT
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L28 ANSWER 2 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The possible use of Ni in Heck reactions was investigated with use of the d. function theory method. It was found that the mechanisms

the Ni- and Pd-catalyzed Heck reactions are quite similar to each other. Nevertheless, oxidative addition and olefin insertion occur

each other. Nevertheless, oxidative addition and olefin insertion occur each other. Nevertheless, oxidative addition and olefin insertion occur lower energy barriers in the Ni system than in the Pd system. Because \( \textit{B} - \text{hydride} \) elimination is more efficient in the Pd system than in the Ni system, there is a poorer selectivity to vinylation over Michael addition, in the Ni system than in the Pd system. In addition, catalyst regeneration through Nr removal is considerably harder to achieve with the Ni system than with the Pd system. Therefore, either a very strong base should be used for the Ni catalysis or a reductive pathway should be designed to remove Nx from the Ni complex. Compared to the Pd system, oxidative addition of an alkenyl or aryl chloride is not much harder than oxidative addition of an alkenyl or aryl chloride is not much harder than oxidative addition of an alkenyl or aryl chloride relatively easily. Also, because \( \text{P-hydride} \) elimination is more difficult in the Ni system than in the Pd system, the Ni-catalyzed check reaction may be applied to aliphatic halides. For an olefin with an electron-donating substituent, the Ni-catalyzed coupling should slightly favor the Markovnikov-type product, if the steric effect is not significant. For an olefin with an electron-withdrawing substituent, the Ni-catalyzed coupling substituent, the Ni-catalyzed coupling should slightly favor the Markovnikov-type product, if the steric effect is not significant. For an olefin with an electron-withdrawing substituent, the Ni-catalyzed coupling should slightly favor the Markovnikov-type product, because of the Ni and Pd catalyzed for the reaction mechanism to the cationic pathway (pre-dissociation of acide-ligand), and anionic pathway (pre-dissociation of acide-ligand), and anionic pathway (pre-dissociation of acide-ligand), and anionic pathway (pre-dissociation of acide-ligand), and for the reactivities of the Ni and Pd catalyzed.

medianism to the maintenance of the Ni and Pd catalysts.

ACCESSION NUMBER: 2004:252965 CAPLUS

DOCUMENT NUMBER: 141:7261

TITLE: Comparing Nickel- and Palladium-Catalyzed Heck Reactions

AUTHOR(S): Lin, Bo-Lin; Liu, Lei; Fu, Yao; Luo, Shi-Wei; Chen, Qian; Guo, Qing-Xiang

CORPORATE SOURCE: Department of Chemistry, University of Science and Technology of China, Hefei, 230026, Peop. Rep. China

SOURCE: Organometallics (2004), 23(9), 2114-2123

CODEN: ORGNOT: ISSN: 0276-7333

PUBLISHER: American Chemical Society

Journal

LANGUAGE: English

REFERENCE COUNT: 69 THERE ARE 69 CITED REFERENCES AVAILABLE FOR

THIS

FORMAT

RECORD. ALL CITATIONS AVAILABLE IN THE RE

Page 15

OTHER SOURCE(S):

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ANSWER 5 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
A method for regenerating spent supported metal catalysts comprising
treating the spent catalyst with an organo-metallic complex forming agent
having an ionization constant pK 1 of at least 2.5. The catalyst
having an ionization constant pK 1 of at least 2.3. The catalyst activity is restored to an activity level near to or greater than the fresh catalyst. The regeneration method is particularly useful for regenerating spent palladium catalysts on an alumina support as utilized for the hydrogenation of Et anthraquinone (EAQ) in the production of hydrogen peroxide.

ACCESSION NUMBER: 2003:912824 CAPLUS
 DOCUMENT NUMBER
                                                         140:9216
                                                       140:9216
Regeneration of spent supported metal catalysts
Zhou, Bing; Rueter, Michael
 TITLE:
 TITLE:
INVENTOR(S):
                                                       Zhou, Hing; Nuclet, Michael
U.S. Pat. Appl. Publ., 8 pp., Cont.-in-part of U.S.
Ser. No. 745,510.
CODEN: USXXCO
Patent
 PATENT ASSIGNEE (S):
SOURCE:
 DOCUMENT TYPE:
LANGUAGE:
                                                      English
2
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                                                                     DATE
                                                                                                  APPLICATION NO.
           PATENT NO.
                                                        KIND DATE
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A 20021220

US 2002-326042

L28 ANSWER 6 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN US 2000-620865 A 20000721
W0 2001-US22733 W 20010719
REFERENCE COUNT: 34 THERE ARE 34 CITED REFERENCES AVAILABLE FOR THIS
RECORD. ALL CITATIONS AVAILABLE IN THE RE

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ANSWER 6 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

Spent metal oxide desulfurization sorbents, in petroleum desulfurization units, are regenerated by oxidizing the sulfided metal oxides (i.e., at 200-850'), followed by reduction, especially with hydrogen gas.

The sorbents are derived from oxides of Fe, Ni, Cu, Co, and Zn (preferably Cu, Ni, or Co), and can contain one or more secondary metals to increase regeneration efficiency or capacity. Suitable regeneration rate enhancing components are Group VIII noble metals, which decreases the regeneration period by 230%. In addition, additives for suppression of hydrocarbon cracking can be added (e.g., 1-10
meight Cu, Ag, Au, Sn, and Pb, and Group IB, Group IVA, Group VIA
elements).

The method can be used also for regeneration of spent
hydrodesulfurization
catalysts, especially those derived from Mo, W, Fe, Co, Ni, Pt, Pd,
ICI, and Rh.
ACCESSION NUMBER: 2003:903193 CAPLUS
DOCUMENT NUMBER: 139:367283
TITLE: Regeneration of
catalysts
          hydrodesulfurization
                                                                                                                                   by oxidation and reduction steps
Chen, Jingquang G.; Brown, Leo D.; Baird, William C.,
Jr.; Mcvicker, Gary B.; Ellis, Edward S.; Touvelle,
Michele S.; Klein, Darryl P.; Vaughan, David E. W.
EXXONMObil Research and Engineering Company, USA
U.S., 7 pp., Cont.-in-part of U.S. Ser. No. 326,827.
CODEN: USXXAM
Patent
English
10
        INVENTOR (S):
          PATENT ASSIGNEE(S):
SOURCE:
          DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                                                                                       DATE
                            PATENT NO. KIND DATE APPLICATION NO. DATE

US 6649043 B1 20031118 US 2000-620865 20000721
US 5935420 A 19990810 US 1997-918641 19970822
US 6221240 B1 20010424 US 1999-316647 19970822
WC 2002008160 A1 20020131 WC 2001-US22733 20010719
WC AE, AL, AM, AT, AU, AZ, BA, BB, BB, GB, BR, YC, AC, HC, KC, UC, CZ, DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, II, IN, IS, JF, KE, KG, KF, KR, KZ, LC, LK, LK, LS, LT, LU, LV, MD, MG, MK, MN, MM, MC, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, RW, TM, TM, TT, UT, UA, UG, UZ, VN, YU, 2A, ZW, AM, AZ, BY, KG, KZ, MD, RU, TD, EG, GK, GK, CK, CM, DE, DK, ES, FI, FR, GB, GR, IE, TT, LU, MC, NL, PF, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, N, TD, TG
EF 1309528 A1 20030514 EP 2001-957188 20010719
RC 2013000289 A 20030319 NO 2003-289 20030120
ORITY APPLN. INFO:
                                   PATENT NO.
                                                                                                                                       KIND
                                                                                                                                                                                                                                      APPLICATION NO.
                                                                                                                                                                                                                                                                                                                                                         DATE
            PRIORITY APPLN. INFO.:
                                                                                                                                                                                                                                      US 1997-918641
                                                                                                                                                                                                                                                                                                                                        A2 19970822
                                                                                                                                                                                                                                       IIS 1999-326827
                                                                                                                                                                                                                                                                                                                                         A2 19990607
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L28 ANSWER 7 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Synthesis gas (CO/H2) is produced by passing light hydrocarbons
Containing 50
volumeN of methane over a catalyst in a reaction zone in the absence of
gaseous oxygen by a net catalytic partial oxidation (CPOX) reaction. Ac
catalyst composition contains an oxidatively reducible source of lattice
oxygen
in the form of a transition metal oxide, a lanthanide oxide, or their
mixts. and a catalytically active compound deposited on the reducible
metal
oxide. The transition metal of the reducible metal oxide can be Zr, V,
W,
Mo, Mn, Ti, Cu, Re, and Ru. The lanthanide element is Ce or La. The
catalytically active compound is Rh, Pt, Pd, Ru, Ir, Os, Ni, Co,
Fe, Mn, or Cr. The reduced metal and/or lower valence metal oxide are
reoxidized in a sep. reaction zone or by alternating syngas and
regenerant
processes. The partially reduced catalyst can be regenerated to the
active (more oxidized) oxidation state in a second reaction zone using
oxygen, atmospheric air or process steam. The second reaction zone is
purged
with N2, Ar, or He prior to re-oxidizing the catalyst. The partial
oxidation
proceas to produce syngas is carried out at 200-10,000 kPa and
350-2000°C whereby the gas is passed over the catalyst at a gas
hourly space velocity of 1,000,000/ with a catalyst residence time of
$\leq 5$ ms. A combustible gas is added to the reaction mixture
sufficient to initiate an exothermic catalytic partial oxidation
reaction.
The reaction mixture is preheated to 30-750°C.
ACCESSION NUMBER:
2003:696026 CAPLUS

DOCUMENT TYPE:
DOCUMENT TYPE:
PATENT ASSIGNEE(S):
CONCOL II. APPLICATION NO.

DATE

MO 2003072490 All 20030904 WO 2002-US40244 2002116
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CK, CN,
CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GG, GG, GH,
CM, HR, HU, ID. II. IN, IS, JP, KE, KG, KR, PK, KZ, LC, LK, LR,
LS, LT, LU, LV, MA, MD, MG, MK, MN, MM, MN, MZ, NO, NZ, OM, PH,
PL, PT, RO, RU, SD, SE, SG, SK, SL, TJ, TH, TH, TT, TT, TU, HC,
CH, CY, CZ, DE, DK
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PRIORITY APPLN. INFO.:

US 2002-359225P P 20020222

L28 ANSWER 7 OF 258 CAPLUS COPYRIGHT 2004 ACS ON STN (Continued)
REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS
RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

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ANSWER 9 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
A process for regenerating the activity of used-metal catalysts for the
hydrogenation of carbon monoxide is described comprising
decreasing the hydrocarbon (i.e., dewaxing) content of the spent
     catalyst,
calcining the dewaxed catalyst under an oxident-containing atmospheric,
     impregnating the calcined catalyst with a solution of at least one of a metal compound,
                           recalcining the metal salt-impregnated catalyst under an
     Oxidant-containing at activating the recalcined catalyst by contacting it
                       a hydrogen-contacting gas at elevated temps. The process regenerates and enhances both supported and dispersed active metal (DAM) catalysts. Used catalysts enhanced by the process are initially treated to decrease their hydrocarbon content. The treatment may be carried out in a single reactor, or by carrying out up to all steps after catalysts may be withdrawn from a reactor and returned to "at least one reactor, both preferably during operation thereof. Up to all steps may be effected in
    a subsequent reactor, or in specialized apparatus
ACCESSION NUMBER: 2003:590845 CAPLUS
DOCUMENT NUMBER: 139:135213
TITLE: hydrogenation catalysts
INVENTOR(S): Daage, Michel; Koveal, Russell John; Chang, Min
PATENT ASSIGNEE(S): U.S. Pat. Appl. Publ., 10 pp.
CODEN: USXXCO
DOCUMENT TYPE: CODEN: USXXCO
DOCUMENT TYPE: Patent
English
FAMILY ACC. NUM. COUNT: 1
FAMILY ACC. NUM. COUNT: 1
FAMILY ACC. NUM. COUNT: 1
      LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
PATENT NO. KIND DATE APPLICATION NO. DATE

US 2003144366 Al 20030731 US 2002-59926 20020129
WO 2003064034 A2 20030807 WO 2003-US150 20030103
WO 2003064034 A3 20031127
W' AR, AG, AL, AN, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, II, IN, IS, DF, KE, KG, KP, KR, KZ, LC, LK, LL, LT, LU, LV, MA, MD, MG, MK, MN, MM, MX, MZ, NO, NZ, PL, FT, RO, RU, SD, SE, SG, SK, SL, TJ, TH, TR, TT, TZ, UA, UG, UZ, VL, YU, ZA, ZW, AH, AZ, BY, KG, KZ, MD, RU, TJ, TM

RN: GH, GM, KE, LS, MW, MZ, SD, ST, SZ, TZ, UG, ZM, ZW, AT, BE, BG, CH, CY, C2, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, ML, PT, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GO, CP, ML, MR, NE, SN, TD, TG

PRIORITY APPLN. INFO:: US 2002-59926 A 20020120
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L28 ANSWER 8 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB There is provided a process for hydrocarbon synthesis wherein a supported metal catalyst for hydrogenating carbon monoxide to form a mixture of hydrocarbons is regenerated by decreasing its hydrocarbon content, impregnating under a non-oxidative atmospheric with a solution of at
                    cone
member of the group of ammonium salts (e.g., ammonium acetate),
alkylammonium salts, and weak organic acids, optionally including
                    nia, oxidizing with a gaseous oxidant (e.g., air) in the presence of the impregnating solution, activating the catalyst by reduction with hydrogen at elevated temps. and reusing the catalyst. The treatment may be carried out in a single reactor, or by carring out all of the steps after catalyst has been withdrawn therefrom and returned to at least one reactor. Up to all steps subsequent to decreasing the hydrocarbon
 content steps subsequent to decreasing the hydrocarbon may be affected in a subsequent reactor, or in specialized apparatus ACCESSION NUMBER: 2003:610393 CAPLUS DOCUMENT NUMBER: 139:151409
                                                                                          CAPLUS

139:151409

Frocess for Fischer-Tropsch catalyst
regeneration in the manufacture of C10+
hydrocarbons from synthesis gas
Koveal, Russell John; Daage, Michel; Shen, Eric
Baochun
ExxonMobil Research and Engineering Company, USA
PCT Int. Appl., 50 pp.
CODEN: PIXXD2
Patent
English
1
  INVENTOR(S):
   PATENT ASSIGNEE(S):
SOURCE:
 DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                                                                                      APPLICATION NO.
                      PATENT NO.
                                                                                                KIND
                                                                                                                    DATE
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WO 2003064356

W: AE, AG, AL,
CO, CR, CU,
CM, HR, HU,
LS, LT, LU,
PL, PT, RO,
UG, UZ, VN,
RW: CH, GM, KE,
CH, CY, CZ,
NL, PT, SC,
US 2003166451
US 6733334
PRIORITY APPLN. INFO.:
                                                                                                   Al
                                                                                                                                                                       WO 2003-US202
                                                                                                                         20030807
                                                                                                                                                                                                                                                            20030103
                                                                                             A1 20030807 W0 2003-US202 20030103
AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CZ, DE, DK, DM, DZ, EC, EE, ES, FT, GB, GD, EG, GH, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LV, NA, MD, MG, MK, MN, MW, KM, M, N, M, NZ, OX, MP, HR, U, SD, SE, SG, SK, SL, TJ, TM, TM, TR, TT, TZ, UA, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG, DE, DK, EE, ES, FT, FR, GB, CR, HU, IE, IT, LU, MC, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, SN, TD, TG
B1 20040622 US 2002-59916 20020129
                                                                                                                                                                      US 2002-59916
                                                                                                                       THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE
 REFERENCE COUNT:
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L28 ANSWER 10 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Regeneration of spent supported metal catalysts for CO
hydrogenation reactions (e.g., Co, Ni, Cu, Ru, Rh, Re,
Pd, Pt, Os, and Ir) ls carried out by a sequential process consisting of:
(1) decreasing the hydrocarbon content on the catalyst, (2) impregnating,
under non-oxidative atmospheric, with a solution of a weak organic acid,
until the
catalyst has absorbed >10 volume% (preferably 10-1000%) of its
calculated pore
volume, (3) oxidizing with a gaseous oxidant in the presence of the weak
organic acid, and (4) reducing the catalyst with a hydrogen-containing
gas to
                   to form the active catalyst. Hydrocarbons on the catalyst (especially for Fischer-Tropsch catalysts) can be removed by contacting the catalyst with one or more of the following: (1) a hydrogen-containing gas, (2) a
                   supercrit. fluid, or (3) an oxygen-containing gas or steam. Optionally,
                   catalyst is calcined after the oxidation step, and passivated after the activation step. A preferred means of decreasing the hydrocarbon content of the catalyst is contacting it with a hydrogen-containing gas at high
temps. The organic acid has the general formula R(COOH)n (n = 1-3; R is a ^{\circ}
The organic acid has the general formula R(COOH)n (n = 1-3; R is a cyclic, aliphatic, saturated, or unsatd. moiety substituted with one or more NO2, NH2, OH, and alkoxy).

ACCESSION NUMBER: 2003:590798 CAPLUS
DOCUMENT NUMBER: 139:119731

TITLE: Combined solvent extraction-oxidation-
                                                                                             139:119731
Combined solvent extraction-oxidation-reduction-acid washing for regeneration of used supported catalysts for carbon monoxide hydrogenation
Clark, Janet Renee; Koveal, Russell John; Daage,
 INVENTOR(S):
                                                                                             Michel
                                                                                             Michel
Exxonmobil Research and Engineering Co., USA
U.S. Pat. Appl. Publ., 10 pp.
CODEN: USXXCO
 PATENT ASSIGNEE (S):
SOURCE:
                                                                                             Patent
English
 LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                                     DATE
                                                                                                                                                                 APPLICATION NO.
US 2002-59917
                   PATENT NO.
                                                                                             KIND
                                                                                                                                                                                                                                                       DATE
                 US 2003144129
US 6753286
WO 2003064033
W: AE, AG,
CO, CR,
GM, HR,
LS, LT,
PL, PT,
UG, UZ,
RW: GH, GM,
CH, CY,
PT, SE,
MR, NE,
RITY APPLN. IMFO.
                                                                                          A1 20030731 US 2002-59917 20020129
B2 20046622
A1 20030807 WO 2002-US41512
AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, ID, III, III, III, SI, JP, KE, KG, KY, NR, KZ, LC, LK, LR, LV, HA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, RU, SD, SE, SG, SK, SL, TJ, TM, TM, TR, TT, TZ, UA, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, LS, MW, MZ, SD, SI, SZ, TZ, UG, ZM, ZW, AT, BE, BG, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, TD, TG
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US 2002-59917

A 20020129

PRIORITY APPLN.

L28 ANSWER 10 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN (Continued) OTHER SOURCE(S): MARPAT 139:119731

34 THERE ARE 34 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

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AB The combustion process is carried out in presence of a primary NOX reducing catalyst coating which is treated when required by means of cerium/europium compound in presence of water vapor and advantageously in presence of a nanoparticulate refractory clay material, such as hydrated aluminum silicate. The primary NOX reducing catalyst comprises at least one atom selected from the group consisting of Ce, Pt, Pd, Cu, W, Pr, Sc, Si, Ga, Ru, Mo, Ni, Ti, Co, Rh, Mg, Ca, La, Eu, Y, Yh, Si and mixts. thereof, preferably Ce, Eu, Pr or Yb. The process of the invention can be operated in a closed chamber (with intermittent opening of door(s) or valve(s)) or in an open chamber, such as a boiler, reformer, fuel cells, kiln, incinerator, ovens, steel works, power station, explosion chamber, turbines, burner, central heating system, grass cutting machine, HCCI (heterogeneous charge compression ignition), CAI or homeogeneous combustion engine, rocket, guns, flat engines, space shuttle, air planes, furnaces, propellant, pulsation combustion, auto inflammation engine,
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.., pulsation combustion 2003:300941 CAPLUS 138:308425 Combustion process Bosteels, Dominique Ire. PCT Int. Appl., 84 pp. CODEN: PIXXD2 Patent English etc.
ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE:
INVENTOR(S):
PATENT ASSIGNEE(S):
SOURCE:

DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

Page 18

PAT	ENT	NO.			KIN	D	DATE					ION				ATE	
	2003				A2 A3			0417 0318	1			BE15					
	W:	AE,	AG,	AL,	AM.	AT.	AU,	AZ,	BA,	BB.	BG,	BR.	BY.	BZ.	CA.	CH.	CN.
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		GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KP,	KR.	KZ,	LC.	LK,	LR.
		LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW.	MX.	MZ,	NO,	NZ,	OM.	PH.
		PL,	PT,	RO,	RU,	SD,	SE,	SG,	SI,	SK,	SL,	TJ,	TM,	TN,	TR,	TT,	TZ,
		UA,	UG,	US,	UZ,	vc,	VN,	YU,	ZA,	ZM,	ZW,	AM,	AZ,	BY,	KG,	KZ,	MD,
			TJ,														
	RW:	GH,	GM,	ΚE,	LS,	MW,	MZ,	SD,	SL,	SZ,	TZ,	υG,	ZM,	ZW,	AT,	BE,	BG,
		CH,	CY,	CZ,	DE,	DK,	ΕE,	ES,	FI,	FR,	GB,	GR,	IE,	IT,	LU,	MC,	NL,
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		G₩,															

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L28 ANSWER 11 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Supported metal catalysts for the hydrogenation of carbon monoxide to form a mixture of hydrocarbons are regenerated by decreasing
    the
               hydrocarbon content of the catalyst, preferably by contact with hydrogen-containing gas at elevated temps., impregnating under a
   non-oxidizing atmospheric with a solution of at least one of an ammonium salt and an alkyl
   alkyl
ammonium salt, optionally in combination with up to five moles of ammonia
per L of solution to the point where it has absorbed a volume of
solution equal to
at least apprx.10% of its calculated pore volume; oxidizing the
catalyst with a
gaseous oxidant in the presence of the impregnating solution and
    activating the catalyst by reduction with hydrogen at elevated temps. Optionally, the catalyst is calcined after the oxidation step, and
  PASSIVATED AT SEP.
ACCESSION NUMBER: 2003:590797 CAPLUS
DOCUMENT NUMBER: 199:119730
TITLE: Supported catalyst treatment
INVENTOR(s): Daage, Michel: Koveal, Russell John; Clark, Janet
Renee; Marler, David Owen
USA
SOURCE: USA
DOCUMENT TYPE: Patch
LANGUIAGE: Patch
LANGUIAGE: Patch
LANGUIAGE: Patch
PATENT INFORMATION: 1
PATENT INFORMATION:
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L28 ANSWER 12 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN WO 2002-BE156 (Continued) A 20021010 L28 ANSWER 13 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Spent nickel catalysts (SNC) used in the edible oil industry to
hydrogenate liquid oil have the potential of polluting the
environment in a number of ways. The recovery of fat from the SNC could
prevent environmental pollution and reduce the oil loss. Hexane was the
solvent of choice for oil extraction Alternative solvents that are considered

safer were evaluated. Hexane, isopropanol, ethanol, and heptane were examined using soxhlet extraction While hexane was more efficient in oil recovery from SNC, isopropanol proved to be very efficient in separating

from the waste material and it could be more easily recovered compared to other solvents. Cooled isopropanol extraction provided separation of miscella into

miscella into
two phases: lower oil-rich and an upper solvent-rich. An aqueous
extraction
process assisted by an immiscible solvent was tested.
ACCESSION NUMBER: 2003:98914 CAPLUS
DOCUMENT NUMBER: 138:342986

DOCUMENT NUMBER: TITLE:

138:342986
Processing of spent nickel catalyst for fat recovery
Nasir, Mohammad Ibrahim
Research Department, Jahan Vegetable Oil Co., Tehran, 19697, Iran
Grasas y Aceites (Sevilla, Spain) (2002), 53(2), 213-217
CODEN: GRACAN, Year. Acc. AUTHOR(S): CORPORATE SOURCE:

source:

CODEN: GRACAN; ISSN: 0017-3495 Instituto de la Grasa

PUBLISHER: DOCUMENT TYPE: LANGUAGE: REFERENCE COUNT:

English
21 THERE ARE 21 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

ANSWER 15 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN A two-stage method for contacting of gases and solids in a bubbling fluidized-bed reactor (for catalytic and noncatalytic reactions) was developed in which the first stage involves fluidization with bubbling, and the second stage involves the formation of the bubbling bed. In the first stage, a primary gas, containing the reactant(s), is introduced into reactor (with bed length to bed diameter ratio .ltorsim.5.0:1) through a primary gas distributor located at the reactor bottom at a superficial velocity, Up, that is close or equivalent to the min. fluidization velocity,

Umf, required to obtain an emulsion phase with little or no formation of gas bubbles. In the second stage, gas bubbles in the incipiently fluidized bed (formed in stage 1) are formed by introducing a secondary gas through a secondary gas distributor located immediately above the primary gas distributor. This secondary gas selected from one of the reactants which is used in excess of that required for reaction stoichiometry (e.g., steam), at a superficial gas velocity, Us. Us is related to the Up (of the primary gas) such that a Us/Up is 0.5-10.0:1, preferably 1-5:1. Typical reactions that can be handled by the bubbling fluidized bed include vapor-phase hydrogenation of nitrobenzene and nitrotoluene isomers to aniline and the corresponding toluidine, methane conversion to synthesis gas, semmoxidh. of propylene to acrylonitrile, propylene oxidation to acrolein, oxidation of acrolein to acrylic velocity. Up, that is close or equivalent to the min. fluidization acrylic acid, regeneration of coked hydrocarbon cracking catalyst, ethane oxychlorination, Fischer-Tropsch reaction, and heavy oil hydrocracking.

ACCESSION NUMBER: 2002:927867 CAPLUS
DOCUMENT NUMBER: 138:26375
TITLE: Two-stage restricts. Two-stage method for gas-solid contact in bubbling fluidized-bed reactors for catalytic and non-catalytic reactions Choudhary, Vasant Ramchandra; Choudhary, Tushar INVENTOR (S): Vasant PATENT ASSIGNEE(S): SOURCE: Council of Scientific & Industrial Research, India U.S. Pat. Appl. Publ., 10 pp. CODEN: USXXCO DOCUMENT TYPE: Patent English LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: APPLICATION NO. PATENT NO. KIND DATE US 2001-817744 US 2003-725723 US 2001-817744 20010326 20031202 A3 20010326 US 2002179489 US 2004122116 PRIORITY APPLN. INFO.: 20021205

L28 ANSWER 14 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The stability of a previously developed stationary-layer modified

Ni catalyst for hydrogenation of cottonseed oil was
confirmed by monitoring catalytic activity over 350 h at 220°. The
selection of conditions for catalyst reactivation are outlined, including ACCESSION NUMBER: 2003:62861 CAPLUS
DOCUMENT NUMBER: 138:254086
TITLE: Time-dependent activity of a catalyst
Karimkulova, M. P.; Iskandarov, Sh. A.; Abidova, M. CORPORATE SOURCE: Uzb. Nauchno-Issled. Khim.-Farm. Inst. im. A. Sultanova, Uzbekistan O'zbekiston Kimyo Jurnali (2002), (4), 67-71 CODEN: OKZZAG; ISSN: 0042-1707 Izdatel'stvo Fan PUBLISHER: DOCUMENT TYPE: LANGUAGE: L28 ANSWER 16 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The article is focused on potential utilization of catalysts in tar
removal from biomass gases. In the theor. part it deals with the use
catalytic purification and application of reforming and prereforming catalysts.

The processes occurring on the catalyst and their mechanisms including carbon formation and catalyst deactivation are described and minimization of the undesirable processes is discussed. A laboratory method for selection of ction of an appropriate catalyst is described in the exptl. part. Several com. available reforming and prereforming nickel catalysts were tested. In application of classic reforming catalysts, temps. higher than 700°C and steam excess must be used to minimize deactivation. If the gas does not contain sulfur compds. and catalyst
regeneration proceeds in the catalytic unit, active prereforming
catalysts can be already used at 450-550°C.
SSION NUMBER: 2002:880370 CAPLUS ACCESSION NUMBER: DOCUMENT NUMBER: TITLE: 138:290381 Application of catalysts for tar removal during Application of catalysts for tar removal during biomass gasification of marsak, J.; Skoblja, S. Department of Gas Manufacture, Institute of Chemical Technology, Prague, Czech Rep. Chemicke Listy (2002), 96(10), 00581: CHSAC: ISSN: 0009-2770 Ceska Spolecnost Chemicka Journal AUTHOR(S): CORPORATE SOURCE: PUBLISHER: DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 17 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The title regeneration process for petroleum refining hydrogenation catalysts with groups VIB, VIIB, and VIII metals as active components comprises reducing the deactivated catalyst with H2 at 200-350°C to remove sulfur, and oxidizing to remove carbon at 220-250°C for 2-6 h, at 280-350°C for 1-4 h, and at 480-550°C for 2-4 h. The process is simple and can restore the catalyst activity.

ACCESSION NUMBER: 2002:801730 CAPLUS DOCUMENT NUMBER: 137:284883 137:284883

Zhang, Xiwen: Sun, Wanfu; Zhao, Changzhi; Zhang, Shumei; Wang, Shaojun
China Petrochemical Corp., Ltd., Peop. Rep. China; Fushun Research Institute of Petroleum Processing, TITLE: INVENTOR (S): PATENT ASSIGNEE (S): Simopec Faming Zhuanli Shenqing Gongkai Shuomingshu, 9 pp. CODEN: CNXXEV Patent Chinese SOURCE: DOCUMENT TYPE: LANGUAGE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE CN 1335202 CN 1120051 PRIORITY APPLN. INFO.: 20020213 20000724 CN 2000-110718 20030903 20000724 CN 2000-110718

L28 ANSWER 19 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Fischer-Tropsch catalysts for hydrocarbon manufacture are dispersed

metal-type catalysts that are not immobilized, and consist of one or more Group VIII metals and copper, in which the catalyst activity is enhanced during reactor operation by: (1) withdrawing a mixture of hydrocarbons

catalyst from the reactor, (2) reducing the hydrocarbon content, (3) heating the mixture in a non-oxidizing atmospheric to a temperature

heating the mixture in a non-oxidizing atmospheric to a temperature the heating the mixture of at least one of the catalyst metals to remove non-metallic impurities and to form a slag of refractory metal oxides, (4) removing any

slag, if present, (5) cooling and solidifying the melt, (6) size reduction of the melt to a fine powder, and (7) returning the catalyst to the reactor. If the catalyst is a Raney catalyst, a leachable metal

added to the hydrocarbon-depleted mixture or melt under non-oxidizing conditions and extracting the leachable metal with caustic after step. The catalysts can be passivated prior to being returned to the reactor. The activity of the catalyst may be modified or enhanced by addition of

promoter metals or, prior to being returned to the reactor, by slurry

low-temperature
oxidation followed by reduction at elevated temperature The catalysts

passivated prior to being returned to the reactor.

ACCESSION NUMBER: 2002:185040 CAPLUS

DOCUMENT NUMBER: 136:234529

TITLE: allovs

Melt formation with caustic extraction of Raney in method for reactivation of spent and deactivated

INVENTOR (S):

in method for reactivation of spent and deactivated Fischer-Tropsch catalysts
Daage, Michel A.; Koweal, Russell John; Long, David Chester; Clavenna, Leroy Russell; Ramanarayanan, Trikur Anantharaman; Mumford, James Dirickson; Culross, Claude Clarence
ExxonMobil Research and Engineering Company, USA PCT Int. Appl., 31 pp.
CODEN: PIXXD2
Patent
English

PATENT ASSIGNEE(S): SOURCE:

DOCUMENT TYPE:

PA	TENT	NO.			KIN	D	DATE			APPL	ICAT	ION	NO.		D	ATE	
						-									-		
WO	2002	0204	41		A1	A1 20020314			1	WO 2	001-	US25	156		20010810		
	W:	ΑE,	AG,	AL,	AM,	AT,	AU,	AZ,	BA,	BB,	BG,	BR,	BY,	BZ,	CA,	CH,	CN,
		CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EE,	£S,	FI,	GB,	GD,	GE,	GH,	GM,	HR,
		HU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KP,	KR,	KZ,	LC,	LK,	LR,	LS,	LT,
		LU,	LV,	MA,	MD,	MG,	MK.	MN,	MW,	MX,	MZ,	NO,	NZ.	PL,	PT,	RO,	RU,
		SD,	SE,	SG,	SI,	SK,	SL,	TJ,	TM.	TR,	TT,	TZ.	UA,	ŲG,	UZ,	VN,	YU,
		ZA,	ZW,	AM,	AZ,	BY,	KG,	KZ,	MD,	RU,	TJ,	TM					
	RW:	GH.	GM,	KE,	LS,	MW,	MZ.	SD,	SL,	sz,	TZ.	UG,	ZW,	AT,	BE,	CH,	CY,
		DE.	DK.	ES,	FI.	FR.	GB,	GR,	IE.	IT,	LU.	MC,	NL,	PT.	SE,	TR.	BF,
		BJ.	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML.	MR,	NE,	SN,	TD,	TG	
US	6624	204			В1	٠.	2003	0923		US 2	000-	6539	14		21	0000	901
110	2001	0812															

FAMILY ACC. NUM. COUNT: PATENT INFORMATION: AU 2001-81234 L28 ANSWER 18 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

Process is disclosed for regeneration of solid catalysts or solid adsorbents wherein the catalyst or the absorbent is used in the form of a bed moving in a regeneration zone. The process contains ≥1 heating step under a reducing atmospheric (e.g., a H-containing atmospheric) at 250-650° 250-650° (preferably at 300-550°). Typically, the procedure is suitable for regeneration of hydrogenation catalysts and adsorbents for their protection (e.g., s traps, S guard beds).

ACCESSION NUMBER: 2002:304691 CAPLUS
DOCUMENT NUMBER: 137:64897 DOCUMENT NUMBER: TITLE: Regeneration method for heterogeneous catalysts and adsorbents adsorbents Robinson, James; Brahma, Nilanjan; Mendakis, Georges; Locatelli, Francois; Dufresne, Pierre Europeenne de Retraitement de Catalyseurs Eurecat, INVENTOR (S): PATENT ASSIGNEE (S): PCT Int. Appl., 15 pp. CODEN: PIXXD2 Patent SOURCE: DOCUMENT TYPE: English LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE WO 2002051548 A1 20020704 WO 2000-EP13237 20001222

W: CA, CN, JP, SG
RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
PT, SE, TR

EP 1345693 A1 20030924 EP 2000-985253 20001222

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
IE, SI, LT, LV, FI, RO, MK, CY, AL, TR

JP 2004516927 T2 20040610 JP 2002-552682 20001222

US 200209971 A1 20020725 US 2001-764147 20010119

US 2004043890 A1 20040304 US 2003-434365 20030509

RTTY APPLIN, INFO:: WO 2000-EP13237 W 20001223 US 2004043890 PRIORITY APPLN. INFO.: US 2003-434365 WO 2000-EP13237 20030509 W 20001222 US 2001-764147 B1 20010119 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE REFERENÇE COUNT: FORMAT

L28 ANSWER 19 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN (Continued)
EP 1313686 A1 20030528 EP 2001-959707 20010810
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
PRIORITY APPLM. INFO: US 2000-653914 A 20000901 A 20000901

WO 2001-US25156 W 20010810

REFERENCE COUNT: THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

Page 20

ANSWER 20 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
Spent nickel catalyst (SNC) has the potential of insulting the
quality of the environment in a number of ways. Its disposal has a
ution ution ...

effect. Optimum recovery of fat from SNC, could save the environment and reduce the oil loss. Hexane has been the solvent of choice for oil action

Alternative solvents considered to have been safer have been evaluated. Hexane, isopropanol, ethanol and heptane were examined using soxhlet nexame, isopropanol, ethanol and neptone were examined using usual action
While hexame is more efficient in oil recovery from SNC, isopropanol
proved to be very good in clear separation of oil from waste material
also aiso provides high solvent recovery compared to other solvents. Isopropanol extraction with chill separation of miscella into lower oil-rich phase, and an upper, solvent-rich recyclable phase save much energy of vaporization for distilling An aqueous extraction process with immiscible solvent assisted was tested.

Solvent like hexane added to SNC, and water added later with continuous stirring. The mixture was stirred for about 30 min, prior to centrifugation. Aqueous process extracted less amount of oil compared to solvent extraction

ACCESSION NUMBER: 2001:620412 CAPLUS DOCUMENT NUMBER: 135:343576

Processing of spectation. 2001:620412 CAPLUS
135:343576
Processing of spent nickel catalyst for fat
recovery
Nasir, Mohammad Ibraim
Research Department, Jahan Vegetable Oil Co., Tehran, AUTHOR(S): CORPORATE SOURCE: 19697, Iran Ciencia e Tecnologia de Alimentos (2001), 21(1), CODEN: CTALDN; ISSN: 0101-2061 Sociedade Brasileira de Ciencia e Tecnologia de Alimentos PUBLISHER:

> THERE ARE 18 CITED REFERENCES AVAILABLE FOR RECORD. ALL CITATIONS AVAILABLE IN THE RE

English

ANSWER 22 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN NOx emissions contribute for acid rain, greenhouse effect and formation

of ozone in the atmospheric Because of that, environmental demands are forcing the reduction of NOx emissions in mobile and static sources. On the other hand, due to the increase of cracking of heavy feed stock, oil refineries are emitting more sulfur and nitrogen compds. Hence, the reduction and understanding of conditions that affect NOx emissions has a great importance. This work evaluates the effect of contaminant metals in catalyst and the regeneration conditions on the NOx emission in the conditions of FCC process. Typical resid cracking catalyst containing contaminant metals were tested. Different metal levels were obtained by d. separation of an equilibrium catalyst. A simulated equilibrium catalyst was also tested for comparison. NO reduction was made in typical corditions.

Lyst was also tested for comparison. No reduction was made in typical conditions of an FCC regenerator. Partial and total combustion were simulated varying relative amount of CO and O2. No reduction to N2 varied with the type of combustion and amount of N1 an V in the catalyst. NO reduction in total combustion was very low. For partial combustion, the amount of N0 reduced varied according to the amount of contaminant

the amount of NO reduced varied according to the amount of contaminan
in the catalyst. This way, evaluating of NOx additives must be done
taking into account catalyst regemeration condition in
the unit and the amount of Ni and V in the equilibrium catalyst.

ACCESSION NUMBER:

AUTHOR(S):

AUTHOR(S):

CORPORATE SOURCE:

SOURCE:

SOURCE:

SOURCE:

PEUBLISHER:

PUBLISHER:

PUBLISHER:

COCUMENT TYPE:

LANGUAGE:

AND A Reductions under fluid catalytic
cracking conditions
Roncolatto, Rodolfo; Mello, Leonardo
Catalysis, Centro Pesquisas e Desenvolvimento da
Petrobras, Rio de Janeiro, N/A, Brazil
Abstracts of Papers - American Chemical Society
(2001), 221st, CATI-007
COODEN: ACSRAL; ISSN: 0055-7727
American Chemical Society
Journal; Meeting Abstract
English

L28 ANSWER 21 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The exhaust gas-purifing catalyst is that prepared by applying a
composition
containing a metal catalyst-bearing support, an electron-conducting rial, and an ion-conducting material on a substrate. Alternatively, the catalyst is made of a composition containing the metal catalyst-bearing support and an ion-conducting material, which is applied on an electron-conducting substrate. The catalyst, after S-poisoning, is desulfurized by heating that reduction of the S compound, adsorbed on the metal catalyst, by a reducing agent and oxidation of the reducing agent are electrochem. performed
as a result of transportation of electrons through the electron-conducting
material and of transportation of ions through the ion-conducting material.
ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE: 134:371126
Exhaust gas purifying catalyst and regeneration of sulfur-poisoned catalyst by electrochemical process Mitsuda, Noriaki; Katashiba, Hideaki; Ouchi, Hiroshi; Kishimoto, Yuji
Mitsubishi Electric Corp., Japan
Jpn. Kokai Tokkyo Koho, 8 pp.
CODEN: JKXXAF
Patent
Japanese
1 2001:388668 CAPLUS 134:371126 INVENTOR (S): PATENT ASSIGNEE(S): DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: KIND DATE APPLICATION NO. DATE JP 2001145837 PRIORITY APPLN. INFO.: 20010529

ANSWER 23 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Regeneration of a thiophene-poisoned Ni-supported catalyst was carried out by supercrit. CO2 extraction The catalyst activity was

red in the hydrogenation of 2-butanone to 2-butanol at 373 K and 1.7 MPa. The supercrit. extraction was tested over a range of operating conditions. Regeneration at 313 K and 41 MPa for 16 h completely

restored
the catalyst activity. Other methods cited in the literature were less effective in regeneration of Ni catalysts.

ACCESSION NUMBER: 2001:133167 CAPLUS
DOCUMENT NUMBER: 134:328183
TITLE: Regeneration of poisoned nickel catalyst by supercritical CO2 extraction

AUTHOR(S): Vadman, L.; Herskowitz, M.; Korin, E.; Wisniak, J. Blechner Center for Industrial Catalysis and Development and Chemical Engineering Department, Ben-Gurion University of the Negev, Beer Sheva,

84105,

Israel
Industrial & Engineering Chemistry Research (2001), 40(7), 1589-1590
CODEN: IECRED: ISSN: 0888-5885
American Chemical Society
Journal
English
8 THERE are SOURCE:

PUBLISHER: DOCUMENT TYPE: LANGUAGE: REFERENCE COUNT:

IN THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

DOCUMENT TYPE:

FORMAT

REFERENCE COUNT:

ANSWER 24 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

This paper focuses on ex situ methods for the regeneration of selective hydrogenation catalysts for olefin streams for downstream processing. Case studies presented include: (1) C2 front end acetylene hydrogenation; (2) C3 tail end acetylene hydrogenation; and (3) first stage pygas (pyrolysis gasoline) hydrogenation.

ACCESSION NUMBER: 2000:885911 CAPLUS
DOCUMENT NUMBER: 134:118154

Hydrogenation catalysts regeneration

AUTHOR(S): Van Leirsburg, Dean; van Keulen, Nick
CORPORATE SOURCE: CRI International, Inc., USA
Hydrocarbon Engineering (2000), 5(11), 46-48,50
CODEN: HYENFS

PUBLISHER: Palladian Publications Ltd
Journal
LANGUAGE: English

L28 ANSWER 25 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The amorphous NiB/SiO2 catalyst was deactivated in partial

hydrogenation of cyclopentadiene (CPD) containing 1000 ppm of carbon
disulfide. After the reactivity had vanished, the catalyst was almost
completely recovered by sequential base-wash and H2-sweep treatments.

base was alkaline ethanol. The results of XRD, DSC, XPS, EXAFS, IR,

SEM-EDX,
and H2 uptake measurements on the regenerated amorphous catalyst were
almost the same as those on original counterparts. The regeneration
operation was repeated several times, but its structural and chemical
properties remained.
ACCESSION NUMBER: 2000:604557 CAPLUS
DOCUMENT NUMBER: 133:301753
TITLE: Regeneration of the amorphous NiB/SiO2 catalyst
poisoned by carbon disulfide in cyclopentadiene

Zapurisous; Carpus 133:301753 Regeneration of the amorphous NiB/SiO2 catalyst poisoned by carbon disulfide in cyclopentadiene hydrogenation Wang, W.-O.; Li, H.; Li, H.-X.; Li, Y.-J.; Deng,

AUTHOR (S):

Department of Chemistry, Fudan University, Shanghai, 200433, Peop. Rep. China Applied Catalysis, A: General (2000), 203(2), 301-306 CODEN: ACAGE4; ISSN: 0926-860X Elsevier Science B.V. J.-F. CORPORATE SOURCE:

SOURCE:

PUBLISHER: DOCUMENT TYPE: LANGUAGE: REFERENCE COUNT:

Journal
English
35 THERE ARE 35 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

AB NH3 is treated with Ni catalysts for removal of O, CO, and CO2 simultaneous with reduction and regeneration of the Ni catalysts. The system does not require feeding of H and N for require feeding of H and N for require feeding of H and N for regeneration of the Ni catalysts and gives high-purity NH3.

ACCESSION NUMBER: 2000:405691 CAPEUS
DOCUMENT NUMBER: 133:45710 Furification of ammonia with nickel catalysts
INVENTOR(S): Hyano, Yasusada; Otsuka, Kenji: Waki, Hiroshi Japan Pionics, Ltd., Japan Jon. Kokai Tokkyo Koho, 5 pp.
CODEN: JEXNAF
DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

KIND DATE PATENT NO. APPLICATION NO. DATE JP 2000169138 PRIORITY APPLN. INFO.: 19981130 JP 1998-340248 JP 1998-340248 A2 20000620

L28 ANSWER 27 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The process comprises activating and reducing the toxic metal compound

AB The process comprises activating and reducing the toxic metal compound with activating gas and reductive gas at 300-600° and 0.1-1.0 MPa for 0.5-12 h to obtain metal, carbonylation with 20-991 CO gas at 70-150° and 0.65-0.95 MPa for 2-24 h to obtain gaseous metal oxo-compound for discharging, decomposing at 200-300° and 0.03-0.5 MPa for 0.5-12 h, collecting metal, and feeding CO back to the carbonylation process. The toxic metal is Ni and/or Pe. The activating gas is selected from HZS, and SOZ; and the reductive gas from HZ, and SOZ; and the reductive gas from HZ, and SOZ; and the reductive gas from HZ, and SOZ; and the reductive gas is (0.5-10):[90-99-5), preferably (2-6):[94-98).

ACCESSION NUMBER: 2000:201185 CAPLUS 200

DOCUMENT TYPE:

Patent Chinese

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE	
CN 1198366	A	19981111	CN 1997-109778	19970506	
CN 1099318	В	20030122			
US 6063721	A	20000516	US 1998-73586	19980506	
PRIORITY APPLN. INFO.:			CN 1997-109778 A	19970605	

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ANSWER 28 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
An amorphous NiB/SiO2 catalyst, deactivated in the partial
hydrogenation of cyclopentadinen to cyclopentene, was regenerated
by an oxidation-reduction method. After regeneration, the activity was
recovered and the selectivity almost recovered. The data of BET, H2
uptake and XPS measurements on the regenerated catalyst are almost the
same as the original counterpart. XRD and DSC tests revealed that some
crystallization had occurred over the regenerated catalyst, which led to
the selectivity decreasing slightly. This regenerated catalyst, which led to selectivity decreasing slightly. This regeneration was repeated several times, but the structural and chemical properties were nearly the same as after the first operation.

ACCESSION NUMBER: 2000:131074 CAPLUS DOCUMENT NUMBER: 132:142555
                                                                                                                                                                                                                                                   132:142585
Regeneration of amorphous NiB/SiO2 catalysts
deactivated in cyclopentadiene hydroganation
wang, Wei-Jiang; Li, Ne-Xing; Deng, Jing-Fa
Department of Chemistry, Fudan University, Shanghai,
200433, Peop. Rep. China
Journal of Chemical Technology 6 Biotechnology
       AUTHOR(S):
CORPORATE SOURCE:
                                                                                                                                                                                                                                                   75(2), 147-151
CODEN: JCTBED; ISSN: 0268-2575
John Wiley & Sons Ltd.
Journal
Journal
South State State
  PUBLISHER:
DOCUMENT TYPE:
LANGUAGE:
REFERENCE COUNT:
THIS
                                                                                                                                                                                                                                                                                                                  RECORD. ALL CITATIONS AVAILABLE IN THE RE
    FORMAT
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L28 ANSWER 30 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Raney catalysts for hydrogenation of nitriles to amines are
regenerated by mixing with aqueous base of basic ion concentration >0.01

regenerated by mixing with aqueous base of basic ion concentration >0.
mol/L,
heating the mixture at <130\*, and washing the treated catalyst with
water or base such that the final rinse water has a pH of 12-13. The
heating may be done under H and the degree of regeneration is 90-1001.
Examples employing soda or NAOH are given for Raney Ni catalyst
used to hydrogenate adiponitrile to hexamethylenediamine and/or
6-aminocapronitrile.
ACCESSION NUMBER: 120:65730
TITLE: Regeneration of catalyst used to hydrogenate
nitriles and its use
INVENTOR(S): Boschat, Vincent; Leconte, Philippe
Rhodia Piber and Resin Intermediates, Fr.
SOURCE: FI. Demande, 13 pp.
CODEN: FRXXBL
DOCUMENT TYPE: Patent

DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

APPLICATION NO. PATENT NO. KIND DATE DATE FATENT NO. KIND DATE APPLICATION NO. DATE

FR 2773086 A1 19990702 FR 1997-16832 19971229
FR 2773086 B1 20000211
CA 2316821 AA 19990708 CA 1998-2316821 19981223
W0 9933561 A1 19990708 W0 1998-FR2856 19981223
W1 BR, BY, CA, CN, CZ, ID, JP, KR, PL, RO, RU, SG, SK, UA, US, VN
RY, AT, BB, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
PT, SE
P 1042068 A1 20001011 EP 1998-963616 19981223
R: BE, DE, ES, FR, GB, IT, NL
BR 9614561 A 20011016 BR 1998-14561 19981223
JF 2001526956 T2 20011225 JP 2000-526298 19981223
RU 21904699 C2 20021010 RU 2000-120191 19981223
RU 2190469 C3 20030211 US 2001-120191 19981223
PRIORITY APPIN. INFO:: FR 1997-16832 A 19971229 WO 1998-FR2856 W 19981223

L28 ANSWER 29 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The invention relates to a method for regenerating a catalyst such as a
plate or honeycomb catalyst that is at least partially deactivated as a
result of toxic substances. The catalyst, e.g., a selective catalytic
reduction catalyst for nitrogen oxides removal from exhaust gases, is
treated with a gaseous reducing agent and a polyfunctional complex treated with a gaseous reducing agent and
forming
agent to eliminate said toxic substances.
ACCESSION NUMBER: 2000:66390 CAPLUS
DOCUMENT NUMBER: 132:82747
ITILE: Method for regenerati:
INVENTOR(S): Newfert, Ronald Method for regenerating a deactivated catalyst Neufert, Ronald Siemens Aktiengesellschaft, Germany PCT Int. Appl., 11 pp. CODEN: PIXXD2 Patent PATENT ASSIGNEE (S): SOURCE: DOCUMENT TYPE: LANGUAGE: FAMILY ACC, NUM. COUNT: PATENT INFORMATION: German 1 APPLICATION NO. KIND DATE PATENT NO. DATE W0 2000003804 A2 20000127 W0 1999-DE2067 19990705
W1 JP, US
RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE
DE 19832057 C1 20000316 DE 1998-19832057 19980716
EP 1098704 A2 20010516 EP 1999-945907 19990705
EP 1098704 B1 20030423
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, FI
JP 2002520153 T2 20020709 JP 2000-559935 19990705
T 238100 E 20030515 AT 1999-945907 19990705 T2 20020709 E 20030515 A1 20010607 B2 20030722 JP 2000-559935 AT 1999-945907 US 2001-761811 AT 238100 US 2001003116 US 6596661 PRIORITY APPLN. INFO.: 19990705 20010116

A 19980716

W 19990705

DE 1998-19832057 WO 1999-DE2067

L28 ANSWER 31 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
Thermogravimetry was used to follow the oxidation-reduction as well as
the regeneration processes of deactivated NiO/P-Al203 catalysts.
The catalysts, whose deactivation was caused by the sintering of metallic
Ni and by the forming of NiAl204 species, could be regenerated by
reduction-oxidation Especially, the regeneration of NiAl204 to NiO was

more
than 90%.
ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE:

2000:9975 CAPLUS
132:95492
Study on the regeneration of deactivated
N10/y-R1203 catalyst for the conversion of
natural gas with CO2 to synthesis gas by TG
Jinxiang, L.; Hengyong, X.
Dalian Institute of Chemical Physics, Chinese Academy
of Sciences, Dalian, Peop. Rep. China
Thermochimica Acta (2000), 343(1-2), 99-104
CODEN: THACAS; ISSN: 0040-6031
Elsevier Science B.V.
Journal AUTHOR(S): CORPORATE SOURCE:

SOURCE:

PUBLISHER: DOCUMENT TYPE: LANGUAGE: REFERENCE COUNT:

English
3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

L28 ANSWER 32 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The com. test of the selective hydrogenation catalyst RDD-1 at
Liaoyang Chemical Fiber Company is introduced. The results indicated that the catalyst was good in activity, selectivity and stability. The catalyst regeneration can be carried with air and steam for burning off the coke, and the activity of the regenerated catalyst was

similar to that of fresh catalyst. SSION NUMBER: 1999:424301 CAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: TITLE:

AUTHOR (S):

1999:424301 CAPLUS
131:9924
Commercial application of selective
hydrogenation catalyst RDD-1
Sun, Lianxia; Sun, Mingyong; Zhang, Dongping; Huang,
Xitai CORPORATE SOURCE: Research Institute of Petroleum, Beijing, 100083,

Research institute of Petroleum, Beljing, 10000 Peop. Rep. China Shiyou Lianzhi Yu Huaqong (1999), 30(5), 10-12 CODEN: SLYHEE; ISSN: 1005-2399 Shiyou Lianzhi Yu Huagong Zazhishe Journal Chinese

PUBLISHER: DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 34 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The hydrodesulfurization (HDS) catalytic activity of spent industrial catalyst NiMo/y-Al203 can be regenerated through oxidation below 723 K, calcination in air at 753 K for 3 h, and reduction at 673 K in H2 for 2 h and oxidation at 723 K in air for 4 h successively. The activity for

HDS of some regenerated samples can even be higher than that of the fresh catalyst sample.

ACCESSION NUMBER: 1999:46558 CAPLUS

DOCUMENT NUMBER: 130:141450

TITLE: Regeneration of The Accession of the Acce

AUTHOR (S):

130:141450
Regeneration of spent industrial NiMo/y-Al2O3
catallysts for HDS
Su, Jixin; Xiao, Tiancun; Wang, Haitao; Lu, Yuli;
Zhou, Changli; Li, Shuben; Zhang, Kongyuan
Department of Environmental Engineering, Shandong
University, Jinan, 250100, Peop. Rep. China
Yingyong Huaxue (1998), 15(6), 11-15
CODEN: YIHUED; ISSN: 1000-0518
Yingyong Huaxue Bianji Weiyuanhui
Journal PUBLISHER

DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 33 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

The deactivated amorphous NiB/SiO2 catalyst in the selective 
hydrogenation of cyclopentadiene (CPB) to cyclopentene (CPE) was 
regenerated by a new method, base-wash method at room temperature. The

was alkaline-ethanol solution After regeneration, the activity and selectivity were completely recovered. The results of XRD and DSC measurements showed

the amorphous structure of the regenerated amorphous catalyst is almost the same as that of the original counterpart. STEM-EDMX anal. revealed that the surface of the deactivated catalyst was covered by carbon. After

regeneration, the carbon was removed. BET, H2 chemisorption and XPS

indicated that the surface areas and chemical state were held on the indicated that the surface areas and chemical state were neid on the renewal catalyst. Such regeneration operation can be repeated several times without significant changes in the structural and chemical properties.

ACCESSION NUMBER: 1999:356468 CAPLUS

DOCUMENT NUMBER: 131:50101

TITLE: Regeneration at room temperature for amorphous NiB/SiO2 catalyst deactivated in cyclopentadiene hydrogenation

AUTHOR(S): Wang, Wei-Jiang; Li, He-Xing; Xie, Song-Hai; Li, Yong-Jiang; Deng, Jing-Pa

CORPORATE SOURCE: Applied Catalysis, Pudan University, Shanghai, SOURCE: Applied Catalysis, A: General (1999), 184(1), 33-39 COUNTENT TYPE: Elsevier Science B.V.

Journal

PUBLISHER: DOCUMENT TYPE: LANGUAGE: REFERENCE COUNT: English

THERE ARE 11 CITED REFERENCES AVAILABLE FOR

RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

L28 ANSWER 35 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The catalysts for hydrotreating process are widely used in petroleum
industry and more and more studies are focused on the reuse of spent
catalysts because of the environmental regulation and the high catalyst
consumption. NiMo/y-Al203 is one of the most important catalysts
for the hydrodesulfurization (HDS)/hydrodenitrogenation (HDN) in
petroleum

sleum industry. In this study, the effect of regeneration conditions on the physico-chemical properties and catalytic performance of the spent

ysts
was investigated. The regenerated samples were characterized by

was investigated. The Asymptotic Properties of regenerating nitrogen adsorption, XRD, SEM-EDS, IR and TPR techniques. The catalytic activities of regenerated samples were evaluated by catalytic hydrodesulfurization of diesel oil. The results show that for the

regenerated at a suitable temperature, there is only a little change in

textural structure, the aggregate of active phase in the oxidized samples may change the coordination states of some active elements, decrease the reduction temperature and improve the HDS activity of the regenerated samples. The presence of sulfate ions does not affect the HDS activity

samples. The profit of the regenerated samples.

ACCESSION NUMBER: 1598:777327 CAPLUS

DOCUMENT NUMBER: 130:68840

TITLE: Effect of regeneration process on structure and hydrodesulfurization activity of industrial NiMo/y-Al203 catalyst

AUTHOR(S): Su, Jixin; Xiao, Tiancun; Wang, Haitao; Lu, Yuli; Li, Shuben

Dep. Environ. Eng., Shandong Univ., Jinan, 250100, Shuben
Dep. Environ. Eng., Shandong Univ., Jinan, 250100,
Peop. Rep. China
Fenzi Cuihua (1998), 12(5), 362-366
CODEN: FECUEN; ISSN: 1001-3555
Zhongguo Kexueyuan Lanzhou Huaxue Wuli Yanjiuso

SOURCE:

PUBLISHER: DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 36 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Vanadium was recovered from spent catalyst by leaching with oxalic acid
and a oxidizing agent selected from Fe(NO3), Al (NO3), and H2O2. The
ebulated bed operation mod was more efficient then the fixed bed in
leaching of vanadium. The possibility of recovering of other metals (
NI, Mo, and CO) by leaching as well as the simultaneous
catalyst regeneration is also considered.
ACCESSION NUMBER: 1988:563347 CAPLUS
DOCUMENT MINERS.

ACCESSION NUMBER: DOCUMENT NUMBER: TITLE:

129:26230
Comparison between fixed-bed and ebullated-bed modes of reactor operations in leaching metals from spent residue hydroprocessing catalysts Stanislaus, Marafi A.; Jassem, F. Petroleum Technology Department, Kuwait Institute for Scientific Research, Safat, Kuwait Preprints - American Chemical Society, Division of Petroleum Chemistry (1998), 43(3), 486-490 CODEN: ACCAT: ISSN. 0569-3799 American Chemical Society, Division of Petroleum Chemistry Unurnal English

AUTHOR (S): CORPORATE SOURCE:

SOURCE:

PUBLISHER:

DOCUMENT TYPE: English

LANGUAGE: REFERENCE COUNT: THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

L28 ANSWER 38 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Title only translated.
ACCESSION NUMBER: 1998:331755 CAPLUS
DOCUMENT NUMBER: 128:313470
Method of regeneration of exhausted nickel—containing hydrogenation catalyst
Novocherkasski] Zavod Sinteticheskikh Produktov,
Russia
SOURCE: Russ. Prom: Tzobreteniva 1997, (36), 206.

Russ. From: Izobreteniya 1997, (36), 206. CODEN: RUXXE7

DOCUMENT TYPE:

Patent Russian 1 LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE RU 2100071 PRIORITY APPLN. INFO.: C1 19971227

L28 ANSWER 37 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Reactivation of spent Ni-Mo, Co-Mo, and Ni-W
hydrorefining, hydrotreating, hydrodesulfurization, and hydrocracking
catalysts is done in 2 stages. In the lst stage, the used catalysts are
washed with a mixture of a petroleum solvent (e.g., gasoline, diesel

light oil) containing 10-75 volume% aromatic hydrocarbons and a

light oil) containing 10-75 volumes aromatic hydrocarbons and a H-containing gas at a ratio of 100-3,000 m3/m3, 250-350\*, 5-200 bar, and space velocity of (0.5-2.5)/h. In the 2nd stage, the catalysts are subjected to hydrocracking to remove coke deposits by using a circulating gas containing 40-1001 Hz for 8-16 h at 380-450\* (preferably at 420\*), 5-200 bar (preferably 50-60 bar), and Hz/catalyst volume ratio of 500-1,000 m3/m3.

ACCESSION NUMBER: 1998:435928 CAPLUS

DOCUMENT NUMBER: 129:43154

TITLE: Reactivation of catalysts used in hydrocraption processes of petroleum fractions

1998:435926 CAPLUS
129:43154 Reactivation of catalysts used in hydrogenation processes of petroleum fractions Savu, Constantin; Ungureanu, Tefan; Nastasi, Adrian; Ghiula, Gheorghe; Avram, Ion; Erban, Mihai Institutul de Cercetari pentru Rafinarii si Petrochimie, S.A., Ploiesti, Rom. Rom., 7 pp.
CODEN: RUXXA3
Patent INVENTOR (S):

PATENT ASSIGNEE(S):

DOCUMENT TYPE: Patent Romanian LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE RO 1991-148554 RO 1991-148554 RO 106966 PRIORITY APPLN. INFO.: В1 19930830

L28 ANSWER 39 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN AB The denitration catalysts (containing active substances selected from

ates
of Ni, Fe, Co, Mn, and Zn) which are used for removal of NOx by
catalytic reaction of waste gases with NH3 at 80-180°, are
regenerated by washing the catalysts with active substance-containing

aqueous solns or H2O. The sulfate catalysts, which are active at low temps. are easily regenerated by the process.

ACCESSION NUMBER: 1998:157531 CAPLUS
DOCUMENT NUMBER: 1298:221047
TITLE: Regeneration of low-temperature denitration catalysts
TOYAO, Mamoruv Mitsuwa, Masaru; Suzumura, Hiroshi;
Tanaka, Hiroshi; Noshima, Shigeru; Nagayasu, Tachito
Chubu Electric Power Co., Inc., Japan; Mitsubishi
Heavy Industries, Ltd.
Jpn. Kokai Tokkyo Koho, 5 pp.
CODEN: JKXXAF

DOCUMENT TYPE: Patent

DOCUMENT TYPE: Patent

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: Japanese

PATENT NO. KIND DATE APPLICATION NO. DATE JP 10066875 PRIORITY APPLN. INFO.: A2 19960829 19960829 19980310 JP 1996-227963 JP 1996-227963

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ANSWER 40 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
Both Co-Mo and Ni-Co-Mo catalyst supported over y-alumina
were synthesized and compulsively deactivated by coking from
1,5-hexadiene. The coked catalysts, containing .apprx.8.8 wt% carbon and
.appxx.7 wt% sulfur, were regenerated by oxidative burnoff at various
temps. in the range of 300-700°C. The catalyst regenerated at each
temperature was characterized by various techniques; i.e., BET, XRD,
and
                    temperature was characterized by various techniques; i.e., BET, XRD, and
TPR. During regeneration, physicochem, properties such as surface area, crystallinity, reducibility, and metal distribution changed significantly with the regeneration temperature Increase in the dispersion of promoter species (cobalt or nickel) was observed in the catalysts regenerated at low temps. and this gave rise to the enhancement of activity in comparison to the fresh catalyst. On the other hand, promoters migrated into the sublayer of alumina support at higher temps. and thus resulted in the formation of PA1204 (P = Co or Ni) phases. Consequently, the crystallinity of the catalyst increased while the reducibility decreased as the regeneration temperature increased. brupt
An abrupt
                       increase in Mo dispersion and a decrease in surface area were observed
                    the coked catalyst was regenerated at 700°C.

SSION NUMBER: 1998:30391 CAPLUS

E: 128:77282

Physicochemical changes in hydrodesulfurization catalysts during oxidative regeneration

OR(S): Oh, Eun-Suok; Park, Yong-Chul; Lee, In-Chul; Rhee, Hyun-Ku

ORATE SOURCE: Department of Chemical Engineering, Seoul National University, Seoul, 151-742, S. Korea

CE: Journal of Catalysis (1997), 172(2), 314-321

CODEN: JCTLAS; ISSN: 0021-9517

Academic Press

Journal
when
ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE:
AUTHOR (S):
CORPORATE SOURCE:
 SOURCE:
  PUBLISHER:
 DOCUMENT TYPE:
LANGUAGE:
REFERENCE COUNT:
                                                                                                                English
17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR
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RECORD. ALL CITATIONS AVAILABLE IN THE RE

catalyst
Ramirez des Agudelo, Magdalena; Hernandez de Godoy,
Zaida; Navarro, Raul; Guerra, Julia
Intevep S.A., Venez.
Ger. Offen., 11 pp.
CODEN: GMXXBX
Patent
German DOCUMENT TYPE: German 6 FAMILY ACC. NUM. COUNT: PATENT INFORMATION: PATENT NO. APPLICATION NO. DATE KIND DATE 19971009 19990729 19981006 19971003 19991019 19971003 19990706 19971003 20000114 19981006 DE 1996-19634880 DE 19634880 A1 C2 19960829 DE 19634880 US 5817589 CA 2175327 CA 2175327 NL 1003041 19960402 19960429 US 1996-631605 CA 1996-2175327 A C A1 C2 A1 B1 NL 1996-1003041 19960506 NL 1003041 FR 2746673 FR 2746673 19960530 BR 1996-2559 JP 1996-169432 19960531 19960628 BR 9602559 JP 09271675 JP 2850304 19971021 19990127 US 1996-631605 PRIORITY APPLN. INFO.: A 19960402

L28 ANSWER 41 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The spent hydrogenation catalyst, which comprises a support of
an inorg oxide-reolite compound, carbon, and zeolites and a
catalytically

DOCUMENT NUMBER: TITLE:

INVENTOR(S): PATENT ASSIGNEE(S): SOURCE:

catalytically
active metal phase of partially reduced Group IB metal and/or completely
reduced Group VIII metals, is first treated with an inert gas (e.g., N,
He, Ar, methane, ethane, propane, or HI to remove hydrocarbon traces and
is then treated with hydrogen to recovery the diolefin
hydrogenation capacity.

ACCESSION HWHSER: 1397.6728322 CAPLUS

Regeneration of a petroleum hydrogenation catalyst

ANSWER 42 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

The present invention relates to processes of regenerating Ni catalysts which had been used in a hydrogenation of unsatd. fatty oil or petroleum resin, which comprise separating the Ni-extracted solution and support by extracting the pretreated Ni catalysts with an acid, preparing support-containing solution by burning the separated support in the flow of air or oxygen diluted with nitrogen at the temperature of 300° to 800° C. for 5 to 15 h and adding deionized water to the support, preparing a catalyst precursor by dropping the Ni-extracted solution and the mixed solution of a basic compound and a compound with free oxygen in the support-containing solution during agitation so as to keep pH of the ion at 9

to 13, whereby nickel oxide ppts. on the support, carrying out a
step consisting of aging, washing, filtering and drying the catalyst
precursor, and stabilizing the dried catalyst precursor by reducing with
hydrogen and passing in nitrogen diluted with oxygen or an organic solution material. ACCESSION NUMBER: . 1997:656888 CAPLUS DOCUMENT NUMBER: Processes of regenerating Ni catalysts and

INVENTOR (S):

Processes or regenerating Mx Catalysts and of preparing Ni Catalysts
Lee, Ro-in: Moon, Sang-heup: Hwang, Gyo-hyun; Coh,
Byung-youl; Hur, Seung-hyun; Han, Sung-hee: Park,
Heung-sun; Lee, Jong-hae
Lucky Engineering Co., Ltd., S. Korea; Seoul National
University

PATENT ASSIGNEE(S):

FORMAT

U.S., 8 pp. CODEN: USXXAM

Patent English 1

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE US 5674796 PRIORITY APPLN. INFO.: 19971007 US 1994-362845 KR 1994-17482

L28 ANSWER 43 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

Unpromoted and K-promoted Ni/Al203 catalysts were prepared,
calcined at 400 or 700 °C, and subsequently reduced at 500 or 800

°C. cyclopentane hydrogenolysis was carried out within the 370-500

°C temperature range to obtain a carbon deposit. After coking, the
catalysts were regenerated by carbon gasification in H2 at 800 °C.

TPO of the carbon deposits was carried out, and the filaments were examined
by SEM and TEM. The structural parameters (particle size, degree of nickel reduction, promotion by K) as well as the hydrogen flow rate showed marked effects on both the coking and the regeneration. The catalysts coked at high temperature (\*480 °C) and a high N2 flow rate. On the contrary, those which are coked at a lower temperature (\*420-460 °C) form filaments much more readily gasified by H2.

ACCESSION NUMBER: 1997:47341 CAPLUS
DOCUMENT NUMBER: 127:71341

TITLE: Regeneration of Nickel Catalysts Deactivated by Filamentous Carbon 1997: 473341 CAPIUS
127:71341
Regeneration of Nickal Catalysts Deactivated
by Filamentous Carbon
Duprez, D.: Fadili, K.: Barbier, J.
Laboratoire de Catalyse en Chimie Organique URA CNRS
350, Poitiers, 86022, Fr.
Industrial & Engineering Chemistry Research (1997),
36(8), 3180-3187
CODEN: IECRED: ISSN: 0888-5885
American Chemical Society
Journal AUTHOR(S): CORPORATE SOURCE: PUBLISHER: DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 44 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Cycloolefins are prepared by continuous partial hydrogenation of monocyclic aromatic hydrocarbons in the presence of Ru catalysts, metal salta, and R20 in a reactor whose wetted part is made of Ni -containing materials, and at least a part of the catalysts is extracted, washed with acids, and recycled in the reaction. The catalysts are preferably washed until the pH of the washing water used becomes 6-7, followed by reduction before recycling. Benzene was continuously fed to a reactor whose wetted part comprised Ni-Cr-Mo steel containing R20, ZnSO4, and Ru-Zn/S1O2 modified with ZcO3 (preparation given) under H at 150 and 5.0 MPa to give cyclohexene (1) at conversion 36: and selectivity 75% after 3 had 30 and 72%, resp., after 87.5%. The used catalyst Reparated rated from the catalyst slurry by filtration was repeatedly washed with H2O to remove Zn. The catalyst was further washed with an aqueous H2SO4 remove Zn. The Catalyst and tenses to solution to remove Ni and Fe, reduced under H, and reused to give I at conversion 20.0% and selectivity 90.2% after 75 min.

ACCESSION NUMBER: 1997:336156 CAPLUS
DOCUMENT NUMBER: 127:17439
Presention of cycloolefins by ruthenium 127:17439
Preparation of cycloolefins by ruthenium-catalyzed partial hydrogenation of monocyclic aromatic hydrocarbons
Suzuki, Toshuki; Matsuoka, Takeshi
Mitsubishi Chemical Industries Ltd., Japan
Jpn. Kokai Tokkyo Koho, 5 pp.
CODEN: JKKXAF
Patent INVENTOR (5): INVENTOR(5):
PATENT ASSIGNEE(S):
SOURCE:

DOCUMENT TYPE: Patent Japanese FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE A2 JP 09087209 PRIORITY APPLN, INFO.: 19970331 JP 1995-246057 JP 1995-246057

L28 ANSWER 46 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN AB The structural and morphol. changes of the metal phases which occur

ng regeneration in hydrogen and air/hydrogen cycles have been studied for Ni-USY catalysts which had been deactivated during the hydrogenation of benzene. Independent of nickel loading in the zeolite, regeneration in hydrogen alone was not sufficient to restore the initial activity. Regeneration in air/hydrogen cycles was more successful for catalysts with higher Ni loadings, but structural transformations of the metal particles occurred during the oxidative regeneration process. Temperature programmed reduction (TPR), photoelectron spectroscopy (XPS) and FTIR of adsorbed CO have been used

identify these modifications.

ACCESSION NUMBER: DOCUMENT NUMBER: TITLE:

AUTHOR (S):

fications.
1996:594892 CAPLUS
125:285933
Regeneration of Ni-USY catalysts used in
benzene hydrogenation
Pawelec, B.; Daza, L.; Fierro, J. L. G.; Anderson, J.

CORPORATE SOURCE:

A. Instituto de Catalisis y Petroleoquimica, CSIC, UAM, Cantoblanco, Madrid, 28049, Spain Applied Catalysis, A: General (1996), 145(1-2), 307-322 CODEN: ACAGE4; ISSN: 0926-860X Elsevier

SOURCE:

PUBLISHER: DOCUMENT TYPE: LANGUAGE: Journal English L28 ANSWER 45 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The process comprises regeneration of solid catalysts, that contain
\$\times 1\$ elements selected from Ti, Si, Al, Zr, Mn, Fe, Co, Ni,
N, Ce, Cu, Ag, Au, Pt, Pd, Rh, Ru, and Ir and whose activities are
lowered

by poisoning by alkaline earth metals or oxidation by excess O2 in wastewater wet
oxidation processes, with liqs. (pH 3-10) containing (NH4)2504, NH4C1,

(NH4)2CO3 or both the ammonium salts and NH3 practically in the absence

(NN4)2CO3 or both the ammonium salts and NN3 practically in the liquid phase.

The process provides efficient regeneration of catalysts by removal of Ca catalysts not catalysts by removal of Ca catalysts in the liquid phase.

ACCESSION NUMBER: 1997.186382 CAPLUS
126:176305
Regeneration of catalysts for wet oxidation treatment of wastewater
Ameda, Shinji; Ikeda, Mitsuaki; Ishii, Tooru; Mitsui, Kiichiro
Nippon Catalytic Chem Ind, Japan Jpn. Kokai Tokkyo Koho, 7 pp.
CODEN: JOXXAF
DOCUMENT TYPE: Patent
LANGUAGE: Japanese
1
PATENT INFORMATION: 1

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE JP 09010602 JP 3083463 PRIORITY APPLN. INFO.: 19970114 JP 1995-162475 19950628 JP 1995-162475 19950628

L28 ANSWER 47 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

A8 The catalysts containing 0.1-50% ≥1 of metals of 4th period and

0.01-10% ≥1 of Pt-group metals are regenerated by (1) heating at

100-250\* in 0-containing atmospheric, or (2) alternatively heating at

300-1000\* in steam and/or CO2-containing atmospheric A process for

300-1000\* in steam and/or CO2-containing atmospheric A process for preparation of C12C:CC12 (II) with the regenerated catalysts is also claimed. A spent catalyst of Cu- and Pt-loaded activated C was heated at 200° in air for 15 h, which was packed in a reactor, then fed with IT-H at 340° for 1 h to give I in 94.9% selectivity at 50.6% conversion.

ACCESSION NUMBER: 1996:497013 CAPLUS
DOCUMENT NUMBER: 125:142101
TITLE: Regeneration of catalysts and preparation of triblocarchylana wing them.

125:142101
Regeneration of catalysts and preparation of trichloroethylene using them
Suzuta, Tetsuya; Yokoi, Tatsuo; Ito, Naokazu;

INVENTOR(S):

Teruo Toa Gosei Kk, Japan Jpn. Kokai Tokkyo Koho, 6 pp. CODEN: JKXXAF Patent Japanese 1 PATENT ASSIGNEE(S): SOURCE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

DATE APPLICATION NO. KIND DATE JP 08155305 PRIORITY APPLN. INFO.: 19960618

ANSWER 48 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN The electrochem. reduction of organic compds. containing C(Hal)2, PCl2,

SiCl2 groups by using NA(0) complexes with PPh3, 2,2'-bipyridine, and 1,10-phenanthroline was studied. Mechanisms of the process are suggested, and effective rate consts. for catalyst regeneration are evaluated. SINT NUMBER: 1996:488194 CAPLUS MENT NUMBER: 126:12441

DOCUMENT NUMBER:

TITLE:

126:12401

Metal-complex catalysis in organic electrosynthesis.

Nickel (II) complexes with PPN3,

2,2'-bipyridine, and 1,10-phenanthroline, their redox
properties, and catalysis of reduction
reactions of dihalo organic compounds

Budnikova, Yu. G.; Petrukhina, O. E.; Kargin, Yu. M.

Arbuzov. A.E., Institut Organicheskoi Flzicheskoi

Khimii, Russia

Zhurnal Obshchei Khimii (1996), 66(4), 610-614

CODEN: ZOKHA4; ISSN: 0044-460X AUTHOR(S): CORPORATE SOURCE:

SOURCE:

PUBLISHER: DOCUMENT TYPE: LANGUAGE: Journal L28 ANSWER 49 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
Hydrocarbon oils containing asphaltene, S, and/or metal components are
hydroganated in the presence of catalyst, where the process
comprises treating deteriorated catalysts by reverse flow of the
hydrocarbon oils according to the deterioration of catalyst performance.
The hydrocarbon oils may be crude oils, naphthar-removed crude oils,
reduced crude, or vacuum distillation crude. The catalysts may contain alumina

alumina
supports loaded with group 6, 8, 9, and/or 10 metals, preferably
Ni-Mo, Co-Mo, Ni-W, or Ni-Co-Mo. Optionally,
the catalysts contain P or B.
ACCESSION NUMBER: 1996:191646 CAPLUS
DOCUMENT NUMBER: 124:236892

DOCUMENT NUMBER: TITLE:

124:236892
Rydrogenation of hydrocarbon oils for prolonging catalysts life
Iwamoto, Ryuichiro
Idemitsu Kogan Co, Japan
Jpn. Kokai Tokkyo Koho, 8 pp.
CODEN: JKXXAF
Patent
Japanese
1 INVENTOR(S): PATENT ASSIGNEE(S): SOURCE:

DOCUMENT TYPE: LANGUAGE: FAMILY ACC, NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE JP 07331254 PRIORITY APPLN. INFO.: 19940606 19940606 A2 19951219

ANSWER 50 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
The poisoning by thiophene and the regeneration of a Ni/Al203
catalyst during the selective hydrogenation of styrene at 353 K
was studied. It was determined by XPS and Fourier transform IR

spectroscopy
that the sulfur species adsorbed during the catalyst poisoning is
thiophene, which is adsorbed coplanarly to the surface. The catalyst was
regenerated by H2 treatments for different times and at several
pressures.

Part of the sulfur remained irreversibly adsorbed after regeneration, and hence the activity and selectivity values corresponding to the

nonpoisoned arter regeneration, and catalyst were never recovered. After the regeneration treatments at 473 K, a modification on the adsorbed sulfur electronic state was detected, which can be ascribed to thiophene hydrogenolysis, thus producing superficial sulfide adsorbed species.

ACCESSION NUMBER: 1995:875119 CAPLUS DOCUMENT NUMBER: 123:255977

TITLE: Regeneration of MICES.

143:2009//
Regeneration of Ni/Al203 Poisoned by
Thiophene during the Selective Hydrogenation
of Styrene

of Styrene L'Argentiere, Pablo C.; Liprandi, Domingo A.; Figoli,

AUTHOR (S):

L'Argentlere, Paolo C.; Liprandi, Domingo A.; Figoli Nora S. Instituto de Investigaciones en Catalisis y Petroquimica, Santa Pe, 3000, Argent. Industrial & Engineering Chemistry Research (1995), 34(11), 3713-17 CODEN: IECRED; ISSN: 0880-5885 American Chemical Society

CORPORATE SOURCE:

SOURCE:

PUBLISHER:

DOCUMENT TYPE: LANGUAGE:

Journal English

L28 ANSWER 51 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Title only translated.

ACCESSION NUMBER: 1995:721224 CAPLUS

DOCUMENT NUMBER: 123:86563

ITILE: Reactivation of catalyst for hydrogenation of vegetable oil

NVENTOR(S): Melamud, Naum L.; Noskova, Nina F.; Ryzhova, Rozaliya Ya.; Savelev, Sergej R.; Korneev, Nikolaj N.; Khrapova, Irina M.

PATENT ASSIGNEE(S): Gosudarstvennyj Nauchno-Issledovatelskij Institut Khimii i Tekhnologii El Amicheskikh Soedinenii, Russia; Institut Organicheskogo Kataliza i Elektrokhimii im.D.V. Sokolskogo

SOURCE: RUXXE7

DOCUMENT TYPE: Patent

DOCUMENT TYPE:

Patent Russian

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. APPLICATION NO. KIND DATE DATE RU 2001941 PRIORITY APPLN. INFO.: C1 19931030

L28 ANSWER 52 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Catalytic composites of the reaction product of a metal halide (e.g.,
AlCl3) having Friedel-Crafts activity with the bound surface OH groups of
inorg. oxides (e.g., alumina) and containing a zero-valent metal (e.g., Pt)

with hydrogenation activity, deactivated during use as Catalysts in the manufacture of alkylates (e.g., from 2-butene-isobutane mixture) 

DOCUMENT TYPE: Patent English 2

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PR

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
			*			
US 5391527	A	19950221	US 1993-172960	19931227		
US 5310713	А	19940510	US 1993-43954	19930405		
AT 169843	E	19980915	AT 1994-303093	19940428		
ES 2119082	Т3	19981001	ES 1994-303093	19940428		
RIORITY APPLN. INFO.:			US 1993-43954 A	2 19930405		

EP 1994-303093 A 19940428

L28 ANSWER 54 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Gas containing NOx and entrained catalyst fines is passed up through a

catalyst at a sufficient velocity to expand and fluidize the catalyst

a catalyst are abraded or elutriated awa low pressure drop design, with vertical, perforated gas distributors covered by the bed.

ACCESSION NUMBER: 1995:599691 CAPLUS
DOCUMENT NUMBER: 123:36930
TITLE: Final-Particulates deposited on the catalyst are abraded or elutriated away by fluidization, preventing fouling of the DeNOx catalyst. The reactor has

1995:599691 CAPLUS 123:36930 Fines tolerant SCR reactor in FCC process for flue

cleanup Chou, Tai Sheng Mobil Oil Corp., USA U.S., 9 pp. CODEN: USXXAM Patent English INVENTOR(S):
PATENT ASSIGNEE(S):
SOURCE:

DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

APPLICATION NO. PATENT NO. KIND DATE DATE US 5413699 PRIORITY APPLN. INFO.: 19950509 US 1993-136053 US 1993-136053 19931014 19931014 L28 ANSWER 53 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The title system comprises means for removing and concentrating CO2 from the

polluted respiratory air from human or animals, means for supplying a portion of the concentrated CO2 to plants and recovering the generated

portion of the concentrated CO2 to plants and recovering the generated O2 for human or animals, means for introducing the remaining portion of CO2 into a lat catalytic reactor to convert CO2 over a NA catalyst at apprx. 400° to form CI4 and H2O, means for electrolyzing the H2O into H12 and O2, means for reacting the formed CH4 over an Fe catalyst on refractory oxide support at apprx. 1000° in a 2nd catalyst reactor to convert CH4 into C and H2. means for utilizing the formed H2 to regenerate the spent catalyst in the lat catalytic reactor, means for combusting the formed C with O2 from the electrolysis cell to form CO2 in the closed cycle. The system is useful in manned space vehicles.

DOCUMENT NUMBER: 123:91967 close-type environmental control system and regeneration of catalysts for carbon dioxide removal Sawada, Toyoo; Kajima, Kazuhiro; Octauji, Kaoru Hitsubishi Heavy Industries, Ltd., Japan Jon. Kokai Tokkyo Koho, 6 pp. CODDE: JOXXAF

DOCUMENT TYPE: Patent Japanese

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE JP 07047229 PRIORITY APPLN. INFO.: A2 19950221 JP 1993-195996 JP 1993-195996 19930806

L28 ANSWER 55 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Cyclic process for catalytically upgrading feedstock and for upgrading
the activity of aged catalyst comprising the steps (1) of supplying feedstock and olefin at a paraffin to olefin ratio greater than 5 volume/volume to

reactor containing a solid acid catalyst and removing effluent comprising upgraded product and (b) of exposing the catalyst to a hydrogeneting medium and removing a hydrocarbonaceous effluent wherein the catalyst comprises zeolite beta crystals.

ACCESSION NUMBER: 1995:551081 CAPLUS

DOCUMENT NUMBER: 122:269848

TITLE: Process for alkylating a paraffinic feedstock by condensation of paraffins with olefins

INVENTOR(S): van Brugge, Paulus Theordorus Maria; de Groot, Christoffel; Mesters, Carolus Matthias Anna;

Danny Gaston Rene Shell Internationale Research Maatschappij B. V., PATENT ASSIGNEE(S):

Shell Internationale R Neth. Eur. Pat. Appl., 9 pp. CODEN: EPXXDW Patent SOURCE:

DOCUMENT TYPE:

English

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO		KIND	DATE	APPLICATION NO.	DATE
EP 640575		A1	19950301	EP 1994-202421	19940824
EP 640575		B1	19970514		
R: A	r, BE, CH,	DE, DE	, ES, FR,	GB, GR, IT, LI, LU, N	L, SE
CA 212979	7	AA	19950227	CA 1994-2129797	19940809
AU 947023	9	A1	19950309	AU 1994-70239	19940811
AU 673445		B2	19961107		
JP 070821	76	A2	19950328	JP 1994-220828	19940824
AT 153014		E	19970515	AT 1994-202421	19940824
ES 210276	6	T3	19970801	ES 1994-202421	19940824
FI 940392	4	A	19950227	FI 1994-3924	19940826
PRIORITY APPLN	. INFO.:			EP 1993-202515	19930826

L28 ANSWER 56 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
As cheme of production of nickel catalysts using as a substrate spent catalysts for oil hydrogenation is provided. It involves leaching in nitric acid in a presence of oxygen and separation of reaction
mixture The part of the solution containing mickel nitrate and silica is
divided using a centrifuge. The fatty layer is added back to the fat
reservoir. The nickel nitrate could be used for the production of
nickel catalysts by precipitation
SSION NUMBER: 1995:455377 CAPLUS
MENT NUMBER: 122:223777
E: 0il hydrogenation spent catalysts as a raw
material for mickel catalyst production
OR(S): Golebiowski, Andrzej; Gorecki, Andrzej; Gorska, Anna;
Lewandowski, Tadeusz; Jesiolowski, Jerzy; Giryn,
Krzysztof ACCESSION NUMBER: DOCUMENT NUMBER: TITLE: AUTHOR (S): Krzysztof Inst. Nawozow Sztucznych, Pulawy, Pol. Chemik (1994), 47(12, BTS), 391 CODEN: CHGLAY; ISSN: 0009-2886 Wydawnictwo SIGMA-NOT CORPORATE SOURCE: PUBLISHER: DOCUMENT TYPE: LANGUAGE: Journal Polish

ANSWER 58 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN A process for the reactivation, or rejuvenation of a nickel—alumina catalyst employed in the production of a gas comprised of an admixt of hydrogen and carbon monoxide, or synthesis gas, by the conversion, in reactor, or reaction zone, of light hydrocarbons in a fluidized bed of

catalyst at elevated temperature, in the presence of steam and oxygen.

Catalyst at executed the catalyst reactivation is accomplished by withdrawing a portion of the catalyst

the fluidized bed of the reactor and treating the catalyst in an

the fluidized bed of the reactor and treating the country of the nickel zone at temperature sufficient to oxidize and convert the nickel component of the catalyst to nickel aluminate and disperse said nickel aluminate within the alumina support, and then recycling the treated catalyst to the reactor, or reaction zone, to reactivate and increase the activity of the catalyst. The catalyst, on reduction in the reactor, or reaction zone, is provided an addnl. boost in activity by washing, treating or contacting the catalyst from the elevated

erature oxidation zone with an acid sufficient to remove trace impurities without removing any substantial amount of the nickel aluminate, and without forming a residue on the catalyst surface. The normal tendency

the catalyst to agglomerate at reaction conditions is also reduced by the

acid treatment.
ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE:

1995:227425 CAPLUS
122:13547
Process for the reactivation of nickel
-alumina catalysts
Clavenna, Leroy R.; Davis, Stephen M.; Beasley, Brent INVENTOR (S):

E. Exxon Research and Engineering Co., USA PATENT ASSIGNEE(S): SOURCE:

U.S., 9 pp. CODEN: USXXAM Patent English 1 DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	US 5356845	A	19941018	US 1993-60333	19930511
	CA 2121942	AA	19941112	CA 1994-2121942	19940422
	CA 2121942	С	20001121		
	NO 9401715	A	19941114	NO 1994-1715	19940509
	AU 9461982	A1	19941117	AU 1994-61982	19940509
	AU 664527	B2	19951116		
	EP 624400	A1	19941117	EP 1994-303329	19940509
	EP 624400	В1	19970910		
	R: DE, FR, GE	, IT, NI			
PRIC	RITY APPLN. INFO.:			US 1993-60333 A	19930511

L28 ANSMER 57 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Transition metal-containing organic solvents separated from olefin
polymerization systems
are regenerated by hydrogenation in the presence of transition
metal catalysts supported by alkali or alkaline earth metal compds.

hexane from an olefin polymerization system was hydrogenated in the presence of K2CO3-supported Pd/alumina and used in polymerization of

propylene in the presence of Ti catalysts, AlEt3, and Me p-toluate at 60° under 7 kg/cm2 gauge for 2 h to prepare polypropylene with polymerization activity 306 kg-polymer/g-Ti.

ACCESSION NUMBER: 1995:300329 CAPLUS DOCUMENT NUMBER: 122:161741

TITLE: Catalysts and process for

1995:300329 CAPLUS
122:161741
Catalysts and process for regeneration of organic solvents for olefin polymerization
Torihata, Takashi: Nishimura, Satoyuki
Mitsui Petrochemical Industries, Co., Ltd., Japan
Jpn. Kokai Tokkyo Koho, 9 pp.
CODEN: JKXXAF
Patent
Japanese 1

INVENTOR(S): PATENT ASSIGNEE(S): SOURCE:

DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE JP 06293808 PRIORITY APPLN. INFO.: A2 19941021 JP 1994-17328 JP 1993-25503 19940214

L28 ANSWER 59 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The activity of catalysts based on the oxides of Cu, Fe, Co, Ni,
Cr, Mn, and V applied on γ- and α-Al2O3 in NO removal from
flue and waste gases was studied under nonstationary chemisorptionregeneration conditions of the process based on the time-separation of regeneration conditions of the process based on the time-separation or stages of NO heterogeneous catalytic decomposition. The effect of the catalyst promotion by Pd at the gas purification stage and during catalyst regeneration by reduction with H is discussed.

ACCESSION NUMBER: 1994-611848 CAPLUS

DOCUMENT NUMBER: 121:211848

TITLE: Activity of transition metal oxide based catalysts in gas purification by nonstationary process for NO removal.

AUTHOR(S): Chernobaev, I. I.; Kuznetsov, V. A.; Vlasenko, V. M. CORPORATE SOURCE: Ukrainakii Khimicheskii Zhurnal (Russian Edition) (1993), 59(9), 939-44

CODEN: UKZHAU; ISSN: 0041-6045

DOCUMENT TYPE: Journal

LANGUAGE: Russian

L28 ANSWER 60 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The extent of hydrogenation of because The extent of hydrogenation of benzene over catalysts obtained by depositing Ni on the zeolite USNY depends on the the amount of exchanged Ni. Ni(0) is the active species. The catalyst was regenerated to a great extent in an air atmospheric, but

not under hydrogen.
ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE:

SOURCE:

1994:604885 CAPLUS

121:204885 Regeneration of Ni-USY catalysts used for

Megeneration of MI-UST Catalysts used for benzene hydrogenation.
Anderson, J.A.; Daza, L.; Fierro, J.L.G.; Pawelec, B. Instituto de Catalisis y Petroleoquimica, S.S.I.C., Madrid, Spain
Revista de la Real Academia de Ciencias Exactas, Fisicas y Naturales de Madrid (1993), 87(1), 139-46
CODEN: RCENAT; ISSN: 0034-0596 AUTHOR(S): CORPORATE SOURCE:

DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 61 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Catalytic composites of the reaction product of a metal halide having Friedel-Crafts activity with the bound surface hydroxyl group of inorg. oxides and containing a zerovalent metal with hydrogenation activity, often are effective catalysts in motor fuel alkylation which, however, undergo rapid deactivation. Deactivated catalysts are readily regenerable by treating the composite from which alkylate feedstock was removed with hydrogen at temps. in the range of 10 to 300°.

Multiple regenerations are possible without appreciable activity loss.

ACCESSION NUMBER: 1994:439011 CAPLUS
DOCUMENT NUMBER: 121:39011
TITLE: Regeneration of an alkylation catalyst with hydrogen INVENTOR(S): Kojima, Masami, Kocal, Joseph A.

PATENT ASSIGNEE(S): U.S., 5 pp.
CODEM: USXXAM
DOCUMENT TYPE: Patent
LANGUNGE: English
PAMILY ACC. NUM. COUNT: 2
PATENT INFORMATION: 2

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 5310713	A	19940510	US 1993-43954	19930405
US 5391527	A	19950221	US 1993-172960	19931227
CA 2122220	AA	19951027	CA 1994-2122220	19940426
CA 2122220	С	20040629		
AU 668288	B2	19960426	AU 1994-60705	19940427
AU 9460705	A1	19951116		
EP 679437	A1	19951102	EP 1994-303093	19940428
EP 679437	B1	19980819		
R: AT, BE, CH,	DE, DK	, ES, FR, GE	3, GR, IT, LI, NL, PT.	. SE
AT 169843	E	19980915	AT 1994-303093	19940428
ES 2119082	т3	19981001	ES 1994-303093	19940428
CZ 285674	В6	19991013	CZ 1994-1045	19940429
2A 9403013	А	19950127	2A 1994-3013	19940502
RU 2128549	C1	19990410	RU 1994-19493	19940506
CN 1144141	A	19970305	CN 1994-105385	19940507
CN 1068245	В	20010711		200.000.
JP 07299363	Ã2	19951114	JP 1994-94671	19940509
KR 9711082	Bl	19970707	KR 1994-10063	19940509
PRIORITY APPLN. INFO.:	<i>D</i> 1	155.0707	US 1993-43954	A2 19930405
RIGHTI AFFIN. INFO			02 1333-43334	AZ 19930403

EP 1994-303093

A 19940428

L28 ANSWER 62 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Activation and regeneration procedures applied to a Ni-Mo/Al2O3
catalyst, both fresh and spent, were tested by hydrodesulfurization of
thiophene. Characterization techniques used included

thiophene. Characterization techniques used included temperature-programmed reduction and state of the programmed reduction and state of the programmed reduction and state of the programmed reduction and state of the state of

AUTHOR (S):

Martin CORPORATE SOURCE:

Universidade Federal do Rio de Janeiro, Rio de

DOCUMENT TYPE: LANGUAGE:

121:38857 Activation and Regeneration of a NiMo/Al2O3 Hydrotreatment Catalyst Teixeira da Silva, V. L. S.; Frety, R.; Schmal,

Universidade Federal do Rio de Janeiro, Rio de Janeiro, Brazil Industrial & Engineering Chemistry Research (1994), 33(7), 1692-9 CODEN: IECRED; ISSN: 0888-5885 Journal English

L28 ANSWER 63 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Ni-W hydrotreating catalysts for heavy-oil refining were studied
with respect to deactivation and regeneration. The catalysts deactivated
by surface accumulation of Si, V, S, and C components and by the
resulting
pore blockage. Temperature-programmed oxidation (TPO) of the
deactivated catalysts
showed that three kinds of the sulfur and one kind of the carbon species
with different oxidation activity existed on the catalyst surface. The
s and

C species were removed in two steps during oxidation in a dilute O at 450° (i.e., a rapid removal during the first hour and a slow and steady removal thereafter). The surface species were removed more easily when the oxygen content in the regeneration stream was 0.05 atm instead

when the oxygen content in the regeneration stream was 0.05 atm instead of 0.13 atmospheric The activity of the regenerated catalyst in propylene hydrogenation was also higher when the regeneration gas stream contained a lower amount of oxygen, in accordance with the result of surface species removal. The activity difference was small between the two catalysts regenerated for either 1 h or 7 h because most of the surface species were removed during the initial 1 h. The phys. properties of the regenerated catalysts (e.g., the BET surface area, the pore volume, the average pore diameter and the pore size distribution) were correlated with the extent of activity recovery.

ACCESSION NUMBER: 1994:327136 CAPLUS
DOCUMENT NUMBER: 120:327136
TITLE: Regeneration of nickel-tungsten hydrotreating catalysts
AUTHOR(S): Kim, 11 Sup; Han, Jeong Hyun; Moon, Sang Heup Dep. Chem. Eng., Seoul Natl. Univ., S. Korea Numbak Konghak (1993), 31(4), 483-91 CODEN: HHKHAT: ISSN: 0304-128X
DOCUMENT TYPE: Journal Korean

L28 ANSWER 64 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The waste catalyst in residual catalytic cracking (RCC) can be regenerated by using a chemical transport process. Thermodn. anal. and exploratory tests

for the removal of the nickel deposit on poisoned catalyst are
conducted. Reduction of nickel oxides and carbonylation
of nickel are carried out consequently in fluidization process.
Effects of degree of nickel compound reduction and
carbonylation pressure on the activation of RCC catalysts are examined
ACCESSION NUMBER: 1994:302926 CAPLUS
DCCUMENT NUMBER: 120:302926
TITLE: themical transport: study on regeneration of RCC
DDISONGE CAPLUS
DDISONGE CAPLUS poisoned catalyst Luo, Baolin; Sun, Kuiyuan; Yang, Chengyan; Xu, AUTHOR (S): Junding CORPORATE SOURCE: Inst. Chem. Metall., Chin. Acad. Sci., Beijing, 100080, Peop. Rep. China Huagong Yejin (1993), 14(4), 370-5 CODEN: NUYEEF; ISSN: 1001-2052 Journal Chinese

ANSWER 66 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
A method is provided for regenerating a mol. sieve-free resid
hydroprocessing catalyst for use with an ebullated bed reaction process
comprising 21 hydrogenation metal and at least one Group
IIA metal deposited on an inorg. oxide support wherein the catalyst
contains a pore volume of pores having a diameter >1200 Å of >0.05 mi/g.
The method comprises the steps of contacting the mol. sieve-free resid
hydroprocessing catalyst with a contaminant metal-containing hydrocarbon
feedstream in a first contacting step at conditions sufficient to deposit
contaminant metals and coke onto the catalyst, and contacting the
coke-deactivated, contaminant metal-containing, mol. sieve-free catalyst ... an oxygen-containing gas in a second contacting step at oxidation conditions conditions

sufficient to remove a substantial amount of the coke from the coke-deactivated, contaminant metal-containing, mol. sieve-free catalyst. The loss on attrition of the regenerated mol. sieve-free resid hydroprocessing catalyst after the second contacting step is <10% by weight of fines per day calculated based on a regeneration temperature of .apprx.900° F.
ACCESSION NUMBER: 1994:111519 CAPLUS
DOCUMENT NUMBER: 120:111519
TITLE: Process for regenerating a spent resid catalyst using a Group IIA metal Clark, Frederick T.; Hensley, Albert L., Jr. Amoco Corp., USA U.S., 14 pp. Cont.-in-part of U.S. Ser. No. 785,452, abandoned. hydroprocessing INVENTOR(S): PATENT ASSIGNEE(S): SOURCE:

PATENT NO. KIND DATE APPLICATION NO. DATE 19921002 19911031 US 5275990 19940104 US 1992-955496 PRIORITY APPLN. INFO.:

CODEN: USXXAM Patent

English

L28 ANSWER 65 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The oxidative regeneration of spent cobalt-molybdate and nickel
-tungstate catalysts from hydrotreatment of petroleum vacuum gas oil and
coal-derived oil was carried out in a fixed bed reactor.
Temperature-programmed
oxidation studies revealed that oxidation proceeded mainly in two steps removal of sulfidic sulfur as SO2 at around 500-600 K and removal of carbon as CO2 and CO at around 650-850 K). Carbonaceous materials on the spent Ni-W catalyst were less aromatic than those on the spent Co-Mo catalyst, but more severe oxidation conditions were needed for the Ni-W catalyst because of the lower oxidation activity of NiO and WO3 compared with Co304 and WO3. For the Ni-W catalyst, EXAR'S data revealed that WS2-like structures, which were laterally grown during the hydrotreatment run, were redispersed to nearly the same level as that of the fresh catalysts when carefully controlled oxidizing conditions were used (1.5% O). XPS data showed that surface compns. of Ni and W were recovered to almost the level of fresh catalysts, but the Ni -W ratio was slightly less than that of the fresh ones. Catalytic activities and selectivities were successfully recovered by temperature

activities and selectivities were successfully recovered by

low-temperature
oxidation On the contrary, for the Co-Mo catalyst on which MoS2-like
sulfides were laterally grown, some of the Co aggregated to Co9S8, and
small ants. of Ni, Fe, and V were deposited, it was not possible
to recover the same level of structural properties as those of the fresh
catalysts. The catalytic activities and selectivities were almost
recovered by low-temperature oxidation, while at higher regeneration

temps. there
was a slight loss of hydrogenation activity and a large increase
in the hydrocracking activity.

ACCESSION NUMBER:
1994:168314 CAPLUS

DOCUMENT NUMBER:
117ITE:
120:168314

AUTHOR(S):
70shimura, Y.: Sato, T.: Shimada, H.: Matsubayashi,
N.: Imamura, M.: Nishijima, A.: Yoshitomi, S.:
Kameoka, T.: Yanase, H.
CORPORATE SOURCE:
Research,
Tsukuba. 305, Japan

Tsukuba, 305, Japan Energy & Fuels (1994), 8(2), 435-45 CODEN: ENFUEM; ISSN: 0887-0624 Journal English

DOCUMENT TYPE:

contact with a hyd hydrocarbons from the off-gas. ACCESSION NUMBER: DOCUMENT NUMBER: TITLE: INVENTOR(S): PATENT ASSIGNEE(S): SOURCE: 1994:34361 CAPLUS
120:34361 Nitrogen oxide removal process
Carnell, Peter John Herbert
Imperial Chemical Industries PLC, UK
EUR. Pat. Appl., 5 pp.
CODEN: EPXXDW
Patent
English 1 DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: KIND DATE PATENT NO. APPLICATION NO. DATE EP 571130 A1
EP 571130 B1
R: DE, FR, GB, IT, NL
CA 2096774 AA
US 5340554 A
PRIORITY APPLN. INFO.: 19931124 19960320 EP 1993-303676 19930512 CA 1993-2096774 US 1993-65391 GB 1992-10926 19931123 19940823 19930521 19930521 19920522

GB 1992-24659

19921125

L28 ANSWER 67 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN AB A refinery catalyst is regenerated by burning off coke in an air stream give an off-gas; nitrogen oxides (NOx) are removed from the off-gas by contact with a hydrogen catalyst before cryogenic recovery of

DOCUMENT TYPE:

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

ANSWER 68 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

In the title method, the feed oil containing 1-methylnaphthalene (I) is
subjected to hydrogenation. The resulting oil is then subjected
to the isomerization process. Thus I was fed to a hydrogenation
reactor containing Mi and H2. The product contained mainly I and
0.21 1-methyltetralin (II). The product was fed to a reactor containing
protonically exchanged Y zeolite at 450° to give
2-methylnaphthalene with 70% conversion and 90% selectivity, vs. 25%
conversion in a process using a feed oil which does not contain II. II
restored the activity of the isomerization catalyst.
SSION NUMBER: 1993:673869 CAPLUS
MENT NUMBER: 119:273869

E: Method for preparation of 2-methylnaphthalene by

ACCESSION NUMBER:

DOCUMENT NUMBER: TITLE:

Method for preparation of 2-methylnaphthalene by isomerization of 1-methylnaphthalene Suzuki, Toshihide; Nobusawa, Tatsuya; Takagi, Yoshinori.

INVENTOR (S):

PATENT ASSIGNEE(S): SOURCE:

TOSNINOTI Kawasaki Steel Co, Japan Jpn. Kokai Tokkyo Koho, 5 pp. CODEN: JKXXAF Patent

DOCUMENT TYPE: Japanese LANGUAGE:

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE JP 1992-50907 JP 1992-50907 19920309 JP 05246907 PRIORITY APPLN. INFO.: A2 19930924 19920309

ANSWER 70 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
A method for regenerating a contaminant metal-containing, residual oil
hydroprocessing mol. sieve-free catalyst for use with an ebullated bed
reaction process and having ≥1 hydrogenation metal and
≥1 Group IV metal deposited on an inorg. oxide support comprises
contacting the catalyst with an O-containing gas under conditions

sufficient
to remove a substantial amount of coke from the catalyst. The
catalyst regeneration process involves an initial
partial coking step where the spent catalyst is contacted with an
O-containing
gas at 400-700's, followed by the addition of a Group IV metal such
that the partially decoked catalyst contains 0.1-20 weight of the Group
IV

metal calculated as the oxide and based on the fresh weight of the

catalyst, a final decoking step at 600-1400°F. The presence of the Group IV metal on the catalyst during regeneration increases the

attrition resistance of the catalyst.
ACCESSION NUMBER: 1993:563819 CAPLUS
POCUMENT NUMBER: 119:163819 119:163819

DOCUMENT NUMBER: TITLE: hydroprocessing

Process for regenerating a spent resid

catalyst using a Group IV metal Clark, Frederick T.; Springman, Mary C. Amoco Corp., USA U.S., 15 pp. CODEN: USXXXAM Patent English

INVENTOR(S):
PATENT ASSIGNEE(S):
SOURCE:

DOCUMENT TYPE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE US 5232885 PRIORITY APPLN. INFO.: US 1991-812131 US 1991-812131 A 19930803

L28 ANSWER 69 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB In processes for the regeneration of catalysts, such as

hydrogenation catalysts, and the defouling of objects, which are

contaminated with carbonaceous deposits, the catalyst or object is

contacted with water, oxygen-containing gas, and at least one alkaline

material maintained at a pH of at least 7.0, and at a temperature and pressure maintained at a pH of at least 7.0, and at a temperature an sufficient to combust at least a portion of the carbonaceous deposits. ACCESSION NUMBER: 1993:568503 CAPLUS DOCUMENT NUMBER: 119:168503 TITLE: Processes for regenerating catalysts contains the company of the carbonaceous deposits.

Processes for regenerating catalysts contaminated

Carbonaceous materials Van Driesen, Roger ABB Lummus Crest Inc., USA U.S., 4 pp. CODEN: USXXAM INVENTOR(S):
PATENT ASSIGNEE(S):
SOURCE:

DOCUMENT TYPE:

Patent English LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

with

PATENT NO. DATE KIND APPLICATION NO. DATE US 5217935 PRIORITY APPLN. INFO.: А 19930608 US 1992-877231 US 1992-877231

L28 ANSWER 71 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB A process for regenerating a hydrogenation catalyst for hydrocarbon oils (e.g., Arabian heavy vacuum residue) comprising a

ort
of an inorg, substance and an active metal component is carried out by
oxidative burning of the coke deposited on the catalyst so that the
regenerated catalyst has a residual coke content of 0.5-10.0 weights; the
residual coke content is expressed by the weight loss of the regenerated
catalyst caused by calcining the regenerated catalyst in an atmospheric

catalyst caused by calcining the regenerated catalyst in an atmospheric of air at 550° for 8 h. The process recovers the catalytic activities of the catalyst sufficiently, extends the lifetime of the catalyst in repeated use, prevents the catalyst from damage due to the regeneration, and, in hydrodesulfurization processes, permits omission of presulfurization of the regenerated catalyst.

ACCESSION NUMBER: 1993:476185 CAPLUS
DOCUMENT NUMBER: 119:76185
INVENTOR(S): A process for regenerating hydrocarbon oil hydrogenation catalysts
NOURCE: Identicus Kosan Coo, Ltd., Japan
EUR. Pat. Appl., 13 pp.
COOLINET TYPE: PATENT PATENT SPEXEDW
Patent

Patent English

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE EP 541994 Al 19930519 EP 1992-117833 R: BE, CH, DE, FR, GB, IT, LI, NL, SE JP 05123586 A2 19930521 JP 1991-315188 JP 3110525 B2 20001120 CA 2081718 AA 19930506 CA 1992-2081718 19921019 19911105 JP 3110525 CA 2081718 PRIORITY APPLN. INFO.: CA 1992-2081718 JP 1991-315188 19921029 A 19911105

ANSWER 72 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
Regeneration of Ni-Mo catalysts, for heavy vacuum residues, was
studied under various conditions. Coke combustion during catalyst
regeneration consisted of 3 steps. The first step was controlled
by the amount of 0, the second by 0 diffusion in catalyst pores, and the
third by the amount of 5 compds., which are difficult to burn. Catalyst
surface area and Mo dispersion on the catalyst were influenced strongly by

steam. When regeneration was stopped before complete C oxidation and the residual coke content was 1.0-3.0 weight%, V on catalyst was tetravalent

inactive. The amount of Mo on the catalyst, catalyst surface area, and hydrogenation activity decreased slightly.
ACCESSION NUMBER: 1993:257784 CAPLUS
DOCUMENT NUMBER: 118:257784

TITLE: AUTHOR(S): CORPORATE SOURCE: 118:257784
Regeneration of heavy oil hydrotreating catalyst Noguchi, Y.; Itoh, T.; Obayashi, Y.; Komine, K. Cent. Res. Lab., Idemitu Kosan Co., Ltd., Sodegaura, 299-02, Japan Preprints - American Chemical Society, Division of Petroleum Chemistry (1993), 38(1), 50-3 CODEN: ACCAT: ISSN: 0569-3799
JOURNAL EDITION OF THE PROPERTY OF THE PROPERTY OF T

SOURCE:

DOCUMENT TYPE: LANGUAGE:

ANSWER 74 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Catalytic activity of Ni for C gasification is initially high at 500-700°, but it disappears when the reaction is repeated or the specimen is preheated in an inert atmospheric >700°. The deactivated catalyst was regenerated by a treatment with steam at 700-800° and subsequent reduction at 350°. The surface state of Ni on C was inferred by temperature-programmed desorption (TPD) of H, and a correlation was found between the TPD patterns and the catalytic activity for gasification. Catalytic activity of the Ni for CO2 conversion in 18 CO2 + 998 HZ mixture deactivated, and the catalyst was regenerated in a similar way as that observed for gasification. Redispersion
of NA and, possibly, removal of contaminants are thought to be responsible for the observed catalyst regeneration.

ACCESSION NUMBER: 1993:133020 CAPLUS
DOCUMENT NUMBER: 118:133020 CAPLUS
TITLE: Regeneration of nickel catalyst on carbon AUTHOR(S): Haga, Tetsuys; Nishiyama, Yoshiyuki
CORPORATE SOURCE: Inst. Chem. React. Sci., Tohoku Univ., Sendai, 980,

Japan Journal of Catalysis (1993), 140(1), 168-72 CODEN: JCTLAS; ISSN: 0021-9517 Journal English SOURCE:

DOCUMENT TYPE: LANGUAGE:

228 ANSWER 73 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Different methods of artificially depositing contaminant metals on FCC
(fluid catalytic cracking) catalysts are compared. The properties and
distribution of the deposited metals were studied by XFS, electron probe
microanal. (EPMA), O adsorption, and temperature programmed reduction
(TPR). Cyclic deposition and aging of the metals in a fixed
reactor using a severe hydrothermal environment desired.

dized-bed reactor using a severe hydrothermal environment during catalyst regeneration provides the best comparison to equilibrium catalysts. The metals initially deposit on the outer surface of the catalyst particles, giving a shell impregnation profile. Under severe

The metals initially depusation and the particles, giving a shell impregnation profile. Under severe hydrothermal conditions, Mi and V migrate into the bulk of the catalyst particles. This process occurs simultaneously with zeolite dealumination and the formation of mesopores in the catalyst. Also this procedure produces selectivities approaching those of equilibrium catalysts, not only for coke and H, but also for all other products. Comparison of several metal resistant catalysts reveal that different rankings are obtained depending on the method used to deposit the metals. ACCESSION NUMBER: 1993:172150 CAPIUS DOCUMENT NUMBER: 1993:172150 CAPIUS DOCUMENT NUMBER: 118:172150 TITLE: Evaluation of metals-contaminated FCC catalysts AUTIOR(S): Haas, A.; Suarez, W.; Young, G. W. CORPORATE SOURCE: Grack, GmbH, Mours, Germany AICHE Symposium Series (1992), 291(Advanced Fluid Catalytic Cracking Technology), 133-42 CODEN: ACSSCQ: ISN: 0065-8812 DOCUMENT TYPE: Journal LANGUAGE: English

L28 ANSWER 75 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB A prolonged lifetime of a NA-faujasite zeolite methanation
catalyst and a stable rate of methanation can be achieved by operating

the fluidized catalyst bed under unsteady-state conditions. A higher rate of methanation is accompanied by a shift in selectivity compared to a steady state operation due to an enhanced disproportionation of CO.

ACCESSION NUMBER: 1993:83216 CAPLUS

DOCUMENT NUMBER: 118:83216

TITLE: In situ perpetual

In situ perpetual regeneration of a N4
-faujasite methanation catalyst
Jaeger, N. I.; Schulz-Ekloff, G.; Kapicka, Jiri
Inst. Angew. Phys. Chem., Univ. Bremen, Bremen,
D-2800/33, Germany
Catalysis Letters (1992), 14(3-4), 331-7
CODEN: CALEER; ISSN: 1011-372X
JOURNAL
English

AUTHOR(S): CORPORATE SOURCE: SOURCE: DOCUMENT TYPE: LANGUAGE:

Page 34

ANSWER 76 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The alkylation and dealkylation activities of Ni-loaded Y
zeolites, prepared by ion exchange and impregnation, are related to the
surface Broensted acidity. The effect on the level of protonic activity
of varying the alkali metal cocation (Lit, Na+, K+, Rb+, or Cs+) and the
inclusion of Ce3+ ions is discussed. Catalyst precursor reduction
temperature, catalyst precalcination, reaction temperature, acid
poisoning by pyridine
adsorption, coke deposition, and catalyst regeneration
were catalytic efficiency parameters. The presence of Ni metal
on the support enhanced the ethylation and cracking activities by
converting coke precursors. In the case of C6H6 ethylation, Et2C6H4 were
formed over the most active samples; the relative distributions of the
ortho- para-, and meta-isomers are reported. Data on C6H6 ethylation

OVER

NI-IMPREGNATED SIO2 and AL203 catalysts are also compared.

ACCESSION NUMBER: 1993:59085 CAPLUS
DOCUMENT NUMBER: 118:59085
Benzene ethylation and cumene dealkylation over nickel-loaded Y zeolites
AUTHOR(S): Coughlan, Brendan: Keane, Mark A.
COURDIAN, Erndan: Keane, Mark A.
JOURNAI of Catalysis (1992), 138(1), 164-78
CODEN: JCTLAS; ISSN: 0021-9517
DOCUMENT TYPE: Journal

DOCUMENT TYPE:

English CASREACT 118:59085 OTHER SOURCE(S):

ANSWER 78 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
The effect of reaction environment on physicochem. characteristics of a
Mi-containing oxide-type catalyst of thermooxidative cracking of
petroleum residue (obtained based on a Mi-Al alloy, promoted
with In and Ga) was studied. Cracking of petroleum distillation residue

to redox processes, and transitions of NiAl204 and NiO to Ni were observed Beside, recrystn. of highly disperse Al particles,

were observed Beside, recrystn. of highly disperse Al particles, present in the original catalyst, occurs. A strong tendency of Ni and Al for colocation was also observed Gasification of coke deposites on the catalyst leads to further reduction of Nio to Ni and In2O3 to In, during which part of the Nio and In2O3 phases becomes amorphous. Catalyst regeneration restores physicochem. characteristics.

ACCESSION NUMBER: 1992:554166 CAPLUS

DOCUMENT NUMBER:

117:154166

AUTHOR (S):

117:154166
Effect of reaction environment on the characteristics of a nickel-containing oxide catalyst for thermooxidative cracking of residue
Akhverdiev, R. B.; Mirzaeva, L. M.; Guseinova, A. D.; Namedova, A. Kh.; Gadzhi-Kasumov, V. S.; Yakerson, V. I.; Sarmurzina, R. G.; Guseinova, M. G.; Timakov, I.

CORPORATE SOURCE: Inst. Neftekhim. Protsess. im. Mamedalieva, Baku,

SOURCE:

Inst. Neftekhim. Protsess. im. Mai Azerbaijan Neftekhimiya (1992), 32(2), 125-9 CODEN: NEFTAH; ISSN: 0028-2421 Journal Russian

DOCUMENT TYPE: LANGUAGE:

L28 ANSMER 77 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Two spent catalysts, obtained from different hydrodemetalation operations,
were regenerated by 2 different treatments, viz. 2 volt 0 in N and air.
One spent catalyst (B), contained 3 weight V and 15 weight C, while

(H) contained 10 weight% V, 14 weight% C and 8 weight% Fe. After

(H) contained 10 weights V, 14 weights C and 8 weights Fe. After regeneration in the O-N stream, catalyst B showed essentially complete recovery of its original surface area, whereas catalyst B showed only 701 recovery. Both catalysts showed substantial losses in surface area by the air treatment. Catalytic activity tests on the regenerated catalysts for hydrodesulfurization of thiophene and for hydrogenation of 1-hexene showed low recovery of activities, even for the regenerated catalyst in which the surface area had been completely recovered. X-ray diffraction analyses of the spent-regenerated catalysts revealed substantial changes in catalyst structure. Surface area and catalytic activity results were qual. explained by these catalyst structural changes.

ACCESSION NUMBER:

1993:41953 CAPLUS

TITLE:

Structure and catalytic activity of regenerated spent hydrotreating catalytes

L28 ANSWER 79 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The hydrogenation of C6H6 to cyclohexane was investigated over a range of NA-exchanged and mickel-impregnated Y zeolites, varying the NA content and the nature of the alkali metal cocation (Li+, Na+, K+, Rh+, or Cs+). With a view to optimizing C6H6 conversion levels, the following catalytic parameters were studied: reaction temperature, reaction time, C6H6 flow rate, and coke deposition. The observed catalytic activities were correlated with previously reported

phys.

Characterizations. C6H6 hydrogenation increased in the order:
NiLiY < NiNAY < NiRNAY < NiGNAY. Catalyst deactivation resulted from the deposition of involatile coke on the catalyst surface, promoted by increasing zeolite acidity. The effects of poisoning the surface Broensted acid sites by adsorption of ammonia onto the activated reduced zeolites were considered. The results of catalyst regeneration by high-temperature oxidation of the coke deposits were also reported.

ACCESSION NUMBER: 1992:108776 CAPLUS DOCUMENT NUMBER:

116:108776

116:108776
Catalyst deactivation during the hydrogenation of benzene ever nickel-loaded Y zeoltes
Coughlan, Brendan; Keane, Mark A,
Phys. Chem. Lab., Univ. Coll., Galway, Ire.
Journal of Molecular Catalysis (1992), 71(1), 93-109
CODEN: JMCADS: ISSN: 0304-5102

1993:41953 CAPLUS
118:41953
Structure and catalytic activity of regenerated spent hydrotreating catalysts
Kim, C. S.; Massoth, F. E.; Furimsky, Edward
Dep. Fuels Eng., Univ. Utah, Salt Lake City, UT,
84112, USA
Fuel Processing Technology (1992), 32(1-2), 39-46
CODEN: PTPUTY; ISSN: 0378-3820
Journal AUTHOR(S): CORPORATE SOURCE: SOURCE:

DOCUMENT TYPE: LANGUAGE:

DOCUMENT NUMBER:

DOCUMENT TYPE: LANGUAGE:

TITLE: AUTHOR (S): CORPORATE SOURCE:

SOURCE:

Page 35

ANSWER 80 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN A detailed examination of 2 regenerated Na-Mo/Al203 catalysts, withdrawn from a lube hydrofinishing reactor after 8 and 12 yr operation, was carried out. The decrease in surface area, pore volume, and

molybdena
dispersion, or the decline in thiophene hydrodesulfurization activity, of
the samples, compared with fresh catalyst, are not related to
deactivation. Temperature-programmed reduction results point to the
increased difficulty of reduction of Ni(Mo) species at low
temperature in the regenerated samples. X-ray photoelectron
spectroscopic data
industed declining NI-Mo atomic ratios on the exterior of the
dataset due to probable more-extensive migration of Ni then Mo
dating continuous use and repeated regenerations.
ACCESSION NUMBER: 1991:562615 CAPLUS
CONTENSIVE MUMBER: 115:162615

115:162615

DOCUMENT NUMBER: TITLE:

115:162615
Monitoring the performance of nickel
-molybdenum-alumina catalyst in lube hydrofinishing
through catalyst characterization
Ramaswamy, A. V.; Sharma, L. D.
Natl. Chem. Lab., Pune, 411 008, India
Studies in Surface Science and Catalysis (1991),
68 (Catal. Deact. 1991), 707-15
CODEN: SSCTDM; ISSN: 0167-2991

AUTHOR(S): CORPORATE SOURCE: SOURCE:

DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 81 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The title apparatus comprises a 1st fixed bed of oxidation catalysts
containing
≥1 of Pt. Pd. and Rh to combust CO and the unburned hydrocarbons
upstream of the flue ducts, and a 2nd fixed bed of denitration catalysts
containing ≥1 oxides of Ti, V, Mo, W, Fe, Ni and Sn located
downstream, means for monitoring boiler load and the concns. of NOX in outlet and inlet of the denitration catalyst bed, means for optimizing

amts. of NH3 injection into the inlet side of the denitration catalyst bed, and means for controlling the amount of  ${\rm SO3}$  or  ${\rm SO2}$  gas injection

bed, and means for contribing to the definition of spent denitration catalysts based on the feedback signals from the monitors. The apparatus extends the service life of denitration catalysts and has lower construction cost.

ACCESSION NUMBER: 1991:253289 CAPLUS

DOCUMENT NUMBER: 114:253289

TITLE: Apparatus for treatment of boiler flue gases and the reconstruction of spent catalysts

regeneration of spent catalysts Mitsumura, Hiroyuki; Ikeshoji, Susumu; Myata,

INVENTOR(S): Kazuaki;

PATENT ASSIGNEE(S): SOURCE:

Nagamine, Shogo: Murakami, Toshio Babcock-Hitachi K. K., Japan Jpn. Kokai Tokkyo Koho, 5 pp. CODEN: JKXXAF

DOCUMENT TYPE: LANGUAGE: Patent Japanese

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND APPLICATION NO. DATE DATE JP 03052626 PRIORITY APPLN. INFO.: A2 19910306

L28 ANSWER 82 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Methods for regenerating S-poisoned Fe-group catalysts, especially

orted Ni catalysts, are described which entail bringing the catalyst into contact with a fermentation mixture comprising a suspension of a into contact with a fermentation mixture comprising a suspension of a growing or living resting sulfide-oxidizing bacteria. The bacteria may be Ectothiorhodospira shaposchnikovii (especially strain WKM W 1525 D).

ACCESSION NUMBER: 1991:235840 CAPIUS
DOCUMENT NUMBER: 114:235840
ITITLE: Method for regenerating catalysts poisoned by sulfur Beck, Dieter; Heinritz, Hans Joerg; Wiessner, Arndt; Anders, Klaus; Wilde, Michael; Geyer, Reinhard; Lambrecht, Wolfgang
PATENT ASSIGNEE(S): Akademie der Wissenschaften der DDR, Institut fuer Biotechnologie, Ger. Dem. Rep.

SOURCE: GERSTEN GEXXAB
POCUMENT TYPE: Patent

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: German 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DD 285024	A5	19901205	DD 1989-329834	19890622
RIORITY APPLN. INFO.:			DD 1989-329834	19890622

L28 ANSWER 83 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

A process for regeneration of cracking catalyst while minimizing NOX
emisions is disclosed. An additive is present in an amount which reduces
NOX emissions, but does not passivate metals (such as Ni and V)
deposited on the catalyst during the cracking reaction nor CO combustion
promoter which may be present. Relatively small amts. of 5b oxides
impregnated on a sep. support having little or no cracking activity are
preferred. preferred. ACCESSION NUMBER:

1991:146752 CAPLUS

DOCUMENT NUMBER: TITLE:

114:146752
Reducing nitrogen oxides emissions with antimony

Additive
Chin, Arthur A.
Mobil Oil Corp., USA
U.S., 7 pp.
CODEN: USXXAM INVENTOR(S):
PATENT ASSIGNEE(S):
SOURCE:

DOCUMENT TYPE: Patent English

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE US 4988432 PRIORITY APPLN. INFO.: 19891228 19891228 A 19910129 US 1989-458052 US 1989-458052

L28 ANSWER 84 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB A Raney mickel catalyst for reduction of
tris(hydroxymethyl)nitromethane (I) by H2 is regenerated using ammonium
formate, or an aliphatic amine formate (produced in situ from
stoichiometric chiometric amms. of HCO2H and an amine base). Thus, a MeOH solution of Raney nickel was saturated with H2, and mixed with I. After 2.5 h reduction, the postreaction liqs. and the catalyst were separated The catalyst was washed, then mixed with MeOH. This solution was saturated with H2, added with Et3N and 30% aqueous HCO2H (in the amount stoichiometric with respect ect
to Et3N), and with MeOH solution of I. The above process was repeated 5
times; Raney midwel showed no reduction of its catalytic
activity. The yield of the product tris(hydroxymethyl)aminomethane
produced from the combined postreaction liqs. was 83.7% (of the theor. value). ACCESSION NUMBER: 1991:130228 CAPLUS 1991:190228 CAPUS 114:130228 Method of regenerating Raney nickel catalyst for reduction of tris-(hydroxymethyl) nitromethane Mazurkiewloz, Roman: Szeja, Wieslaw; Kiersznicki, DOCUMENT NUMBER: TITLE: INVENTOR (S): Tadeusz Politechnika Slaska, Pol. PATENT ASSIGNEE(S): Pol., 2 pp. CODEN: POXXA7 DOCUMENT TYPE: Patent Polish 1

DATE 19831230 PL 138692 PRIORITY APPLN. INFO.: В1 19861031 CASREACT 114:130228

APPLICATION NO.

DATE

OTHER SOURCE(S):

KIND

FAMILY ACC. NUM. COUNT: PATENT INFORMATION: PATENT NO.

ABS Spent Ni hydrogenation catalyst was processed into fresh catalyst by extracting the fat, leaching Ni as sulfate, converting the sulfate to formate, and reduction of the Ni to metal. A stainless atcel activation unit with a capacity of 150 kg/batch was used to degrade Ni formate at 230°. A mixture of saturated fat and Ni formate was heated from 150° to 230° at 0.5° min during H gas bubbling, allowed to stand for 1 h, and then cooled with H gas bubbling to 150°. Diatomaceous earth was added to give a catalyst containing fat 61, Ni 21, and diatomaceous earth 12 kg/100 kg. The catalyst is suitable for preparation of margarine, shortening, and related products.

ACCESSION NUMBER: 1990:439109 CAPLUS
DOCUMENT NUMBER: 113:39109
TITLE: Regeneration of nickel catalyst
AUTHOR(S): Raie M. A.; A Rhand, Manzoor
CORPORATE SOURCE: PCSIR Lab., Lahore, Pak.
SOURCE: Pakistan Journal of Scientific and Industrial Research

SOURCE: Research

(1989), 32(12), 837-9 CODEN: PSIRAA; ISSN: 0030-9885 Journal English DOCUMENT TYPE: LANGUAGE:

ANSWER 85 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN  $Ni-W/\Lambda 1203$  used for hydrotreating of petroleum for >3 yr were regenerated by 2 calcination methods. One is 1-step calcination at 500° in a stream of an O/N mixture (21 volume& 0). The other is stepwise calcination at 300, 400, and 500° in a stream of an O/N mixture (2 volume& 0). The latter method was better in restoring both

mixture (2 volume% 0). The latter method was better in restoring both the hydrogenation and hydrocracking activities. To clarify structural change of active metals in regeneration and mechanisms of deactivation, oxide and sulfide catalysts were investigated by EXAFS of NI K-edge and W LIII-edge. Ni and W aggregate in 1-step calcination. This aggregation is caused by rapid burning of carbonaceous deposits. The aggregation has great influence on deactivation of catalysts. Stepwise calcination was effective in removal of carbonaceous deposits without aggregation of active metals.

ACCESSION NUMBER: 1991:46143 CAPLUS
TITLE: Language 2 EXAFS study on regeneration of nickel - tungsten/alumina catalysts

AUTHOR(S): Matsubayashi, Nobuyuki; Shimada, Hiromichi; Sato, Toshio; Yoshimura, Yuji; Nishijima, Akio; Tamayama, Masaki; Mori, Yusuke; Kawamata, Hiroyuki; Abe, Masahiko; Ogino, Keizo

CORPORATE SOURCE: Shokubai (1990), 32(6), 354-7

CODEN: SHKUAJ; ISSN: 0559-8958

Journal Japanese

L28 ANSWER 87 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

Ni-containing Al203- or Si02-supported catalysts were nearly
completely regenerated in the continuous process of suspension
hydrogenation of 1,4-butynediol solns. by a temporary redn
. of the working pressure of H followed by reestablishment of the
original original
pressure. Thus, a 1.44 M aqueous 1,4-butynediol solution was
hydrogenated at 353 K and 1.5 MPa H at the rate of 1 L/h in the
presence of a Ni-containing SiO2-supported catalyst suspension whose
activity dropped 37% after 50 h. H stream was turned off, the reactor
pressure was reduced to, and kept at 0.1 MPa for 300 s, and the working
pressure reestablished which resulted in an essentially complete
restoration of the original catalyst activity.

ACCESSION NUMBER: 1990:157673 CAPLUS
DOCUMENT NUMBER: 112:157673
TITLE: The regeneration of Ni contains the TAPLUS

112:157673
The regeneration of Ni catalyst in hydrogenation of 1,4-butynediol Goetz, Reiner; Martens, Hans Werner; Turek, Fritz; Otto, Burkardt; Sieble, Holger; Weisbach, Lothar; Ulrich, Alfred; Busse, Dorls VEB Chemische Werk, Ger. Dem. Rep. Ger. (East), 7 pp.
CODEN: GEXXA8
Patent
German
1 INVENTOR (S): PATENT ASSIGNEE(S): SOURCE: DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE DD 272644 PRIORITY APPLN. INFO.: Al 19891018

OTHER SOURCE(S): CASREACT 112:157673

ANSWER 88 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
The oxidative regeneration of N1 molybdate and Co molybdate
catalysts from hydrotreatment of coal-derived cils and petroleum vacuum
gas oil was carried out in thermogravimetric and fixed bed reactors.
During the temperature-programmed burnoff, O uptake by carbonaceous
salts was pits was followed by decomposition of surface complexes, yielding CO2 as the main product. The oxidation of metal sulfides to corresponding oxides was accompanied by SO2 formations, which exhibited 2 maximum; one maximum oxidation of sulfidic S and the other from decomposition of sulfate as well as the oxidation of organic S. XPS data showed that small amts. of S remained regenerated catalysts in the form of sulfates. Refractory nature for C removal correlated with the aromaticities of the carbonaceous deposits. With the recoveries of fine structures of supported metals, hydrogenation and hydrocracking activities were almost recovered to the level of the fresh catalysts. Deposited metals, such as Ni, V, and Fe, affected the fine structure redispersion. They also contributed to the increase in hydrocracking activity, due to synergistic effects. effects. ACCESSION NUMBER: 1989:460778 CAPLUS DOCUMENT NUMBER: 111:60778 Oxidative regeneration of nickel-molybdate and cobalt-molybdate hydrotreating catalysts. Yoshimura, Yuji; Furimsky, Edward; Sato, Tosl Shimada, Hiromichi: Matsubayashi, Nobuyuki: AUTHOR (S) : Nishiima. Akio Natl. Chem. Lab. Ind., Tsukuba, 305, Japan Proc. - Int. Congr. Catal., 9th (1988), Volume 1, 136-43. Editor(s): Phillips, M. J.; Ternan, M. CORPORATE SOURCE: SOURCE :

Inst. Can.: Ottawa, Ont. CODEN: 56NZA9

Conference English

ANSWER 90 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Spent Shell 424 NL molybdate catalyst supported on Y-Al203, which was used to hydrotreat a synfuel distillate, was regenerated by burnoff with a wide-O-concentration-range medium. A

r correlation was established between O concentration and the AT parameter (difference between the maximum temperature and the bed temperature e the admission

re the admission of oxidizing gas) reflecting the temperature runaway on the oxidizing gas admission. The addition of steam or CO2 resulted in a dilution effect only.

Catalyst pretreatment by extraction and reduction resulted in a dilution effect a significant increase in the initial rate of SO2 in CO and CO2 yields.

ACCESSION NUMBER: 1989:98370 CAPLUS

DOCUMENT NUMBER: 110:98370

TITLE: Effect of

Effect of oxygen concentration on temperature runaway during regeneration of hydrotreating catalyst Furimsky, Edward Energy Res. Lab., Canada Cent. Miner. Energy

AUTHOR(S): CORPORATE SOURCE:

Technol.,

Ottawa, ON, K1A 0G1, Can. Applied Catalysis (1988), 44(1-2), 189-98 CODEN: APCADI; ISSN: 0166-9834

DOCUMENT TYPE: LANGUAGE:

Chem.

DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 89 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The kinetics and mechanisms of leaching of metals (V and Ni)
from spent atmospheric residue hydrodesulfurization catalysts and from
pure metal

metal sulfides (V2S3 and **Ni8**) were studied using oxalic acid. An increase in temperature from 25-75° increased the extent of leaching significantly. Acid concns. of 0.33-1.0M had little effect on the rate

metal leaching. The kinetics of leaching of pure metal sulfides by oxalic acid was controlled by a chemical reaction, the rate-determining step

acid was controlled by a Cnemical results, ...
being the
formation of metal ions from metal sulfides. Conversely, with spent
catalyst, mass transfer in the C matrix deposits was rate controlling.
Build-up of a passivating layer of the product was observed after some
leaching. Subsequent leaching was dictated by the slow removal of this
film followed by a relatively fast attack on exposed sulfides. Metal
leaching resulted in substantial increases in the surface area and
hydrodesulfurization activity.
ACCESSION NUMBER: 1969:413036 CAPLUS
111:3036

TITLE:

111:13036
Regeneration of spent hydroprocessing catalysts: metals removal
Marafi, M.; Stanislaus, A.; Mumford, C. J.; Fahim, M.
Pet. Technol. Dep., Kuwait Inst. Sci. Res., Kuwait
Applied Catalysis (1989), 47(1), 85-96
CODEN: APCADI; ISSN: 0166-9831 AUTHOR (S): CORPORATE SOURCE:

DOCUMENT TYPE:

SOURCE:

L28 ANSWER 91 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The adsorption properties and activity of Co and Ni catalysts
were studied by temperature-programmed desorption (TPD) of CO and by
methanation methanation

test. C accumulated on the catalyst surface CO by disproportionation of CO at elevated temps. Catalysts with surface CO deposited at 450° deactivated more rapidly than those with C deposited at 300°. The former are more easily regenerated by H treatment than the latter. CO undergoes more severe deactivation than Ni by surface C accumulation, but it is readily regenerated by H treatment. Catalyst deactivation by surface C could be monitored by changes in the TPD chromatogram, but the catalyst regeneration process could not be monitored, suggesting that the structure of the catalysts were modified during the deactivation and regeneration processes.

ACCLESSION NUMBER: 1988:119754 CAPLUS

DOCUMENT NUMBER: 108:119754 DOCUMENT NUMBER: TITLE: 108:119754 108:119754
Carbon-deactivation and regeneration of supported cobalt and nickel catalysts
Moon, Sang H.; Lee, Sang W.; Sung, Byuck P.
Coll. Eng., Secul Natl. Univ., S. Korea
Hwahak Konghak (1987), 25(5), 468-76
CODEN: HHKHRT: ISSN: 0304-128X AUTHOR (S); CORPORATE SOURCE: SOURCE: DOCUMENT TYPE: Journal

L28 ANSWER 92 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The metal-containing spinels for reducing the SOx and/or NOx emitted during regeneration of a hydrocarbon cracking unit have a 1st metal, e.g. Mg,

a 2nd metal having a valence higher than the valence of the 1st metal, e.g. Al, a minor amount of  $\geq 1$  3rd metal, e.g., group 1B, IIB, or VIA metals, rare earth metals, Pt group metals or mixts. thereof, and a 4th metal, e.g. Fe, Ni, Ti, Cr, Mn, Co, Be, Sn, Bi, Mo, Sb, V, and mixts. therof. The 3rd metal promotes SO2 oxidation to SO3 and the 4th metal

promotes reduction of the 1st metal sulfate. Thus, Mg-Al-containing spinels containing 5% Sn and 5% Ce had substantial SOx removal activity and

ACCESSION NUMBER: DOCUMENT NUMBER:

reduced NOx emissions.
SSION NUMBER: 1988:61733 CAPLUS
MENT NUMBER: 108:61733
E: Metal-containing spinel composition and process of TITLE:

Metal-containing spinel composition and process of using same Yoo, Jin Sun; Radlowski, Cecelia Ann; Karch, John Albert; Bhattacharyya, Alakananda Katalistiks International, Inc., USA PCT Int. Appl., 54 pp. CODEN: PIXXD2 INVENTOR (5):

PATENT ASSIGNEE(S): SOURCE:

DOCUMENT TYPE: Patent

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: English

	PAT	PENT	NO.			KIN		DAT	E		API	PLICATION NO.	 DATE
	WO		157 AU,					198	71022		WO	1987-US750	19870407
						FR.	GB	ΤT	, NL,	SE			
	US	4790			J.,						us	1986-848954	19860407
												1987-72815	19870407
		5947											
											EP	1987-903035	19870407
	EP	2631	71			В1		199	21125				
		R:	AT,	BE,	DE,	FR,	GB,	IT	, NL,	SE			
	BR	8707	261			Α		198	80419		BR	1987-7261	19870407
											JP	1987-502371	19870407
	JP	0702	9054			В4		199	50405				
	CA	1293	967						20107			1987-534085	19870407
	AT	8270	6						21215			1987-903035	19870407
	US	4963	520					199	01016		US	1988-180577	19880801
	ΑU	8945						199	00405		ΑU	1989-45836	19891205
	ΑU	6201	78			B2		199	20213				
	JP	0621	0177			A2		199	40802		JΡ	1993-278924	19931001
RIO	RIT	APP	LN.	INFO	. :						US	1986-848954	19860407
											EP	1987-903035	19870407
											wo	1987-US750	19870407

L28 ANSWER 94 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB A process for reactivating spent catalysts containing zeolite and for

ARBA A process for reactivating spent catalysts containing zeolite and for removing

NH3 or its precursors from the zeolite comprises (a) passing H over the spent catalyst composition at 850-950°F and 200-1000 psig, and then (b) contacting the resulting catalyst with a polar hydrocarbon solvent (mol. weight ≤200) at 25-500°C and 1 atm-1000 psig for a time sufficient to desorb residual N on the catalyst after the H contact. Thus, in hydrodewaxing of a lubricating oil base stock, spent catalysts (containing 1.18 NH-ZSM 5) were regenerated by passing over H at 900°F and 400 psig, and then contacting with THF at ≤350°C, resulting in effective removal of NH3 from the spent zeolite.

ACCLESSION NUMBER: 1987:179597 CAPLUS

TITLE: Process for hydrogenative reactivation of spent catalysts

PR

106:179597
Process for hydrogenative reactivation of spent catalysts
Degnan, Thomas F., Jr.; Chen, Nai Y.
Mobil Oil Corp., USA
U.S., 10 pp.
CODEN: USXXAM
Patent INVENTOR (S)

PATENT ASSIGNEE(S): SOURCE:

DOCUMENT TYPE: Patent English

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
110 4640107		19870310	US 1985-690065	19850109
US 4649127 PRIORITY APPLN. INFO.:	A	198/0310	US 1985-690065	19850109

L28 ANSWER 93 OF 258 CAPLUS COPYRIGHT 2004 ACS on.STN
AB The title catalysts are regenerated at high temperature in a gas

AB The title catalysts are regenerated at high temperature in a gas containing 0, N compds. in the form of amines and/or N oxides, and, optionally, steam. A spent catalyst containing Pd 6, V 3, and Pb 6 g/L alumina, used to reduce PhNO2 to PhNH2, was regenerated at <150 with a gas mixture containing 12.5 g/h 5% aqueous NH3, 4 L/h air, and 8 L/h N (reduced to 4 L/h during regeneration) until CO2 was not present in the outlet gas. The regeneration until CO2 was not present in the outlet gas. The PhNH2 initially and 88% PhNO2 conversion with 88% selectivity to PhNH2 initially and 88% PhNO2 conversion with 90% selectivity to PhNH2 actor 200 h.

ACCESSION NUMBER: 1987.409324 CAPLUS
DOCUMENT NUMBER: 107:9324
TITLE: Regeneration of catalysts for gas-phase reduction of aromatic nitro compounds

1987:409324 CAPLUS
107:9324
Regeneration of catalysts for gas-phase reduction of aromatic nitro compounds
Stammann, Guenter; Kricsfalussy, Zoltan; Waldmann, Helmut; Schneider, Joachim; Medem, Harald
Bayer A.-G., Fed. Rep. Ger.
Ger. Offen., 6 pp.
CODEN: GWXXEX
Patent
German INVENTOR (5):

PATENT ASSIGNEE (5):

DOCUMENT TYPE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: German 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
		~		
DE 3530820	A1	19870305	DE 1985-3530820	19850829
DE 3530820	C2	19880721		
US 4714689	A	19871222	US 1986-894581	19860807
EP 212602	A1	19870304	EP 1986-111377	19860818
EP 212602	B1	19880921		
R: BE, DE, FR,	GB, IT			
JP 62053745	A2	19870309	JP 1986-194108	19860821
BR 8604109	A	19870422	BR 1986-4109	19860828
PRIORITY APPLN. INFO.:			DE 1985-3530820	19850829

ANSWER 95 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB A conventional SiO2-supported Mi catalyst had almost double the increased activity in the hydrogenation of fats when the Ni component was prepared from a HCI leachate of a spent Mi catalyst. Thus, a regenerated NiCl2 solution containing 22 g Mi/L was stirred with kieselguhr at 80-100° with gradual feeding of an aqueous Na2CO3 solution to obtain a molar Na/Mi ratio of 2:2.9. The precipitate was filtered, dried, calcined at 350-400° and reduced with H at 430-480°. The catalyst had 1.9 times higher activity than a fresh catalyst.

ACCESSION NUMBER: 1987:23907 CAPLUS
DOCUMENT NUMBER: 106:23907 TAPLUS
TITLE: Highly active nickel catalysts on carriers
Highly active nickel catalysts on carriers
Kudlacek, Rostislav; Pospisil, Milan; Mucka, Viliam; Cabicar, Josef

Highly active nickel catalysts on carriers Kudlacek, Rostislav; Pospisil, Milan; Mucka, Viliam; Cabicar, Josef

PATENT ASSIGNEE(S): Czech. Czech., 5 pp. CODEN: CZXXA9

DOCUMENT TYPE:

Patent Czech 1

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE CS 229154 PRIORITY APPLN. INFO.: В 19840618

ANSWER 96 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

A cylindrical container, walled with Nylon cloth and having helicoidally distributed shafts, is adapted to a concrete mixer and rotated. There, regeneration of Alundum balle is done by abrasion with a mixture of water and thick sand and that of Co/Mo or Ni/Mo catalyst rods with gasoline and friction against the Nylon wall. With this process, regeneration losses are 0.3%.

ACCESSION NUMBER: 1986:597842 CAPLUS
DOCUMENT NUMBER: 105:197842

TITLE: Process and apparatus for washing There,

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BR 8405200 PRIORITY APPLN. INFO.: 19841016 А 19860520 BR 1984-5200 BR 1984-5200

L28 ANSWER 97 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Catalysts for coal liquefaction, hydrogenation of coal liqs.,
and hydroforming of petroleum, tar sand bitumen, or shale oils are
regenerated by removing metals (e.g., Fe, Ti, Ca, Na, Ni, and
V), by converting them to sulfates or oxysulfate compds. using acids
(e.g., H2SO4). Suitable treatments for oil-free catalysts are: 5-50
weight8 weight%

weight% aqueous H2SO4 at 60-250°F (with or without ≤10 weight% NH4+) or 5-20 weight% dilute aqueous (NH4)28208 at 60-150°F.

ACCESSION NUMBER: 1966:152259 CAPLUS

DOCUMENT NUMBER: 104:152259

TITLE:

104:152259
Regeneration of a catalyst
Van Partha, S. Ganguli
HRI, Inc., USA
Belg., 18 pp.
CODEN: BEXXAL
Patent
Dut-b INVENTOR(S): PATENT ASSIGNEE(S): SOURCE:

DOCUMENT TYPE:

MANGUAGE: Dutch
FAMILY ACC. NUM. COUNT: 2
PATENT INFORMATION:

APPLICATION NO. PATENT NO. DATE KIND DATE BE 1985-60671 US 1984-605537 ZA 1985-1011 DE 1985-3508028 GB 1985-6453 JP 1985-81536 US 1984-605537 A2 A A A1 A1 A2 19850816 19860617 19850925 19851031 BE 902270 19850425 US 4595666 2A 8501011 DE 3508028 GB 2157968 19840430 19850211 19850307 19850313 JP 60235645 19851122 PRIORITY APPLN. INFO.: 19840430

US 1981-317216

19811102

L28 ANSWER 98 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Attempts were made to regenerate 2 N1/S102 Catalysts which had
been exposed to S-containing compds. Attempted regeneration was by
oxidation oxidation
followed by reduction The amount of S present after these treatments
was determined by chemical anal, and XPS. Oxidation and reduction were
successful in removing a large part of the S, but catalytic activity for
benzene hydrogenation could not be restored. The small amts, of
S which remained after treatment in H were located largely in the bulk
rather than at the surface of the Ni particles.
ACCESSION NUMBER: 1986:136681 CAPIJUS
COUNDER HUMBER: 1986:136681 1986:136681 CAPAUS
104:136681
X-ray photoelectron spectroscopy of sulfur containing
nickel-silicon dioxide catalysts
Montes, M.; Genet, M.; Hodnett, B. K.; Stone, W. E.;
Delmon, B.
Groupe Phys.-Chim. Miner. Catal., Univ. Cathol.
Louvain, Louvain-la-Neuve, 1348, Belg.
Bulletin des Societes Chimiques Belges (1986), 95(1),
1-12 DOCUMENT NUMBER: AUTHOR(S):

CORPORATE SOURCE:

SOURCE:

CODEN: BSCBAG; ISSN: 0037-9646

DOCUMENT TYPE: LANGUAGE: English L28 ANSWER 99 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

A method is given to regenerate spent catalysts for hydrotreating processes for heavy hydrocarbon oils when the catalytic activity is reduced as a result of accumulation of heavy metals (e.g., V and M1) involving immersing the catalysts in alc. at 0-200° for 5-100h after removing coke accumulated on them by combustion at 400-600° in an atmospheric of O-containing gas. The catalysts to which

regeneration method is applied are used in hydrodesulfurization, hydrodenitrogenation, hydrodemetalization, and hydrocracking. Thus, a presulfided Co-Mo-Al2O3 catalyst used in hydrodesulfurization,

presulfided Co-No-Alzus causiyus and an and containing Moo3

15.0 and CoO 5.0 weights, was immersed in iso-Pr alc. at 90° for 20 h, or ethylene glycol at 100° for 40 h, to selectively extract V after combustion at 500° for 30 h in air.

ACCESSION NUMBER: 1985:618103 CAPLUS
DOCUMENT NUMBER: 103:218103

Regeneration of spent hydrotreating catalysts
Takase, Shinji; Inoue, Akira; Kawakatsu, Ken; Ino,
Takashi; Miyajima, Nobuyuki
Nippon Oil Co., ttd. , Japan
Eur. Pat. Appl., 13 pp.
CODEN: EPRXEDW
Patent
English INVENTOR(S):

PATENT ASSIGNEE(S):

SOURCE:

DOCUMENT TYPE: English

LANGUAGE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE 19851002 EP 156226 A2 A3 EP 1985-102733 19850311 EP 156226 19870114 R: DE, GB, NL JP 60190241 PRIORITY APPLN. INFO.: A2 19850927 JP 1984-46699 19840312

JP 1984-46699

Page 40

L28 ANSWER 100 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB A kinetic study of the regeneration of a series of AlPO4, SiO2, and
AlPO4-SiO2 supported NA catalysts poisoned with BUSH was carried
out by using 1-hexene hydrogenation as the test reaction.
ACCESSION NUMBER: 1985:601540 CAPLUS
DOCUMENT NUMBER: 103:201540
TITLE: A kinetic study of the regeneration of new alumi

103:201540
A kinetic study of the regeneration of new aluminum phosphate-supported nickel catalysts
Campelo, J. M.; García, A.; Luna, D.; Marinas, J. M. Fac. Sci., Cordoba Univ., Cordoba, 14005, Spain
Reaction Kinetics and Catalysia Letters (1985),

AUTHOR (S): CORPORATE SOURCE: SOURCE:

CODEN: RKCLAU; ISSN: 0304-4122

DOCUMENT TYPE:

LANGUAGE:

L28 ANSWER 101 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The metal-passivation agents for aluminosilicate catalysts contaminated with and deactivated by Mi, V, Fe, Co, and Cu in cracking of distillation residues contain 0.002-1.6 weight (relative to catalyst) mixture 8-14:1 (weight ratio) Sb-Sn; the reactivated catalyst is then treated in

reductive atmospheric before use. Carbonaceous deposits on the heavy metal-contaminated catalysts are removed by combustion with a min. amount of

amount of
O-containing gas before passivation. Thus, equilibrated
Al203-sio2-zeolite
Cracking catalysts treated with Sb was treated in air at 1250°F to
remove coke, impregnated with Thermolite 31 (I) [25168-24-5] in a
solvent, dried, aged in 10 cycles of heating for 10 min at 1200°F
and at 900°F, cooled, and then treated with H at 1250°F
before reuse. Gasoline yield in cracking (950°F) of an atm distillation
residue with only the removal of carbonaceous deposits, with I without H
treatment, and with both I and H treatment were 55.0, 56.9, and 60.5
volume\*, resp.

ACCESSION NUMBER: 1985:424730 CAPLUS
DOCUMENT NUMBER: 103:24730
Passivation agents for cracking catalysts

1985:424730 CAPUUS 103:24730 Passivation agents for cracking catalysts Phillips Petroleum Co., USA Jpn. Kokai Tokkyo Koho, 6 pp. CODEN: JKXXAF Patent

TITLE: PATENT ASSIGNEE(S): SOURCE:

DOCUMENT TYPE: LANGUAGE: Japanese FAMILY ACC, NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 60071042	A2	19850422	JP 1984-162770	19840801
CA 1223856	A1	19870707	CA 1984-454706	19840518
NO 8403811	A	19850329	NO 1984-3811	19840921
NO 160333	В	19890102		
NO 160333	С	19890412		
EP 141988	A1	19850522	EP 1984-111536	19840927
EP 141988	В1	19880518		
R: AT, BE, CH,	DE, FI	R, GB, IT,	LI, LU, NL, SE	
ES 536300	A1	19860401	ES 1984-536300	19840927
AT 34404	E	19880615	AT 1984-111536	19840927
PRIORITY APPLN. INFO.:			US 1983-536820	19830928
			EP 1984-111536	19840927

ANSWER 102 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Conventional, oil-industry hydrotreating catalysts such as sulfided Ni molybdate on alumina were used to upgrade coal-derived ligs. obtained from a 1 kg/h bench-scale continuous coal hydrogenation unit. Catalyst activity was not seriously affected by contamination with Fe, and when the catalyst was deactivated by C deposition it could be easily regenerated by heating in air. All of the higher boiling, more polar products of coal hydrogenation can readily be converted into lower-mol.-weight, less polar products (oils) by heating with H in a H-donor solvent, especially in the presence of a hydrotreating catalyst.

se results may be very significant in the design of a coal liquefaction facility in that coal could be dissolved in a vehicle with the aid of an Fe-based catalyst and the resulting product immediately stabilized,

mainly
as a desirable oil fraction, by passing the material over a hydrotreating
catalyst such as supported sulfided Ni molybdate.
ACCESSION MUMBER: 1985:424683 CAPLUS

DOCUMENT NUMBER: TITLE:

AUTHOR(S): CORPORATE SOURCE:

1985:424683 CAPLUS
103:24683
Regeneration of catalysts for use in continuous flow coal hydrogenation
Jackson, W. R.; Larkins, F. P.; Herton, P. A.
Dev. Demonstration Counc., Natl. Energy Res.,
Camberra, Australia
Report (1983), NERDDP-EG-83-130, 338 pp. Avail.:
Natl. Energy Res., Dev. Demonstr. Counc., Canberra,
Aust.

Aust. Fram: Energy Res. Abstr. 1985, 10(7), Abstr. No.

10847 DOCUMENT TYPE:

SOURCE:

Report English LANGUAGE:

L28 ANSWER 103 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB A spent flue gas denitration catalyst (to reduce Nox with NH3) is washed
with aqueous salts of ≥1 of V, Fe, NI, Mo, Co, Cu, Cr, W, Sn,
and U (as active components), preferably after water or dilute acid

washing.

Thus, a spent plate catalyst (used for heavy oil combustion in boiler

gas) was immersed in 2 vols. aqueous VOSO4 49 g/L for 20 min (or after immersing in water for 20 min 3 times) and dried at 180° for 2 h.
Gas containing NOS 200, NH3 240, SO2 500 ppm, O2 3, CO2 12, HZO 12%, and balance NZ was passed over at 15 m/h and 350°. The denitration was 97, 83, and 97 (or 97.5%) with the fresh, spent, and regenerated one.
ACCESSION NUMBER: 1885-410778 CAPLUS

DOCUMENT NUMBER: 103:10778

TITLE: PATENT ASSIGNEE(S): SOURCE:

Regeneration of spent flue gas denitration catalysts Babcock-Hitachi K. K., Japan Jpn. Kokai Tokkyo Koho, 3 pp. CODEN: JRXXAF

DOCUMENT TYPE:

Japanese

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 60034743	A2	19850222	JP 1983-143666	19830808
JP 04046621	B4	19920730		
PRIORITY APPLN. INFO.:			JP 1983-143666	19830808

ANSWER 104 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
The vapor-phase hydrogenation of acetylenic impurities in the
presence of other hydrocathons in an industrial butadiene feedstock was
studied in the presence of Mi-Cu, Cu, Ni, and Raney Cu
catalysts and various supports. Comparisons of all catalysts showed that
activity and selectivity were not related, but Cu-based catalysts gave

highest activity and the highest selectivity. With the exception of

Raney
Cu, the activity of Cu-based catalysts increased with increasing total surface area. Sintering during catalyst regeneration may affect catalyst life.

ACCESSION NUMBER: 1985:7984 CAPLUS

DOCUMENT NUMBER: TITLE:

1985:7984 CAPLUS
102:7984 Selection of catalysts for the selective
bydrogenation of acetylenic impurities in an
industrial butadiene stream
Kummerow, 1.7 frimm, D. L.; Wainwright, M. S.
Sch. Chem. Eng. Ind. Chem., Univ. New South Wales,
Kensington, 2033, Australia
Actas Simp. Iberoam. Catal., 9th (1984), Volume 1,
797-806. Soc. Iberoam. Catal.: Lisbon, Port.
CODEN: SZTUAU AUTHOR(S): CORPORATE SOURCE:

DOCUMENT TYPE: LANGUAGE: Conference English

SOURCE:

L28 ANSWER 105 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Ni catalyst used in hydrogenation is treated with
chelate resin or its mixture with adsorbent to remove completely impurities,
such as Fe, Zn, and Cu, contained in the catalyst. Thus, Ni
catalyst used in hydroganation of soybean oil was heated with
H2SO4 to give NiSO4 solution The solution was passed upstream through

packed with chelate resin (phenol-HCHO resin containing iminodiacetic acid

group) and activated C, resp. The effluent was mixed with diatom earth and NaZCO3, then the solid and liquid were separated Solid (regenerated Ni catalyst) contained no Zn. Cu. P, or S and compared well in catalytic activity with a new Ni catalyst.

SION NUMBER: 1995:4731 CAPLUS
MENT NUMBER: 102:4731

ACCESSION NUMBER: DOCUMENT NUMBER:

102:4731
Regeneration of nickel catalyst
Miyoshi Oil and Fat Co., Ltd., Japan
Jpn. Kokai Tokkyo Koho, 4 pp.
CODEN: JKXXAF
Fatent TITLE: PATENT ASSIGNEE(S):

SOURCE:

DOCUMENT TYPE:

Japanese LANGUAGE:

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 59162950	A2	19840913	JP 1963-37978	19830308
JP 04036741	В4	19920617		
PRIORITY APPLN. INFO.:			JP 1983-37978	19830308

ANSWER 106 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
The recovery of Mi, Co, or Rh catalysts from C3-10 carboxylic
acid mixts. produced by the hydrocarboxylation of olefins consists of
distilling off the free acids to leave an acid salt concentrate which is

used for further hydrocarboxylations. Thus, after hydrocarboxylation of ethylene in a Ni salt containing propionic acid solution, the products were removed, the free acid mixture distilled, and fresh propionic acid added

added
to reconstitute the catalyst.
ACCESSION NUMBER: 1984:216461 CAPLUS
DOCUMENT NUMBER: 100:216461

TITLE:

100:216461

Recovery and re-use of nickel, cobalt or rhodium from mixtures containing carboxylic acids Broellos, Klaus: Lindner, Alfred; Irnich, Rudolf; Hornberger, Paul BASF A.-G., Fed. Rep. Ger. Ger. offen., 11 pp. CODEN: GMXXEX PATENT GEXTER COMPANY C INVENTOR (S):

PATENT ASSIGNEE (S):

DOCUMENT TYPE: LANGUAGE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

SOURCE:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 3238653	A1	19840419	DE 1982-3238653	19821019
EP 106271	Al	19840425	EP 1983-109954	19831005
EP 106271	B1	19860115		
R: BE, DE, FR,	GB, I'	r, NL		
PRIORITY APPLN. INFO.:			DE 1982-3238653	19821019

L28 ANSWER 107 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The spent catalyst from heavy-hydrocarbon-oil demetalization is
regenerated continuously in an integrated fluidized-bed apparatus
comprising a
reducing tower, a Cl-series chemical treating reactor, and a
polar-solvent polar-solvent
scrubber. Thus, 500 g/h of a spent zeolite cracking catalyst
containing 0.3
weights Ni and 0.5 weights V was regenerated in the apparatus by
reduction at 750°, contact with CCl4 [56-23-5] in the presence
of H at 375°, and scrubbing with MeOH [67-56-1].
ACCESSION NUMBER: 1984:19912 CAPLUS
DOCUMENT NUMBER: 100:194912 1984:194912 CAPLUS 100:194912 Fluidized-bed apparatus for regeneration of demetalization catalyst Jushitsuyu Taisaku Gijutsu Kenkyu Kumiai, Japan Jpn. Kokai Tokkyo Koho, 6 pp. CODEN: JXXXAF TITLE: PATENT ASSIGNEE(S): SOURCE: DOCUMENT TYPE: Japanese 1 FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 59010347	A2	19840119	JP 1982-116942	19820707
JP 63039290	В4	19880804		
PRIORITY APPLN. INFO.:			JP 1982-116942	19820707

ANSWER 108 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN A spent hydrogenation catalyst for coal and/or its liquefaction products is washed with a 100: (2-25) mixture of hydrocarbons (containing 280% aroms.) and PhOH and/or alkylphenols at 100-250° and 340 kg/cm2 gage. Thus, a 211 mixture of solvent (hydrogenated coal-tar oil distillate b. 2250°) and bituminous coal (-100 mesh) was heated at 430°, the solid material was removed, and the oil was hydrogenated with a sulfided Ni-Mo catalyst at 380°, 100 kg/cm2 gage, and space velocity 1.0 h-1 for 24 h. The catalyst was washed with a 100:10 mixture of petroleum alkylbenzene solvent containing 96% aroms. (b.p., initial, 10,

and 90%; 165°, 170°, 185°, and 191°, resp.)
and cresol [1319-77-3] at 200°, 20 kg/cm2 gage (H2), and space
velocity 3 h-1 for 1 h. The product consisted of gas 2.1, naphtha
(C5-160°) 6.5, kerosine (160-240°) 10.3, light oil
(240-360°) 19.0, and heavy oil (2360°) 62.1% (containing
S 0.6 and N 1.3%) 2 h before washing; and 1.7, 8.2, 14.6, 22.0, and 53.5
(0.3 and 0.7), resp., 2 h after washing. Without cresol in the washing
step, the latter data were 2.1, 6.0, 10.5, 18.5, and 62.9 (0.7 and 1.4)
resp.

ACCESSION NUMBER:

1984:142086 CAPLUS 100:142086

DOCUMENT NUMBER: TITLE:

Regeneration of spent coal hydroliquefaction catalyst

PATENT ASSIGNEE(S): SOURCE:

by washing
Mitsubishi Oil Co., Ltd., Japan
Jpn. Kokai Tokkyo Koho, 7 pp.
CODEN: JKXXAF

DOCUMENT TYPE:

Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

APPLICATION NO. PATENT NO. DATE KIND DATE JP 58223442 JP 1982-103710 19820618

19831226 19910326 JP 03022214 PRIORITY APPLN. INFO.: JP 1982-103710

L28 ANSWER 110 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN AB Title only translated. ACCESSION NUMBER: 1984:51168 CAPLUS

DOCUMENT NUMBER: 100:51168

Regeneration of Raney-type catalysts TITLE: Butenko, T. R.

AUTHOR(S): CORPORATE SOURCE: USSR

USSK Vopr. Kinet. i Kataliza, Ivanovo (1982) 73-5 From: Ref. Zh., Khim. 1983, Abstr. No. 17N174 SOURCE:

DOCUMENT TYPE: LANGUAGE: Journal

ANSWER 109 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Spent Rh tertiary phosphine complex catalysts are hydrogenated in organic solvents in the presence of NA, Pd, Pt, Co, and/or Rh catalysts at 30-150° and 570 kg/cm2 gage. Thus, 100 mL spent HRh(CO) (PPh3) [17185-29-4] catalyst solution in PhMe (from the hydroformylation of allyl alc.) was stirred with 2 g Raney Ni at 90° and 1 kg/cm2 H for 2 h to give a product giving a yield of 99%, compared with 40 without Raney Ni.

ACCESSION NUMBER: 1984:123169 CAPLUS
DOCUMENT NUMBER: 100:123169
EQHENT ASSIGNEE(S): Daicel Chemical Industries, Ltd., Japan Jpn. Kokai Tokkyo Koho, 4 pp.

DOCUMENT TYPE: Patent

DOCUMENT TYPE: Patent Japanese

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

APPLICATION NO. PATENT NO. KIND DATE DATE JP 58186443 JP 02037213 PRIORITY APPLN. INFO.: 19831031 19900823 JP 1982-69876 19820426 JP 1982-69876 19820426

L28 ANSWER 111 OF 258 CAPLUS COPYRIGHT 2004 ACS on STM
AB Spent catalyst (loaded with ≥1 of Fe, Co, Mi, Pt-group
metals, Cu, Au, and W) is contacted with ≥1 of aqueous HCO2H [64-18-6]
and H2c204 [144-62-7], and is optionally heated further or mixed with
strong acid. Thus, coke-oven gas liquor containing total NH3 3000, total N 4000, and Fe + Cu + Mg 1500 (originally 15) ppm was adjusted to pH 10 with

4000, and Fe + Cu + Mg 1500 (originally 15) ppm was adjusted to pH 10

aqueous NaOH, fed along with air at space velocities of 1.0 and 65 L/h,
250°, and 75 kg/cm² (gage) into a tower containing 1% Ir/TiO2 catalyst.

The activity index (given as NH3 removal) was 71 relative to the original
100. The spent catalyst was soaked in IN HCO2H at 60° for 2 h and
washed with water for 1 h to give a catalyst with an index of 99. When
spent 2% Ru/TiO2 catalyst (index 69) was heated at 90° in IN HCO2H,
the regenerated catalyst had an index of 98.

ACCESSION NUMBER: 1984:24327 CAPLUS

DOCUMENT NUMBER: 100:24327
TITLE: Regeneration of wet oxidation catalysts for coke oven
gas liquor
OSAK Gas Co., Ltd., Japan
Jpn. Kokai Tokkyo Koho, 6 pp.
CODEN: JKXXAP

DOCUMENT TYPE: Patent
LANGUAGE: 400.

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: Japanese 1

PATENT NO. KIND DATE APPLICATION NO. DATE JP 58114733 JP 1981-215854 JP 04045214 PRIORITY APPLN. INFO.:

Page 43

ANSWER 112 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
Spent oxidation catalyst (loaded with ≥1 of Fe, Co, Mi,
Pt-group metals, Cu, Au, and M) is contacted with ≥1 of dilute inorg,
and organic acids and then with ≥1 aqueous reductants or with
gas containing ≥1 of H or CO. The reductants include
N2H4.H2O, HCHO [50-00-0], NaBH4, LiAlH4, Na tartrate [868-18-8],
DESS.

ose [50-99-7], HCO2K, and HCO2Na. Thus, coke-oven gas liquor containing

3000, total N 4000, and Fe + Ca + Mg 1500 (originally 15) ppm was

pH 10 with aqueous NaOH, fed along with air at space velocities of

1.0 and
65 L/h, 250°, and 75 kg/cm2 (gage) into a tower containing 1% Ir/TiO2 catalyst. The activity index (given as NH3 removal) was 71 relative to the original 100. The spent catalyst was soaked in IN H3PO4 at 80° for 1 h, washed with water for 1 h, soaked in 18 N2H4.H20 at 60° for 1.5 h, and washed yo give a catalyst with an index of 94.

ACCESSION NUMBER: 1984:24326 CAPLUS
DOCUMENT NUMBER: 100:24326

DOCUMENT NUMBER:

100:24326
Regeneration of wet oxidation catalysts for coke oven gas liquor
Osaka Gas Co., Ltd., Japan
Jpn. Kokai Tokkyo Koho, 12 pp.
CODEN: JKXXAF
Patent
Japanese

PATENT ASSIGNEE (S): SOURCE:

DOCUMENT TYPE:

FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 58114731	A2	19830708	JP 1981-212949	19811229
JP 03066018	B4	19911015		
PRIORITY APPLN. INFO.:			JP 1981-212949	19811229

L28 ANSWER 114 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB A process is disclosed for decreasing the amount of coke deposited on catalysts in the cracking of hydrocarbon feedstocks containing ≥2 of Ni, V, and Fe. At least a portion of the catalyst is passed from the reaction zone through a regeneration zone operated under net reducing conditions and through a reduction zone maintained at ≥600° to at least partially passivate the metal contaminants on the catalyst. The reducing environment in the reduction zone is maintained by addition of H, CO, or their mixts. The passivated catalyst is 

zeolite. The catalyst had been contaminated with the metals by

reclite. The catalyst had peen concurred to the catalyst had been catalyst to the catalyst had been catalyst catalyst had been catalyst had been catalyst catalyst had been catal

English LANGUAGE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 4409093	A	19831011	US 1981-285737	19810722
US 4280895	A	19810728	US 1979-108395	19791231
US 4372841	A	19830208	US 1981-260191	19810504
EP 70681	A2	19830126	EP 1982-303694	19820714
EP 70681	A3	19830330		
EP 70681	B1	19850918		
R: BE, DE, FR,	GB, I	T, NL		
CA 1190170	A1	19850709	CA 1982-407490	19820716
JP 58037087	A2	19830304	JP 1982-125932	19820721
PRIORITY APPLN. INFO.:			US 1979-108395	19791231
			US 1981-260191	19810504
			US 1981-285737	19810722

L28 ANSWER 113 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The HCl wastewater containing <600 g/L Nicl2 and MnCl2 from inorg.

synthesis processes with Ni-Mn catalysts were treated for the regeneration of HCl, Ni, and Mn. Nicl2 and MnCl2 were salted-out from the wastewater by saturation with gaseous HCl, the chloridme.

salted-out from the wastewater by saturation with gaseous HCI, the chlorides were hydrolyzed in an air-steam at 600° to Mn2O3 and NiO, and the oxides were reduced at 1300-1400° with graphite to a Ni.

-Mn alloy, which was remelted in an induction furnace and reused as a catalyst.

ACCESSION HUMBER: 1984:10439 CAPLUS

1984:10439 CAPLUS 100:10439

DOCUMENT NUMBER: TITLE:

Treatment of highly concentrated hydrochloric acid wastewaters containing nickel and manganese Mil'ner, A. A.: Putivl'skii, V. V.; Zapol'skii, A. K. Inst. Kolloidn. Khim. Khim. Vody, Kiev, USSR Khimiya i Tekhnologiya Vody (1983), 5(4), 343-7 CODEN: KTVODL; ISSN: 0204-3556
JOURNAL REVOLL; AUTHOR (S): CORPORATE SOURCE: SOURCE:

DOCUMENT TYPE:

LANGUAGE:

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L28 ANSWER 115 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Spent denitration catalyst (for reduction of NOx with NH3) containing
V205 and Ti02(optionally 2-1 of W, Mo, Fe, Cu, Cr, Ni, Co,
and Ce oxides) is washed with water or aqueous inorg, acid, soaked in

tungstate, and calcined at 400-650° for regeneration. Thus, a honeycomb of V205 3.0% and balance T102 was used for boiler flue gas 1000 m3/h containing NOX 200 and SOX 800 ppm and mixed with 200 ppm NN3 at 300-400° and space velocity 6500 h-1. The denitration was 95.5 at 1st and 97.0% after 12,000 h and SO2 oxidation 2.1 and 4.8%, resp. The

spent. t catalyst was washed with water with supersonic wave application for 1 h, loaded with WO3 1 kg/m3, and calcined at 550 $^{\circ}$  for 3 h. After regeneration of the catalyst, denitration was 96.0 and SO2 oxidation was 1.8%.

1983:131635 CAPLUS

1.8%.
ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE:
PATENT ASSIGNEE(S):
SOURCE:

Regeneration of denitration catalyst NGK Insulators, Ltd., Japan Jpn. Kokai Tokkyo Koho, 4 pp. CODEN: JKXXAF

DOCUMENT TYPE: Patent Japanese

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 58000247	A2	19830105	JP 1981-98361	19810626
JP 62048537	B4	19871014		
PRIORITY APPLN. INFO.:			JP 1981-98361	19810626

ANSWER 116 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
Spent laterite catalyst fixed with coke and FeS is redn
.-roasted, the waste gas is mixed with H2, contacted with a catalyst at
200-400° to reduce So2 partially into H2S (mixed with H2S if
necessary) to give a H2S:SO2 mol ratio 2:1, and contacted with a catalyst
at 200-305° to obtain S. Thus, pellets were prepared from laterite
containing Fe 55.1, Ni 1.26, Mgo 2.6, SiO2 4.7, and Al2O3 2.81% with
a diameter of 0.2 mm, calcined to have sp. surface area 3.2 m2/g, and the
catalyst (A) 20 kg/h was circulated between a 12.7 diameter + 160 cm
stainless steel decomposition tower (X) and a 15.1 diameter + 180 cm
regeneration tower (Y). Low-pressure residual oil (B) 40 and steam 2.0
kg/h were fed into tower X at 540° and 1 kg/cm2 gage holding 13 kg
catalyst A, and oil B 0.06 kg and air 5 m3/h into tower Y at 850°
holding 20 kg catalyst A. Spent catalyst A containing 4.78 S1 kg/h was
removed from tower Y, 6 kg/h was heated at 750° in an air stream at
2.1 m3/h, and 1 kg/h containing 1.58 S was returned to tower Y. The

waste gas  $1.95~\mathrm{m3/h}$  containing SO2 7.4, O2 0.3, CO2 5.38 and H2 0.05 volumet was

with 1.5 m3/h gas containing 35 volume% H2, contacted with a 1.7 L 4 mm

diameter

CO-Mo catalyst column at 330°, then with a 3 L 5 mm diameter Al203 catalyst column at 160°, and cooled. The S yield was 0.25 kg/h and the (inal gas contained 6100 volume ppm SO2.

ACCESSION NUMBER: 1983:56527 CAPLUS

DOCUMENT NUMBER: 98:56527

Recovery of sulfur in reduction-roasting waste gases of spent laterite catalyst for heavy oil hydrogenation-decomposition

PATENT ASSIGNEE(S): Jushitsuyu Taisaku Gijutsu Kenkyu Kumiai, Japan Jpn. Kookai Tokkyo Koho, 7 pp.

DOCUMENT TYPE: Patent

DOCUMENT TYPE: Patent

Japanese LANGUAGE: FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PR

KIND	DATE	APPLICATION NO.	DATE
A2	19820821	JP 1981-20709	19810213
B4	19890922		
		JP 1981-20709	19810213
	A2	A2 19820821	AZ 19820821 JP 1981-20709 B4 19890922

L28 ANSWER 117 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
A method is described for regenerating the activity of deactivated
supported Ni catalysts that have lost activity after use as a
hydrogenation catalyst in the finishing stage of the process of
conversion of butynediol to butanediol. The method comprises contacting
deactivated catalyst with H at 200-500 to convert surface polymers
to gaseous compds. and removing these gaseous compds. from the vicinity contacting

the treated catalyst. In the preferred form of the invention, in a single

step, a slow stream of H is passed through the deactivated catalyst in situ as a fixed bed and the gaseous compds. are removed simultaneously in the H stream.

1983:4292 CAPLUS ACCESSION NUMBER: DOCUMENT NUMBER:

98:4292
Regeneration of supported nickel catalysts
Hort, Eugene V.; Dethomas, Waldo
GAF Corp., USA
EUr. Pat. Appl., 15 pp.
CODEN: EPXXDW
Patent TITLE: INVENTOR(S):

PATENT ASSIGNEE (5): SOURCE:

DOCUMENT TYPE:

English LANGUAGE:

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 61042	A1	19820929	EP 1982-101750	19820305
R: DE, GB				
US 4361495	A	19821130	US 1981-243315	19810313
JP 57167735	A2	19821015	JP 1982-38232	19820312
PRIORITY APPLN. INFO.:			US 1981-243315	19810313

ANSWER 118 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Spent HRI 3830 (Co-Mn) catalyst from the H-Coal process was regenerated

removing C and metal compds. Tests showed that the various activities of the regenerated catalyst were either superior or equivalent to those of

fresh catalyst. Spent Co-Mo and Ni-Mo catalysts from H-Oil process employing heavy petroleum residue were regenerated by the same procedures. Useful V and Ni compds. cauld be reclaimed by an addnl. chemical treatment step. Optimization and engineering anal. of

regeneration process for com. practice are underway. SION NUMBER: 1982:547372 CAPLUS MENT NUMBER: 97:147372

ACCESSION NUMBER: DOCUMENT NUMBER:

TITLE: Regeneration of spent catalysts from coal liquefaction

AUTHOR (S):

and petroleum residua processing operations Ganguli, Partha Hydrocarbon Res., Inc., Lawrenceville, NJ, 08648, USA Proceedings of the Intersociety Energy Conversion Engineering Conference (1982), 17th(Vol. 2), 825-30 CODEN: PIECDE; ISSN: 0146-355X CORPORATE SOURCE: SOURCE:

DOCUMENT TYPE: LANGUAGE:

ANSWER 119 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN The Br number of diisoamyl ether (I) [544-01-4], reflecting the content

unsatd. compds. undesirable for subsequent use of I in the preparation of polymerization catalysts, was decreased from 14 to 0.5 g/100 mL I by hydrogenation in 3 successive apparatus containing NL-Cr catalyst at space velocity of I 2 h-1 and NZ 1500 h-1, 200° and pressure 1.2-1.5 MPa. The activation of the catalyst was carried out with N2 at 200° and space velocity 40-50 h-1 for 2-4 h, followed by treatment with H2 for 4 h under the same conditions. The procedure of catalyst regeneration was similar to that of activation but continued for shorter time.

SSION NUMBER: 982:346537 CAPLUS

MENT NUMBER: 97:146537

E: Removal of unsaturated compounds from dissoamyl ether

ACCESSION NUMBER:

DOCUMENT NUMBER:

97:146537 Removal of unsaturated compounds from diisoamyl ether Timofeev, A. V.; Romanchenko, L. Ya.; Shumovskii, V. G.; Vol'f, Yu. I.; Zhukov, V. I.; Shestak, N. P.; Min'ko, T. G.; Koval'skaya, E. V. TITLE: AUTHOR (S):

CORPORATE SOURCE:

SOURCE:

USSR Khimicheskaya Promyshlennost (Moscow, Russian Federation) (1982), (7), 401-2 CODEN: KPRMAW; ISSN: 0023-110X Journal

DOCUMENT TYPE:

LANGUAGE:

ANSWER 120 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
A steam-reforming catalyst is reconditioned by H reduction followed
by contact with a 1.0-1.5:1 mixture of steam and C1-4 hydrocarbons at
650-950° with space velocity 15-800 L/h. When the volume ratio of
the product gas and CH4, C2H6, C3H8, or C4H10 (i.e., the feed stream) is
23.2:1, 5.6:1, 8.0:1, or 10.4:1, resp., the catalyst is ready for
further use. Thus, 0.6 L catalyst (14 mm diameter) containing 13% NA
after 8000 h on stream was heated at 850° in 650 L flowing H for 30
min before a 1:1.3 steam-C3H8 mixture was passed over at 750° with
space velocity 200 h-1. When the product gas-C3H8 volume ratio was
1.

the catalyst had 91% of its initial activity and was used for an addnl. 1982:426236 CAPLUS
97:26236
Estimation of activity of steam-reforming catalysts for hydrocarbons
Osaka Gas Co., Ltd., Japan
Jpn. Kokai Tokkyo Koho, 3 pp.
CODEN: JOKKZAF
Patent 4000 h.
ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE:

PATENT ASSIGNEE(S): SOURCE:

DOCUMENT TYPE:

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KTND DATE APPLICATION NO. DATE JP 57042337 JP 60021773 PRIORITY APPLN. INFO.: A2 B4 19820309 JP 1980-118824 19800827 19850529

JP 1980-118824

19800827

ANSWER 122 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN A process is described for fluid catalytic cracking of residuum and other heavy oils, gas oil, petroleum residue, reduced and whole crudes, and shale oils with high metals content. The coke deposits on the used cracking catalyst are reduced in amount by regeneration and the aminant

mmnant metals e.g., Ni, V, Cu, and Fe are deactivated in sufficient amount to reduce H and coke formation during cracking. Thus, the

lyst particles are contacted with fresh feed and associated recycle feed, and wherein the feed is cracked in a cracking zone. The used catalyst particles are subjected to alternate exposures of \$30 mins in duration of conditions comprising an oxidizing zone at >900°F wherein 02 in flue gas emitted from the oxidizing zone is >0.1 volume%,

and
a reducing zone at 900-1450°F, wherein the reducing atmospheric is a
material selected from the group, consisting of H, hydrocarbons, CO, and
mixts. thereof and is present in a concentration of 4-100 volume%. The
regenerated
acctalyst can be returned to the cracking zone.
ACCESSION NUMBER: 1982:22288 CAPLUS
DOCUMENT NUMBER: 958:222288
TITLE: Fluid catalytic cracking of heavy petroleum fractions
INVENTOR(S): FATENT ASSIGNEE(S): Standard Oil Co., USA
SOURCE: USXXXMM
DOCUMENT TYPE: Patent

Patent English DOCUMENT TYPE: LANGUAGE:

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE 19791105 US 4298459 PRIORITY APPLN, INFO.: 19811103 US 1979-91455 US 1979-91455 А

L28 ANSWER 121 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB A nonoxidative regeneration of hydrocracking catalysts consists of hydrostripping of the deposited coke by passage of H and petroleum middle distillate over the catalysts. This process also gives a hydrorefined middle distillate. Thus, a spent catalyst (Nio-Moo3/Al203-Sio2

containing 14.5 weight% C after hydrocracking of vacuum distillate) was

hydrostripped of coke by treatment with middle distillate and H at 673 K and 24.0 MPa, to yield an essentially N- and S-free product and a catalyst containing 5.5 weights

weight%
C (vs. 4.0 weight% C on fresh catalyst). The regenerated catalyst had a hydrocracking activity comparable to that of fresh catalyst.
ACCESSION NUMBER: 1982:126091 CAPLUS

DOCUMENT NUMBER: 96:126091

Reactivation or utilization of spent hydrocracking TITLE:

Reactivation or utilization of spent hydrocracking catalysts
Buerkner, Helmut; Mueller, Jochen; Pfeiffer, Fritz;
Pietz, Gerd; Printz, Dietmar; Rentzsch, Gabriele;
Richter, Fritz; Schneider, Wolfgang; Spindler,
Herbert; et al.
VEB Leuna-Werke "Walter Ulbricht", Ger. Dem. Rep. INVENTOR (S):

PATENT ASSIGNEE(S): SOURCE:

Ger. (East), 9 pp. CODEN: GEXXA8 Patent

DOCUMENT TYPE:

German

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE 19800624 DD 151764 PRIORITY APPLN. INFO.: z DD 1980-222105 DD 1980-222105 19811104

L28 ANSWER 123 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Requisite conditions and a flow sheet are given for the combustion of C
deposits in tubular reactors which inactivate Cu- and Ni-based
catalysts. The reactor is flushed with a gas at 280-400°, containing
gradually increasing O concos. from 0.5 to 5%, first at a rate of 0.2-2
m/s outside the pipes and then into the pipes charged with catalyst.

arrangement increases the heat transfer to the reactor sleeve and prevents

prevents

pipe cracking owing to excessive temperature difference. Thus, in the treatment

of 3 m3 of spent Cu catalyst after reduction of PhNO2 by H with
3,000 m3/h gas containing 469 m3/h steam for 50 h, the C content in the catalyst dropped from 17% to 3% on the average and Cu was oxidized to CuO.

Reduction by H at 180-220° was necessary before the next use.

SSION NUMBER: 1982:21909 CAPLUS
MENT NUMBER: 96:21909

ACCESSION NUMBER:

DOCUMENT NUMBER:

TITLE: INVENTOR(S):

96:21909
Regeneration of catalysts
Pasek, Josef; Pexidr, Vaclav; Dvorak, Bohumir;
Ruzicka, Vlastimil; Kocarek, Petr

PATENT ASSIGNEE (S): Czech. Czech., 4 pp. CODEN: CZXXA9 SOURCE: DOCUMENT TYPE:

Patent LANGUAGE: Czech

LANGUAGE: FAMILY ACC, NUM, COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE 19770524 19770524 19790330 CS 1977-3402 CS 1977-3402 CS 188729 PRIORITY APPLN. INFO.: В

ANSWER 124 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

As pent catalyst is hydrogenated in a mineral oil or its hydrogenation product under H at lower temperature and higher pressure than the hydrogenation, sometimes in the presence of another catalyst. Thus, Al203 loaded with 15% Mo03 and 3.5% CoO was ball-milled, mixed with kerosine, vacuum-dried at 80-90°, and used for hydrogenation of heavy oil containing S 5.1%, heptane-insol. asphaltenes 11.5, and Conradson C 15.9%, 1130 ppm V, and 106 ppm Ni at 50 kg/cm2 (H) and 430°. The spent catalyst was mixed with a 10-fold amount of light oil containing 3.0% S and 1.5-mm (diameter) catalyst.

(Al203 loaded with 14.1% Mo03 and 7.3% Nio) at 160 kg/cm2 (H) and 410° for 3 h, separated from the latter, filtered, washed with PhMe, and used again. The catalyst contained Al 12.9, C 61.4, H 4.45, S 4.19, V 1.30, and Ni 0.10% before treatment, and 26.7, 29.4, 2.49, 5.61, 2.49, and 0.27%, resp., of these components after treatment. To recover Z.49, and U.271, resp., of these components after treatment. To recove

and Ni, the catalyst was suspended in a 20-fold amount of H2O and injected with Cl at 30° for 1 h to contain 27.6, 31.0, -, 3.97, 0.95, and 0.121, resp., of the above components.

ACCESSION NUMBER: 1991:535657 CAPLUS

DOCUMENT NUMBER: 95:135657 CAPLUS

PATENT ASSIGNEE(S): Removal of precipitated carbon over hydrogenation catalyst for heavy oil Chiyoda Chemical Engineering and Construction Co., Ltd., Japan Jpn. Kokai Tokkyo Koho, 6 pp. CODEN: JCKXAP Patent LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION: 1

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. JP 56070840 JP 63045860 PRIORITY APPLN. INFO.:

KIND DATE DATE JP 1979-145498 19791112 JP 1979-145498 19791112

ANSWER 126 of 258 CAPLUS COPYRIGHT 2004 ACS on STN
The catalyst is finely dispersed Ni deposited upon clay. During
the reduction of nitrobenzene, the catalyst continually looses
activity due to the formation of organic resinous products on the

activaty due to the formation of organic resinous products on the Surface, blocking the active centers. The regeneration is accomplished by oxidation at 400-550 for 3 h, and H2 reduction of the oxide at 350-400 for 3 h.

ACCESSION NUMBER: 95:13447 CAPLUS

DOCUMENT NUMBER: 95:13447

TITLE: Study of the regeneration of nickel catalysts used in an illne production.

AUTHOR(S): CORPORATE SOURCE:

1981:413447 CAPLUS
95:13447
Study of the regeneration of nickel
catalysts used in aniline production
Gramatikov, K.: Kotseva, S.
Inst. Malotonazhn. Khim. Proizvod., Dimitrovgrad,
Bulg.
Khimiya i Industriya (1922-1988) (1981), (2), 54-6
CODEN: KINSAF; ISSN: 0368-5764
Journal
Bulgarian

DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 125 OF 250 CAPLUS COPYRIGHT 2004 ACS on STN
AB NOx is removed from flue gases by reduction with NH3 over a
regenerated hydrodesulfurization catalyst. This catalyst is obtained from

netroleum or coal processing plants and is regenerated by oxidation in

presence of O to burn off the coke coated on the catalyst and to convert metal sulfides to oxides. Thus, NOx was 100% removed from a flue gas-NH3 mixture (containing NO 195, NO2 5, NH3 250, and SO2 1000-60 ppm, and 5 volumes O) at 300-400° using a regenerated V-Ni-Co-Mo/y-Al203

1981:467209 CAPLUS

95:67209 GPRIDS 95:67209 SPRIDS PS:67209 GPRIDS PS:67209 GPRIDS Nagao Soda Co., Ltd., Japan Jpn. Kokai Tokkyo Koho, 5 pp. CODEN: JKXXAF

catalyst.
ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE:
PATENT ASSIGNEE(S):
SOURCE:

Patent Japanese 1

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 55167025	A2	19801226	JP 1979-74597	19790615
JP 61008733	B4	19860317		
IORITY APPLN, INFO.:			JP 1979-74597	19790615

L28 ANSWER 127 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The spent catalyst, containing TiO2 250% and 21 of Fe, Mo, V, W,
N1, Cr, Cu, Mn, Co. Bi, and Sn oxides is heated in HZSO4 and
hydrolyzed and (or) neutralized and regenerated. Thus, 300 g spent
granular catalyst containing TiO2 75, Fe2O3 15, and V2O5 7% (used for gas denitration) was boiled in 300 mL 45%  $\rm H2SO4$  for 3 h, cooled to 70°, and neutralized to pH 7 with aqueous NH3; the precipitate was filtered,

ed,
dried, mixed with a small amount of binder, pelletized to 10 diameter + 7
mm, and calcined at 500° for 2 h. The regenerated catalyst containing
TiO2 74.6, Fe203 15.2, and V205 6.9% had compressive strength 12 kg and
gave No removals of 82 or 91% at 250° or 300°, resp., when
crushed to 10-20 mesh and used for gas containing 300 ppm No, 500 ppm

crushed to 10-20 mean and uses as a few mixing with NH3 at space social and 0.0 4, CO2 13, and water 15% balance N after mixing with NH3 at space velocity 10,000 h-1. The data for the fresh catalyst had 81 and 91% removal efficiencies, resp.

ACCESSION NUMBER: 1981:412043 CAPLUS
DOCUMENT NUMBER: 95;12043
Treatment of spent titania-support flue gas denitration catalyst
PATENT ASSIGNEE(S): Ishihara Sangyo Kaisha, Ltd., Japan; Babcock-Hitachi K. K.

K. K.
Jpn. Kokai Tokkyo Koho, 5 pp.
CODEN: JKXXAF Patent Japanese 1

DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE JP 55145532 JP 58029143 PRIORITY APPLN. INFO.: 19801113 JP 1979-52379 19790427 19790427 L28 ANSWER 128 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Reaction products were extracted from spent catalysts by use of gases in AB the

supercrit. state as extraction agents. The supercrit. gas was then from the extraction reaction products products by expansion to a gaseous

state
and then recompressed for further use. The method is particularly applicable for treatment of used Ni and Cu (or Cu chromite) catalysts for hydrogenation of fats. The preferred supercrit gas was CO2.

ACCESSION NUMBER: 1980:502079 CAPLUS DOCUMENT NUMBER: 93:102079 CAPLUS TITLE: Extraction of reaction products from spent cataly Number (S): Abel, Heinz Number (S): VARTA Batterie A.-G., Fed. Rep. Ger. Offen., 10 pp. CODEN: GMXXRX CODEN: GMXXRX

1980:502079 CAPLUS
93:102079
Extraction of reaction products from spent catalysts Abel, Heinz
VARTA Batterie A.-G., Fed. Rep. Ger.
Ger. Offen., 10 pp.
CODEN: GWXXBX
Patent
German
1

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 2853065 PRIORITY APPLN. INFO.:	Al	19800619	DE 1978-2853065 DE 1978-2853065	19781208 19781208

AB ANSWER 130 OF 258 CAPLUS COPYRIGHT 2004 ACS ON STN

AB An used cracking catalyst containing a zeolite and heavy metals is impregnated with 0.1-1.5% B to increase its activity in the cracking of hydrocarbon oils. Thus, 50g used catalyst containing a zeolite and 0.13% Mi, 0.11% V, and 0.66% Fe was impregnated with 15 mL of an aqueous solution containing 0.43 to give a regenerated catalyst containing 0.43% B. When the B-containing catalyst was used in the cracking (480°) of a gas oil, the gasoline yield was 55% (based on the gas oil) and the coke yield was 55%, va. 51.2 and 8.6%, resp. when using the B-free used catalyst.

ACCESSION NUMBER: 1980:61576 CAPLUS 92:61576

ITILE: Catalyst for cracking of hydrocarbon oils Shell Internationale Research Maatschappij B. V., Neth. Jpn. Kokai Tokkyo Koho, 4 pp. CODEN: JKXXAF

DOCUMENT TYPE: Patent
LANGUAGE: Japaneae

FAMTLY ACC. NUM. COUNT: 2

PATENT INFORMATION:

DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 54122692	A2	19790922	JP 1979-17359	19790219
JP 54122692	B4	19871121		
FR 2417540	A1	19790914	FR 1978-4719	19780220
FR 2417540	Bl	19810904		
CA 1132122	A1	19820921	CA 1979-320417	19790129
IN 151186	А	19830305	IN 1979-DE63	19790129
BE 874053	A2	19790809	BE 1979-9271	19790209
ES 477845	Al	19790801	ES 1979-477845	19790219
NL 7901288	А	19790822	NL 1979-1288	19790219
AU 7944363	A1	19790830	AU 1979-44363	19790219
AU 532790	B2	19831013		
GB 2014468	А	19790830	GB 1979-5706	19790219
GB 2014468	B2	19820825		
BR 7901042	A	19791002	BR 1979-1042	19790219
ZA 7900748	А	19800227	ZA 1979-748	19790219
PRIORITY APPLN. INFO.:			FR 1978-4719	19780220

ANSWER 129 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

The spent catalyst [for γ-butyrolactone (I) and/or THF production from maleic anhydride (II) and/or its partial hydrogenation products) is heated at 200-900° in a H stream. Thus, the spent catalyst of ratios SiO2-Al2O3/N1 2(weight), Re/N1 0.03, and Mo/N1 0.07(atomic) was centrifuged, a 40 g portion of several mm diameter as a 50 mm high column in a 30 mm diameter vertical quartz tube was heated to and at 650° within and for 1 h each in a downward H stream at superficial velocity 1.4 cm/s, cooled, and reserved in N. When the catalyst 1.5 and a 1:4 mixture 150 g of I and II in an autoclawe with an electromagnetic stirer was pressured at 240° with H at 50 kg/cm2 gage pressure and 150 L/h under reflux at 5° for 3 days, and THF was condensed, the activity was 71.5 and selectivity 89 mol %, vs. 100(standard) and 86 with the fresh one, or 41.0 and 81 with the spent one.

one. ACCESSION NUMBER: DOCUMENT NUMBER: TITLE:

1980:170024 CAPLUS
92:170024
Regeneration of nickel-base
hydrogenation catalyst
Takeshita, Soichi; Ayuzawa, Tadashi
Mitsubishi Petrochemical Co., Ltd., Japan
Jpn. Kokai Tokkyo Koho, 6 pp.
CODEN: JXXXAF
Patent
Japanese
1 INVENTOR(S): PATENT ASSIGNEE(S): SOURCE:

DOCUMENT TYPE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE JP 1978-73908 JP 1978-73908 JP 55001832 PRIORITY APPLN. INFO.: A2 19780619 19780619 19800109

L28 ANSWER 131 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The deteriorated granular catalyst is regenerated by washing with C2-12 olefin and/or paraffin (containing no aromatic hydrocarbon) at -30° to +170° and 0-100 kg/cm2 gauge. The catalysts are Si02-, Al203-, MgO-, 2r02-, Ti02- or zeolite-supported Co, Mo, Ni, Cr, and Pd, and are used for gas-phase polymerization, hydroganation, thermal decomposition, dehydrogenation, delkylation, disproportionation, and isomerization of hydrocarbons, especially for hydroganation of acetylenes and/or C2-18 diolefins in olefins and/or paraffina, and in removal of S and/or N compds. in C7-18 kerosine. Thus, Al203-supported Pd

was used for hydrogenation of C2H2 in C2H4 prepared by naphtha decomposition The hydrogenation of C2H4 containing 1200 ppm C2H2 over the Pd catalyst lowered the C2H2 content to <1 ppm at 1st, and to 5-50

The Pd catalyst lowered the C2HZ content to <1 ppm at 1st, and to 5-50 ppm af 1st, and to 5-50 ppm af

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 54087693 PRIORITY APPLN. INFO.:	A2	19790712	JP 1977-155398 JP 1977-155398	19771223 19771223

L28 ANSWER 132 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Deteriorated catalyst due to precipitation of V is contacted with Cl or

Assume the state of the state o

Lyst containing V 6.3, Ni 2.2, Mo 6.4, Co 2.5, C 6.4, and S 9.1% 50 was mixed with sea sand 150 mL, filled in a glass tube, treated with Cl 600-700 mL/min at 150° for 15 min to remove all V and trace Mo, heated to 400° in a He stream, treated with steam 50 mL(as H2O)/h for 1 h, heated to 550°, and treated with air 500-600 mL/min for 3 h. The crushing strength was 2.2 kg, sp. surface area 235 m2/g, and pore volume 0.41 mL/g, vs. 2.0, 281, and 0.49 with fresh catalyst. When

Volume 0.41 mL/g, vs. 2.0, 281, and 0.49 with fresh catalyst. when regidual oil containing S 2.62, asphaltene 3.0%, and v 130 ppm was passed over the regenerated catalyst at 100 kg/cm2, 400°, and space velocity 1.0 h-1, the product after 100 h contained S 0.21% and V 41 ppm, vs. 0.20 and 39 with fresh catalyst.

ACCESSION NUMBER: 1979:617584 CAPLUS DOCUMENT NUMBER: 91:217584 Regeneration of catalyst coated with vanadium Invertor(S): 100 km Assatada; Shimizu, Toshiharu Chiyoda Chemical Engineering and Construction Co., Ltd., Japan SOURCE: Japan SOURCE: Japan SOURCE: JAPAN GORDEN JAXXAF Patent LANGUAGE: 14 Japanese 1 Japanese 2 Japa

DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 54099790	A2	19790806	JP 1978-6151	19780125
JP 57052863	B4	19821110		
PRIORITY APPLN. INFO.:			JP 1978-6151	19780125

L28 ANSWER 134 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

ABh -containing coal is liquefied by catalytic hydrogenation at

700-825° and >3500 psi in the presence of an aromatic solvent, and
the life of the catalyst (e.g., Co-Mo-MR/A1203) is extended by
removing it periodically for the separation of ash-containing
particulate matter.

The process includes a heating zone (750-90°F) and a dissoln. zone
(800-900°F) in advance of the hydrogenation zone.

Deactivation of the catalyst occurs when undissolved coal and ash are
deposited on the surface of the catalyst. To reverse this deactivation,
the catalyst is washed in place by an aromatic liquid, dried. removed
from the
reactor, crumbled, and sifted on a wire mesh screen. The material
passing

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 4157291	А	19790605	US 1978-878019	19780215
AU 7725922	A1	19781214	AU 1977-25922	19770608
GB 1584583	A	19810211	GB 1977-24922	19770615
ZA 7703679	A	19780530	ZA 1977-3679	19770620
JP 53077203	A2	19780708	JP 1977-142872	19771130
ORITY APPLN. INFO.:			us 1976-746179	19761130

L28 ANSWER 133 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB By use of the modified computer programs of S. Gordon and B. J. McBride
[1971] for the determination of heterogeneous phase and chemical equilibrium

preassigned temps. (300-1100 K) and atmospheric pressure (101,325 N m-2),

preassigned temps. (300-1100 K) and atmospheric pressure (107,323 M), the oxidation with O of S-poisoned Raney Ni catalyst and subsequent reduction with H are modeled thermodn. by using the properties of bulk Ni compds. An alternate process, the direct reduction with H of the sulfidized Ni, is also modeled and arguments are advanced for the further investigation of this thermodynamically favored 2nd process. In both processes the mole ratios of reactants, H/NiSO4 and H/Ni352, presp., for complete disappearance of the last product to be reduced, Ni352, increase markedly as the desired temperature for complete thermodn. reduction decreases. These ratios and the equilibrium activity quotient PH/PHIS are given as quant. functions of this critical reductions of this critical reductions.

the
reaction of O with mixts. of Ni and Ni3S2 is developed. From
the equilibrium calcns., Ni3S2 is relatively more stable both to
oxidation with O
than is Ni and to reduction with H than is NiO. One point
of modest connection with experiment is presented for the reduction

1979:582003 CAPLUS

processes.
ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE:

1979:582003 CAPLUS
91:182003
Thermodynamic model for the regeneration of sulfur-poisoned michel catalyst. 1. Using thermodynamic properties of bulk michel compounds only Chughtai, A. R.; Riter, J. R., Jr. Chem. Dep., Univ. Denver, Denver, CO, 80208, USA Journal of Physical Chemistry (1979), 83(21), 2771-3 CODEN: JPCHAX; ISSN: 0022-3654
English

AUTHOR(S): CORPORATE SOURCE: SOURCE:

reactor, crumbled, and sifted on a wire mesh screen. The material passing through the screen is discarded, and the purified catalyst particles are returned to the reactor.

ACCESSION NUMBER: 1979:560236 CAPLUS

DOCUMENT NUMBER: 91:160236

ITITLE: Extending life of coal liquefaction catalyst

INVENTOR(S): Paraskos, John A.; Taylor, Herman, Jr.

Gulf Research and Development Co., USA

SOUNCE: U.S., 9 pp.

DOCUMENT TYPE: Patert

LANGUAGE: Patername LANGUAGE: 2

PAMILY ACC. NUM. COUNT: 2 DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

L28 ANSWER 135 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Spent Ni catalyst from an oil hydrogenation plant was
regenerated by converting the Ni content of slurry into NiO and
Ni(OH)2, and subsequently reducing these compds. by H at 450 and
290°, resp. Lab and plant apparatus for reduction are designed.
Activity of regenerated catalyst was measured in comparison with
reference reference
catalyst (G15). Pretreatment of discarded catalyst prior to redn

1979:454808 CAPLUS

CATCATYST (GIS).
. is described.
ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE:
AUTHOR (S):
CORPORATE SOURCE:
SOURCE:

1979:454808 CAPLUS
91:54808
Regeneration of used nickel catalyst
Massoumi, A.; Kajbaf, M. A.
Dep. Chem., Shiraz Univ., Shiraz, Iran
Journal of the American Oil Chemists' Society (1979),
56(5), 565-8
CODEN: JACCA7; ISSN: 0003-021X
JOURNAL
English

DOCUMENT TYPE: LANGUAGE:

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ANSWER 136 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Spent catalyst, SOX-resistant support carrying metal oxide to reduce NOX with NH3, is contacted with gas containing NO at 2200°. The catalyst is of TiO2 or SiO2 support and metal oxide: Cu, Fe, Co, Ni, Cr, Mn, or V oxide. Thus, TiO2 support carrying V205 used for gas containing NOX 110-70, SO2 110-50, SO3 1-3 ppm, 0 0.6-2.5, CO2 13,
  H20
               12%, and dust 5-10 mg/m3 from a boiler and deteriorated from 91 to 71%
  NOX
                removal (S content 0.3 to 1.2%) was heated in a gas stream containing NO
or 320, SO2 165 ppm, O 4, CO2 12, and H2O 10% at 270° and space velocity 10,000 h-1 for 2.3 or 1.4 h to recover the activity 100%.

ACCESSION NUMBER: 1979:210813 CAPLUS
SOCUMENT NUMBERS: 90:210813 CAPLUS
SOURCE: Regeneration of a denitration catalyst Hino, Masaor Mitsucka, Shigetoshi Mitsubishi Heavy Industries, Ltd., Japan Jpn, Kokai Tokkyo Koho, 4 pp.
CODEN: JKXXAF
DOCUMENT TYPE: Patent
  200
  DOCUMENT TYPE:
                                                                           Patent
Japanese
 FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
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DATE PATENT NO. KIND APPLICATION NO. DATE 19790206 19850913 JP 54016394 JP 60040895 PRIORITY APPLN. INFO.: JP 1977-81017 19770708 JP 1977-81017

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: 19770708 PATENT NO. DATE APPLICATION NO. DATE KIND FR 2354814
FR 2354814
US 4101444
US 4102811
US 4243550
CA 1087591
CA 1087592
BR 7703818
US 4163710
PRIORITY APPLN. INFO.: 19780113 19840210 19780718 19780725 19810106 19801014 19801014 19780418 19790807 FR 1977-17853 19770610 19760614 19760614 19760614 19770526 19770527 19770613 19780213 19760614 us 1976-695662 19760614 US 1976-695687 19760614 US 1977-695687 19770614

DOCUMENT NUMBER: TITLE: INVENTOR (S):

DOCUMENT TYPE:

PATENT ASSIGNEE(S): SOURCE:

ANSWER 138 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
The activity of the catalyst was completely restored by treating it with
an alkaline solution and a nonionic detergent containing flotation and an alkaline solution and a nonionic detergent containing flotation and dispersion
additives consisting of Na metasilicate 1.5-2, Na tripolyphosphate 0.3-0.5, and calcined soda 0.5-1 weight x, with the remainder being H2O.
ACCESSION NUMBER: 1978:553422 CAPFUS
DOCUMENT NUMBER: 89:153422
FUTITE: 50:153422
FUTITE: for hydrogenation of a nickel-ceramic catalyst for hydrogenation of vegetable oils and fats
Raemskaya, N. P.; Filimonova, Z. V.; Gorenshtein, B. M.; Shmidt, A. A.; Aznaur'yan, M. P.; But, N. D.; Konevets, E. M.
PATENT ASSIGNEE(S): All-Union Scientific-Research Institute of Fats, Moscow, USSR, Saratov Fats Combine
U.S.S.R. From: Othrytiya, Izobret., Prom. Obraztsy, Tovarnyc Znaki 1978, 55(28), 12.
CODEN: URXXAF
PAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE SU 617066 PRIORITY APPLN. INFO.: 19740728 19780730

L28 ANSWER 139 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB In the title process, a slurry of ash-containing coal in a solvent is preheated and hydrogenated over a supported Group VI or VIII metal catalyst at 370-440'/>21.7 MPa. When the catalyst becomes deactivated, metallic impurities are removed mech. Thus, a slurry of coal
in a coal liquefaction vacuum distillate is preheated to 421°, held
in a dissolver at 482°, and hydrogenated over a Co-MoNI/Al203 catalyst at 987° and weight space velocity 1.34/h to
qive a 94.33 conversion, with a 58.33 yield of liquid b. <415° and a
92.68 recycle of calcium distillate (b. 232-412°) as solvent.

ACCESSION NUMBER: 1978:549398 CAPLUS

DOCUMENT NUMBER: 89:149398

Liquefaction of coal
INVENTOR(S): Paraskos, John Angelo; Taylor, Herman, Jr.
GUIF Research and Development Co., USA
GUIF RESEARCH AND PROCUMENT TYPE: Patent

DOCUMENT TYPE: Patent

Paten DOCUMENT TYPE: LANGUAGE: Patent German 2 FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

L28 ANSWER 137 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Hydrocarbon conversion catalysts which have been poisoned by metals in feed are regenerated by washing with a reducing solution, optionally

min at

10°, rinsed, and treated twice as a 20% aqueous slurry with 4.54-9.07

kg H202/ton at 80° for 3 min, decreasing the NA, Fe, and
U contents by 82, 40, and 45%, resp., and restoring activity to 75.6.

ACCESSION NUMBER: 1978:582205 CAPLUS

Sun, Jui-Yuan Atlantic Richfield Co., USA Fr. Demande, 51 pp. CODEN: FRXXBL

owed by washing with an oxidizing solution Thus, a zeolite cracking catalyst (initial activity rating 80) is used until the metal contents are Ni 2500, V 7500, and Fe 6800 ppm and the activity falls to 59.1. The catalyst is then oxidized to remove C, heated 4 h at 732 with 2.75% (based on catalyst) HZS/min (diluted with N), stabilized in N at 316°, heated with 8.0 L air/min-kg at 332-60° for 25 min, cooled, treated 3 times as a 20% aqueous slurry with SO2 (pH 2.0) for 3

Burk, Emmett H., Jr.; Yoo, Jin S.; Karch, John A.;

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 2728611	A1	19780601	DE 1977-2728611	19770624
AU 7725922	A1	19781214	AU 1977-25922	19770608
GB 1584583	A	19810211	GB 1977-24922	19770615
ZA 7703679	A	19780530	ZA 1977-3679	19770620
JP 53077203	A2	19780708	JP 1977-142872	19771130
PRIORITY APPLN. INFO.:			US 1976-746179	19761130
AU 7725922 GB 1584583 ZA 7703679 JP 53077203	A1 A A	19781214 19810211 19780530	AU 1977-25922 GB 1977-24922 ZA 1977-3679 JP 1977-142872	197706 197706 197706 197711

ANSWER 140 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Spent catalysts for the hydrogenation of isobutylene (I) oligomers are regenerated by treatment with hydrocarbon oils free of asphaltenes at 250-330°, 20-50 kg/cm2, and 1.5-3.5 h-1 (liquid space velocity) in the presence of 100-200 times the catalyst volume of H for

Thus, a H2S-treated catalyst containing 3.0% Ni and 15.0% Mo was h. Thus, a MZS-treated catalyst containing 3.01 Ni and 15.01 No was used for the hydrogenation of I oligomers. The spent catalyst was regenerated with 3 h-1 (liquid space velocity) of light oil at 300° and 30 kg/cm2 in the presence of 100 times the catalyst volume of H for 20 h. The product of I oligomer hydrogenation after the catalyst regeneration contained isobutant 2.5, Ce alkane 82.2, and Cl2 alkane 15.3 weightt, compared with 7.2, 84, and 8.5 weightt, resp., for a similar product before catalyst regeneration.

ACCESSION NUMBER: 1978:477485 CAPLUS
DOCUMENT NUMBER: 89:77485
TITLE: Regeneration of hydrogenation catalysts for isobutylene oligomers

INVENTOR (S):

89:77485
Regeneration of hydrogenation catalysts for isobutylene oligomers
Kosugi, Masakata; Sekine, Sachio; Takizawa, Haruo
Maruzen Oil Co., Ltd., Japan
Jpn. Kokai Tokkyo Koho, 5 pp.
CODEN: JKXXAF
Patent
Japanese PATENT ASSIGNEE(S): SOURCE:

DOCUMENT TYPE: Japanese LANGUAGE:

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

APPLICATION NO. DATE PATENT NO. KIND DATE JP 53011894 PRIORITY APPLN. INFO.: A2 19780202

ANSWER 141 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

A process is described for regenerating a Ni catalyst used for reduction of polyenes and acetylenes to alkenes. The process consists of removal of volatiles by treatment with an inert carrier gas such as steam at 120-350 °, elimination of deposited hydrocarbons by oxidation with 0 and a carrier gas at 350-450 °, reduction with H and a carrier gas at 350-450 °, and a final treatment with H at 350-450 °. The catalyst can be washed with an azomatic solvent prior to treatment. The process serves to remove S compds. For example, a catalyst of 10 weight % Ni on sepiolite was washed with benzene, treated with steam at 330 ° for 20 h, then with a mixture of O and steam for 2 h, slowly increasing the O content to 1.6 % and the temperature to temperature to
450°, followed by a mixture of 20-40% H and steam for 5 h at 400°, and finally a 20 h treatment with H at 450°. The regenerated catalyst performed better than a freshly prepared one.

ACCESSION NUMBER: 1978:142285 CRPLUS
DOCUMENT NUMBER: 88:142285 CRPLUS
ITILE: Regeneration of catalysts containing mickel
Carruthers, John: Olive, Martin Frederick
FATENT ASSIGNEE(S): British Petroleum Co. Ltd., UK
FOLOMENT TYPE: Patent
LANGUAGE: FRXXBL

DOCUMENT TYPE: Patent
LANGUAGE: FRXCBL

ACTION TO THE PATENT ASSIGNETION: 1

PATENT INFORMATION: 1

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
FR 2329347	A1	19770527	FR 1975-33236	19751030
FR 2329347	B1	19790504		
PRIORITY APPLN. INFO.:			FR 1975-33236	19751030

twe regenerated catalysts useful for the hydrogenation of hydrocarbons. Thus, 100g waste catalyst containing V 9.5, No 6.2, Ni 2.0, Co 2.9, S 2.0, and C 10.5% was ground with 100 g zeolite to <200°  $\mu$ , 100 g of a 5% solids Al203 sol solution was added, and the mixture was

extruded, pelletized, dried, calcined 2 h at 550°, immersed 2 h in
500 H2O at 80°, and calcined again at 300° for 1 h. A
petroleum distillation residue containing 2.87% S and 150 ppm V was
hydrorefined in
the presence of the regenerated catalyst at 400° and 140 kg/cm2 (H
pressure) to give a product oil containing 1.4% S and 45 ppm V, compared 2.2% and 110 ppm, resp., when the residue was hydrorefined in the 2.2% amo no ppm, arrivers.

Presence
of a similar catalyst without regeneration.

ACCESSION NUMBER: 1977:554774 CAPLUS

DOCUMENT NUMBER: 87:154774

TITLE: Regeneration of hydrogenation catalysts for petroleum residues Inooka, Masayoshi; Wakabayashi, Motonobu; Nakamura, INVENTOR (S): Inooka, Masayoshi; Wakapayashi, Motonobu; Nakamura Munekazu; Morimoto, Tatsuo Chiyoda Chemical Engineering and Construction Co., Ltd., Japan Jpn. Kokai Tokkyo Koho, 5 pp. CODEN: JKXXAF

ANSWER 142 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Waste catalysts containing Mo, W, Co, N1, and (or) V are ground to 1-200  $\mu$ , pelletized, calcined, and extracted with H2O or aqueous solns.

PATENT ASSIGNEE (S):

SOURCE:

DOCUMENT TYPE: Patent.

Japanese 1

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE DATE 19770608 19820501 JP 52068890 JP 1975-143932 19751205 JP 57020852 PRIORITY APPLN. INFO.: JP 1975-143932 19751205

L28 ANSWER 143 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

A8 A catalyst comprising 10% Ni on sepiclite having S-Ni
atomic ratio 0.092:1, which had been in continuous use for 9150 h in the
selective hydrogenation of steam cracker gasoline was
regenerated using the following processes, in which all liquid and gas
components were in downflow. Washing with benzene 45 h at 200°, 34
bars (gage), space velocity 2 volume/volume/h, and H recycle rate 90 removed 5.5% of gum after which the catalyst was purged with steam 20 h  $350^{\circ}$ , 1 bar, and 1200 volume/volume/h. The catalyst was then regenerated by mixing air with the steam flow to give an O content of 0.2 increasing to 1.6 volume % as the inlet temperature rose from 350 to

The catalyst bed was purged of residual O by continuing the steam flow

 ${\bf 2}$  h, and then the bed was reduced by an H-steam mixture initially for  ${\bf 5}$ 1 bar, H flow 200, and steam flow 300 volume/volume/h. The steam

concentration was
tapered off to 0 over the 5 h, and the catalyst was finally reduced 20 h
at 1 bar with H at 1000 volume/volume/h, maximum catalyst temperature
450°.

1977:158123 CAPLUS
86:158123
Improvements relating to the regeneration of support nickel catalysts
Carruthers, John: Olive, Martin F.
British Petroleum Co. Ltd., UK
Brit., 6 pp.
CODEN: BRXXAA
Patent
English
1 450°. ACCESSION NUMBER: DOCUMENT NUMBER: TITLE:

INVENTOR(S): PATENT ASSIGNEE(S): SOURCE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

KIND APPLICATION NO. DATE DATE GB 1455030 PRIORITY APPLN. INFO.: 19761110 GB 1974-18121 GB 1974-18121 19750409 L28 ANSWER 144 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Deteriorated Raney-type catalyst is washed with warm H2O and/or aqueous

and heated in aqueous alkali with a 1/20-1/200 amount of fresh Raney

alloy at 40-90° for 15-90 min. Thus, Raney Ni catalyst (used 50 times) 500 l. (1250 kg) was stirred with H20 500 l. at 50°, then with 2% NaOH 500 l., and then with 2% NaOH 325 l. and fresh alloy 10.8

suspended in 50 1. H20 and added in 20 min. at <50°, heated at 60 ± 2° for 1 hr, attrred with 5 + 2000 1. H20 to obtain regenerated catalyst 550 1. (1260 kg). When used for reduction of glucose, mannitol 0.4% was produced, vs. 3.4% with the used Catalyst. DOCUMENT NUMBER: 1976:567302 CAPLUS

DOCUMENT NUMBER: TITLE: INVENTOR(S): 85:167302
Regeneration of Raney-type catalyst
Yamanoi, Hidesuke: Masuda, Tamee
Nikken Chemicals Co., Ltd., Japan
Jpn. Kokai Tokkyo Koho, 4 pp.
CODEN: JKXXAF PATENT ASSIGNEE(S): SOURCE:

Japanese

FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

DOCUMENT TYPE:

PATENT NO. APPLICATION NO. DATE KIND DATE A2 JP 51084792 PRIORITY APPLN. INFO.: 19750124 19750124 19760724

L28 ANSWER 146 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB In the determination of coke on the title catalysts (by observing weight

on regeneration in air at 600°), the catalysts were preheated in He at 600° for 2 hr, to introduced a correction for the adsorbed gases and moisture; also, a correction was used for the amount of O used to oxidized the reduced form of the catalyst. The coke contents thus,

oxidized the reduced form of the catalyst. The coke contents thus, determine were lower by a factor of 2 than those obtained by conventional methods. A Mo-Ni/Al2O3 catalyst contained 3-4 times more coke than a Mo-Ni/bentonite catalyst, when each was used in hydrorefining gasoline-ligroin at 400° and 40 atm for >100 hr. The gases formed in regeneration of catalysts contained SO2, CO2, and CO.

ACCESSION NUMBER: 1976:480594 CAPLUS

DOCUMENT NUMBER: TITLE: 85:80594 Study of the regeneration process for

hydrogenation-refining catalysts Ismailov, T. S.; Sharipov, A. Kh.; Talipov, G. Sh.

AUTHOR (S): CORPORATE SOURCE: Deposited Doc. (1973), VINITI 6495-73, 10 pp.

SOURCE: Avail.:

VINITI

DOCUMENT TYPE: LANGUAGE: Report Russian

L28 ANSWER 145 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB A mixture of 2-isoamylanthraquinone and
tetrahydro-2-isoamylanthraquinone in
an organic solvent is hydrogenated in the presence of a Ni
catalyst heated previously to 120-160° in an alkali medium. With
200-300 g tetrahydro-2-isoamylanthraquinone/1. the degree of
hydrogenation of the 2-isoamylanthraquinone is 55-601. After
removal of the Ni catalyst, oxidation of the hydrogenated
isoamylanthraquinone, and extraction of the oxidized product with water an aqueous solution containing 30-44 g H2O2/1. is obtained. Thus, a solvent of equal vols.

of tert-butyltoluene and 2-octanol containing 2-isoamylanthraquinone 250 tetrahydro-2-isoamylanthraquinone 250 g/l. was hydrogenated at 60-65° in the presence of a Ni catalyst treated previously at 120-160° for 1-3 hr in 20-51 aqueous NaOH with 25.4 l. H/l. The Ni catalyst was removed and the solution treated 0 or an 0-containing gas and then the H2O2 extracted with water. The product O-containing gas and then the new new was an aqueous solution containing 38.6 g H202/1.

ACCESSION NUMBER: 1976:26755 CAPLUS
DOCUMENT NUMBER: 85:126755
ITILE: Hydroyen peroxide by the anthraquinone process Hydroyen peroxide by the anthraquinone process Hydroyen peroxide by the anthraquinone process Hydroyen Counter Tranchuk, V. 1.; Ovchinnikova, L. I.; Kosareva, V. F. USSR
SOURCE: Ger., 3 pp.
CODEN: GRXXAW
DOCUMENT TYPE: Patent
LANGUAGE: PAHLLY ACC. NUM. COUNT: 1

APPLICATION NO. DATE PATENT NO. KIND DATE 19710616 19751120 19760930 DE 1667515 DE 1667515 DE 1667515 PRIORITY APPLN. INFO.: DE 1968-F54579 19680118 DE 1968-F54579 19680118

AB Spent hydroalkylation catalysts, containing active metal such as Co or Ni on an acidic oxide support, are regenerated by heating to regeneration temperature in the presence of H. For example, a zeolite Y catalyst containing 6% Ni and 19% W was used at 325°F to react undried benzene plus H to give 16.1% cyclohexylbenzene, which over 82 hr dropped to 6.8% yield. The catalyst unit was then flushed with benzene at 325°F, N at 325°F, then N at 900°F, and finally H at 900°F, After cooling, the regenerated catalyst now gave 20.8% cyclohexylbenzene under the initial conditions, a yield >200% better than with the used catalyst and 30% better than with the initial catalyst.

ACCESSION NUMBER: 1976:141304 CAPLUS
DOCUMENT NUMBER: 8141304
HYDROALSHALLS
SURGET: Suggitt, Robert N.; Crone, John N., Jr.
PATENT ASSIGNEE(S): Texaco, Inc., USA
DOCUMENT TYPE: CODEN: USX/AM
PATENT INFORMATION: 2

DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

APPLICATION NO. PATENT NO. DATE KIND DATE US 3869523 US 3926842 PRIORITY APPLN. INFO.: US 1973-425403 US 1973-320215 US 1973-320215 19750304 19731217 19730102 19730102

L28 ANSWER 148 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB \$102 or Al203-supported catalysts containing Mo and Mi and(or) Co
used for hydrogenation desulfurization of petroleum residues and
contaminated with V are treated with aqueous basic solns. (to selectively
remove V and Mo) and impregnated with Mo salts to supplement Mo and
regenerate the catalysts. Thus, a used catalyst was washed with kerosine
and calcined at 550°. The catalyst (100 g) was immersed 45 min in
2 l. of 1% NaON at 80°, washed with H2O, dried, immersed of 200 ml
of an aqueous solution containing 21 g (NH4)2NoO4, dried at 150°, and
calcined at 550°. The regenerated catalyst having compressive strength

ined 3 hr at 550°. The regenerated catalyst having compressive strength 1.6 kg, sp. surface area 272 m2/g, and voids 0.69 cm3/g was used 100 hr

to
treat a petroleum residue containing 4.19% S at 390°, 100 kg/cm2, and 1
hr-1 to remove 75% of S, compared with 1.6, 85, 0.3, and 55, resp., for a
similar used catalyst regenerated by calcination, and 2.3, 300, 0.95, and
80, resp., for a similar fresh catalyst.
ACCESSION NUMBER:
1976:138256 CAPLUS
DOCUMENT NUMBER:
17TITLE:
Regeneration of desulfurization catalysts
Sugawara, Yujiro; Nakazawa, Tadahiaa: New York 100 kg.

84:138256
Regeneration of desulfurization catalysts
Sugawara, Yujiro; Nakazawa, Tadahisa; Usui, Koichi;
Ogawa, Masahide; Komatsu, Masaie
Mizusawa Industrial Chemicals, Ltd., Japan
Jpn. Kokai Tokkyo Koho, 17 pp.
CODEN: JKXKAF

PATENT ASSIGNEE(S): SOURCE:

DOCUMENT TYPE:

LANGUAGE: J.
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 50075185	A2	19750620	JP 1973-124002	19731106
JP 54009995	В4	19790428		
PRIORITY APPLN. INFO.:			JP 1973-124002	19731106

ANSWER 150 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB NOX [11104-93-1] is continuously removed from waste gases by using catalysts such as Cu and Mi oxides in the presence of reducing gases in a series of chambers, which allows catalyst, regeneration. Thus, waste gas from a boiler was first treated to remove dust, heated to 300-400°, the O2 content brought to <21, a reducing agent, e.g. H2 [1333-74-0] added, catalytically reduced, and residual reducing agent burned.

ACCESSION NUMBER: 976:64786 CAPLUS

DOCUMENT NUMBER: 916:64786 CAPLUS

ACCESSION NUMBER: 916:4786 CAPLUS

BYTCHE: Nitrogen oxide removal from waste gas

ILO, Hayami

FATENT ASSIGNEE(S): Kawasaki Heavy Industries, Ltd., Japan

JOD. COLORNI TYPE: Patent

LANGUAGE: JAPANES

JAPANES

Japanese

FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 50045771 PRIORITY APPLN. INFO.:	A2	19750424	JP 1973-95982 JP 1973-95982	19730827 19730827

L28 ANSWER 149 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
A8 pent catalyst containing Ni and/or Co, W, and a zeolite dispersed
in a silica-alumina matrix was heated in situ in a stream of H2 gas at
temps. >310 for 1-24 hr. The regenerative gas should contain at
least 30 mole % H2. Tests showed that the activity of the regenerated
catalyst was restored to the initial level, and in some cases in excess

ACCESSION NUMBER: DOCUMENT NUMBER:

the initial activity.
SSION NUMBER: 1976:96082 CAPLUS
84:96082
84:96082 Regenerating spent hydroalkylation catalyst TITLE:

containing

an oxidizable group VIII metal Suggitt, Robert M.; Crone, John M., Jr. Texaco Inc., USA U.S., 6 pp. CODEN: USXXAM INVENTOR (S):

PATENT ASSIGNEE (S): SOURCE:

DOCUMENT TYPE: LANGUAGE: Patent English 2

LANGUAGE: FAMILY ACC, NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 3926842	А	19751216	US 1973-320215	19730102
US 3869523	A	19750304	US 1973-425403	19731217
PRIORITY APPLN. INFO.:			US 1973-320215	19730102

L28 ANSWER 151 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Petroleum hydrorefining catalysts were freed of coke and S [7704-34-9] by
spreading the particles in a shallow bed (55 particles deep),
flooding the bed with air through its shortest dimension, heating the
particles uniformly, and reducing the temperature of the particles to the sintering temperature. The high temps, used avoided the formation of toxic

products, e.g., Ni carbonyls. The catalyst layer was carried on
a perforated belt through a series of 9 or 10 chambers, in which the
regeneration conditions were separately controlled. The equipment was
described.

ACCESSION NUMBER: 1975:596119 CAPLUS DOCUMENT NUMBER: TITLE: INVENTOR(S): 83:196119 Regeneration of catalysts used in petroleum refining Hickman, Leo M.; Miller, Howard L.; Bacic, Thomas A. PATENT ASSIGNEE (S): SOURCE: USA Ger. Offen., 37 pp. CODEN: GWXXBX DOCUMENT TYPE: Patent LANGUAGE: German 1 FAMILY ACC. NUM. COUNT: PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE DE 1974-2461067 NL 1974-16823 FR 1974-42908 JP 1975-4218 US 1973-428666 19741223 19741224 19741226 19741227 19731227 A1 A A1 A2 19750717 DE 2461067 PRIORITY APPLN. INFO.:

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L28 ANSWER 152 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The replacement of descending flow by ascending flow of a gas-liquid reaction mixture through a fixed catalyst bed was examined in the
evacuation
of liquid phase when passing from processing to catalyst
regeneration. The reactor was operated with Al-Ni
[7440-02-0]-Mo [7439-98-7] catalyst to hydrofine vacuum distillate at
340-70. The regeneration in the reactor with ascending flow
burned off the coke in 14 hr (compared with 28 hr for descending flow).
No complications or temperature flash were produced during coke
combustion in
combustion in the regeneration zone; thus, reactors with ascending flow were usable for hydrogenation.

ACCESSION NUMBER: 1975:596031 CAPLUS
                                                                                         S3:196031
Testing the practice of catalyst
Testing the practice of catalyst
Tegeneration in reaction vessels for
liquid-phase hydrogenation processes with
ascending flow of the reacting mixture
Solov'ev, V. G.; Ermilov, G. T.; Tremasov, V. A.;
Terekhin, E. M.; Vasciko, A. I.; Shevelev, Yu. V.;
Kurganov, V. M.; Kislov, V. D.
Vses. Nauchno-Issled. Inst. Neft. Prom., Moscow, USSR
Nettepererabotka i Neftekhimiya (Moscow, Russian
Federation) (1975), (7), 8-10
CODEN: NNNSAF; ISSN: 0233-5727
Journal
Russian
DOCUMENT NUMBER:
TITLE:
                                                                                            83:196031
AUTHOR (5):
CORPORATE SOURCE:
DOCUMENT TYPE:
LANGUAGE:
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ANSWER 153 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The activity of a zeolite catalyst used in the alkylation of hydrocarbons (e.g., isobutane with butene) was maintained by adding MA, Pt, Pd, Ru, or Rh to the catalyst and hydrocarating periodically with a solution of H in a hydrocarbon solvent.

ACCESSION NUMBER: 1975:97649 CAPLUS

DOCUMENT NUMBER: 25:97649 CAPLUS

INVENTOR(S): PATENT ASSIGNEE(S): Union Carbide Corp.

U.S., Spp.

CODEN: USXXXAM

DOCUMENT TYPE: Patent
 DOCUMENT TYPE:
                                                                                                Patent
English
 LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                     PATENT NO.
                                                                                                                                                                       APPLICATION NO.
                                                                                                                        DATE
                                                                                                                                                                                                                                                             DATE
                                                                                                 KIND
                                                                                                                                                                                                                                                            19730927
19740926
19740926
19740926
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19740926
19740926
19730927
US 3851004
BE 820397
NL 7412731
DE 2445927
FR 2245586
JP 50059301
GB 1483939
PRIORITY APPLN. INFO.:
                                                                                                                           19741126
                                                                                                                                                                       us 1973-401178
                                                                                                  A
A1
A
A1
A1
A2
                                                                                                                                                                      OS 1973-401178
BE 1974-148949
NL 1974-12731
DE 1974-2445927
FR 1974-232508
JP 1974-110073
GB 1974-41844
US 1973-401178
                                                                                                                           19750326
19750402
19750410
19750425
                                                                                                                           19750522
19770824
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L28 ANSWER 154 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Rancy Ni catalysts, used in the hydrogenation of
invert august to sorbitol and mannitol, are regenerated by the addition of Mg of Mg
metal powder to an acidified mixture of catalyst and invert sugar
solution,
prior to hydrogenation, forming a soluble Mg salt. The
hydrogenation is conducted at 400-2000 psig. and pH 5-5. The
process produces a high purity hydrogenated invert sugar containing
< 0.3% nonreducing sugars and < 1% impurities. The catalyst is
separated by C 0.3% nonreducing sugars and < 1% impurities. The Garage Separated by filtration and used in the next batch. The Mg ions may be removed by ion exchange. In an example, the procedure was repeated 20 times without any apparent loss in catalyst activity.

ACCESSION NUMBER: 1975:90551 CAPPLUS
DOCUMENT NUMBER: 82:90651
Raney nickel or cobalt catalyst regeneration
INVENTOR(S): Krzeminski, Zbigniew S.
C.S.R. Chemicals Ltd.
SOURCE: Pat. Specif. (Aust.), 12 pp.
CODEN. ALXXAP
DOCUMENT TYPE: Patent Patent English LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: PATENT NO. DATE APPLICATION NO. DATE

19741016

AU 1971-32711

and/or						
C1-C4 carboxylic a	nid eal	te of alkali	and alkalia	a earth met	21 e 20	
20-200° under an i					als at	
30 times to give a	n equal	vield when	stirred with	100 ml 0.5	N NaOH at	
90° for 3 hr under	Hina	n autoclave	after each			
hydrogenation of is	oprene	peroxide po	vmer to obt	ain diols.		
ACCESSION NUMBER:						
DOCUMENT NUMBER:	B2:64957					
	Regen	Regeneration of Raney mickel				
INVENTOR (S):	Mabuchi, Shunsuke; Tsuzuki, Kenji; Matsunaga,					
Hideaki;	Habas	iii, biiuiibun	, ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	caja, adaba	gu,	
HIGGSKI,	Chimi	zu, Sadami;	Cumita Make	at o		
PATENT ASSIGNEE(S):		Soda Mfg. Co		,,,,		
SOURCE:						
SOURCE.	Jpn. Kokai Tokkyo Koho, 4 pp. CODEN: JKXXAF					
DOCUMENT TYPE:	Paten					
LANGUAGE:	Japanese					
FAMILY ACC. NUM. COUNT:		ese				
PATENT INFORMATION:	2					
PATENT INFORMATION:						
PATENT NO.	KIND	DATE	APPLICATIO	N NO.	DATE	
JP 49086290	A2	19740819	JP 1972-12	29219	19721225	
JP 53038719	B4	19781017				
DE 2362253	A1	19740704	DE 1973-23	362253	19731214	
DE 2362253	C3	19790301				
GB 1400340	A			9098	19731220	
FR 2211287	A1	19740719 19750722	FR 1973-45	980	19731221	
US 3896051	A	19750722	US 1973-42	7683	19731226	
US 3980720	A		US 1974-53	5125		
PRIORITY APPLN. INFO.:				9219		
			US 1973-42	7683	19731226	
					22.31220	

L28 ANSWER 155 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Raney Ni used in hydrogenation of organic peroxide
polymer is washed with 0.1-5N solution of NH3, hydroxides, carbonates,
and/or

AU 454487

L28 ANSWER 156 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB A Raney Ni catalyst employed in hydrogenation of organic
peroxides is regenerated by washing with a 0.15N solution at the performers is regarded over a manner of the solution contains at least 1 compound selected from NH3, hydroxide or carbonate of alkali metal or alkaline line earth metal, and alkaline earth salts of carbonic acids. Thus, a butadieneperoxide polymer was hydrogenated at 70° and H pressure 50 kg/cm2 with 10g of Raney Ni and the yield of 1,4-butanediol was 564. Then the catalyst used was regenerated with 100 ml of 0.5N NaOH solution at 90° for 3 hr under H atmospheric, and the separated catalyst was employed for the hydrogenation again. The lowering of the catalytic activity could not be observed after 100 repetitions of regeneration. ACCESSION NUMBER: 1975:48142 CAPLUS DOCUMENT NUMBER: TITLE: 82:48142 82:48142
Regeneration of Raney nickel catalyst
Mabuchi, Shunsuku
Toyo Soda Mfg. Co., Ltd.
Jpn. Kokai Tokkyo Koho, 5 pp.
CODEN: JKXXAF TITLE: INVENTOR(S): PATENT ASSIGNEE (S): SOURCE: DOCUMENT TYPE: Patent Japanese FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE JP 49066000 JP 54013877 PRIORITY APPLN. INFO.: 19721031 A2 B4 19740626 JP 1972-108548 19790602 19721031 JP 1972-108548

ANSWER 158 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
For catalytic hydrogenation desulfurizing and demetallizing
hydrocarbon residual oils containing V and M1 without
catalyst regeneration a catalyst is used which fulfills
specific conditions of particle size, pore diam, and pose distribution.
The catalyst comprises an Al203 or Si02-Al203 carrier impregnated with Co
and Mo, Mi, or Ni and Mo. Catalysts that had
characteristics specified in the patent had alonger service life than did
those not having those specified.
SSSION NUMBER: 1974:493863 CAPLUS
NEWS NUMBER: 1918:493863 CAPLUS ACCESSION NUMBER: 81:93863 DOCUMENT NUMBER:

81:93863
Catalytic hydrogenation desulfurization of hydrocarbon residual oils Pronk, Karel M. A.; Sie, Swan Tiong Shell Internationale Research Maatschappij B. V. Ger. Offen., 60 pp. COODEN: GWKENS TITLE:

INVENTOR (S):

PATENT ASSIGNEE(S): SOURCE:

DOCUMENT TYPE: Patent

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. APPLICATION NO. KIND DATE DATE 19740425 DE 1973-2351136 19731011 DE 2351136 DE 2351136 A1 C2 19841122 19**7**50924 GB 1972-47299
CA 1973-177438
BE 1973-1005388
NL 1973-13986
FR 1973-36334
JP 1973-113339 19721013 GB 1407610 CA 1012478 BE 805370 A A1 A2 A A1 A2 B4 19721013 19730726 19730927 19731011 19731011 19770621 19740327 NL 7313986 FR 2202931 JP 49074204 JP 59047715 19740416 19740510 19740717 49074204 59047719 7307909 19841121 19740828 19731011 19731011 19731011 19731011 A A1 A C B 19750417 19751120 19781220 AU 1973-61292 IT 1973-30015 NO 1973-3944 AU IT 7361292 995791 139002 NO NO FI 139002 19780911 FI 1973-3145 19731011 59118 19810227 19810610 PRIORITY APPLN. INFO.: GB 1972-47299 19721013 L28 ANSWER 157 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

A8 A solid catalyst is regenerated by drastically decreasing the reactor
pressure at 1/8 of the total per hr. Thus, when the color of the product
indicated deactivation of the Mi-Mo catalyst during wax production,
the pressure was reduced from 600 to 200 psig, at 100 psig/hr and then
returned to 600 psig. The color d. of the product decreased for 22.1
days, indicating catalyst reactivation.
ACCESSION NUMBER: 1975:33200 CAPLUS
DOCUMENTS MARGES

DOCUMENT NUMBER:

TITLE: INVENTOR(S):

82:33200
Desorbing petroleum hydrogenation catalysts
Kochie, Joseph E.
Foster Wheeler Corp.
U.S., 3 pp.
CODEN: USXXAM
Parent PATENT ASSIGNEE(S): SOURCE:

CODEN: 4
Patent
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO. APPLICATION NO. KIND DATE DATE US 3823085 PRIORITY APPLN. INFO.: 19740709 US 1973-345827 US 1973-345827 A

L28 ANSWER 159 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Aged catalysts were regenerated by successive washings with aqueous solns. . or organic acids and bases. Thus, 1 g. of aged catalyst was added to a solution of

tion of
1 g. lactic acid in 50 ml. dist. H2O. After stirring for 10 min. at
50°, the mixture was titrated with 0.05 N NaOH to pH. apprx.7. Ice
was added and the mixture cooled to 0°. The pH was adjusted to 7.5
with NaOH. The liquid was decanted, and the catalyst was washed with 500
ml. H2O, then with 500 ml. absolute EtOH. The regenerated catalyst had

activity of 250 compared to 100 for fresh catalyst. Other organic acids such

ACCESSION NUMBER: DOCUMENT NUMBER:

TITLE:

as formic, tartaric, and acetic gave similar results.
SSION NUMBER: 1974:442059 CAPLUS
HENT NUMBER: 91:42058
E: Regenerating an aged Raney nickal catalyst
with carboxylic acid and base treatment
NTOR(S): Pleters, William J. M.; Freel, John; Anderson, Robert INVENTOR (S):

Bernard PATENT ASSIGNEE(S):

W. R. Grace and Co. U.S., 3 pp. CODEN: USXXAM

DOCUMENT TYPE: Patent English

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE US 3796670 CA 973864 PRIORITY APPLN. INFO.: US 1970-82832 CA 1971-125495 US 1970-82832 19740312 19701021 A A1 19750902 19711019 19701021 L28 ANSWER 160 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB An Al-Mo-Ni catalyst used for hydrocracking and hydrofining is regenerated by burning off the coke with an air-steam mixture The

uced gases containing CO2, SO2, and SO3, are cooled and contacted with a circulating solution of 6-8% aqueous Na2CO3. The S oxides are absorbed and the CO2-containing gas is vented to the atmospheric ACCESSION NUMBER: 1974:122994 CAPLUS
TITLE: Gas-air 1974:122994

AUTHOR (S):

80:122994
Gas-air regeneration of catalysts for hydrogenation processes with recirculation of a soda solution Rogov, S. P.: Perezhigina, I. Ya.; Finelonov, V. P.; Kislov, V. D.; Khavkin, V. A.; Kozlov, I. T.; Osipov, L. N.; Tremasov, V. A.; Baulin, N. F.; et al. USSR

CORPORATE SOURCE:

DOCUMENT TYPE:

LANGUAGE:

USSR Neftepererabotka i Neftekhimiya (Moscow, Russian Federation) (1973), (10), 10-14 CODEN: NNNSAF; ISSN: 0233-5727

ANSWER 162 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
The activity and lifetime of the catalyst are improved by treatment with

F-containing gas during regeneration. In hydrocracking processes where

at least a portion of the higher-boiling fraction of the product stream of the 2nd step is recycled, the catalyst bed for the 2nd step is deactivated

tivated by formation of condensed polyarom. compds. in the final products on exposure to 0. The deactivated catalyst, consisting of at least 1 hydrogenating metal on an acidic amorphous support, is freed of hydrocarbons by stripping with H followed by an inert gas. The stripped catalyst is regenerated by treatment with air diluted with N, then, successively or simultaneously, with a H20/H mixture containing 0.1-20

successively or simultaneously, will a market market volume %

H23 at ≥345 until ≥1% S is taken up, and with H
containing 0.02 volume & difluoroethane at 175-600\*.

ACCESSION NUMBER: 79:94446 CAPLUS
DOCUMENT NUMBER: 79:94446
INVENTOR(S): Regeneration of hydrocracking catalysts
SCHULT, Hans Uwe
PATENT ASSIGNEE(S): Schult, Hans Uwe
PATENT ASSIGNEE(S): Schult, Hans Uwe
DOCUMENT TYPE: CODEN: GWXXBX

DOCUMENT TYPE: Patent
LANGUAGE: German

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: German 2

KIND DATE APPLICATION NO. DATE DE 1972-2249892 US 1971-188408 US 1971-188408 DE 2249892 US 3725244 PRIORITY APPLN. INFO.: A1 A 19721011 19711013 19711013 L28 ANSWER 161 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN AB A catalyst containing NiO 5.3, MoO3 12.4% and a trace of CoO on Al2O3

AR A catalyst containing NiO 5.3, MeO3 12.4% and a trace of CoO on Al203
was used

in a hydrofiner for 1 yr for the 275-430° distillate from bitumen.

After shutdown, the catalyst was rinsed with kerosine. In situ treatment
for 8 hr at 427° with either air in flue gas or air in steam
completely removed. The catalysta; the 7% S and 8% C were almost
chan for the one regenerated in a rotary kiln.

ACCESSION NUMBER: 974:61867 CAPLUS
DOCUMENT NUMBER: 80:61867
Constant low oxygen concentration gas regeneration of
refining catalysts
Tae, Harold F.; Seitzer, H.
SUNCE: U.S., 3 pp.
CODEN: USXXXAM
DOCUMENT TYPE: Parent.

DOCUMENT TYPE: Patent English

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE US 3764558 CA 966823 PRIORITY APPLN. INFO.: US 1971-156571 CA 1972-140615 US 1971-156571 A Al 19731009 19710624 19750429

L28 ANSWER 163 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB A Ni catalyst on active Cu (stable to deactivation by S compds.)
from an aqueous solution containing 15% Ni(OAc)2, 5% Co(OAc)2, and 5%
Mo(SO4)2. The mode of catalyst regeneration is given.
The catalyst activity in the low-temperature hydrofining of high-S petroleum spirit decreased slightly after catalyst
regeneration. The S content of gasoline decreased from 20 to
2.7-3 ppm, acid number from 0.43 to 0.29-0.34, and Br number from 0.083
to 0.047 1973:494294 CAPLUS 79:94294
Synthesis of hydrofining catalyst and optimization of the low-temperature hydrofining of gasoline with it Orizarski, I.

AUTHOR(S): CORPORATE SOURCE: SOURCE:

Dalig. 1. Bulg. 1. Bu

DOCUMENT TYPE: LANGUAGE:

Bulgarian

ANSWER 164 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
The maximum conversion of asphaltic charge stocks to distillable,
lower-bolining hydrocarbon products is effected by mixing with a finely
divided metal oxide catalyst selected from the metals of Groups IVB, VB,
VIB, and the iron group of the periodic table. Thus, suitable catalytic
composites are the unsupported oxides of Ti, Zr, Hf, V, Ta, Cr, Mo, W,

Co, and Ni as well as intimate phys. mixts. of 2 or more of these oxides. The total charge to the reaction chamber is heated to 600-600°F. The reaction chamber is maintained at 1500-3500 psig. The residence or contact time within the reaction chamber is generally 30 secs. to 2 mins. The product effluent is initially separated to provide

hyddrogen-rich gaseous phase, ultimately to be recycled to the reaction chamber in admixt. With a hydrocarbon charge stock, and a normally liquid hydrocarbon-metal sulfide slurry. The slurry is passed into a H2S stripping zone. The stripped catalyst-hydrocarbon slurry is filtered. The sludge is regenerated by contact with air, resulting in the production

of SO2 and a metal oxide. The SO2 and H2S are then reacted at an elevated temperature to produce water and elemental S.

ACCESSION NUMBER: 1973:18635 CAPLUS
DOCUMENT NUMBER: 76:18635
ITITLE: 49th Coarbon desulfurization, sulfur recovery, and catalyst regeneration
INVENTOR(S): Watkins, Charles H.
PATENT ASSIGNEE(S): Universal Oil Products Co.

SOURCE: U.S., 4 pp.
CODEN: USXKAW
PATENT DOCUMENT TYPE: Patent
EMBILY ACC. NUM. COUNT: 1
PATENT INFORMATION: 1
PATENT INFORMATION:

elevated

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

APPLICATION NO. PATENT NO. KIND DATE DATE A 19701026 19701026 US 3687629 PRIORITY APPLN. INFO.: 19720829

L28 ANSWER 166 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Reprocessing of a worked out catalyst, containing W and Ni sulfides, included the reduction melting of the charge containing the worked out catalyst with an additive of a Fe-containing alloy. To increase the recovery of the valuable S-free components, calcium-ferrosilicon 40-50 weight %

of the valuable S-free components, calcium-ferrosilicon 40-50 weight % of the Charge was added for the Fe-containing alloy.

ACCESSION NUMBER: 1971:453885 CAPLUS

DOCUMENT NUMBER: 75:53885

TITLE: Reprocessing of a worked out catalyst
INVENTOR(S): Gusarov, V. N.; Pigasov, S. E.; Ryss, M. A.; Zaiko, V.

PATENT ASSIGNEE(S): SOURCE:

r. Chelyabinsk Electrometallurgical Combine U.S.S.R. From: Otkrytiya, Izobret., Prom. Obraztsy, Tovarnye Znaki 1971, 48(9), 33. CODEN: URXKAF

DOCUMENT TYPE:

Patent Russian 1

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. DATE APPLICATION NO. DATE KIND su .... SU 296583 19710302

L28 ANSWER 165 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The catalyst was solvent washed to speed up decoking and to insure an even temperature increase in the reactor. When the temperature attained 250-300°, 300', steam inlet was stopped and the system was cooled by air. The regeneration took .apprx.70-8 hr. The mech. strength and the structure of
the catalyst remained unchanged.
ACCESSION NUMBER: 1973:18516 CAPLUS
DOCUMENT NUMBER: 78:18516
TITLE: Experimental vapor-air regeneration of aluminum-cobalt-molybdenum and aluminum-nickel
-molybdenum oxide catalysis on industrial hydrofining -molybdenum oxide catalysis on industrial hydrofining apparatus grants. W. M.; Kushner, B. E.; Finelonov, V. P.; Agafonov, A. V.; Mynova, Z. A.; Ermilov, G. T.; Kislov, V. D.; Baulin, N. F. Vses. Nauchnon-Issled. Inst. Neft. Prom., Moscow, USSR Neftepererabotka i Neftekhimiya (Moscow, Russian Federation) (1972), (8), 1-3 CODEN: NNNSAF; ISSN: 0233-5727 Journal Russian AUTHOR(S):

CORPORATE SOURCE:

DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 167 OF 258	CAPLUS	COPYRIGHT 2	004 ACS on STN	
			oln. in H2504 and (or) 1	HCl.
			wder, and fusion of the	
precipitated	•	•		
			eated with NaOH to give	
			catalysts. Thus, 450	
			solved in 3 1. H2SO4.	
	H 1 was	obtained wi	th concentrated H2SO4.	Al powder
400 TV	100		with vigorous stirring	- :- 15 -:-
			The solution, quickly	
			3 times with 300 ml be	
and		orda, madrice		, , , , , , , , , , , , , , , , , , ,
dried in a nonoxidi	zing at	mospheric to	give a powder contain:	ing 99.3 g Ni
and 28.2 g Al.	_	-		
ACCESSION NUMBER:		16485 CAPLU	IS	
DOCUMENT NUMBER:	74:116			
TITLE:	Regeneration of Raney nickel catalysts			
INVENTOR (S):	Gandon, Louis; Bozec, Christian; Laville, Jean M.; Solar, Serge			
PATENT ASSIGNEE (S):			ral Itd	
SOURCE:	International Nickel Ltd. Ger. Offen., 12 pp.			
SOURCE:		GWXXBX	··	
DOCUMENT TYPE:	Patent			
LANGUAGE:	German			
FAMILY ACC. NUM. COUNT:	1			
PATENT INFORMATION:				
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
FAILHI NO.	A1110	DATE	AFFBICATION NO.	
DE 2038753	A	19710304	DE 1970-2038753	19700804
FR 2055967	A5	19710514	FR 1969-27705	19690812
GB 1313574		19730411		19700730
AT 301496		19720911		19700807
CH 517523	A	19720115	CH 1970-517523	19700811
NL 7011894	A	19710216	NL 1970-11894 JP 1970-70136	19700812
JP 49007320 PRIORITY APPLN. INFO.:	B4	19740219	FR 1969-27705	19700812 19690812
FRIGRIII MPPDN. INIU.:			EN 1303-21103	13030017

L28 ANSWER 168 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB A previous study of the chemisorption of H2S on a Ni catalyst
has been extended. Steam has no influence on the chemisorption
equilibrium
Regeneration by means of desorption of H2S may be slow process.
Principles for the regeneration of Ni catalysts by means of
oxidation and reduction have been studied. The expts, have included
unpromoted and promoted catalysts and indicate great differences in their
behavior.
ACCESSION NUMBER: 1971:116373 CAPLUS
TITLE: Principles relating to the regeneration of
sulfur-poisoned mickel catalyses.

1971:116373 CAPLUS
74:116373
Principles relating to the regeneration of
sulfur-poisoned mickel catalyst
Rostrup-Nielsen, Jens R.
Haldor Topsoe Res. Lab., Vedbaek, Den.
Journal of Catalysis (1971), 21(2), 171-8
CODEN: JCTLA5; ISSN: 0021-9517
Journal Journal AUTHOR(S): CORPORATE SOURCE: SOURCE:

DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 169 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB NL/SiO2 or Co/SiO2 catalysts, used in the hydrogenation of aromatic or olefinic hydrocarbons and deactivated especially by their

of aromatic or olefinic hydrocarbons and deactivated especially by to simpurities, were regenerated in the solid bed reactor by heating at 300-600 under o or air, cooling, treating with liquid H2O at increased temps. and subsequently with H at 200-500°. Thus, a 20% NM/SiO2 catalyst of 200 m2/g sp. surface and 0.7 m1/g total pore volume, which was deactivated in a C6H6 hydrogenation, was heated 2 hr at 500° and 250 l. air/l. catalyst hr at 2 aim, cooled to 80° under air treated 8 hr with 2 l. H2O/l. catalyst hr at 80°, dried at 200°, and treated 12 hr with 200 l. H/l. catalyst hr at 400° and 5 atm to give a regenerated catalyst, over which a charge may hydrogenated at 100° with 59% conversion.

ACCESSION NUMBER: 1971:35152 CAPLUS
DOCUMENT NUMBER: 1971:35152 CAPLUS
TA:35152
INVENTOR(S): Regeneration of hydrogenation catalysts
Cosyms, Jean; Duhaut, Pierre
Institut Francais du Petrole, des Carburants et Lubrifiants

1971:35152 CAPLUS
74:35152
Regeneration of hydrogenation catalysts
Cosyns, Jean; Duhaut, Pierre
Institut Francais du Petrole, des Carburants et
Lubrifiants
Ger Offen 24 np.

SOURCE:

Ger. Offen., 24 pp. CODEN: GWXXBX Patent DOCUMENT TYPE:

FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

DATE APPLICATION NO. DATE PATENT NO. KIND DE 1970-2018244 FR 1969-12376 DE 2018244
PRIORITY APPLN. INFO.: 19700416 19690418 A 19701112

ANSWER 170 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB A Co-Ni hydrogenation catalyst was regenerated with
steam (containing air equivalent to 0.5 mole % 0) at 1 lb steam/hr-lb

catalyst at
apprx.100 psig and 600°F. After coke burnoff was initiated, the
temperature of the flame front was maintained at 750°F. Coke burnoff was
completed by bringing the flame front temperature to .apprx.800°F. After
coke burnoff, the steam flow was terminated and air flow continued to

COORDITION, the steam flow was terminated and all flow Continued to color the catalyst to .apprx.350°F. The regenerated catalyst might be sulfided. The catalyst was stored in the regeneration vessel.

ACCESSION NUMBER: 1971:24081 CAPPUS

TITLE: 1971:24081 CAPPUS

TITLE: 1971:24081 CAPPUS

TITLE: 1971:24081 CAPPUS

NUMENTOR(S): PATENT ASSIGNEE(S): Cities Service Research and Development Co. U.S., 5 pp.

DOCUMENT TYPE: Patent

DOCUMENT TYPE: Patent

DOCUMENT TYPE: English

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE US 1967-694561 US 1967-694561 19671229 19701117 US 3541002 PRIORITY APPLN. INFO.: А

L28 ANSWER 171 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB A Co-Ni-Mo-Al203-Sio2 hydrorefining catalyst was contaminated by
apprx.6.1% C and 6.1% S in a com. operation. The deactivated catalyst
was regenerated in 3 stages with 1:99 (volume%) O-N at 650°F for 3
hr; with 1:99 (volume%) O-N at 825°F for 1.5 hr; and with 4:96
(volume%) O-N at 1075°F for 4 hr. The regenerated catalyst had a
relative activity coefficient of 266, with no C deposition and 0.12% S.

relative activity coeffs. of the deactivated catalyst and the freshly prepared catalyst were 100 and 285, resp. Prior to regeneration, the

charge
heater was isolated from the conversion zone. The scale deposits of metallic disulfides were removed from the heater to prevent sulfate formation on the catalyst.

ACCESSION NUMBER: 1970:532728 CAPLUS
DOCUMENT NUMBER: 73:132728
TITLE: Regeneration of sulfur-contaminated carbonized catalyst subsequent to sulfide scale removal from heater

INVENTOR(S): PATENT ASSIGNEE(S):

Weinert, Peter C. Universal Oil Products Co. U.S., 10 pp. CODEN: USXXAM SOURCE:

DOCUMENT TYPE:

Patent English 1 LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. APPLICATION NO. KIND DATE DATE US 1967-697248 US 1967-697248 19671006 19671006 US 3533960 PRIORITY APPLN. INFO.: 19701013 А

ANSWER 172 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
The Cr203 content of catalysts, normally lost on regeneration, is
maintained by spraying a dilute solution of M2Cr04 into the superheated
steam-air mixture used for regeneration. The preferred amount added is
0.1-0.15 lbs. (as Cr03)/1000 ft3 of catalyst. The M2Cr04 forms finely
divided Cr03 which is deposited on the catalyst and converted to Cr203
either by reaction with the carbonaceous deposits formed on the catalyst
during tegeneration or by reaction with H present during the
dehydrogenation cycle. The method has the added advantage of speeding

regeneration process.

ACCESSION NUMBER:

DOCUMENT NUMBER:

5. 1970:502483 CAPLUS 73:102483 Regenerating of chromium oxide-containing dehydrogenation catalysts Frevel, Ludo K. Dow Chemical Co. TITLE:

INVENTOR(S):

PATENT ASSIGNEE (S): SOURCE: U.S., 2 pp. CODEN: USXXAM

English

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

DOCUMENT TYPE:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
us 3524821	А	19700818	US 1967-633711	19670426
BE 750819	A	19701123	BE 1970-750819	19700522
PRIORITY APPLN. INFO.:			US 1967-633711	19670426

L28 ANSWER 174 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN AB The title process included desorption of volatile hydrocarbons,

ustion

of coke and S, stripping the catalyst pellets to remove the poisons (Fe,
Y, As) accumulated on the external layer, grinding, mixing the powder with

with

graphite, tabletting, and sulfiding the tablets. The oxidative
regeneration started in hot N containing 1-2% 0 and hot air was gradually
added, avoiding heating over 500°. The process was complete when
the outlet gas contained no CO2 or SO2.
ACCESSION NUMBER: 1970:437114 CAPLUS
DOCUMENT NUMBER: 73:37114
TITLE: Regeneration of the aluminonickel-tungsten catalyst
8376 on experimental-industrial apparatus
AUTHOR(S): Valitov, N. Kh.; Panchenkov, G. M.; Zakharov, M. S.;
Poteryakhin, V. A.; Tanatarov, M. A.
CORPORATE SOURCE: USSR
SOURCE: Neftenerabotka i Neftekhimiya (Moscow, Russian

USSR Neftepererabotka i Neftekhimiya (Moscow, Russian Federation) (1970), (5), 12-14 CODEN: NNNSAF; ISSN: 0233-5727 Journal Russian SOURCE:

DOCUMENT TYPE: LANGUAGE:

ANSWER 173 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Spent hydrogenation catalysts were regenerated by treating them with reducing gases. Mi, Co, and Fe were recovered by treating them the catalysts with H at 300-600° and 1-20 atm; then with Co at 50-200°, 1-100 atm, and volume rate of CO supply 100-1000/hr-1.

ACCESSION NUMBER: 1970:502480 CAPLUS

DOCUMENT NUMBER: 73:102480

Regeneration of worked out catalysts

Berg, G. A.: Danilova, R. A.: Dubinina, G. G.

U.S.S.R. From: Othrytiya, Izobret., Prom. Obraztsy, Tovarnye Znaki 1970, 47(19), 23.

CODEN: URKMAF

DOCUMENT TYPE: Patent

LANGUAGE: Response Regeneration Response Regeneration of Worked Out Catalysts

Regenerat

DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

SU 272281

PATENT NO. APPLICATION NO. DATE KIND

19700603

su

19690107

L28 ANSWER 175 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Supported metal oxide catalysts (Mo, Co, Ni, or W oxides), which were contaminated with C-, V- and Ni-containing deposits during the hydrogenation of crude hydrocarbon fractions, were regenerated in a 2-step process. First, the C-containing impurities were burned with containing O and then the metallic impurities were removed by washing containing O and then the mediant companies with aqueous oxalic acid. Thus, a 12:3:85 MoO3-COO-Al2O3 catalyst of 240 m2/g sp. surface and 0.53 cm3/g pore volume was contaminated in the desulfurization and demetallation of Kuwait residue to give a catalyst containing 6% V and demetallation of Kuwait residue to give a catalyst containing 6% V and

0.9% Ni and having 80 m2/g sp. surface and 0.33 cm3/g pore volume
This catalyst was washed with gas oil, heated at 550° in N containing
<ili>0, and washed 1 hr at 80° with 90 g/l. oxalic acid to give
complete regeneration.

ACCESSION NUMBER: 1970:427311 CAPLUS
DOCUMENT NUMBER: 73:27311 Regeneration of supported group VI or VIII catalysts
for the hydrogenation of hydrocarbons
INVENTOR(S): Page, Jean F. U.; Baumgartner, Pierre; Duhaut, Pierre
FATENT ASSIGNEE(S): Institut Francais du Petrole, des Carburants et
Lubrifiants
SOURCE: Ger. Offen., 11 pp.
CODEN: GMXXBX
DOCUMENT TYPE: Patent
FAMILY ACC, NUM, COUNT: 1 LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

APPLICATION NO.

DATE

DATE

KIND

PATENT NO.

ANSWER 176 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
Regeneration of a (Nis-WS2)/Al203 hydrogenation
catalyst by burning off the coke and excess 8, removing surface layers of
Fe, V, and As compds. in a ball mill, grinding the pellets to 315 μ
particle size, mixing the resulting powder with graphite, pelletizing the
mixture, and sulfidizing it to 7% S content restored almost all of the
original activity. By hydrogenating a straight distillation
153-253\* kerosine fraction introduced at 140 ml/hr with 70 1. H/hr
on 94 g of fresh and regenerated catalysts, resp., at 300 t 10 atm and
360 t 1.5\*, the total S content was reduced from 0.63 (20 t 1.5\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.63 (20 t 1.8\* the total S content was reduced from 0.65 (20 t 1.8\* the total S content was reduced from 0.65 (20 t 1.8\* the total S content was reduced from 0.65 (20 t 1.8\* the total S content was reduced from 0.65 (20 t 1.8\* the total S content was reduced from 0.65 (20 t 1.8\* the total S content was reduced from 0.65 (20 t 1.8\* the total S content was reduced from 0.65 (20 t 1.8\* the total S content was reduced from 0.65 (20 t 1.8\* the total S content was reduced from 0.65 (20 t 1.8\* the total S content was reduced from 0.65 (20 t 1.8\* the total S content was reduced from 0.65 (20 t 1.8\* the total S conten 73:5674
Regeneration of alumina-nickel-tungsten
sulfide hydrogenation catalysts
Valitov, N. Kh.; Panchenkov, G. M.; Balandina, K. L.
MINKKGP Im. Gubkina, Moacow, USSR
Neftepererabotka i Neftekhimiya (Moscow, Russian
Federation) (1970), (3), 8-9
CODEM: NNNSAF; ISSN: 0233-5727
JOURNAI
Russian DOCUMENT NUMBER: TITLE: AUTHOR(S): CORPORATE SOURCE: SOURCE:

DOCUMENT TYPE:

L28 ANSWER 178 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The catalysts were regenerated by treatment with acetylacetonatonickel, acetyl-acetonatomolybdenum (I), or SnCl4 in Me2S and reduction with 7.5-10 volume 8 H2S-H for 40-50 hr at 482°, 500°, or 982° and 49.5, 65.2, or 126 atmospheric Thus, a SiO2-Al2O3-supported catalyst containing Ni 4.7, Ni 3.2, and F 3.2% and deactivated by the deposition of 5% C as coke and 0.13% N was impregnated with a 1.3% Mo-containing I solution in Me2S and reduced with 10 volume % H2S-H mixture for 40 hr at 982°, 65 atm, and space velocity 950 vols./volume/hr. After hydrocracking of gas oil for 80 hr at 105 atm, 293°, and space velocity 0.67 volume/volume/hr and a 67% conversion to gasoline, the C and N velocity 0.67 volume/volume/hr and a 6/% conversion or grand N
deposition on the catalyst were 0.7% and 0.01%, resp.
ACCESSION NUMBER: 1970:34057 CAPLUS
DOCUMENT NUMBER: 72:34057
TITLE: Regeneration of nickel molybdenum
hydrocracking catalysts
INVENTOR(S): Schutt, Hans U.
PATENT ASSIGNEE(S): Shell Internationale Research Maatschappij N. V.
SOUNCE: Ger. Offen., 20 pp.
CODEN: GWXXBX
DOCUMENT TYPE: Patent
LANGUAGE: Patent
German
FAMILY ACC. NUM. COUNT: 1
DARPHYT INFORMATION: DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. DATE APPLICATION NO. DATE DE 1921693 US 3558514 NL 6906503 FR 2007189 PRIORITY APPLN. INFO.: DE 1969-1921693 US 1968-725194 NL 1969-6503 FR 1969-13402 US 1968-725194 19691120 19710126 19691031 19700102 19690428 19680429 19690428 19690428

L28 ANSWER 17 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB A reforming catalyst consisting of Ni, Co, or Fe, activated by
K, and containing oxides of metals, such as Al, Mg, Th, Zr, or Si, is
poisoned oned
by S present in the material being reformed after a period of use. This
polsoned catalyst is regenerated in the reforming apparatus by firstpassing through the catalyst an oxidizing mixture of steam and (or) O or through the catalyst an UNIVERSE METERS WITH THE CONTROL OF SUPERAM PROPERTY OF SUPERA times the necessary stoichiometric amount; in the reducing stage, an H2O-H2 ratio of 30-400:1, preferably 50-300:1, was used; but if the catalyst contained N1, the H2O-H2 mole ratio was 20-5:1, preferably 10:1. Thus, 62 g of a poisoned catalyst containing N1 activated with K was subjected at 30 bars first to an oxidative treatment during 2 hr with SUBjected at 30 bars first to an extendive treatment during 2 hr wish moles 0 steam/hr at 700° and then to a reducing treatment during 6 hr at 800° with 55.5 moles H2O/hr containing H2 in a H2O:H2 ratio c 250:1. An extent of regeneration of 90-58 was obtained.

ACCESSION NUMBER: 1970:81178 CAPLUS

DOCUMENT NUMBER: 72:81178

Reforming process catalyst regeneration

INVENTOR(S): Topsee, Haldor F. A.

FOURCE: Pr., 13 pp.

CODEN: FRXYAK

DOCUMENT TYPE: Patent

LANGUAGE: Patent

FRANILY ACC. NUM. COUNT: 1

PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. FR 1568433 PRIORITY APPLN. INFO.: 19690523 FR 1968-148912 DK 1967-2137 19680422 A 19670420

ANSWER 179 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB An effective stationary N1 catalyst, in various modifications, was developed for hydrogenation of olls. Powder presintering techniques were used to impart porosity to the catalyst. Catalyst regeneration and activation methods were developed and the catalyst was thoroughly tested during hydrogenation of oils.

ACCESSION NUMBER: 1969:89356 CAPLUS
TOCUMENT NUMBER: 70:89356
TITLE: New type catalysts and principles of their production Dubrovskii, A. P.: Bakarinova, V. I.; Shmidt, A. A.; Geishina, K. V.; Raemskaya, N. P.

CORPORATE SOURCE: Porosh. Met. Novoi Tekh. (1968) 41-52
From: Ref. Zh., Met., 1968, Abstr. No. 8G356

DOCUMENT TYPE: Journal Russian

L28 ANSWER 180 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The oxidation rates of N1, Cu, and N1-Cu powders obtained
in various ways were studied and compared with the oxidation rates of
fat-free metal catalysts. The roles of the formation of an oxide film

of O diffusion through the film in the catalyst
regeneration were discussed. The combustion of residual organic
compds. in fat-free catalysts allows the decrease of the heating
temperature

rature down to 250°. The effects of heating at high temperature and of fat removal by Me2CO on the solubility of the catalysts in H2SO4 were

studied. ACCESSION NUMBER: DOCUMENT NUMBER:

1969:23257 CAPLUS

70:23257

TITLE: AUTHOR (S): CORPORATE SOURCE:

Air-oxidation tests on fat-free catalyst metals Mishkareva, L. V.; Sterlin, B. Ya. USSR

USSR Trudy, Vsesoyuznyi Nauchno-Issledovatel'skii Institut Zhirov (1967), No. 26, 322-9 CODEN: TVZHAS; ISSN: 0372-3259 SOURCE:

DOCUMENT TYPE:

LANGUAGE:

L28 ANSWER 181 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Fat removal by Me2CO from metal catalysts was studied in the ratio
solvent: catalyst of 10 to 1 and compared to removal by C6H6. A single
and short duration extraction allows the removal of 95% of the fat. The
catalyst-Me2CO suspension sep. in 2 layers, a 20-5% fatty one and another
containing the catalysts; the 2nd fraction, by washing with Me2CO, gives the

fat-free catalyst and an Me2CO solution which may be used to extract a new

catalyst fraction. ACCESSION NUMBER:

1969:23234 CAPLUS DOCUMENT NUMBER:

TITLE:

750:23234 Change 70:23234 Removal of fat from the spent catalyst as the preliminary step in metal catalyst regeneration Mishkareva, L. V.; Sterlin, B. Ya.; Charikova, M. S.

AUTHOR (S):

CORPORATE SOURCE: USSR USSR Trudy, Vsesoyuznyi Nauchno-Issledovatel'skii Institut Ehirov (1967), No. 26, 317-21 CODEN: TVZHAS; ISSN: 0372-3259 Journal SOURCE:

DOCUMENT TYPE:

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L28 ANSWER 182 OF 258 CAPLUS COPYRIGHT 2004 ACS ON STN
AB Continuous hydrogenation of cottonseed oil on a stationary
catalyst under H pressure to 10 atmospheric was studied in a laboratory
                 or and on an enlarged industrial apparatus A granulated, alloyed Ni catalyst with an addition agent was used as the catalyst; this catalyst was regenerated directly in the reaction apparatus by leaching with a
                ion or NaOH
with subsequent drying in a H stream at 110°.
Hydrogenation was conducted at 80-100, 120, 140, and 160°
in alc. or n-hexane or in the absence of a solvent. The optimum
  temperature,
               prature,
pressure, and spatial hydrogenation velocity conditions were
determined which result in the production of hydrogenated fat for
food purposes, toilet or household soap, and thoroughly
hydrogenated fat. Catalyst regeneration is
discussed in relation to operation in the laboratory apparatus and in
 the enlarged apparatus Ni consumption was 0.02% for 50 days of operation in the enlarged apparatus. The catalyst is suitable for com. hydrogenation of cottonseed oil with or without a solvent. The advantages of conducting
apparatus

apparatus

is given along with detailed characteristics of the resulting hydrogenated fats.

ACCESSION NUMBER: 1969:18942 CAPLUS DOCUMENT NUMBER: 70:18942

TITLE:
                the process in a flow-type system are listed. A line diagram of the
                                                                      70:18942
Pydrogenation of cottonseed oil over a stationary catalyst in a flow-type apparatus Shlyakhov, V. 1; Sokol'skii, D. V.; Golodov, F. G.; Orlov, V. K.
USSR
Gidrirovanie Zhirov, Sakharov Furfurola (1967) 117-24
From: Ref. Zh., Khim. 1968, Abstr. No. 10R435
Journal
Russian
  AUTHOR (S):
  CORPORATE SOURCE:
  DOCUMENT TYPE:
LANGUAGE:
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ANSWER 183 OF 258 CAPLUS COPYRIGHT 2004 ACS on STM
The title process is applicable to the catalytic hydrorefining of
naphthenes containing feedstocks, such as crude oil and atmospheric and

The title process is applicable to the tatalytic nyuntaning or naphthenes containing feedstocks, such as crude oil and atmospheric and um tower bottoms. It consists of collecting from the hydrorefining unit under normal operation a first cycle oil fraction b. >650°F, and switching from the normal feed to this first cycle oil during the regeneration phase. A second cycle oil b. >650°F, is thus obtained which is later joined with the initial feedstock in the next normal phase of operation. This first cycle oil effects the dissolm of the asphaltenic material from the catalyst. It is capable of transferring H from itself to the naphthenes so as to convert them to pentane-soluble products. The operating temperature in the normal hydrorefining phase is 425-500°, and during the regeneration stage 300-425°. All other conditions remain the same during both phases. For instance, a topped Wyoming sour crude was hydrorefined. Its gravity was 19.5° A.P.I. at 60°F, and contained 3 weight % S. 2900 ppm. N, 105 ppm. combined NM and V, and 8.5 weight % of a pentaneinsol. asphaltenic fraction. The catalyst was a composite of 2 weight % NA and 16% Mo on a carrier of Al203 68, \$102 10, and B phosphate 22% by weight The let.

crude was introduced at a rate of 200 g./hr. into a reactor containing 100 g.

catalyst in the form of spheres of 20-150µ. The inlet temperature was 790°F., the H:oil ratio 50,000 standard ft.3/bbl., and the pressure was 3000 psig. After 36 hrs., a total of 7200 g. oil was processed. Inspection of the catalyst indicated 1.03% of the oil to be deposited on the catalyst as C, while the catalyst was bonded together in one mass. The liquid product from this period had a gravity of 33.1° A.P.I. at 60°F. In a second test, the same crude was processed in the same conditions for 18 hrs. to yield a total liquid product of 32.9° gravity, of which the heavy cycle oil fraction had a 21.7° gravity. Then the feed was switched from crude to this heavy cycle oil for the

next
9 hrs., then back to crude for 18 hrs. and finally to cycle oil again for another 9 hrs. A total of 7200 g. crude was thus processed as well as 3600 g. of the cycle oil. Inspection at that moment showed that 91% of the catalyst passed 60-mesh and that 0.55% of the total hydrocarbons treated remained on the catalyst as C. The total liquid product from the crude had a gravity of 32.9°, while that from the cycle oil was 27.5°.

ACCESSION NUMBER: 1968:445101 CAPLUS
DOCUMENT NUMBER: 69:45101
TITLE: BENGERIAL OF ASSAULTED AND CATALUST.

TITLE: INVENTOR(S): PATENT ASSIGNEE(S): SOURCE:

69:45101
Regeneration of asphaltene hydrorefining catalyst Gleim, William K. T.; O'Hara, Mark J. Universal Oil Products Co.

U.S., 4 pp. CODEN: USXXAM

DOCUMENT TYPE: Patent English

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE US 3389077 PRIORITY APPLN. INFO.: 19680618 US 1965-511975 US 1965-511975 19651206 19651206

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L28 ANSWER 184 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The title catalyst is used for the hydrogenation of PhOH to
cyclohexanol (I) and is regenerated by washing with an organic solvent.
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To improve the quality of I and to increase the life-time of the catalyst rectified cyclohexanone is used as the solvent.

ACCESSION NUMBER: 1968-4816594 CAPLUS

DOCUMENT NUMBER:

TITLE: INVENTOR(S):

--- CAPLUS
69:18694
Regeneration of mickel-chromium catalyst
Kervalishvili, Z. Ya.; Gvaliya, N. K.
U.S.S.R. From: Izobret., Prom. Obraztsy, Tovarnye
Znaki 1967, 44(16), 16.
CODEN: URXXAF
Patent SOURCE:

DOCUMENT TYPE: LANGUAGE: Russian

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE 19670713 SU 199838 su 19651022

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L28 ANSWER 186 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB A catalyst containing 0.5-25 weight % Ni deposited on an acidic,
refractory oxide catalyst (e.g. sio2 + Al203, MgO, Tio2, or Zr203), which
had been deactivated by the deposition of N compds., was regenerated with
a gaseous mixture comprising H and 5-20 volume % HZS at 427-593 and
35-175 atmospheric The method is suitable for processes where the C
 deposition
 deposition
is not too high, such as hydrofining and hydrocracking. Since the
depositions are only partially removed, an air regeneration may be
necessary if the C content of the catalyst becomes too high.
ACCESSION NUMBER: 1968:70995 CAPLUS
DOCUMENT NUMBER: TITLE:
                                                                      68:70995
                                                                   68:7095)
Nickal catalyst
regemeration
Shell Internationale Research Maatschappij N. V.
Neth. Appl., 9 pp.
CODEN: NAXXAN
 PATENT ASSIGNEE(S):
 DOCUMENT TYPE:
                                                                    Dutch
 PATENT INFORMATION:
              PATENT NO.
                                                                    KIND
                                                                                      DATE
                                                                                                                       APPLICATION NO.
                                                                                                                                                                                     DATE
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19671113

19690000

GB US US

19660511

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L28 ANSWER 187 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

Feed-stocks with an initial b.p. ≥204* (of which 50% b.
≥260*) and a final b.p. ≥316* can he
hydrocracked at <421*, preferably 316-400*, and 35-140 atmospheric
over a crystalline rare earth aluminosilicate catalyst containing Ni-W
sulfides, sulfurized Co-Mo oxides, or Pt. The catalyst is regenerated by
heating in an O-free H atmospheric to 204-650*, the regeneration
                      erature being always 22-252° above the operating temperature If the hydrocracking temperature is >421°, the C deposits on the catalyst cannot be removed in this way. Thus, a 13X Na aluminosilicate mol. sieve was exchanged with an aqueous solution of rare earth chlorides containing, wide.
                      xides,
La203 24, Ce203 48, Pr6011 5, Nd203 17, Sm203 3, Gd203 2, Y203 0.2, and
unidentified rare earth oxides 0.8% by weight After washing and drying
                      104\,^{\circ} for 20 hrs., the Na-free product was ground to 0.71-1.41 mm. diameter and calcined 10 hrs. at 538 ^{\circ} to give I containing 27.4% by
 weight
rare earth oxides, after which 111.4 g. of I was impregnated with 66 cc.
of an aqueous solution of NH4 tungstates containing 0.158 g. of W/cc.,
adjusted to pH
6.5 with citric acid, and dried 16 hrs. at 110°, after which the
impregnation was repeated by using 15.3 cc. of the same solution and the
product dried as before and calcined in an atmospheric of N with 2% by
  product dried as before and calcined in an atmospheric of N with 2% by volume of 0
at 538 for 24 hrs. The calcined product was then impregnated with an aqueous solution of Ni(NO3)2 containing 0.04 g. Ni/cc., dried as before, and calcined 3 hrs. at 538 to give a product containing 9.8% wand 3.8% Ni by weight This was then heated to 427 and an aqueous. In the containing 9.8% and an equivol. mixture of H and H2S blown through it at 200 ml./min./100 ml. catalyst to give II, containing 3.8% by weight of S. The preparation of other
                     aration of other catalysts on a base identical with or similar to I is described. A feed gas oil of 0.894 sp. gr., 32° pour point, 86° aniline point, b. 304 to > 507°, containing N 0.06, S 0.57, and 0 0.77% by weight was hydrocracked over II at 140 atmospheric, 0.5 volume/volume/hr., and
                      with 0.534 m.3 of H/l. of feed to give 40% by volume conversion to
 products
b. 77-199*. After 6 days, the catalyst was regenerated in the
conventional manner at 399* by using a mixture of 590 l. N/100 ml.
catalyst/min. and 12 ml. O/100 ml./min., giving a combustion-zone
  temperature of

427-82°, followed by air at 538°, after which the catalyst
was sulfurized and returned to service. Regeneration had to be repeated
approx. every 3 days, and after the fifth regeneration the operating
approx. every 3 days, and after the fifth regeneration the operature
was 395-400° for the same total liquid product yield, with a rise
in gas production from 2.0 to 2.6% by weight of the feed and a rwdn
to 30.3% in the product b. 77-199°.

ACCESSION NUMBER: 1967:404656 CAPLUS
DOCUMENT NUMBER: 67:4656
TITLE: Low-temperature hydrocracking process and non-oxidative catalyst regeneration
Mobil 011 Corp.
SOURCE: Fr. 17 pp.
COODEN: FRXXAK
DOCUMENT TYPE: Patent
LANGUAGE: Prench
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L28 ANSWER 185 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB In the preparation of lean gas and town gas by naphtha reforming, the r  $\nu$  with Ni on  $\alpha$ -Al203 improved the life of the catalyst and the S resistance, and maintained the reaction equilibrium during the

The U and Ni form a complex Ni uranate, and many physicochem. data are presented. Less K is needed to make a neutral catalyst, and the normal use of y-Al203 is eliminated. The nature of the coke deposit is described in detail, and the U reduces the crystallite size of the Ni. Analyses of the nitrate or acetate forms of the Ni and U used for impregnation show that 400° is sufficient for decomposition but that 580° is required to form the oxides. The Lewis acid content was measured by a nonaq, butylamine titration by using a series of Hammett indicators. No mated

acid content could be found, but the Lewis acidity was checked independently by adsorption of various bases. This acidity was neutralized by addition of only 0.15-0.22% K to the Al203. A modified

surface area was determined after each impregnation of the Al2O3. The U Ni data showed greater area than that of the individual metals.
Reduction of the catalyst proceeds 4 times as rapidly with CO as with
H. NiUO4 is reduced to NiO and U4O9, and the latter to UO2. About 4500
hrs. of pilot-plant operation (lean gas) are described and compared with

2nd operation in which town gas was made for 815 hrs. The carryover of K was nil, and KOAc was found to be the best source of K. Equilibrium obtained for the reaction 2CO .dblarw. CO2 + C in a CO atmospheric and of CO and CO2 at approx. equilibrium composition for 3 catalysts at 350, 400, 500, 600, 700, and 800°, also calculated values on the basis of C = graphite. For C4H10 reforming, the addition of U converts the kinetics

from d(C4H10)/dt = k(H2O)\*(C4H10)1 to d(C4H10)/dt = k(C4H10)0.6 (R2O)0.6. Values of activation energy ranging from 7.5 kcal./g. mole for 9.0% U to 27.5 for 0% U, for catalysts containing 5.5% Ni, were obtained by D.T.A. by using 2 independent methods. C oxidation was the reaction studied, and a lower catalyst regeneration temperature resulted from this work.

ACCESSION NUMBER: 1568:116150 CAPLUS

DOCUMENT NUMBER: 68:116150

Influence of uranium on nickel

Influence of uranium on nickel

CODEN: IGEJA9; ISSN: 0020-3432

Journal English

Antitude of Training of Training of Training of Training (Archive) (Archive)

DOCUMENT NUMBER: TITLE:

DOCUMENT TYPE: LANGUAGE:

AUTHOR (S):

NL 6706444

US 3459675 PRIORITY APPLN. INFO.:

L28 ANSWER 187 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN FAMILY ACC. NUM. COUNT: 1 (Continued) FAMILY ACC. NUM. CO PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE FR 1457382 19661104 FR 19650713

L28 ANSWER 188 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The process consists of adding 1-10% (based on the weight of the metal) organometallic compound, such as Mo(CO)6, to a heavy hydrocarbon charge stock for removal of N and S compds. and to convert the C5H12-insol. asphaltenes into useful C5H12-soluble hydrocarbons. Thus, 450 g. Wyoming sour crude oil (gravity 23.2° APT at 60°F. and containing approx. N 2700, Ni 18, and V 81 ppm., 2.81 S, and 8.391 cpm. C5H12-insol. asphaltenes) was slowly mixed with 23.2 g. Mo(Co)6 and 42 g. vanadyl acetylacetonate in AmoN, heated at 200° to distil the alc., placed under 100 atmospheric H and heated for 6 hrs. at 400° (H pressure 200 atmospheric at 400°) to give a liquid product (after separation from sludge) with API gravity 35.5° at 60°F. and containing N <10, Ni <0.1, and V <0.1 ppm., 0.02% S, and 0.15% CSH12-insol. asphaltenes. The sludge .apprx.(60 g.) containing the organometallic catalyst
was mixed for 0.5 hr. with 60 cc. C6H6 and centrifuged. The sludge (50 g.) was separated and mixed with 500 ml. of a solution of 10 g. S2Cl2 for 1 hr.
at 300° to convert the asphaltic material to coke and liquid hydrocarbons. About 5 g. of solid was separated and the supernatant liquid. liquid along with the C6H6 from the preceding separation, was added to the fresh cru oruse
oriuse
ori obtained from the initial hydrorefining. The crude oil mixture was
heated at 400° under H (200 atmospheric) for 8 hrs. and centrifuged to
give a liquid product containing N <10, Ni <0.1, and V <0.1 ppm.,
<0.10% S, and .apprx.0.02% C5H12-insol. asphaltenes. Phosphomolybdic acid acid gave similar results when used as the catalyst.
ACCESSION NUMBER: 1967:87335 CAPLUS
DOCUMENT NUMBER: 66:87335 66:87335
Regenerative hydrorefining of petroleum Gleim, william K. T.
Universal Oil Products Co. TITLE: INVENTOR(S): PATENT ASSIGNEE(S): SOURCE: U.S., 6 pp. CODEN: USXXAM Patent DOCUMENT TYPE: FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE US 3293172 19661220

US

19640429

ANSWER 189 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN A colloidally dispersed hydrorefining catalyst made up of an organometallic compound with metals from Group VI-B greater than 24

organometallic compound with metals from Group VI-B greater than 24 ic number, Group V-B, or the Fe group, is used in treatment of crude oil to remove undesirable metallics, S, and asphaltene compds. Catalyst is regenerated while in a colloidal suspension in the presence of H. Thus, 23.3 g. molybdenum hexacarbonyl mixed with 200 g. Wyoming sour crude and charged to an autoclave was heated to 250° for 3 hrs.; CO was vented and the mixture brought to 400°/200 atmospheric Effluent from the treatment zone was centrifugally separated. The liquid contained <10 ppm. N, 0.02%

weight S, 0.20% by weight pentane insoluble, <0.02 ppm. Ni, and 0.02 ppm. V. The sludge containing the catalyst (27 g.), is mixed with ppm. V. iodoform up

to about 3.5% by weight of iodine in the mixture. The mixture in an autoclave was

plave was heated to 350°C. and pressured with H to 2000 psi. for 4 hrs. and cooled. Sludge resulting from this treatment was mixed with fresh charge after about 0.1% to 1.0% of solid catalyst particles were removed and the new catalyst added in equivalent amount This mixture was heated to 250°C.

decompose the new catalyst, after which the suspension was ready for recycling.

ACCESSION NUMBER: 1967:87331 CAPLUS
DOCUMENT NUMBER: 66:87331
TITLE: Regementative hydrorefining of hydrocarbon oils
INVENTOR(S): Gatsis, John G.
PATENT ASSIGNEE(S): Universal Oil Products Co.

SOURCE: U.S., 4 pp.

U.S., 4 pp. CODEN: USXXAM Patent English DOCUMENT TYPE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE US 3249556 19660503 19630819

1.28 ANSWER 190 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB A process for hydrorefining an asphaltenic hydrocarbon stock containing organometallic compds. is described. The steps are: (a) heating the with H to a temperature below which thermal cracking of asphaltenes (s; (b) passing the mixture through a bed of an adsorptive hydrogenation catalyst at 725-785°F. and at 500-5000 psig.; (c) when the catalyst accumulates unreacted asphaltenes, the flow of stock is stopped, the flow of H is continued, and the temperature is increased 7785°F. to crack the asphaltenes; (d) when the asphaltenes are removed from the catalyst the temperature is decreased to <785°F. and the flow of charge stock is introduced while the H flow is continued; and (e) the reaction effluent separated to provide a fraction free from asphaltenes and organometallic compds. Such compds. contain: Ni, V, or Fe; Cu, Pb, or Zn.
Other contaminants contain N, S, and O. The catalytic composite may comprise ≥1 metals or compds. of metals from the group of V, Nb,
Ta, Mo, W, Cr, Fe, Co, Na, Pt, Pd, Ir, Oa, Rh, Ru, and mixts. therefrom.
ACCESSION NUMBER: 1967:78094 CAPLUS DOCUMENT NUMBER: 66:78094 bb: 8094
Autoregeneration of hydrofining catalysts
O'Hara, Mark J.; Gatsis, John G.
Universal Oil Products Co. TITLE: INVENTOR (S): PATENT ASSIGNEE (S): SOURCE: U.S., 7 pp. CODEN: USXXAM DOCUMENT TYPE: Patent

PATENT NO. KIND DATE APPLICATION NO. DATE US 3288704 19661129 US 19631226

English

LANGUAGE:

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

L28 ANSWER 191 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB a high-activity sulfided nickel-molybdenum-alumina catalyst is
charged to all reactors except the first one in order to increase feed
throughput and to decrease the time between catalyst
regeneration. A feed containing 308 olefins and 298 aromatics is
charged to the first reactor at about 800 psig, and 575 °F.
Mydrogenation of oefins causes the outlet temperature to be increased to
701 °F. The outlet material is heat exchanged with fresh feed to
cool it to 625 °F. before it enters the 2nd reactor containing the
high-activity catalyst. The product leaves the 2nd reactor at
641 °F. and is charged to the 3rd reactor where the temperature increases
to 651 °F. at the outlet. The catalysts deactivate only slowly at
the low erav. temps. (after 4 months, the outlet temperature from the 3rd
reactor was only 665 °F.). The length of the run is more than twice
that of previous runs even with higher throughput and lower product N
specifications. The dew point of the naphtha-H mixture must be
maintained
below the lowest temperature at the hydrorefining arrangement.

below the lowest temperature at the hydrorefining pressure by monitoring

boiling range of the naphtha feed unless a greater ratio of H can be

used.
ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE:
INVENTOR(S): 1967:39620 CAPLUS 1967:39620 CAPLUS 66:39620 Naphtha hydrofining catalyst Blue, Emanuel M. Chevron Research Co. U.S., 6 pp. PATENT ASSIGNEE(S): SOURCE: DOCUMENT TYPE: English LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION;

PATENT NO. APPLICATION NO. KIND DATE

DATE 19661213 US 19640625 US 3291723

Answer 193 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

In the production of xylitol by hydrogenation of xylose on a skeletal Mi catalyst, the latter becomes inactivated and must be discharged from the reactor for regeneration. The existing method for discharging the catalyst is time-consuming and cumbersome. A new method, developed at the Chimkent hydrolysis plant, utilizes suction. The reaction column is connected by means of a system of tubes with a cyclone and a suction pump, which creates a partial vacuum (350-400 mm. Hg) in

system. In the cyclone, the catalyst granules are separated from the air, and

are collected in a receiving vessel, from which they are directed into

the regeneration equipment. The latter is a vibratory mill in which the inactivating film is mech. removed.

ACCESSION NUMBER: 1966:439707 CAPLUS
DOCUMENT NUMBER: 65:39707
ORIGINAL REFERENCE NO.: 65:7448g-h

TITLE:

Experimental production of xylitol Nemanov, E. A. Gidrolizn. i Lesokhim. Prom. (1966), 19(4), 23 AUTHOR (S):

SOURCE:

DOCUMENT TYPE: Journal

LANGUAGE:

L28 ANSWER 192 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

A method for reactivating a hydrogenation catalyst while the reactor is maintained on stream, and without transporting the catalyst to a different vessel for regeneration, is described. A residual fuel is hydrogenated in a reaction zone with H plus a catalyst such as Co, Fe, Me, Mi, W, or Co2[McO4]3 sufficient to provide 1 lb. of catalyst for a rate of up to 2 bbl. of oil-feed per day. After the activity of the catalyst has decreased due to normal usage, there is substituted for the residual oil-feed a catalytic cycle oil-feed together with H under hydrogenating conditions of temperature and pressure. When the catalyst is regenerated the residual oil-feed is again introduced

introduced
and the catalytic oil-feed stopped. The catalyst bed is usually in the form of an expanded or ebullated bed. Cf. CA 55, 23954g.

ACCESSION NUMBER: 1966:498533 CAPLUS

DOCUMENT NUMBER: 65:98533

ORIGINAL REFERENCE NO.: 65:18397c-e

Regeneration of hydrogenation catalysts
Galbreath, Richmond B.
Cities Service Research and Development Co. TITLE: INVENTOR(S):

PATENT ASSIGNEE(S): SOURCE:

3 pp. Patent DOCUMENT TYPE: EARGUAGE: Unavailable FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

APPLICATION NO. PATENT NO. KIND DATE \_ ...CATIO DATE US 3271301 19660906 19640203

L28 ANSWER 194 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The preparation and use of catalysts that can be regenerated after use in
hydrocracking is described. Thus, a catalyst containing N18 on
\$102-Al203 was prepared by impregnating Si02-Al203 particles with a
solution of

Nydrocracking is westived. Thus, a case-3-constant, solution of Side Al203 was prepared by impregnating Side-Al203 particles with a solution of Ni nitrate in a concentration sufficient to provide 6 weight Ni on a dry basis. The Side-Al203 particles contained 90% Side and had a "Cat A" value (Mills, et al., CA 44, 6611f) 40 before being impregnated with the hydrogenating metal component. After impregnation and drying, the catalyst was thermally activated by contact for 2.2 hrs. with a stream of hot air at 1425°F. Then, the catalyst was sulfided and used for hydrocracking in a reactor for several thousand hrs. on a hydrocarbon feed stock having a total N content of Si ppm. After hydrocracking, the catalyst was treated with enough 2.5M HNO3 to fill the pores, oxidized in dry air at 800°F. for 2 hrs. and at 1000°F. for 2 hrs., and thermally activated for 2 hrs. at 1400°F. in dry air. Then, the catalyst was sulfided at 600°F. With H2S and used to hydrocrack n-CloH22. A conversion of 25.0 mole % was obtained, as compared with 29.4 mole % for a freshly prepared catalyst.

ACCESSION NUMBER: 1966:437738 CAPLUS
DOCUMENT NUMBER: 65:37738

DOCUMENT NUMBER: ORIGINAL REFERENCE NO.:

65:6977g-h,6978a
Hydrocracking catalyst regeneration
Constabaris, George: Unverferth, Jack W.
Chevron Research Co. INVENTOR (S):

PATENT ASSIGNEE (S): SOURCE:

6 pp. Patent DOCUMENT TYPE:

LANGUAGE: Unavailable

FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE US 3256205 19660614 US 19630701

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ANSWER 195 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Hydrocarbons of the gas oil to heavy recycle oil boiling range of >316° are catalytically hydrocracked to give a ≥75 weight% yield of a b. 154-371° distillate. The product is obtained in a continuous cycle system by circulating 800-2135 1. H/1. starting hydrocarbon in the presence of a N compound containing 0.005-5.0
weights, but preferably 0.1-1.0 weights N based on the reacting hydrocarbon. The catalyst
                 yst consists of an acid support impregnated with an oxide or sulfide of NI, W, or Mo, or their mixts. and the reaction proceeds at 371-454° and 35-350 kg./cm.2 at a H feed rate of 0.1-10.0 vols./hr. relative to the liquid state. Inclusion of the N compound in the
cracking
                 ting process is essential to minimize the formation of distillates b. <154°, thereby providing ≥75 weight% yield of 154-371° distillates. The acid catalyst supports are various mixts. of SiO2,
2r02
                Th02, Al203, and (or) MgO. The acidity of the support may be diminished by treatment with H2O or enhanced by treatment with halogens or halogen acids. The catalyst is treated with a S compound before or during the hydrocracking process to yield the metal sulfide. For example, a co-gel of 20% SiO2 and 80% Al203 (0.63-2.00 mm.) is impregnated with an aqueous
solution
of NH4 molybdate to give 12.2 weight Mo based on the support. The
product
at 102 calcined overnight at 482,
                is dried for 6 hrs. at 103°, calcined overnight at 482°, then impregnated with an aqueous solution of Mi nitrate to give 4 weights Mi. After drying for 2 hrs. at 103°, the product is calcined for 1 hr. at 593°, exposed to H at atmospheric pressure for several hrs. at 299°, then at 140 kg./cm.2 for 1.5 hrs. Finally, the catalyst is treated with n-hexane containing 5 weights CS2 at 140 cm.2
the catalyst is tre
kg./cm.2
for 4-6 hrs.
ACCESSION NUMBER:
DOCUMENT NUMBER:
ORIGINAL REFERENCE NO.:
                                                                                  1966:437737 CAPLUS
65:37737
65:6977d-g
Hydrocracking heavy hydrocarbons to
medium-boiling-range distillates
Myers, John W.: Lanning, William C.
Phillips Petroleum Co.
12 pp.
Patent.
Unavailable
1
 TITLE:
INVENTOR(S):
PATENT ASSIGNEE(S):
SOURCE:
DOCUMENT TYPE:
LANGUAGE.
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                 PATENT NO.
                                                                                                                                                   APPLICATION NO.
                                                                                    KIND
                                                                                                          DATE
                                                                                                                                                                                                                                DATE
BE 668580
FR 1441320
PRIORITY APPLN. INFO.:
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19660221

19640928

ANSWER 197 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
The resistance of Ti alloys to corrosion in the presence of C1-containing amine in a H atmospheric (Treseder and Miller, CA 45, 6985f; McQuillan, amine in a H atmospheric (Treseder and Miller, CA 45, 6985f; McQuillan, 5, 6036) was exptl. investigated. A series of tests was carried out under the conditions of catalytic reduction of nitro-chlorobenzene to chloroaniline (USSR 166,037, CA 62, 10368c) at a H pressure of 200 kg,/cm2, 2200°, and Cl- content in the catalyst zone of approx. 3 mol. 8. A sample of 12Kh5MA steel under these conditions was decomposed in 9 hrs., the Cr-Ni-W steels were badly corroded after 90 hrs., while high-quality stainless steels lasted 253 hrs.; the resp. corrosion rates were 19.7, 2.2, and 0.15 g./m.2 hr. Four types of Ti material showed no corrosion after 100 hrs. In more prolonged tests over 1700 hrs. of the same reaction at 140° and 200 kg./cm.2 H pressure, practically no mech. changes were observed. Full details of these tests are tabulated. A study of the microstructure of a reactor wall made from vTl-1 Ti after 2 years under the conditions of the nitrochlorobenzene reduction showed the absence of Ti hydrides. At the Cl- concentration increased to 20 mol. 8 the same material was corroded in the area of previously made incisions after 5500 hrs. of exposure. An exptl. reactor was made from a 30KhMA steel, diameter 30 mm. height 1.5 m., having a thick VT1-1 sleeve, and the thermal zone, catalyst separating sieves, drop to 200\*. In both cases hydrides of Ti were formed, and the Ti material was disintegrating.

ACCESSION NUMBER: 1966:402648 CAPLUS

DOCUMENT NUMBER: 65:2648

ORIGINAL REFERENCE NO: 65:438d-f

TITLE: 5048 of titanium for apparatus for chloroorganic synthesis in a hydrogen atmosphere

AUTHOR(S): Antonovskaya, E. I.: Pozdeva, A. A.

KNIM. Prom. (1966), 42(4), 304-5

JOURNEL LANGUAGE: JOURNEL LANGUAGE: Russian

L28 ANSWER 196 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The Ni-Al and cermet catalyst is regenerated by removing fat
from its surface with a 10% solution of Na tripolyphosphate followed by
leaching of the catalyst.

ACCESSION NUMBER: 1966:421920 CAPLUS
DOCUMENT NUMBER: 65:21920
ORIGINAL REFERENCE NO.: 65:4120b-c
REGENERATION OF STATEMENT 1966:421920 CAPLUS
65:4120b-C
65:4120b-C
Regeneration of stationary catalysts for
hydrogenation of fats
Shlyakhov, V. I.; Klochko, N. D.; Sokol'skii, D. V.;
Golodov, F. G.
Izobret., Prom. Obraztsy, Tovarnye Znaki 43(5),
15(1956)..
Patent
Unavailable
1 INVENTOR (S): SOURCE From: DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: APPLICATION NO. PATENT NO. KIND DATE DATE SU 179280 19660208 SU 19650120

ANSWER 198 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB A hydrocracking catalyst poisoned by NH3 was regenerated by heating it in an inert gas. Thus, 6 weight % NH3 on a \$i02-Al203 support was poisoned by contact with NH3 so that its hydrocracking activity was essentially zero. By treating the catalyst with He at 1 atmospheric at 800°F, at a flow rate of 9400 vols. He per volume of catalyst per hr., full catalytic activity was regained in 2 hrs. At 900°F., 60% of the activity was restored in 40 min.

ACCESSION NUMBER: 1966:83695 CAPLUS

OCCUMENT NUMBER: 64:35652g-h

RIGHAR REFERENCE NO.: 64:15652g-h

Regeneration of catalysts poisoned with nitrogen compounds

INVENTOR(5): Yamamoto, Sachio

Chevron Research Co.

2 pp.

DOCUMENT TYPP: Patent

LANGUAGE: Unavailable

1 Unavailable INVENTOR(S):
PATENT ASSIGNEE(S):
SOURCE:
DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION: PATENT NO. DATE US 3211668 19651012 US 19630130

L28 ANSWER 199 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

Title catalysts can be rejuvenated to a high percentage of their original
activity, following an extended on-stream period of 1000-4000 hrs. under
hydrocracking conditions, without repeating the thermoactivation, by a
method which comprises contacting the deactivated catalyst before removal
of the deposited carbonaceous material (I) with an aqueous (27M of the deposited catorinectae and of forming salts with the catalytic metal, preferably HOAC, for 1-7 hrs. at >100°F.; and after converting the predominant portion of the metal to salts and without washing the salts from the treated catalyst, simultaneously decomposing the salts and from the treated catalyst, simultaneously decomposing the salts and removing
I with a dry oxidizing gas at 800-1200°F. Thus, spent, thermoactivated NLB catalysts supported on silica-alumina, having metal crystallite sizes of 500-2000 A., with I still on the surface, were treated with 29 cc. of aqueous 5-7.08 HOAC/100 g. of spent catalyst at 150°F. Without removing the aqueous solution, the catalysts were dried and then gradually heated in a dry air stream up to 1000-1200°F. for 2-17.5 hrs. Before testing, each catalyst was sulfided at 550°F. A 1-hr. treatment gave results superior to acid treatment for 17.5 hrs., but longer heat treatment at the final high temperature temperature
was preferable with the longer acid treatment. Comparative
hydro-cracking
tests showed that these catalysts had a 5-18% higher activity than that of
 air-regenerated catalysts.
ACCESSION NUMBER: 1966:74959 CAPLUS
DOCUMENT NUMBER: 64:74959
ORIGINAL REFERENCE NO.: 64:13999d-f
TITLE: catalysts
INVENTOR(S): Constabaris, George; Lindquist, Robert H.
PATENT ASSIGNEE(S): Chevron Research Co.
6 pp. 6 INVENTOR(S):
PATENT ASSIGNEE(S):
SOURCE:
DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION: 6 pp. Patent Unavailable 1 APPLICATION NO. PATENT NO. DATE US US 3235486 19660215 19630701

ANSWER 201 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
The catalytic activity and total surface area of a brass catalyst for the production of MeZCO from iso-ProH is increased by simultaneous production of MeZCO from iso-PrOH is increased by simultaneous application of ultrasonic energy and chemical treatment. The chemical agent consists of a solution (I) of H2O 51.6, H2SO4 4.0, HNO3 15.4, and Na dichromate 29.0 weight %. weight %.

A catalyst so treated has an average efficiency of >90% with an estimated life of 6 mos. or more (50%, 3 mos. by conventional treatment). For example, spent S-30% brass catalyst was first immersed for 10 min. in a preclea bath consisting of a 50% aqueous solution of I. The catalyst was then research drained, and immersed in a tank also containing a 50% solution of I. Ultrasonic energy of 20,000 cycles/sec. was applied for 10 min., the temperature of bath being maintained at 75-85  $^{\circ}$ F. The catalyst was then placed in a H2O rinse tank and subjected to ultrasonic energy as before. The a #20 rinse tank and subjected to ultrasonic energy as before. The average efficiency of the catalyst for Me2CO production was 82%, compared to 63% for a portion of the same spent catalyst regenerated by conventional methods, e.g., barrel tumbling. The process is applicable to other types of catalysts as well.

ACCESSION NUMBER: 1966:55829 CAPLUS 64:05829

ORIGINAL REFERENCE NO.: 64:10453a-b
CATIFLE: Catalyst regeneration by ultrasonic and chemical treatment Graves, Charles A.; Steiner, Donald F.; Hirdler, Fairbanks C.

PATENT ASSIGNEE(S): SOURCE: Fairbanks C.

PATENT ASSIGNEE (S): Purex Corp., Ltd.
6 pp.
DOCUMENT TYPE: LANGUAGE: Unavailable 1

LANGUAGE: Unavailable 1

APPLICATION NO

US

DATE

19611204

L28 ANSWER 202 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Hexafluoroantimonic acid (I) can be used for the isomerization of straight, slightly branched, or cyclic hydrocarbons. Gradual AB Hexafluoroantimonic acid (I) can be used for the isometization of straight, slightly branched, or cyclic hydrocarbons. Gradual inactivation of the catalyst takes place as a result of formation of catalytically inert complex compds. These complexes are usually insol. in the hydrocarbon phase and accumulate in the catalyst phase. The spent catalyst is a brown, viscous, oily liquid, and is less viscous in solution in HF. At increased temps. and in presence of H, complete or practically complete decomposition of the I complexes into SbF3 and lower-moll-weight hydrocarbons occurs in the presence of tetrafluoroboric acid (II) (0.01-1 mole per mole complex), which can be recovered practically completely after reaction. The complex is treated in the liquid phase diluted with 1-10 vols. HF per volume complex to reduce its viscosity. The II can be prepared in situ by passing in gaseous BF3. The hydrogenating decomposition is usually carried out at 50-150 atmospheric H pressure and 90-160 for 0.5-20 hrs. The hydrocarbon of the complex is broken down to Cl-5 paraffins. The Sb forms SbF3, which settles as a white powder. When separated and washed, it is treated with Cl, giving SbF3. Cl2, and this with HF gives SbF5 and I with evolution of HCl. The H used need not be pure, and mixts. containing H may be taken; but the gas must be practically free from H20 and H25. From 0.2 to 0.5 m.3 H/Kg. of complex are consumed and 2-6 times the amount consumed should be present. Paraffinic complexes can be decomposed under milder conditions than ring structures. A continuous process is described. As I is extremely corrosive, special materials of construction are required, e.g. Pt and Al, or alloys of Pt and Au, Ni and Mo, Ni and W, or Al and or alloys of Pt and Au, Ni and Mo, Ni and W, or Al and Mg, synthetic materials, such as polytrifluorochloroethylene or polytetrafluoroethylene, as well as modified polymers. For example, a spent I catalyst used for isomerizing a C5-6 hydrocarbon oil fraction contained 50 parts HF and 17 parts hydrocarbon complex containing 6 s Sh. Mydrogenation-decomposition tests were carried out at 95 and 150° for 5 and 20 hrs. with and without addition of II in a magnetically stirred reactor lined with Teflon at 100 atmospheric pressure. In pressure. In the absence of II, the decomposition was very incomplete, but with 13 parts II parts II
formed in situ from BF3 + HF, 80% of the Sb was recovered at 95° and 100% after 20 hrs. at 150° as white powdered SbF3.
ACCESSION NUMBER: 1966:34773 CAPLUS
DOCUMENT NUMBER: 64:34773
ORIGINAL REFERENCE No.: 64:633 64:3173
64:6376c-g
Decomposition of complex compounds of
hexafiluoroantimonic acid and hydrocarbons formed in
isomerizing hydrocarbons
Shell Internationale Research Maatschappij NV PATENT ASSIGNEE (S): DOCUMENT TYPE: LANGUAGE: Unavailable FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE NL 301210 19650927 NL 19631202

L28 ANSWER 200 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Preheated to 90° a 10% aqueous-solution of Na tripolyphosphate (NaSP3010)
was used instead of extraction ligroine for removal of fat from a

1966:69094
64:69094
64:12979h
Regeneration of stationary catalysts for bydrogenation of fats
Shlyakhov, V. I.; Sokol'skii, D. V.; Golodov, F. G.; Klochko, N. D.
Maslozhirovaya Promyshlennost (1966), 32(1), 15-16
CODEN: MZEYFRE; ISSN: 0025-4649
Russian

was used instead of extraction ligroine for removal of fat from a fixed-bed from a NA-Al-Mo alloy. Degrees of reactivation were equal in both cases. The method is preferred for its explosion safety.

ACCESSION NUMBER: 1966:69094 CAPLUS
DOCUMENT NUMBER: 64:69094
ORIGINAL REFERENCE NO.: 64:12979h
REPERENCE NO.: 64:12979h
Repeneration of stationary catalysts for hydrogenation of fats

DOCUMENT TYPE: LANGUAGE:

PATENT INFORMATION

PATENT NO.

US 3231513

KTND DATE

19660125

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L28 ANSWER 204 OP 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The report summarized the work on tests for determining the reforming
activity
of catalysts in which H and CH4-steam mixts. were used in reduction
methods, in the forms of Na found in catalysts, catalyst
regameration, and equipment and procedures used in the redox
tests.
ACCESSION NUMBER: 1966:7731 CAPLUS
DOCUMENT NUMBER: 64:7731
ORIGINAL REFERENCE NO.: 64:1393f
TITLE:
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1906://31 CAPLUS
64:1731
64:1393f
Final report of the catalyst research task group
Cundari, D. F.
Public Serv. Elec. & Gas Co., Harrison, NJ
Proc., Operating Sect., Am. Gas Assoc. (1965), 1965,
186-94 TITLE: AUTHOR(S): CORPORATE SOURCE: SOURCE:

DOCUMENT TYPE: LANGUAGE: Journal English

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ANSWER 203 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Catalysts after long exposure to a hydrocarbon feed under hydrocracking conditions become deactivated and are difficult to regenerate. Such catalysts can be regenerated by contact with dry liquid 32Cl2 or SCl2 at 100° F. up to the b.p. of the liquid for Sl hr. The catalyst is then calcined in a dry oxidizing atmospheric at 700-1600°F, for 21 hr. When a fresh catalyst of NLB on SiO2-Al203 with a relative activity of 100 was compared to a regenerated catalyst treated as above at 1000°F, calcination temperature, the regenerated catalyst had an activity of 60: a nonregenerated batch had an activity of 5. These catalysts are composed of >1 Group VIII metal or compound on a high-surface-area support, and are used in the petroleum industry.

ACCESSION NUMBER: 66:34759
ORIGINAL REFERENCE NO.: 66:6372h, 6373-b
TITLE: Reactivating catalysts
MCClellan, Aubrey L.

AND STORMER ASSIGNEE(S): Scilfornia Research Corp.

SOURCE: WYNE. 4 Pp.
   SOURCE:
DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                                                                       4 pp.
Patent
Unavailable
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                                  PATENT NO.
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ANSWER 205 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

The regeneration of a NLB catalyst which has become deactivated after long exposure to hydrocarbon feed under hydrocracking conditions is described. The procedure comprises removal of the accumulated carbonaceous deposits on the catalyst and oxidizing the hydrogenating metal component in a steppise procedure and then reducing the metal oxide component with <3 mole % H in a mixture with N. Thus, 6% NI on SIO2Al203 used in hydrocracking operations for several thousand hrs. on hydrocrabon feedstock with a total N content of <1 ppm; 750°F. was necessary to maintain 60% hydrocarbon mixture of N and
                           air at 800-1000°F. for 24-80 hrs. The catalyst was then reduced in dry N containing 3% H under 26 in. Hg vacuum at a total gas rate of 6 vols./volume of catalyst/hr. The temperature was raised in increments of 150°F. from 400°F. to 900°F. over 8 hrs. The H content of the dry N-H mixture was raised to 6% and the catalyst was
reduced
for 1 hr. at 900°F. The catalyst was resulfided by adding to the circulating H iso-PrSH at a rate sufficient to give a 2% H2S concentration in the H flowing through the reactor at 450°F, over several hrs. to convert M to NAS. The regenerated catalyst had an 80% relative activity as compared to fresh catalyst.

ACCESSION NUMBER: 63:97106 CAPLUS
DOCUMENT NUMBER: 63:97106
ORIGINAL REFERENCE NO.: 63:17767d-f
TITLE: Regeneration of hydrocracking catalysts
Unverterth, Jack W.
PATEMT ASSIONEE(S): California Research Corp.
7 pp.
  SOURCE: 7 pp.
DOCUMENT TYPE: Patent
LANGUAGE: Unavailable
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
                           PATENT NO.
                                                                                                                                  KIND
                                                                                                                                                             DATE
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                                                                                                                                                                                                                                 APPLICATION NO.
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19651012

19620116

US 3211642

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L28 ANSWER 206 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Used hydrogenation catalysts are regenerated by digesting in
HNO3 and treating the washed precipitate with basic soins. to lower the
residual

5. Thus, 300 kg. of a used Mi-Al oxide catalyst containing 6.9% S
and 4.7% organic matter was dissolved at 90° in 900 1. HNO3 of d.
1.868. This solution was diluted to 1800 1. with distilled H20 and now
contained
77.2 g. Ni/1., 46.3 g. Al203/1., and 26.7 g. S03/1. The filter
cake from this solution was stirred with 3000 1. of 15% Na2003 at 90°.
The solution was filtered and the cake was given a secondary wash with
20 m.3
of a 0.8% Na2CO3 solution and finally washed with H20. After drying for
18
hrs. at 120°, the solid contained 33.8% Ni, 20.2% Al203,
and 0.067% S03. This product was reduced in H2 at 400° and 30 ml.
was employed as a hydrogenation catalyst for a 50-50
phenol-cyclohexanol mixture at 140° at 1 atmospheric at a rate of 100
ml. of
mixture/hr, and 200 1. H/hr. Ninety-six % of the phenol was converted to
cyclohexanol. A comparison treatment, leaving out the step of washing
with 0.8% Na2CO3 solution, yielded a catalyst containing 1.53% S03 and
converted
only 17% of the phenol in an identical reduction experiment Substitution
of 40 l. of 50% NaOH for 10 l. of the secondary carbonate wash step gave
a
catalyst which yielded a 96% phenol conversion. Substitution of 10 m3 of
1% NH4OH for the secondary wash gave a catalyst which converted 97% of
the
phenol to cyclohexanol.
ACCESSION NUMBER:
1965:493743 CAPLUS
63:93743
ONGGINLAR REFRENCE NO.:
63:17207e-f
Regeneration of nickel-aluminum oxide
catalysts
INVENTOR(S):
SUBCE:
Unavailable
FAMILIY ACC. NUM. COUNT: 1
PATENT INFORMATION:

1 Submariable
FAMILIY ACC. NUM. COUNT: 1
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DATE

19650515

DD

APPLICATION NO.

DATE

19640914

AB The addition of Cu, Pb, Ag, or Hg to a hydroreforming catalyst Containing a Group

V, VI, or, preferably, VIII metal on an acidic SiO2-Al2O3 support increases its selectivity for isoparaffins from C4-10 n-olefin feeds. Preferred catalysts contain .apprx.50 of Ni by weight and 1-20% of Hg, Ag, or Pb or 1-10% of Cu on a base of 100-500 m.2/g. cracking catalyst

containing 10-30% Al2O3 by weight Preferred conditions are 450-650°F., S1500 psig., and 1-10 vols./volume/hr. with an olefinic naphtha feed b. 20-350°F. containing a high proportion of C5-6 normal olefins. The preferred catalyst regeneration procedure calls for raduction with H followed by oxidation with gas of low O content followed by air and final raduction with H, all at an unspecified high temperature The initial raduction increases the number of times that the catalyst can be satisfactorily regenerated. Thus, a com. cracking catalyst containing 75% SiO2 and 25% Al2O3 by weight was soaked in an aqueous Ni acetate solution, dried at 400°F., calcined at 1000°F., and reduced under H at 700°F., yielding a product (I) containing 5% Ni by weight Other catalysts were then prepared by soaking I in aqueous solns. of the appropriate metal nitrate, and drying, calcining, and reducing as before. These catalysts were then used in the hydroforming of n-2-pentene at 574-90°F., 1000 psig., 10 moles of H/mole of feed, and 8.7-10 vols./volume/hr., with the results shown in the table. Alternative catalysts containing 0.1-1% of P by weight instead of Ni are referred to, without examples of their preparation or performance. Catalyst, (% by Weight), Volume %, saturates in, the product, Iso-/n-, pentane, mole catio, Relative, selectivity for, isopentane: I, 100, 1.3, 1.0; I + 1% Hg, 100, 2.3, 1.7; I + 17.1% Hg, 94.8, 8.8, 6.7; I + 11% Pb, 100, 5.0, 3.7; I + 17.6% Pb, 16.3, 4.6, 6.2; I + 1% Cu, 9.6, 3.0, 2.3; I + 5.4% Cu, 90.6, 7.0, 5.3; I + 9.2% Ag, 100, 5.5, 4.0;

ACCESSION NUMBER: 63:1640c-f
Hydron Hydro Hydro

KIND DATE

19650504

APPLICATION NO.

US

DATE

19610531

Fe(OH)3 is formed and washed away, thus increasing the Al203 content.

same result is obtained by sulfide-oxidizing bacteria. Fe2(SO4)3 is formed and washed away. When the raw material contains a sulfate, sulfate-reducing bacteria form sulfide, which is hydrolyzed to the hydroxide. In the same way, catalysts containing M16, Al203, CoS, MnS, Fe2S3, V2S3, Mo2S3, NiO, CoO, Mo2O3, Fe2O3, and Fe3O4 can be produced. When a mixture of N1(NO3)2 and NiSO4 is treated, a mixture of N10 and N18 is formed. This mixture is reduced with a reducing gas, forming nickel subsulfide.

ACCESSION NUMBER: 196:135655 CAPLUS
DOCUMENT NUMBER: 63:35655
DOCUMENT NUMBER: 63:35655
DOCUMENT ASSIGNEE(S): 25 pp.
PATENT ASSIGNEE(S): 25 pp.
DOCUMENT TYPE: Patent Unavailable
FAMILY ACC. NUM. COUNT: 1
FAMILY ACC. NUM. COUNT: 1 LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: KIND DATE APPLICATION NO. DATE NL 6411131 PRIORITY APPLN. INFO.: 19650326 19630925 L28 ANSWER 209 OF 258 CAPLUS COPYRIGHT 2004 ACS ON STN
AB A hydrogenation catalyst of Mi on kieselguhr needs to
be regenerated when the Mi has been converted to Mis.
Conventional regeneration is exidation with air followed by reduction
with H. Halogen regeneration with Cl or Br is a 3-step process. Cl is
added at 650°F. at 0.5 lb./lb. of catalyst/hr., for 2 hts.,
followed by a N purge to strip out residual S2C12; then at 700°F.
air is added for 4 hrs. at 20 ft.3/hb. of catalyst/hr., followed by a N
purge; and then the Ni is reduced at 700°F. for 4-8 hrs.
with 30 ft.3 of M/lb. of catalyst/hr. Pressure may be 1-20 atmospheric
ACCESSION NUMBER:
62:78470
ORIGINAL REFERENCE NO.: 62:13903a-b
Regeneration of spent nickel catalyst TITLE: INVENTOR(S): PATENT ASSIGNEE(S): Regeneration of spent nickel catalyst Brooke, Jesse M. Phillips Petroleum Co. SOURCE: DOCUMENT TYPE: LANGUAGE: PAMILY ACC. NUM. COUNT: PATENT INFORMATION: 2 pp. Patent Unavailable

APPLICATION NO.

DATE

19610828

PATENT NO.

US 3172863

KIND DATE

19650309

L28 ANSWER 207 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

The catalysts are prepared or poisoned catalysts are reactivated by bacterial action. The bacterial culture can flow through a fixed bed of the raw material or of the poisoned catalyst continuously or discontinuously. After this treatment the material can be heated (e.g.

500°) to stabilize the structure without endangering its activity. Examples of bacteria which can be used are: sulfate-reducing bacteria (e.g. Desulfovibrio desulfuricans), sulfide-oxidizing bacteria (e.g. Thiobacillus thiooxidans), and Fe-oxidizing bacteria (e.g. Ferrobacillus ferrooxidans). Hauxite can be activated by Fe-oxidizing bacteria. Fe(OH)3 is formed and washed away, thus increasing the Al203 content.

PATENT NO.

US 3182097

PATENT NO.

DD 39114

L28 ANSWER 210 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN AB Laboratory apparatus for studying the catalytic reduction of CO with H designed, fabricated, and operated. Three different Ni-containing catalysts were evaluated. In addition, the effects on catalyst activity determined of several impurities in the reactant gas, and of the determines of soveres any description description achieving the reduction reaction without catalyst. Virtually quant. reduction of CO with H was obtained by using Catalyst C-0765-1001 under these conditions: H2/CO mole ratio, 3:1; space velocity, 1000 under these conditions: H2/CO mole ratio, 3:1; space velocity, hr.-1;
catalyst bed temperature, 250°; and catalyst bed pressure, 6.1 atmospheric
ACCESSION NUMBER: 1965:78415 CAPLUS
CORIGINAL REFERENCE NO.: 62:78415
CATALYSIC reduction of carbon monoxide with budgers. Catalytic reduction of carbon monoxide with Catalytic Feducation of Carpon monoxide with hydrogen Rosenberg, S. D.; Guter, G. A.; Miller, F. E.; Jameson, G. R. Aerojet-Gen. Corp., Azusa, CA (1964), NASA Accession No. N64-25603, Rept. No. NASA-CR-57, 69 pp. Avail: OTS From: Sci. Tech. Aerospace Rept. 2(17), 2224(1964). Report AUTHOR (S): CORPORATE SOURCE:

DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 212 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The C and S deposits after prolonged use of catalysts are burned off by a controlled method. The method is not only applicable to catalysts

controlled method. The method is not only applicable to colling 4-25% Mo and 1-6% ML on Al203/SiO2 carriers, but also to W, Cr, Pe, Co, Pt, or Pd on Th02, B203, Zr02, MgO, or Sr0. A regenerating gas containing 0.5-1.5% O is circulated through the catalyst bed at 371° for 3 days at 6 atmospheric at a rate of 120 m.3/hr., m.3 catalyst. The combustion wave raises the temperature (111°. The SO2/SO3 content of the circulating gas is kept low by passing through a solution of NaOH, Na2CO3, or KOH at pH 8. The O concentration is kept constant by addition of air. H20 Vapor (up to 20%) is not detrimental for restoring catalyst activity by this method.

A 2nd combustion wave, also raising the temperature (111°, is induced by a constant of the temperature is a 200 to 200 to

.. 1 2nd combustion wave, also raising the temperature <111°, is induced by ncreasing the temperature to 371-454° for 2 days. The temperature is

finally
raised to 566°. After 8 hrs., the O concentration is increased to 4% for 8 hrs. The reactivity measured as the ratio of the space velocity for

the regenerated catalyst to that of a standard catalyst, required to leave only 2 ppm. basic N in the feed, is 85%.

ACCESSION NUMBER: 1965:86459 CAPLUS

BOCUMENT NUMBER: 62:8459

CATIONAL REFERENCE NO.: 62:1497D-d

TITLE: Regeneration of hydrorefining and hydrocarcking catalysts

CATALYSISIGNEE(S): Universal Oil Products Co.

BOCUMENT TYPE: 13 pp.

BOCUMENT TYPE: Patent

LANGUAGE: Universal Oil Products Co.

TARGUAGE: Universal Oil Products Co.

LANGUAGE: Universal Oil Products Co.

TARGUAGE: Universal Oil Products Co.

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

APPLICATION NO. DATE PATENT NO. KIND DATE NL 6400750 PRIORITY APPLN. INFO.: 19640803 19630131

L28 ANSWER 211 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB In the catalytic process, a petroleum distillate and H are passed at 500-850°F, and 600-3000 psig. through an isomerization cracking zone. The effluent is freed of a H-rich recycle stream, which is then returned with make-up H to the reaction zone. This process is modified by

removing NH3 from the recycle gas and operating below 750°F. during most of the on-stream time; 5-20 lb. of H20/bbl. of feed is added to the recycle stream which is then cooled and the aqueous NH3 is separated Feedstocks

Feedstacks
may contain 200-500 p.p.m. N (preferably <200 ppm.). The catalyst
consists of Group VI or VIII metals including oxides or sulfides of Mo,
Fe, Ni or Cr. In one example, by operating at 1200 psig.,
775 F., and an hourly liquid space velocity of 2 on a catalyst of
0.3% Pt on SiO2-AlO3, the increase in gasoline yield per pass conversion
was 53%. The feed contained 100 ppm. N. Cf. CA 54, 25751g.
ACCESSION NUMBER: 1965:29040 CAPLUS

DOCUMENT NUMBER: 62:29040
ORIGINAL REFERENCE NO.: 62:5124c-e
TITLE: Catalytic Catalytic upgrading of nitrogen-containing petroleum distillates

INVENTOR (S): Scott, John W., Jr.; Stanton, Lyman S.; Mason, Harold

PATENT ASSIGNEE(S): r. California Research Corp.

SOURCE: DOCUMENT TYPE: LANGUAGE: 5 pp. Patent Unavailable

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. APPLICATION NO. KIND DATE DATE US 3157590 19641117 US 19620511

L28 ANSWER 213 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

During catalytic cracking of hydrocarbons containing Ni, Fe, and
(or) V as impurities, the impurities deposit on the catalyst and reduce
its efficiency. The poisoned catalyst is regenerated by continuously
bleeding a small portion of the catalyst, treating it with the

sulfidizing agent (such as H2S, CS2, or a mercaptan) at 800-1500°F., then treating it with a vapor which is a mixture of 5-10 parts of a

treating it with a vapor which is a mixture of 5-10 parts of a chlorinating agent consisting of either HCl or Cl or their mixture and 1 part of a promoter (a Cl-containing compound of C and S, such as CCl4 or SSCl2) at 700-1000°F. and 0-100 pag. The chlorinator effluent containing excess chlorinating vapor and chlorides of V and Fe is washed with H20. The catalyst is also washed with H20 to remove chlorides of Ni. Both washings are acidic. The reformed gasoline also is washed with NaOH solution of PH 12 to remove compds. such as phenols and H2S. These washings

washings are alkaline and are neutralized with the acidic washings before disposal.

ACCESSION NUMBER: 1965:8458 CAPLUS

DOCUMENT NUMBER: ORIGINAL REFERENCE NO.: 62:8458 62:1497a-b

oc:1979.00
Cracking catalyst demetallization effluent treatment
Smith, kenneth A.; Watson, William B.
Sinclair Research, Inc.
7 pp.
Patent TITLE: INVENTOR(S):

PATENT ASSIGNEE(S):

SOURCE: DOCUMENT TYPE: Unavailable

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE US 3150074 19640922 US 19601128

ANSWER 214 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
High-metal-content Ni-Mo-Al203 hydrofining catalyst is made
containing 20.5% Mo and 6.4% Ni on an Al203 a support and is brought
to 15% S at 450°F. Fresh catalyst reduces 775 p.m. N in a light
cycle oil to 0.7 p.p.m, at 645°F., 800 lb./in.2 gage, liquid hourly
space velocity, and 4000 ft.s H/bbl. Aged catalyst becomes coked and
reduces the N to only 12 p.p.m, at 700°F. or to 0.7 p.p.m, at
750°F. Steam-air regeneration up to 800°F. did not restore
the activity completely. Reduction to <0.1 p.p.m. N at
645°F. is possible with aged catalyst which has been regenerated
with 0.6 volume % O in N at 530°F., gradually increasing to 5 volume %
0 at 800°F., and followed by cooling plus resulfiding by using H
containing 1% Me2S. H2O at 0.6 volume % in the O-N regeneration gas
ults in containing 18 Mezs. HZO at 0.0 volume 1.11 cits 0.1 cyclimes 2.11 cits 0.11 cits 0.1 SOURCE:
DOCUMENT TYPE:
LANGUAGE:
PATENT INFORMATION: DATE PATENT NO. DATE APPLICATION NO. KIND GB 962778 US 3172864 US 3211669 PRIORITY APPLN. INFO.: 19640701

19610921

L28 ANSWER 216 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Hydrocarbon distillates, b. \$1050°F. and of high N content,
are hydrocaracked in the presence of H and a catalyst consisting of a
hydrocymating-dehydrogenating component arranged on a solid,
active acidic support to produce products boiling below the initial b.p.
of the feed. The process has long on-stream periods without need for
catalyst regeneration to produce the desired products
with substantial consumption of H, feed denitrification, and at
<775°F. and 1500-3000 lb./in.2 A light cycle oil comprising a
blend of 19.91 raw cycle oil, 41.38 hydrocflaed cycle oil, and 46.61
recycle bottoms containing 60 p.p.m. of N was hydrocracked and
concurrently
denitrified over a Ni sulfide catalyst on a solid SiO2-active

irrently denitrified over a Ni sulfide catalyst on a solid SiO2-active AlZO3, acid cracking support, accompanied by 6500 ft.3 of H/bbl. The catalyst immediately after being placed on stream was rapidly of feed.

sted with NH3 derived from the N content of the cycle feed oil at 400-750°F., 1800 lb./in.2 gage, and a liquid hourly space velocity of 0.5. After such rapid equilibration of the catalyst with NH3, temperature necessary to maintain a 50% conversion of the feed to

ucts

boiling below the initial b.p. of the feed abruptly leveled off at
.apprx.730°F. The hydrogeneration-dehydrogenation was effective
for approx. 800 hrs. without raising the operating temperature
\$0.033°F./hr. To obtain a 600 conversion, it was necessary to
raise the operating temperature at a greater rate, but the operation
d be

continued for almost 1000 addnl. hrs. without raising the temperature

rapidly than 0.073 $^{\circ}$ F./ hr. During the 50% conversion period, the average temperature was 730 $^{\circ}$ F. and during the 60% conversion period, it

WAS

790 °F. The total length of the run before prohibitive catalyst
fouling took place was approx. 2000 hrs.

ACCESSION NUMBER: 1964:59949 CAPLUS

ORIGINAL REFERENCE NO.: 60:10451d-f
HYdrocracking of nitrogen-containing distillates

INVENTOR(S): Scott, John W.: Mason, Harold F.

PATENT ASSIGNEE(S): California Research Corp.

3 pp.
DOCUMENT TYPE: Patent
LANGUAGE: Unavailable

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. US 3117075 GB 992264 19640107 US GB 19610328

L28 ANSWER 215 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB The Ni is derived from the formate rather than the nitrate. A
108 Ni on sepiolite derived from Ni formate was
activated by treatment in a H atmospheric at 1000 vols. of H/volume catalyst/hr.

for 1 hr. at 250°. The catalyst was then oxidized with air or 0
and sulfurized with thiophene. Subsequent reaction with a 10% isoprene
mixture in n-heptane at 100° and atmospheric pressure indicated 100% 60:13195er
Improvements (in the manufacture of nickel catalysts) to catalysts and the reactors containing these catalysts
Holmes, Peter D.: H.Bourne, Kenneth
British Petroleum Co. Ltd. INVENTOR(S):
PATENT ASSIGNEE(S):
SOURCE:
DOCUMENT TYPE:
LANGUAGE:
PATENT INFORMATION: 14 pp. Patent Unavailable

PATENT NO. KIND DATE APPLICATION NO. DATE BE 629089 GB 1009590 PRIORITY APPLN, INFO.: 19630701 19620302

ANSWER 217 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
The formation of C deposits on the catalyst is prevented and yield
improved if a mixture of steam and 0 or air is passed continuously ugn a dehydrogenation catalyst, preferably Ni or Ca phosphate at 550-700° and atmospheric pressure, the hydrocarbon being admitted intermittently so that the periods of dehydrogenation and regeneration 30-300 sec. each. The preferred proportions are 1-4 parts of O and 150-250 parts of steam per 10 parts by volume of hydrocarbon, and the process is particularly suitable for C4-8 hydrocarbons. Thus a mixture 645 ml. 1-butene and 9.3 g. steam/min. was passed through 500 g.
Ni-Ca phosphate-Cr203 catalyst at 602° initial and
582° final temperature, followed by 15 min. regeneration with steam and 0
before the next 15 min. dehydrogenation period. The yield of
1,3-butadiene was 368 with about 91.78 selectivity. The use of a
continuous stream of the same butene-steam mixture as before plus 1 volume of
O/volume of butene at 602°, with the butene shut off for 1 min. in
each 2 min. gave 43% yield with 90.4% selectivity. An increase of temperature
to 670° improved the yield to 57%, with a reduction in
selectivity to 83%, and shutting off the O during the dehydrogenation
period reduced the yield by 4-5% over the temperature range. Qual. similar results were obtained for the dehydrogenation of 2-methyl-2-butene to results were obtain isoprene. ACCESSION NUMBER: DOCUMENT NUMBER: ORIGINAL REFERENCE NO.: TITLE: INVENTOR(S): PATENT ASSIGNEE (S): SOURCE: DOCUMENT TYPE: LANGUAGE: PATENT INFORMATION: 1964:52311 CAPLUS 60:52311 60:9145g-h,9146a Catalytic dehydrogenation Alexander, Douglas S.: Firko, John Polymer Corp. Ltd. 24 pp. Patent Unavailable PATENT NO. KIND DATE APPLICATION NO. DATE BE 626512 FR 1351509 GB 976913

19630415

19620106

PRIORITY APPLN. INFO.:

L28 ANSWER 218 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
A Ni (9.1 weight %)-on-Al203 hydrogenation catalyst is
used for 2600 hours in the com. H treatment of steam-cracker gasoline.
During this time, the temperature is gradually raised from 111° to
204° to maintain catalyst activity. The activity can be restored
to the 143° level after a 4-hr. H treatment at 250° 1 atmospheric,
and H space velocity 100. An extra 700 hrs. of operating life is

achieved

before the 204' level is again reached. A total production of 17.4

bbl. of feedstock/lb. of catalyst is obtained. Similarly Ni on
sepicilte has its life extended from 4934 hrs. to 26200 hours for a
total production of 51 bbl./lb. of catalyst.
ACCESSION NUMBER: 1964:38031 CAPLUS
DOCUMENT NUMBER: 60:38031
ORIGINAL REFERENCE NO.: 60:6684a-b

Reactivation of gasoline-hydrogenation catalysts white, Peter T.; Olive, Martin F. British Petroleum Co. Ltd.

INVENTOR (S):

PATENT ASSIGNEE(S):

3 pp.
Patent
Unavailable SOURCE: DOCUMENT TYPE:

LANGUAGE: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. 19631203 US US 3113097

ANSWER 220 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Approx. 90% of the original activity can be restored by lowering the content of As by extraction or heating the catalyst in H or HZS as usual

then adjusting the atomic ratio between Ni and As to 3-6:1. The content of Ni must be at least 1.5-2.5% higher than the rest of As. Concentration of the used solns, must be sufficiently high to

after a single treatment such an amount of N1 into the catalyst as requires the given ratio. The same goes for other active components, especially Co.

ACCESSION NUMBER: 1963:458811 CAPLUS

DOCUMENT NUMBER: ORIGINAL REFERENCE NO.: 59:58811 59:10798e-f

Reactivating arsenic-deactivated hydrogenation catalyst containing oxides or sulfides of metals of Groups VI and VIII Svajgl, Oldrich; Jiricek, Bedrich

INVENTOR (S):

SOURCE: DOCUMENT TYPE: LANGUAGE: PATENT INFORMATION: 3 pp.
Patent
Unavailable

PATENT NO. KIND DATE APPLICATION NO. DATE CS 103673 19620515 cs 19610214 L28 ANSWER 219 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Mi catalyst used for the hydrogenation of oils is
extracted from the spent mixts, by means of 10-208 HNO3. The resu
Mi(NO3)2 is converted into NiCO3 by the addition of Na2CO3. Humid
NiCO3 is heated to 85-90° and treated with HCOOH to obtain
Mi formate from which a new catalyst is obtained by thermal
decomposition in oil. The activity of this regenerated catalyst i
satisfactory. Com.-scale expts. led to regeneration of >95% Ni
from spent catalysts.
ACCESSION NUMBER: 1964:2379 CAPLUS
DOCUMENT NUMBER: 60:2379
ORIGINAL REFERENCE NO.: 60:347e-f
TITLE: Regeneration of a nickel catalyst
AUTHOR(S): Malinowski, Stanislaw
CORPORATE SOURCE: Zakl. Przemyslu Tluszczowego, Gdansk, Pol.
SOURCE: Tuszcze i Srodki Piorace (1962), 6(5), 202-9
CODEN: TSRPAP; ISSN: 0495-6672
DOCUMENT TYPE: Journal
LANGUAGE: Unavailable

ANSWER 221 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Straight-run gas oil is extracted by silica gel beds to remove aromatics. This follows alternate adsorption-desorption phases in which the This follows alternate successful and desorbed by an effluent, such as a C6M6-pentane mixture The aromatics must be completely removed prior to hydrocracking with AlBr3-HBr catalyst. The straight-run gas oil, free of aromatics, is mixed with 30-50 weight % of C7-9 naphthene diluent. The hydrocarbons are hydrocracked under 25-500 lb./sq. in. partial pressure H in contact with AIBr3 in the presence of HBr at 25-75°C. The residence time in the reactor zone is regulated so that the major portion of the feed is cracked. The hydrocarbons are stripped of HBr and fractionally distilled to give an isobutane fraction, a C5-7 paraffin fraction, a heavy gasoline fraction, the naphthene diluent fraction in C7-9 range, and the higher-mol.-weight hydrocarbons mixed with AIBr3 as residue. The naphthene diluent fraction and the residual hydrocarbon-AlBr3 mixture are recycled into the reaction zone along with stripped HBr and mixed with fresh feed stock. The extracted aromatics mixed with catalytic gas oil and fuel oil and hydrocracked in the mixed With Catalytic gas off an array mixed with. Nis, CoS, or Cr2S3 as a hydrogenation-type catalyst mixed with. Nis, CoS, or Cr2S3 as a hydrogenation-type catalyst in contact with H at a partial pressure of 500-5000 lb./sq. in. The hydrocracking temperature is 350-1050°F. The H is separated from the hydrocracking product, and fractional distillation gives a C4 or lighter fraction, a gasoline fraction. fractional distillation gives a C4 or lighter fraction, a gasoline fraction, and
a residuum which may be recycled. The high-octane gasoline is a blend of the aromatic-catalytic gas oil-hydrocracked gasoline with the C5-7 paraffins and the heavy gasoline fraction from the AlBr3-HBr hydrocracking. Almost all of the feed is convertible to gasoline.

ACCESSION NUMBER: 1963:80499 CAPLUS
DOCUMENT NUMBER: 55:80499
ORIGINAL REFERENCE NO.: 55:13687c-f
TITLE: Gas-oil hydrocracking process to produce a high-octane gasoline Mertes, Thomas S. Sun Oil Co. 4 pp. Patent Unavailable INVENTOR(S):
PATENT ASSIGNEE(S):
SOURCE:
DOCUMENT TYPE:
LANGUAGE: PATENT INFORMATION: APPLICATION NO. PATENT NO. DATE US 3080311 19630305 US 19600915

L28 ANSWER 222 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

Used Raney N1 from anthraquinone-H202 process is treated with an
alc. solution of an organic acid 1-3 hrs. in an inert or H atmospheric,
then washed

washed with alc. Thus, 50 g. exhausted catalyst was added to a solution

with alc. Thus, 50 g. exhausted catalyst was added to a solution containing 140
ml. MeOH, 60 ml. AcOH, and 20 ml. H2O, agitated 1 hr. at 25° in a N atmospheric, then separated from liquid and washed with MeOH. This procedure reduces
tendency of catalyst to form tetrahydroquinones by excessive hydrogenation.

ACCESSION NUMBER: 1962:414386 CAPLUS

DOCUMENT NUMBER: ORIGINAL REFERENCE NO.: TITLE: 57:14386 57:2906b-c

J::ZYUGD-C Regeneration of Raney-nickel catalyst Hauschild, Ulrich; Nicolaus, Horst Kali-Chemie A.-G. INVENTOR (5):

PATENT ASSIGNEE(S): SOURCE: 3 pp. Patent

DOCUMENT TYPE: LANGUAGE: Unavailable

PATENT INFORMATION:

APPLICATION NO. DATE PATENT NO. KIND DATE 19611228 1965 19600805 DE 1120432 US 3165478

L28 ANSWER 224 OF 258 CAPLUS COPYRIGHT 2004 ACS ON STN

AB Sulfided Ni catalyst from desulfurizing-hydrofining operations
is melted and cast into anodes. The anodes, in cotton-fabric diaphragms,
are electrolyzed by using NACI solution as the electrolyte, a Ni
cathode, and a c.d. of 9-13 amp./sq, ft. S collects in the diaphragm while Mi hydroxide ppts. in the electrolyte. The precipitate is

Separated

from the electrolyte, washed, and converted to catalyst, e.g., by reduction with H (obtained from the cathode).

ACCESSION NUMBER: 1962:52423
ORIGINAL REFERENCE NO: 56:9883f-g
RIVENTOR(S): Geynn, Marion Hayes
DOCUMENT TYPE: Patent
LANGUAGE: Patent
Unavailable
Patent Uncornation:

PATENT INFORMATION:

PATENT NO. DATE DATE KIND US US 3019181 19620130 19590205

L28 ANSWER 223 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB A continuous process and apparatus are described for the manufacture of
high-purity
H by the catalytic reduction of steam in a fluidized bed of
particles of a metal or metal oxide of the Fe group. The improvement
consists in regeneration of the oxidized metal-steam catalyst by sep.
treatment with a reducing gas mixture (mainly H and CO) prepared by
reaction

reaction
of hydrocarbons with steam in the presence of another catalyst of the Fe
group. In an example given, the catalyst is 20% Fe on a SiO2-A12O3 base,
the hydrocarbons are natural gas containing mostly CH4, and the catalyst

preparation of reducing gas is Ni on SiO2-Al2O3. In the H-generation Zone, steam is reduced in a countercurrent bed of an Fe catalyst, the H

collected, and the oxidized Fe catalyst is cycled to a reducing zone, where it is regenerated in a fluidized bed through which the reducing-gas mixture is passed. The Fe catalyst is recycled to the 1st zone and the

gases are partly recycled and partly mixed with air to form a combustible gas mixture. In a reaction zone, natural gas and steam are heated in a countercurrent fluidized bed of Ni catalyst to produce the reducing gas mixture for the Znd zone and a C-laden catalyst. The latter catalyst is fed to a heating zone, where the C is burned off by the

catalyst 18 feu to a measure service mixture
of air and combustible gases from the reducing zone. The regenerated, heated Ni catalyst is recycled to the reaction zone.
ACCESSION NIMMERS: 1962:78026 CAPLUS
DOCUMENT NUMBER: 56:78026
ONIGINAL REFERENCE NO: 56:15152h-i,15153a-b
Steam-iron process for manufacturing high-purity hydrogen

watkins, Charles H.
Universal Oil Products Co.
Patent INVENTOR(S):
PATENT ASSIGNEE(S):
DOCUMENT TYPE:
LANGUAGE:

Unavailable

PATENT INFORMATION:

APPLICATION NO. PATENT NO. KIND DATE DATE 19620327 19591207 US 3027238

ANSWER 225 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
The metal ion of the carrier and that of the active addition should have

same diameter. The mixture of their precipitated hydroxides is converted

Neat (100-200\*) and pressure (10-100 atmospheric) to carbonate crystals, which are then decomposed with H. Thus, 50% of the precipitate from

6 g. NiSO4.7H2O and 24 g. MgO in 900 ml. H2O was treated for 15 hrs. at 150° with CO2 under 80 atmospheric The product contained 17.4% Ni , 18.8% Mg, and 12.68% (calculated 12.82%) C. A portion containing 100

Ni was treated with 20 1./hr. of H during 2.5 hrs. at  $400^{\circ}$ . This catalyst reduced 5 ml. cyclohexene at room temperature in 9 min.;

the

corresponding one without the CO2 treatment took 36 min. For a Co
catalyst, the difference was still greater. A catalyst from Cu(NO3)2 and
CaO used for the reduction of PhNO2 at 270° and 30 ml./hr.
yielded 90% PhNH2 as compared to 27% with the untreated catalyst.
ACCESSION NUMBER: 1961:133182 CAPLUS
DOCUMENT NUMBER: 55:133182
ORIGINAL REFERENCE NO.: 55:25094b-d
TITLE ORIGINAL REFERENCE NO.: 55:25094b-d
TITLE ORIGINAL REFERENCE NO.: 55:25094b-d
TITLE ORIGINAL REFERENCE NO.: 55:25094b-d

33:20948-4 Activation of mixed catalysts Lagenbeck, Wolfgang; Welker, Jurgen; Dreyer, Hans; Nehring, Dietwart; Mahrwald, Richard VEB Leuna-Werke "Walter Ulbricht" INVENTOR(S):

PATENT ASSIGNEE (S):

DOCUMENT TYPE: LANGUAGE: Patent

Unavailable

FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

APPLICATION NO. PATENT NO. KIND DATE DATE

19610112 DE DE 1096878

ANSWER 226 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN A process is described for regenerating a SiO2-Al2O3 catalyst support containing 2-8 by weight of MI which has been activated by reduction, oxidation, and final reduction with pure M. The regeneration process comprises purging the reaction zone with an inert

regeneration process comprises purging the reaction zone with an inert
gas
free of H2O, CO, and H2S at about 600-700°F, and 0-15 lb./sq. in.
gage to strip the catalyst of hydrocarbons. Next, an O-containing gas is
brought into contact with the catalyst at 800-1000°F. until the
reaction ceases. The catalyst is again purged with dry inert gas at
975-1050°F, until dehydration is complete. Then the oxidized
catalyst is reduced with a H-containing gas at 700-75°F, until the
reduction is complete and all by-product water is zemoved. The
regenerated catalyst has about the same efficiency as the original
product. Minor alterations from the above conditions are also claimed.
Cf. U.S. 2,917,565 (see Brit. 787, 904, CA 52, 1014lb); U.S. 2,917,566
(see Brit. 814,708, CA 54, 299b).
ACCESSION NUMBER:
1961:56648
ORIGINAL REFERENCE NO.: 55:10865b-d
RIGINAL REFERENCE NO.: 55:10865b-d
RIGINAL REFERENCE NO.: 55:10865b-d
RIGINAL REFERENCE NO.: 55:10865b-d
RIGINAL REFERENCE NO.: 50:10865b-d
RIGINAL REF

PATENT NO. US 2968631

DATE 19610117

US

APPLICATION NO.

Answer 228 of 258 Capius Copyright 2004 ACS on STN
A process for the continuous production of piperidine (I) by
hydrogenation of pyridine (II) with Ni as the catalyst
was discussed. As Ni reacted with pyridine and thus lost its
catalytic activity, bases were added to free Ni from the compound
The catalytic contained 478 Ni, 528 Al, and 48 Fe. The base
solution was mixed with II and placed in an autoclave where II was
hydrogenated at 160° and 50 atmospheric Better results were
obtained by using NaOH as base instead of Na2CO3. Different expts. with
various amis. of base solution, variations in the catalyst, temps., and H
pressures were discussed. Some side reactions were noticed, such as the
formation of pyrrole and picoline. This could be avoided by
hydrogenating under best conditions in 2 phases. In the 1st
phase, II was hydrogenated to 65-68 and in the 2nd phase to
92-100°. The yield was 91% in respect to II. The catalyst did not
show any loss in activity after 720 hrs. reaction or 4000 hrs. in the

I-II
mixture
ACCESSION NUMBER:
DOCUMENT NUMBER:
ORIGINAL REFERENCE NO.:
TITLE:

AUTHOR(S): SOURCE:

DOCUMENT TYPE: LANGUAGE:

1960:128940 CAPLUS
54:128940
54:24707c-e
A continuous process for the production of piperidine
by catalytic hydrogenation of pyridine
Sultanov, A. S.; Vasil'eva, N. V.; Safaev, A. S.
UZbekskii Khimicheskii Zhurnal (1960), (No. 1), 81-7
CODEN: UZKZAC
JOURNAI
RUSSIAN

L28 ANSWER 227 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The Mi oxide-on-kieselguhr catalyst used in hydrofining wax was reactivated without plant shutdown. By raising the usual reaction temperature of 450-600 to 650-700° F. for 6 hrs. at 2-week intervals, the life of the catalyst was greatly extended. The wax processed before and after treatment of the catalyst showed a 46 and 86%, resp., reduction in absolute color units with 6.2 and 12.2% reduction in optical d. at 330 mm.

ABSOLUTE CONTENT

THE PROPERTY NUMBER: 1961:45726 CAPLUS

DOCUMENT NUMBER: 55:45726

ORIGINAL REFERENCE NO.: 55:8843c-d

Hydrofining of wax

PATENT ASSIGNEE(S): Easo Research and Engineering Co.

DOCUMENT TYPE: Unavailable

Unavailable

TITLE: Hy
PATENT ASSIGNEE(S): E:
DOCUMENT TYPE: P:
LANGUAGE: UT
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

GB 851969 19601019 GB

L28 ANSWER 229 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Pure Ni from reduction of Ni salts loses its
activity owing to absorption of cyclohexane and may be regenerated by
passage of H over the Ni for 3 or more hrs. at the
hydrogenation temperature [120-230°], by storing the Ni
in a H atmospheric at room temperature for 3-4 hrs. or more, by passing

Ar or N over
the Ni at 184°, or by storing the Ni under a N
or other inert gas atmospheric at room temperature

ACCESSION NUMBER:
1960:124194 CAPLUS

OCCIMENT NUMBER:
454:124194

ORIGINAL REFERENCE NO.: 54:23671d-e

Wetallic nickel and nickel on
aluminum oxide as hydrogenation catalysts.
I. The particular reversible change of activity of
metallic nickel catalysts during
hydrogenation of benzene
Alchudchan, A. A.

AUTHOR(S):
(1959),
12 377-88

AUTHOR(S): CORPORATE SOURCE: SOURCE: (1959),

12, 377-88 Journal Russian DOCUMENT TYPE: LANGUAGE:

Page 73

ANSWER 230 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
Regeneration of spent Mi catalysts for the hydrogenation
of glucose to sorbitol was investigated. The spent catalyst could be
regenerated by boiling in 208 NaOH for 1 hr. followed by washing and H2
treatment at 100° and 70-80 atmospheric; repeated similar treatments
damaged the catalyst structure. A catalyst containing Ni 46, Al 52,
and Cr 21 was freshly prepared and was used for repeated
hydrogenations of 168 glucose solution (pH 7.5-8) at 100° and
60 atmospheric It was regenerated after 17 initial and 7, 6, and 5
secutive
runs of 1-hr. duration by H2 treatment in EtOH at 100-20° and 70
atmospheric for 1 hr., but after the third regeneration and the 5 runs a
treatment with 108 NaOH was necessary to remove the polymer film from the
catalyst surface. Two other catalysts also were prepared and studied. Catalyst surface. Two other catalysts also were prepared and studied.

Ni-Al-T1 one was slightly more active than the Ni-Al-Cr catalyst and was successfully regenerated (10 g. catalyst) with H2 in 25 ml. solution containing 12.5 ml. 0.1 N NaOH and 12.5 ml. 961 EtOH. The Ni-Al-Cr-B catalyst was prepared in the 46:521.9:0.1 ratio and was more active and more stable than the other two.

ACCESSION NUMBER: 1960:82559 CAPLUS

DOCUMENT NUMBER: 35:82559

ORIGINAL REFERENCE NO. 54:137622-i

Some methods of nickel-skeleton catalyst regeneration

AUTHOR(S): Scheglov, N. I.; Sokol'skii, D. V.

CORPORATE SOURCE: Inst. Chem. Sci., Acad. Sci. Kazakh. S.S.R., Alma Ata Trudy Inst. Khim. Nauk, Akad. Nauk Kazakh. S.S.R.

DOCUMENT TYPE: Journal Language: Unavailable

DOCUMENT TYPE: Journal Language: Unavailable

L28 ANSWER 232 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Metallic catalysts, e.g., NA, Cu, Co, or Fe, poisoned by S compds. present as impurities in hydrogenation of dehydrogenation processes, are purified by treatment with liquid HCO2H. Ni poisoned by S compds. in the hydrogenation of tech. Cresol is purified, and its activity is restored by the following treatment in the reaction vessel. Add liquid HCO2H in a stream of N at temperature of 85°, allow to remain overnight, add H slowly, and raise the temperature to 120° and finally to 185° to remove excess HCO2H, and H2S formed. Repeat until all H2S is driven off.

ACCESSION NUMBER: 1959:8487 CAPLUS
DOCUMENT NUMBER: 53:8487

ORIGINAL REFERENCE NO: 53:1595a-C
REACTIVATION (S): HORST CONTROLLED TO PROMOTOR SINGLANGUAGE: HORST CONTROLLED TO PATENT TYPE: Fatent Unavailable
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO. DATE APPLICATION NO. DATE GB 797111 19580625 GB

L28 ANSWER 231 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Catalytic reactions which produce C deposits on the catalyst and the subsequent oxidative regeneration are carried out in a set of parallel reactors so that continuous operation results. The reactors contain a of successive catalyst beds. Only the 1st bed is regenerated with fresh steam and a part of the O-containing regeneration gas. The hot Sceam size of personal strength of the strengt

from the 1st (upper) catalyst bed is cooled and diluted within the reactor
with H2O and fresh O-containing gas is injected into the catalyst-free sections and then enters the next (lower) bed. In an example, crude oil containing 4.87% S was desulfurized in the presence of H and a Ni W oxide-Al2O3 catalyst at 433-62° and 70 atmospheric A deposit of 9-10% coke (by weight of catalyst) was formed. After 4 hrs., the reactors were purged with H for 1 hr., subjected to pressure release for 10 min., purged

purged
with steam for 20 min., and regenerated with a gas containing O 3.2, N
19, H2O
76.9, and CO2 0.9% at 399° and 18.3 atmospheric for 2 hrs. The

76.9, and CO2 0.9% at 399 and 10.5 section 10.5 combustion gases were mixed with air and steam and recycled.

ACCESSION NUMBER: 1960:30599 CAPLUS

DOCUMENT NUMBER: 54:30599

ORIGINAL REFERENCE NO: 54:5990c-f
Oxidative regeneration of catalysts loaded with carbonaceous deposits

Glazier, Edwin M.; Hirsch, Joel H.

Gulf Research & Development Co.

PATENT ASSIGNEE(S): Gulf Research & Development Co.

PATENT ASSIGNEE(S): Unavailable

PAMILY ACC. NUM. COUNT: 1

INVENTOR(S):
PATENT ASSIGNEE(S):
DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:

DATE PATENT NO. KIND APPLICATION NO. DATE DE 19570328 DE 1005219

L28 ANSWER 233 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB See U.S. 2,671,763 (C.A. 48, 6620b).
ACCESSION NUMBER: 1958:89587 CAPLUS
DOCUMENT NUMBER: 52:89587
ORIGINAL REFERENCE NO.: 52:15786c-d Catalyst regeneration
Winstrom, Leon O.
Allied Chemical & Dye Corp. TITLE: INVENTOR(S): PATENT ASSIGNEE(S): DOCUMENT TYPE: Patent Unavailable FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

APPLICATION NO. PATENT NO. DATE KIND DATE DE 927508 19550509 DE

ANSWER 234 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

Catalysts containing combined sulfides of metals from the groups Pe, Co,
NI, and Cr, Mo, and W are reactivated by oxidizing the
carbonaceous deposits between 400° and 550°, cooling with
air to <220°, and resulfiding at about 200° with H and H2S
in a mole ratio of 12:1. Thus, an activated Al203 carrier impregnated
with 2.99 NN and 13.68 Mo in the form of their sulfides is
reactivated after use in the hydrogenation of crude
a-methylnaphthalene by oxidizing with air for 16 hrs. at
400°, cooling with air to 20° and treating with H and H2S
in a 12:1 mole ratio for 1 hr. at 200°. The catalyst regained 90%
of its oxiginal activity.

ACCESSION NUMBER: 59:22504
ORIGINAL REFERENCE NO.: 52:22504
CRIGINAL REFERENCE NO.: 52:4069h-i
TITLE:

52:4069n-1 Reactivation of sulfide catalysts Nozaki, Kenzie Shell Development Co. PATENT ASSIGNEE(S);
DOCUMENT TYPE:

Patent Unavailable LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. US 2813835 KIND DATE

APPLICATION NO. 19571119 us

DATE

ANSWER 236 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
Hydrogenation catalysts consisting of or containing Pt, Pd, Rh, Ir,
Ru, Os, Ni, Co, Cu, or Fe, Which have been poisoned in
liquid-phase hydrogenation, dehydrogenation, or
reduction processes by compds. of Se, S, Te, P, As, or Sb, or by S
or P in a free state, or by catalytically toxic metallic ions or metallic
compds., are revivified by desorption or desorptive displacement of the
poison by treating the poisoned catalyst with a solvent for the poison,
but a solvent which does not attack either the catalyst or the poison
chemically. The liquid consists of or contains an unsatd. substance, the
chemical structure of which includes 1 or more acetylenic or ethylenic
s,

, or a set or sets of benzenoid bonds, or 1 or more carbonyl groups, or unsatd. bonds between 2 N atoms or between a C atom and an N atom. T liquid consists or or contains cyclohexene (I), C6H6, ethylenic

generally, or acetone. In an example, Pt black which had been poisoned

generally, or acetome. In an example, Pt black which had been poisoned in the liquid-phase hydrogenation of an unsatd. substance so that its activity for hydrogenation of I under standardized conditions at 30° was less than 7% of its original unpoisoned activity, was regenerated to its original activity by 2 extns., each for 15 min. at room temperature: (1) 400 cc. AcOH/g. Pt, and (2) 400 cc. AcOH containing 10% I/g. Pt. ACCESSION NUMBER: 50:91895
DOCUMENT NUMBER: 50:91895 CAPLUS
DOCUMENT NUMBER: 50:91895
DOCUMENT NUMBER: 50:91895
TITLE: Catalyst regeneration
HXMETOR(S): Catalyst regeneration
HXMETOR(S): University of the Characteristic of the Containing University of the Co

PATENT NO. GB 744049

DATE 19560201 APPLICATION NO. DATE

L28 ANSWER 235 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Impregnation of active C with aqueous caustic containing activators
increases the
magnitude of the partition coefficient of mercaptans in hydrocarbon
distillates. Therefore, they are removed more efficiently. Regeneration
of the active C is accomplished by steaming or air-blowing. The process,
including the regeneration, takes place at 10-5°. For example,
active C having a d. of 330 g./l. is saturated with 1 l./kg. of a 12.5%
aqueous

ous
MaOH solution A straight-run gasoline containing 0.013% mercaptan S is
percolated at a throughput of 13 vols. gasoline/volume C/hr. to give 71

doctor-sweet product per kg. of C. This product contains approx. 0.001% mercaptan S. The ability of the NaOH-impregnated active C to remove mercaptans is increased by use as activators of a variety of compds.

are given on the use of MeOH, ethylene glycol, dimethylamine-HCl, diethanolamine, butyric acid, naphthenic acids, tartaric acid, AcH, acetone, grape sugar, PhOH, m-cresol, mixed phenols, l-naphthol, pytocatechol, resorcinol, hydroquinone, pytogallol, salicylic acid, pytocatechol, resorcylic acid, gallic acid, tannins, p-maninophenol, N-benzyl-p-aminophenol, p-aminosalicylic acid, anthraquinone, alizarin, quinizarin, Na salt of m-nitrobenzenesulfonic acid, picric acid, pytidine, N2H4 hydrate, CuSO4, FeCl3, CrCl3, MnSO4, Pb acetate, Co(NO3)2, NiCl3, and Na

tartrate. ACCESSION NUMBER: 1957:45235 CAPLUS DOCUMENT NUMBER: 51:45235
ORIGINAL REFERENCE NO.: 51:8422g-i

TITLE: PATENT ASSIGNEE(S): Purification of hydrocarbons Gelsenberg Benzin A.-G.

DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: Patent Unavailable

PATENT NO. KIND DATE APPLICATION NO. DATE 19561212 GB 763625 GB

L28 ANSWER 237 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB The catalytic O hydrogenation at 50-100° and at a high
space velocity of 5000-10,000/hr. can be used for the O elimination,
without catalyst regeneration over long time periods.

The equilibrium conditions of the H-O system insures a practically

removal. The only objection to the method is the contamination of the

with H, a small excess of which is required. Both metallic and oxide catalysts can be used, but the oxide catalysts require a much higher

Catalysts can be used, but the object catalysts required in man, again temperature

(300-400°) at the high flow rate. The group VIII catalysts, Pt, Pd, and Ni, can be used, and above 180° their activities differ very little. Ni catalysts are used in gas mixts. containing a large excess of H (H, N + H, etc.), but are not adapted to mixts.

over 0.1% O, or in the inert-gas purification with a H:O proportion close to the stoichiometric, in which case the Pt or Pd catalysts are used.

methods of catalyst preparation are described. The kinetics of the O-hydrogenation reaction is discussed, and the directions are given for the calcn. of required catalyst volume Data are given on the

heating of the catalyst during the reaction and the temperature of the catalyst surface.
ACCESSION NUMBER:

1956:84601 CAPLUS
50:84601
50:15991h-i,15992a
Catalytic purification of gases from admixed oxygen
Boreskov, G. K.; Slin'ko, M. G.
Khim. Prom. (1956) 69-77
Journal DOCUMENT NUMBER: ORIGINAL REFERENCE NO.:

TITLE: AUTHOR(S):

Unavailable

DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 238 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB See U.S. 2,671,763 (C.A. 48, 6620b).
ACCESSION NUMBER: 1954:50635 CAPLUS
COCUMENT NUMBER: 48:50635
ORIGINAL REFERENCE NO.: 48:8990d CALPYET PATENT ASSIGNEE (S): Alied Chemical & Dye Corp.
DOCUMENT TYPE: LANGUAGE: Unavailable
FAMILY ACC. NUM. COUNT: 1

DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. DATE APPLICATION NO. GB 702018 19540106

AB A catalyst comprising Nis associated with a substance of the group consisting of activated Al203 is regenerated by treating the partially spent catalyst with a gas containing O at about 300-700° until al1 carbonaceous deposits on the catalyst have been removed and without any intervening sulfiding step. The oxidized catalyst is treated with a gas containing H at about 200-700° until the catalyst is treated with a gas containing H at about 200-700° until the catalyst is treated with a gas containing H at about 200-700° until the catalyst is treated with a gas containing H at about 200-700° until the catalyst is reated with a gas containing H at about 200-700° until the catalyst has been reactivated. The process is applicable to regenerating the Nis-Al203-type catalysts used in the catalytic hydrogenation of aromatic nitro compds., particularly the vapor-phase catalytic hydrogenation of Procument Number: 48:36975 CAPLUS
DOCUMENT NUMBER: 48:36975 CAPLUS
DOCUMENT NUMBER: 48:36975 CAPLUS
DOCUMENT TYPE: Catalyst regeneration
Winstrom, Leon O.; Harris, Wm. B.
PATENT NO. KIND DATE APPLICATION NO. DATE

PATENT NO. KIND DATE APPLICATION NO. DATE us 2671763 19540309 US

ANSWER 240 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN A catalytic process for desulfurizing heavy petroleum oils is described which reactivation of the spent Ni catalyst is accomplished by oxidation followed by reduction under controlled conditions. S-containing hydrocarbons in the vapor phase are desulfurized by contact

elevated pressure and 820-75°F. with H and a Ni catalyst containing 15-25% Ni deposited on a porous support. Regeneration is accomplished by treatment of the catalyst at 750-80°F. with a mixture of steam and reducing gas (mole ratio 4.5-28:1). The final

mixture of steam and reducing gas (mole ratio 4.5-28:1). The final temperature after reduction is at least 40°F. higher than the initial reduction temperature (820-75°F.). On terminating the flow of steam, the S-containing hydrocarbon vapors are immediately brought into contact with the partially reduced NA catalyst while it is at approx. the final reaction temperature and pressure.

ACCESSION NUMBER: 1953:60227 CAPLUS DOCUMENT NUMBER: 47:6027
ORIGINAL REFERENCE NO.: 47:10215d-f
TITLE: Hydrodesulfurization process
INVENTOR(S): Crawford, Vincent L.
PATENT ASSIGNEE(S): Guif Research & Development Co.
DOCUMENT TYPE: Patent
LANGUAGE: Unavailable
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO. US 2646388

DATE APPLICATION NO. 19530721 US

L28 ANSWER 241 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Highly active hydrogenation or dehydrogenation catalysts are
prepared by dissolving the ineffective components from Raney alloys and
simultaneously precipitating catalytically effective metals, such as Ni,
Co, Cu, or Ag, from appropriate metal solns, by the evolved nascent H.

Metal salt solns, are chosen which will not set free an anion (after the
catalytic metal separation) capable of forming an inorg, acid with water

which
might inactivate or dissolve the catalyst. Therefore complex metal salts
are suitable, especially those with metal components (to be
precipitated) which
form a cation complex with NH3. The reaction may be represented by the
following equations: 2 (Ni-Al) + 6 NaOH + 2 Ni
+ 2 Al (ONa) 3 + 3 H2; 3 H2 + 3 [Cu (NH3) 4] Cl2 + 3 Cu + 6 NH3 + 6
NH4Cl. Fulverulent Al-Mi alloy 100 is suspended in water 1000
at 90°. To this suspension is added a solution of NaOH 240 in water
400 and a solution of Cu chloride 43 and NH4Cl 26 in 12% aqueous NH3 280
parts

parts
which gives after 2 hrs. a highly active Ni-Cu mixture The mixture
may be purified by decanting and washing. The Cu chloride solution may De replaced by a Ni or Co chloride solution or a solution of several metals, such as a mixture of Cu, Ni, Co, and Ag nitrate, with NHANO3.

ACCESSION NUMBER: 1952:52785 CAPLUS
DOCUMENT NUMBER: 46:52785
DOCUMENT NUMBER: 46:52785
DOCUMENT RUMBER: 46:52785
DOCUMENT NUMBER: 46:5278

1952:52785 CAPLUS
46:52785
46:8788h-i,8789a-b
Highly active hydrogenation and
dehydrogenation catalysts
Moldenhauer, Otto; Trautmann, Gunther; Behncke,
Hermann
Phrix-Werke A.-G.
Patent
Unavailable
1

INVENTOR (S):

PATENT ASSIGNEE(S): DOCUMENT TYPE: LANGUAGE:

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. DATE

APPLICATION NO. DATE DE 826134 19511227 DE

L28 ANSWER 242 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN AB A catalyst used in forming CH4 in gases containing CO and H, in the preparation of

which catalyst Ni is precipitated partly as a silicate and partly as an easily reducible basic substance, and which catalyst does not give rise

the formation of silicates to a considerable extent, is regenerated at a temperature of 250-500° and at atmospheric or increased pressure with the H-containing gas obtained in the formation of CH4. In an example a Mi catalyst was prepared by mixing 900 cc. of a boiling solution contains

Na2CO3 with 100 cc. of a solution of Na silicate containing 11.2 g.

SiO2, adding in less than 0.5 min. 1 l. of a boiling solution containing 40 g. Ni, 6 g. MgO, and 2.4 g. ThO2 in the form of nitrates, stirring for 2 min., washing the precipitate for 2 hrs. with 9 l. distilled M2O of 95°,

precipitate for 4 hrs. at 110°, pulverizing and sieving on a sieve with opening of 0.35-mm. diameter, mixing 70 parts by weight of this product with 100

100
parts of Al powder and 1 part of graphite, and pressing the mass into pieces of 5-mm. diameter and 3-mm. length. The catalyst was reduced by

pleces of 5-mm. diameter and 3-mm. length. The catalyst was reduced by Hat

325° for 1 hr. with a velocity of 250 cm./sec. Now 66.4% of the
Ni was in metallic form. Gas mixts. containing H and 2-10% CO were
passed at atmospheric pressure over this catalyst at 210-350° and with
velocities of 1000-3000 l. gas/l. catalyst/hr. After 2000 hrs. the
activity was reduced to 95% of the original one, and a carbonaceous
product was deposited upon the catalyst. This was regenerated with the
gas obtained in the formation of CH4 and containing besides H 0.002

volume % CO
and 2.3 volume % CH4, which gas was passed over the catalyst at 350°
for 24 hrs. with a velocity of 100 1./100 1. catalyst/hr. The catalyst
regained its original activity.

ACCESSION NUMBER: 1952:43752 CAPLUS
DOCUMENT NUMBER: 46:43752

ORIGINAL REFERENCE NO.: 46:7308a-d

Regenerating nickel-containing catalysts

Regenerating nickel-containing catalysts .N. V. de Bataafsche Petroleum Maatschappij Patent Unavailable TITLE: PATENT ASSIGNEE(S):

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE DATE APPLICATION NO. NL 69050 19511215

ANSWER 244 OP 258 CAPLUS COPYRIGHT 2004 ACS on STN

Hydrogenation catalysts consisting of or containing metallic

Ni, Co, Cu, or Fe, which have become poisoned during use in the
reduction of materials containing toxic compds. of S, Se, Te, or P, or
of these elements in a free state, are revivified by treatment with a per
acid of Mo, W, Cr, V, P, or a salt thereof, and washing and drying. In
many cases it is convenient to prepare the. solution containing the per acids or

s or salts by causing H2O2 to react with a suitable acid or salt of the above elements. In an example, a catalyst consisting of Ni supported on klesselguhr had become poisoned so that it possessed about 2% of its original activity, by use in hydrogenation of an impure phenol containing catalytically toxic S compds. The catalyst was washed to

ve adherent phenol, and was then suspended in cold water and revivified by addition of Na2Mol0031 (I) in dilute aqueous solution The I was made by

addition of NaZMo10031 (1) in Galute aqueous occasions allowing NaZMo04, in amount equal to about 1% by weight of the poisoned Ni, to react with an excess of H202. The aqueous suspension of catalyst in the reagent was then heated to 100° in order to destroy the excess of I. The catalyst was then separated, washed, and dried at 100°. After reduction of the revivified catalyst at 300-320°, the original activity was found to have been restored.

ACCESSION NUMBER: 155114652 CAPLUS
DOCUMENT NUMBER: 45:14652

ACCESSION NUMBER:
DOCUMENT NUMBER:
45:14652
ORIGINAL REPERENCE NO:
45:2608b-e
Revivification of catalysts
Maxted, Edward B.
DOCUMENT TYPE:
Patent
Unavailable
1

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE DATE GB 644239 19501004

L28 ANSWER 243 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AS pent Fischer-Tropsch synthesis catalyst is used in such a manner that
in one step it furnishes C to the process for the manufacture of CO and H
synthesis gas, and in a 2nd step furnishes O to the natural-gas oxidation
for producing synthesis gas. Catalysts adaptable to the process contain
Fe, CO, or Ni.
ACCESSION NUMBER: 1952:43751 CAPLUS
DOCUMENT NUMBER: 46:43751

DOCUMENT NUMBER: ORIGINAL REFERENCE NO.:

1952:43751 CAPLUS 46:43751 46:7307i,7308a Fischer-Tropsch catalyst Mayland, Bertrand J. Phillips Petroleum Co. TITLE: INVENTOR(S):

PATENT ASSIGNEE(S): DOCUMENT TYPE: Patent Unavailable

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. APPLICATION NO. DATE KIND DATE US 2598186 19520527

L28 ANSWER 245 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB A catalyst which is used for the conversion of CO and H2 to form
hydrocarbons is decarbonized by bringing it into contact with a metal
oxide which is reducible by C at a temperature below the fusion point of

catalyst, and which does not adversely affect the conversion, at a

temperature
which is conducive to decarbonization of the catalyst, with formation of
CO or CO2. Suitable oxides are those of Fe, Ni, Mn, Ti, Y, Co,
Cr, Mo, Cu, and their mixts. When using Fe203 as the metal oxide for
reducing the C content of an Fe catalyst from about 351 by weight to
about 51

t be by weight, about 2 lb. Fe203 will be required per lb. of catalyst to be decarbonized. The catalyst to be decarbonized in contact with the metal oxide should contain about 5-20% by weight of the catalytically active

metal component in the form of the oxide of that component. Decarbonization temps. suitable for the treatment of Fe catalysts with Fe2O3 are in the range 649-760°.

ACCESSION NUMBER: 1950:34491 CAPLUS

DOCUMENT NUMBER: ORIGINAL REFERENCE NO.: 44:34491 44:6605b-d

44:06U3D-d Regeneration of catalysts Krebs, Robert W. Standard Oil Development Co. Patent Unavailable TITLE: INVENTOR(S):

INVENTOR(S):
PATENT ASSIGNE(S):
DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE US 2506302 19500502 US

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L28 ANSWER 246 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Catalytically cracked gasolines were obvious sources of aromatic compds.
for improvement of rich-mixture ratings of avaitation gasolines. High-S
and
             olefin content of many of these stocks made them unsuitable without
 addnl
             treatment. Conventional refining with H2SO4 was unsuitable because of excessive losses. A process for the controlled hydrogenation of these stocks with a com. W-Mi sulfide catalyst, was developed and used in two com. plants, greatly augmenting supplies of aviation gasoline. Pilot-plant data are given graphically on the effect of: pressure on S retention and catalyst life, space velocity and maximum catalyst temperature on S retention, S content of feed on Br number of
             uct, and b.p. range of feed on relative S and olefin retentions. The pilot-plant reactor was a 12-ft. by 4.23-in. I.D. tube of 46% Cr steel, with elec. heaters to maintain adiabatic conditions. Upflow was used through the 6-ft. catalyst bed containing 0.57 cu. ft. of 0.25-in. catalyst pellets. Catalyst had a bulk density of 130 lb./cu. ft. and contained 40% W and
25%
             Ni. Typical operating conditions were: pressure, 720 lb./sq. in.; mol. ratio H to feed, 7:1; liquid hourly space velocity, 10; liquid feed rate, 42 gal./hr.; temperature, 650°F. Liquid yield was over 100 volume
             because of decrease in liquid density. Increase in total pressure
reduces
             ses

S retention and increases catalyst life. At a given space velocity
conversion is a function of both catalyst age and temperature, but in
             practice a moderate figure is preferrred, 5 to 15. With a fresh
catalyst,
control of extent of reaction requires a low temperature or high space
velocity,
the former being more feasible. With decline in catalyst activity the
temperature is raised. Practical temperature range is 450-800 °F. At
               r
temps. decline in catalyst activity is rapid. Maximum catalyst
temperature gives
the best correlation with extent of reaction, and temperature rise in the adiabatic reactor must be considered. This is about 4°F. for each mol. % olefins hydrogenated. If the reaction proceeds to hydrogenation of aromatic compds., temperature rise will be much greater. Variations of mol. ratio of H to feed above 2:1 do not affect S retention, but maintenance of catalyst activity requires recycle of at least 6000 cu. ft. of gas containing not less than 80% H per barrel of feed.
 feed.
              H2S is removed from recycle gas by scrubbing with NaOH solution, and if
              feed is reasonably pure very little build-up of inert gas occurs. At constant operating conditions S retention is independent of S content of feed. S compds. are hydrogenated more readily than olefins, and olefins more readily than aromatic compds. When the Br number of the
product
             act is not below 3, loss of aromatic compds. is negligible, and at the usual level of 5-10 diolefins are completely hydrogenated. With increase in average b.p. of feed, increased temperature is required to
maintain the
             tain the same S retention. Overheating or failure of H recirculation may cause rapid loss of catalyst activity, but this loss is usually gradual. The catalyst can be reactivated by recirculation of H without charge, but
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ANSWER 247 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Hydrocarbons of more than one C atom are synthesized from CO and H by a catalytic process at a temperature of about 180°. A high-activity, powdered, solid hydrogenation catalyst consisting of NM, Co, Fe, and their oxides is employed. This is suspended in the gases,

an active metal carbide layer forms on the catalyst. Following a period of synthesis reaction with consequent reduction in catalytic activity to an efficiency of about 30% normal activity, the deactivated catalyst is introduced into direct contact with a stream of H gas at a temperature of about 50° for a period of 0.5 to 5.0 hrs. to regenerate the activity of the catalyst without destruction of the carbide layer on the surface of the catalyst without destruction of the carbide layer on

zone. ACCESSION NUMBER: DOCUMENT NUMBER: ORIGINAL REFERENCE NO.:

1949:52757 CAPLUS 43:52757 43:9420h-i

43:942Dh-i Reactivation of catalysts in hydrocarbon synthesis d'Ouville, Edmond L. Koppers Co., Inc. Patent Unavailable INVENTOR (S): PATENT ASSIGNEE(S): DOCUMENT TYPE:

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

APPLICATION NO. DATE PATENT NO. DATE KIND US 2748899 19490816

L28 ANSWER 246 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN (Continued) after 2 or 3 such treatments (15,000-25,000 vols. feed/vol. catalyst) reguneration with steam and air, followed by resulfiding, is required. Complete data are given on treatment of 4 catalytic gasolines, one aromatic distillate, and a high-S straight-run 1950:11245 CAPLUS
44:11245
44:2216a-h
Hydrodesulfurization of gasoline fractions with
tungsten-nickel sulfide catalyst
Cole, Robert M.; Davidson, D. D.
Shell Development Co., Wilmington, CA
Journal of Industrial and Engineering Chemistry
(Washington, D. C.) (1949), 41, 2711-15
CODEN: JIECAD: ISSN: 0095-9014
Unavailable gasoline. ACCESSION NUMBER: DOCUMENT NUMBER: ORIGINAL REFERENCE NO.: TITLE: AUTHOR(S): CORPORATE SOURCE: SOURCE: DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 248 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB When the activity of a catalyst has been reduced by the formation of waxy deposits, regeneration is periodically effected by increasing the temperature so that gaseous hydrocarbons are formed and the waxy deposits on the temperature so that gaseous nyarcoarbons are formed and the waxy sits on the catalyst are removed. E.g., a catalyst containing Co 100, ThO2 18, and kieselguhr 100 parts was used in hydrocarbon synthesis 2 weeks at 246° and 100 lb./sq. in. pressure at an hourly space velocity of 100. At this point a decline in the yield of liquid hydrocarbons was noted due to waxy deposits on the catalyst. The temperature was allowed

use to 308° by reducing the cooling until the effluent hydrocarbons were primarily CH4 and C2H6 and continuing operation at this temperature Were primarity constructions where the process is also applicable to Fe and Ni catalysts.

ACCESSION NUMBER: 1949:52755 CAPLUS

ORIGINAL REFERENCE NO.: 43:9420d-f

Paceperation of catalysts in the production of

DOCUMENT NUMBER: ORIGINAL REFERENCE NO.: TITLE:

43:9420d-t Regeneration of catalysts in the production of hydrocarbons from carbon monoxide and hydrogen Clark, Alfred Phillips Petroleum Co. Patent Unavailable

INVENTOR(S):
PATENT ASSIGNEE(S):
DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:

APPLICATION NO. DATE PATENT NO. DATE US 2479999 19490823 US

Page 78

ANSWER 249 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Metals, such as Mn, U, Ce, Mg, Ni, and Fe deposited in kiesclguhr, are regenerated after use in the catalytic preparation of hydrocarbons by heating in a stream of H2 at 350-450°; the speed of the gas being 500-1000 cu. m. per hr. for each sq. m. of the passage and the gas having less than 2.5 g. of CO and 1 g. H2O per cu. m. The the gas having less than 2.5 g, or CO and 1 g, ..... First control of the gas having less than 2.5 g, or CO and 1 g, .... First control of the gas having less than 2.5 g, or CO and 1 g, .... First control of the gas having and carbon that seems are gas having a control of the gas having and carbon monoxide of the gas having a control of the gas having and carbon monoxide of the gas having a control of the gas having and carbon monoxide of the gas having a control of the gas

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. DATE APPLICATION NO. DATE

FR 861745 19410215 FR L28 ANSWER 250 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Full regeneration is possible in 2 phases, the lst with H2 at 350°
or more to remove high-mol. compds., the 2nd at about 450° till
activity is restored. The method is meant for CO catalyst, also for
Ni or Fe, possibly activated with Th, Mn, U, Ce, or Mg oxide.
ACCESSION NUMBER: 41:19681 CAPLUS
ORIGINAL REFERENCE NO.: 41:39430-d
REGULATION ACCESSION OF ACCESSION

DOCUMENT NUMBER: ORIGINAL REFERENCE NO.: TITLE:

41:3943G-C Regenerating catalyzers from the conversion of carbon monoxide and hydrogen to hydrocarbons N. V. Internationale Koolwaterstoffen Synthese Maatschappij Patent Unavailable PATENT ASSIGNEE (S):

DOCUMENT TYPE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. APPLICATION NO. DATE DATE NL 53298

19421015

L28 ANSWER 251 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Spent hydrogenation catalysts of Ni-Al, Ni
-5i type are regenerated by heating in contact with material in which the
catalyst was previously employed at a temperature of 600-700°F. The
decomposition products are then removed by blowing with an inert gas,
preferably superheated steam, at the same temperature, or by the use of

solvent. The catalyst is then subjected to an activation process, such

as is customarily applied to a freshly prepared catalyst.

ACCESSION NUMBER: 1946:5808 CAPLUS

DOCUMENT NUMBER: 40:5808

ORIGINAL REFERENCE NO: 40:1000g-h

TITLE: Regeneration of nickel alloy cataly

INVENTOR(S): Drew, John

40:1000g-h
Regeneration of nickel alloy catalysts
Drew, John
Hercules Powder Co. PATENT ASSIGNEE(S): DOCUMENT TYPE:

Patent Unavailable

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

APPLICATION NO. DATE PATENT NO. DATE KIND US 2388959 19451113 US

ANSWER 252 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN A process for mfg. a pelleted composite catalyst consisting essentially

a relatively inert carrier, such as kieselguhr, supporting a minor amount of

ne or a catalytically active material, such as reduced Ni, involves mixing the finely divided carrier with a minor amount of a metal soap of

a normally solid organic acid, such as Mi stearate, forming pellets from the mixture and adding a solution of a metal salt, such as Ni carbonate, to form an active catalyst on the pellets, and subsequently calcining in an oxidizing atmospheric and reducing by H.

ACCESSION NUMBER: 1941:3557 CAPLUS

DOCUMENT NUMBER: 35:3557

ORIGINAL REFERENCE NO.: 35:3588d-e

Catalysts suitable for use in hydrogenation, etc.

etc.
Ipatieff, Vladimir N.: Corson, Ben B.
Universal Oil Products Co.
Patent
Unavailable

INVENTOR(S):
PATENT ASSIGNEE(S):
DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE US 2211208 19400813

ANSWER 253 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN Spent catalytic metal is oxidized and the resulting oxide is mixed with a reducing metal (such as Al in the case of Ni oxide) in sufficient amount to reduce the oxide and with an excess to alloy with THE REFERENCE NO.: 33:2099-7

TITLE:

At in the case of Ni oxide) in the case of Ni oxide in the case of Ni oxide) in the case oxide and with an excess to allow with a selective oxide and with an excess to allow with a selective oxide and visit and case oxide and visit and visit

1939:14120 CAPLUS
33:14120
33:2099f-g
Recovering catalytic metals such as nickel
after use
Raney, Murray
Patent
Unavailable
1

INVENTOR(S):
DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:

PATENT NO. DATE APPLICATION NO. DATE US 2139602 19381206

L28 ANSWER 255 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB cf. Troyanovskii, C. A. 29, 8249.8. For the regeneration of Ni
catalyst precipitated on clay, 350-400 kg. of the spent catalyst is
heated, With
stirring, for 30 40 min. with a sufficient amount of 8-10\*B.acte.e.
NAOR to saponify 1/2 of the fat, and then with 30\*B.acte.e. NAOH to
saponify 20-40% of the remaining fat at 90-5\* for 1.5-2 hrs. The
mixture is then treated with 350-400 l. of 5\*B.acte.e. NAOI (containing
1.4% active Cl) and 350-400 l. H20 at 60\* for 1 hr. After dilution
with 2.5-4 vols. of H20, the mixture is brought to a boil and allowed to
settle for 3-4 hrs. The supernatant soap solution and fat are siphoned
off,

the catalyst stirred with 4-5 vols. of hot  $\rm H2O~(80-5^{\circ})$ , the mixture neutralized with  $\rm H2SO4$ , and the catalyst, after filtering and washing to

a neutral reaction, is dried and reactivated in H2 at 450\*.

Nydrogenation of sunflower oil with the regenerated catalyst gave fat mixts., m. 50-55\*. The recovery is effected with a loss of 6-610 Ni, as compared with 40-50% by the ordinary method.

ACCESSION NUMBER: 1936:29952 CAPELUS

ORIGINAL REFERENCE NO: 30:3950h.; 3951a

TITLE: Repensation of spent catalysts

AUTHOR(S): Etinburg, E.; Vengerova, N.; Gol\*dshtein, G.

SOURCE: Masloboino-Zhirovce Delo (1935), 11, 574-5

CODEN: MZHDAD; ISSN: 0369-304X

DOCUMENT TYPE: Journal LANGUAGE: Unavailable

L28 ANSWER 254 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN AB In the recovery of spent Ni formate catalyst with the aid of sulfonic acids, a good emulsification of fat and rapid dissolving of Ni in H2SO4 and separation of oil (2-5 min. in laboratory) are effected.

mixture of 15% excess of H2SO4, 20° B. acte.e., 3% sulfonic acids (based on the spent catalyst) and 100 g. of catalyst (containing 11% Mi) was heated on a water bath with stirring, giving a solution with 10.2 g. Ni. The Ni was recovered from the solution by the usual methods and used in hydrogenation of oils, with good results.

ACCESSION NUMBER: 1936:40417 CAPLUS DOCUMENT NUMBER: 30:40417
ORIGINAL REFERENCE NO.: 30:5371e-g
RECOVERY of nickel catalyst with

30:40417
30:5371e-g
Recovery of nickel catalyst with sulfuric acid, with the addition of sulfonic acids Kazakova, L.
Masioboino-Zhirovoe Delo (1936), 12, 145-6
CODEN: MZHDAD; ISSN: 0369-304X

AUTHOR (S): SOURCE:

DOCUMENT TYPE: LANGUAGE:

L28 ANSWER 256 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Reduced Ni catalysts were made from Ni(OH) 2,
 Ni(OH) 2 after oxidation with Br2, and NiCO3, in suspension in oil.
 Hydrogenation tests showed that NiCO3 yields even more potent
 catalysts than does Ni formate. The results have an important
 bearing on the regeneration of spent catalysts in oil hardening.

ACCESSION NUMBER: 1935:63848 CAPLUS

DOCUMENT NUMBER: 29:63848

ORIGINAL REFERENCE NO.: 29:63748

Regenerating simple nickel compounds
 directly in oil

AUTHOR(S): Zinov'ev, A.; Vinogradova, M.; Popova, O.

SOURCE: Trudui VMIII2h (1934), (No. 3), 24-9

DOCUMENT TYPE: Journal AUTHOR(S): SOURCE: DOCUMENT TYPE: LANGUAGE: Journal Unavailable

L28 ANSWER 257 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN
AB Waste Ni catalyst used for the hardening of oils is dissolved in
H2604 containing HN03. To the solution NoOH or Na2CO3 is added in such
amount that
Ni is not precipitated but most of the Fe and other impurities are
precipitated
After separation from the precipitate, the solution is heated with a
small amount of an
oxidizing agent (such as KMm04 or Na2CO2), by which the remaining Fe is
precipitated From the filtrate, Ni is recovered as hydroxide with
caustic alkali.
ACCESSION NUMBER: 1934:21525 CAPLUS
DOCUMENT NUMBER: 28:21525
ORIGINAL REFERENCE NO.: 28:2559f—h
TITLE: Recovery of waste nickel catalyst
INVENTOR(S): Kimura, Goro
FATENT ASSIGMEE(S): HOKAI Yushi Kogyo K. K.
DOCUMENT TYPE: LANGUAGE: Unavailable
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 99594 19330216 JP

L28 ANSWER 258 OF 258 CAPLUS COPYRIGHT 2004 ACS on STN

AB Bydrogenation-promoting metals such as Ni, Cu and Co
are separated in finely divided form from aqueous solns. of their salts

by the use
Of Fe, Zn and Al (which may be effected in the presence of a non-metallic
carrier such as kieselguhr, silica gel or charcoal) and the catalysts

thus

prepared may be used in the hydrogenation of aromatic hydrocarbons
or their derivs. such as toluene or phenol, in the hydrogenation
of unsatd. compds. such as cinnamic acid or fats in the production of
alcs. from aldehydes and ketones, conversion of quinones to
hydroquinones,
and in reduction of nitriles, nitro, azoxy, azo and hydrazo
compds. to the corresponding amines (these reactions may be effected in
the liquid phase and at comparatively low temps.). If the solution from
which the catalyst is prepared is alkaline the oxide of the more
electropositive
element is deposited and serves as an activator. The catalyst may be
dried in vacuo or in a current of H before use.

ACCESSION NUMBER: 1928:29848 CAPLUS
DOCUMENT NUMBER: 22:29848
ORIGINAL REFERENCE NO. 22:3500b-d
Catalysts
PATENT ASSIGNEE(S): I. G. Farbenindustrie AG
DOCUMENT TYPE: Patent
LANGUAGE: Unavailable
PAMILY ACC. NUM. COUNT: 1
PATENT NOR AND A COUNT: 1
PATENT NOR AND A COUNT: 1

PATENT NO. KIND DATE APPLICATION NO. DATE

GB 281218 19261127 GB

## => d his

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FILE 'CAPLUS' ENTERED AT 16:17:17 ON 14 SEP 2004
        1207996 S CATALY?
L1
L2
        1087400 S REDUCT?
L3
        256764 S HYDROGENAT?
        1299917 S L2 OR L3
L4
L5
        224466 S L1 AND L4
         775627 S NI OR NICKEL
L6
L7
        1091254 S COPPER OR CU
L8
        513944 S CHROMIUM OR CR
L9
          40533 S L5 AND L6
L10
         162143 S REGEN?
L11
           1162 S L9 AND L10
L12
            239 S L11 AND L7
     FILE 'REGISTRY' ENTERED AT 16:21:50 ON 14 SEP 2004
L13
                STRUCTURE UPLOADED
L14
              1 S L13
               STRUCTURE UPLOADED
L15
L16
              1 S L15
               STRUCTURE UPLOADED
L17
L18
              7 S L17
               STRUCTURE UPLOADED
L19
L20
              7 S L19
L21
          1647 S L19 FULL
     FILE 'CAPLUS' ENTERED AT 16:24:50 ON 14 SEP 2004
L22
           3471 S L21
L23
             90 S L22 AND L5
             40 S L23 AND L6
L24
              5 S L24 AND L10
L25
L26
           4927 S (CATALYST REGENERATION)
L27
           1176 S L26 AND L4
L28
           258 S L27 AND L6
=> s amine
        247822 AMINE
        235832 AMINES
L29
        380622 AMINE
                 (AMINE OR AMINES)
=> s ?amine
SYSTEM LIMITS EXCEEDED - SEARCH ENDED
The search profile you entered was too complex or gave too many
answers. Simplify or subdivide the query and try again. If you have
exceeded the answer limit, enter DELETE HISTORY at an arrow prompt
(=>) to remove all previous answers sets and begin at L1. Use the
SAVE command to store any important profiles or answer sets before
using DELETE HISTORY.
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(FILE 'HOME' ENTERED AT 16:17:05 ON 14 SEP 2004)

L30

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11 L28 AND L29

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L30 ANSWER 1 OF 11 CAPLUS COPYRIGHT 2004 ACS on STN
             Xylylenediamines are prepared by hydrogenating dicyanobenzenes in a liquid phase in the presence of a catalyst, where the catalyst having
 its
             activity decreased during the course of the hydrogenation of the
             activity decreased during the course of the hydrogenation of the dicyanobenzenes is contacted with a hydrogen-containing gas at 200-500° while controlling a temperature rise speed of the catalyst to $400°/min, thus regenerating the catalyst which can then be reused in subsequent hydrogenations of the dicyanobenzenes.

SSION NUMBER: 2004:162518 CAPLUS WERN NUMBER: 100:210329
 ACCESSION NUMBER:
 DOCUMENT NUMBER:
TITLE:
                                                                 140:219730
                                                                Hydrogenation process and catalyst for the production of xylylenediamines from dicyanobenzenes and a regeneration process for the catalyst Amakawa, Kazuhiko
 INVENTOR (S):
PATENT ASSIGNEE (S):
SOURCE:
                                                                 Japan
U.S. Pat. Appl. Publ., 7 pp.
CODEN: USXXCO
 DOCUMENT TYPE:
                                                                English
 FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
             PATENT NO.
                                                                 KIND
                                                                                                                 APPLICATION NO.
                                                                                                                                                                             DATE
                                                                              DATE
US 2004039232 A1 20040226 US 2003-643981 20030820
JP 2004107327 A2 20040408 JP 2003-292139 20030812
EP 1394146 A1 20040303 EP 2003-18532 20030816
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LIJ, NI, SE, MC, PT,
IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK
PRIORITY APPLN. INFO:: JP 2002-245222 A 20020826
OTHER SOURCE(S):
                                                                CASREACT 140:219730
 L30 ANSWER 3 OF 11 CAPLUS COPYRIGHT 2004 ACS on STN
             ASSMENT OF ILEMPTON COPINIONI 2004 ALS ON SIN
There is provided a process for hydrocarbon synthesis wherein a supported
metal catalyst for hydrogenating carbon monoxide to form a mixture
of hydrocarbons is regenerated by decreasing its hydrocarbon content,
impregnating under a non-oxidative atmospheric with a solution of at
             member of the group of ammonium salts (e.g., ammonium acetate), alkylammonium salts, and weak organic acids, optionally including
            nia, oxidizing with a gaseous oxidant (e.g., air) in the presence of the impregnating solution, activating the catalyst by reduction with hydrogen at elevated temps., and reusing the catalyst. The treatment may be carried out in a single reactor, or by carring out all of the steps after catalyst has been withdrawn therefrom and returned to at least one react. Up to all steps subsequent to decreasing the hydrocarbon ent.
content
may be affected in a subsequent reactor, or in specialized apparatus
ACCESSION NUMBER: 2003:610393 CAPLUS
TITLE: CAPLUS
DECLINEARY NUMBER: 139:151409
```

rrocess for Fischer-Tropsch Catalyac regeneration in the manufacture of ClO+ hydrocarbons from synthesis gas Koveal, Russell John; Daage, Michel; Shen, Eric

Baochun ExxonMobil Research and Engineering Company, USA PCT Int. Appl., 50 pp. CODEN: PIXXD2

APPLICATION NO.

THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

DATE

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L30 ANSWER 4 OF 11 CAPLUS COPYRIGHT 2004 ACS on STN

AB The invention relates to a method for regenerating a catalyst such as a
plate or honeycomb catalyst that is at least partially deactivated as a
result of toxic substances. The catalyst, e.g., a selective catalytic
reduction catalyst for nitrogen oxides removal from exhaust gases, is
treated with a gaseous reducing agent and a polyfunctional complex
forming agent to eliminate said toxic substances. ACCESSION NUMBER: 2000:68390 CAPLUS DOCUMENT NUMBER: 132:82747
                                                   Method for regenerating a deactivated catalyst
 TITLE:
INVENTOR(S):
                                                   Neufert, Ronald
Siemens Aktiengesellschaft, Germany
PCT Int. Appl., 11 pp.
CODEN: PIXXD2
 PATENT ASSIGNEE(S):
SOURCE:
 DOCUMENT TYPE:
                                                   German
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
          PATENT NO.
                                                   KIND DATE
                                                                                          APPLICATION NO.
                                                                                                                                         DATE
           WO 2000003804
WO 2000003804
                                                    A2
A3
                                                                20000127
                                                                                                                                         19990705
                                                                                          WO 1999-DE2067
          ... 20000038U4 A3 20000420 19990705
W: JP, US
RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
PR, SE
DE 19832057 C1 20000316 DE 1998-19922067
                                                    C1
A2
                                                                                         DE 1998-19832057
EP 1999-945907
          EP 1098704
EP 1098704
                                                     A2
B1
                                                                 20010516
20030423
          R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, FI
JP 2002520153 T2 20020709 JP 2000-559935 19990705
                                                                20030515
20010607
20030722
          AT 238100
US 2001003116
                                                     E
Al
B2
                                                                                         AT 1999-945907
US 2001-761811
                                                                                                                                         19990705
20010116
US 6596661
PRIORITY APPLN. INFO.:
                                                                                          DE 1998-19832057 A 19980716
                                                                                          WO 1999-DE2067
                                                                                                                                  W 19990705
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L30 ANSWER 2 OF 11 CAPLUS COPYRIGHT 2004 ACS on STN

AB A method for regenerating spent supported metal catalysts comprising treating the spent catalyst with an organo-metallic complex forming agent having an ionization constant pK 1 of at least 2.5. The catalyst

is restored to an activity level near to or greater than the fresh catalyst. The regeneration method is particularly useful for

apent palladium catalysts on an alumina support as utilized for the hydrogenation of Et anthraquinone (EAQ) in the production of hydrogen peroxide.

ACCESSION NUMBER: 2003:912824 CAPLUS

140:3216
Regeneration of spent supported metal catalysts
Zhou, Bing; Rueter, Michael
USA
U.S. Pat. Appl. Publ., 8 pp., Cont.-in-part of U.S.
Ser. No. 745,510.
CODEN: USXXCO

APPLICATION NO.

US 2002-326042 US 2000-745510

US 2002-326042

DATE

A 20021220

20021220

2003:912824 CAPLUS 140:9216

Patent English 2

A1

KIND DATE

20031120

US 2002115554 A1 20020822 US 2000-745510 20001222
US 6740615 B2 20040525
W: AT, CA, CN, IN, JP, MX
RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
IT, LU, MC, NL, PT, RO, SE, SI, SK, TR
PRIORITY APPLN. INFO.: US 2000-745510 A2 20001222

regenerating

DOCUMENT NUMBER

PATENT ASSIGNEE(S): SOURCE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

US 2003216245 US 2002115554

PATENT NO.

TITLE: INVENTOR(S):

DOCUMENT TYPE:

REFERENCE COUNT: FORMAT

INVENTOR (S):

PATENT ASSIGNEE(5): SOURCE: DOCUMENT TYPE:

PATENT NO.

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

English I

KIND DATE

PATENT NO. KIND DATE APPLICATION NO. DATE

WO 2003064356 Al 20030807 W0 2003-US202 20030103

W: AR, AG, AL, AM, AT, AU, AZ, RA, BB, BG, BR, BV, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, NA, MD, MG, MK, NH, MW, MX, MZ, NO, NZ, CM, PH, PL, PT, RO, RU, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ, UL, UC, UZ, VN, YU, ZA, ZM, ZW, AN, AZ, BY, KG, KZ, MD, RU, TJ, TM RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, CC, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, ML, MT, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GO, GW, ML, PT, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GO, GW, ML, PT, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GO, GW, MS, 2003166451 Al 2003004 US 2002-59916 20020129

US 2003166451 Al 20030004 B2 2002-59916 A 20020129

on STN

convenation of nitriles to amines

leavenerated by mixing with aqueous base of basic ion concentration

10.01 mol/L,

heating the mixture at (130°, and washing the treated catalyst with

water or base such that the final rinse water has a pH of 12-13. The

heating may be done under H and the degree of regeneration is 90-100%.

Examples employing soda or NaOH are given for Raney Ni catalyst

used to hydrogenate adiponitrile to hexamethylenediamine and/or

6-CE-SMION NUMBER: 2000:43909 CAPLUS

DOCUMENT NUMBER: 132:65730

TITLE: Regeneration L30 ANSWER 5 OF 11 CAPLUS COPYRIGHT 2004 ACS on STN
AB Raney catalysts for hydrogenation of nitriles to amines
are regenerated by mixing with aqueous base of basic ion concentration 132:65730
Regeneration of catalyst used to hydrogenate nitriles and its use
Boschat, Vincent: Leconte, Philippe
Rhodia Fiber and Resin Intermediates, Fr.
Fr. Demande, 13 pp.
CODEN: FRXXBI.
Patent INVENTOR(S): PATENT ASSIGNEE(S): SOURCE: DOCUMENT TYPE: Patent French 1 LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: DATE PATENT NO. APPLICATION NO. DATE KIND FR 2773086 19990702 FR 1997-16832

A1 19971229 T1991/1229

CA 2316821

AA 19990708

WO 9933561

A1 19990708

WO 1998-P2316821

19981223

WO 9833561

WI BR, BY, CA, CN, CZ, ID, JP, KR, PL, RO, RU, SG, SK, UA, US, VN

RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,

EP 1042068

A1 20001011

EP 1000 A PT, SE
EP 1042068 A1 20001011
R: 8E, DE, ES, FR, GB, TT, NL
ER 9814561 A 20011016
JP 2001526956 T2 20011225 BR 1998-14561 JP 2000-526298 RU 2000-120191 US 2001-582657 FR 1997-16832 19981223 19981223 19981223 20010209 A 19971229 A T2 C2 B1 RU 2190469 US 6518449 PRIORITY APPLN. INFO.: WO 1998-FR2856 W 19981223

L30 ANSWER 7 OF 11 CAPLUS COPYRIGHT 2004 ACS on STN
AB The title catalysts are regenerated at high temperature in a gas containing O, N

nining O, N compds. in the form of **amines** and/or N oxides, and, optionally, steam. A spent catalyst containing Pd 6, V 9, and Pb 6 g/L alumina,

to reduce PhNO2 to PhNH2, was regenerated at  $<450^{\circ}$  with a gas mixture containing 12.5 g/h 5% aqueous NH3, 4 L/h air, and 8 L/h N (reduced to 4

L/h

during regeneration) until CO2 was not present in the outlet gas. The
regenerated catalyst gave 90% PhNO2 conversion with 80% selectivity to
PhNH2 initially and 80% PhNO2 conversion with 90% selectivity to PhNH2
after 200 h.
ACCESSION NUMBER: 1987:409324 CAPLUS

DOCUMENT NUMBER: TITLE:

INVENTOR (S):

1987:439324 CAPLOS
107:9324
Regeneration of catalysts for gas-phase reduction of aromatic nitro compounds
Stammann, Guenter: Kricsfalussy, Zoltan; Waldmann, Hclmut; Schneider, Joachim: Medem, Harald
Bayer A.-G., Fed. Rep. Ger.
Ger. Offen, 6 pp.
CODEN: GWXXBX
Patent PATENT ASSIGNEE (S):

DOCUMENT TYPE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

	PAT	ATENT NO.				KIND		DATE	AP	PLICATION NO.	DATE
							-				
	DE	35308	320			A1		19870305	DE	1985-3530820	19850829
	DE	35308	320			C2		19880721			
	US	47146	589			A		19871222	US	1986-894581	19860807
	EΡ	21260	)2			A1		19870304	EP	1986-111377	19860818
	E₽	21260	02			В1		19880921			
		R:	ΒE,	DE,	FR,	GB,	ΙŢ				
	J₽	62053	3745			A2		19870309	JP	1986-194108	19860821
	BR	86043	109			A		19870422	BR	1986-4109	19860828
PRIOR	ITY	APPI	N.	INFO	. :				DE	1985-3530820	19850829

L30 ANSWER 6 OF 11 CAPLUS COPYRIGHT 2004 ACS on STN
AR anny mickel catalyst for reduction of
tris(hydroxymethyl) nitromethane (1) by H2 is regenerated using ammonium
formate, or an aliphatic amine formate (produced in situ from
stoichiometric amits. of HCO2H and an amine base). Thus, a MeOH
solution of Raney nichel was saturated with H2, and mixed with I.
After 2.5 h reduction, the postreaction liqs. and the catalyst were
separated The catalyst was washed, then mixed with MeOH. This solution

saturated with H2, added with Et3N and 30% aqueous HCO2H (in the a stoichiometric with respect to Et3N), and with MeOH solution of I.

above

process was repeated 5 times; Raney nickel showed no
reduction of its catalytic activity. The yield of the product
tris(hydroxymethyl) aminomethane produced from the combined postreaction
liqs. was 83.7% (of the theor. value).
ACCESSION NUMBER: 1991:130228 CAPLUS
DOCUMENT NUMBER: 119:130228
TITLE: Method of regenerating Raney nickel catalyst
for reduction of tris(hydroxymethyl) nitromethane
INVENTOR(S): Mazurkiewicz, Roman; Szeja, Wieslaw; Kiersznicki,
Tadeusz

Tadeusz Politechnika Slaska, Pol. Pol., 2 pp. CODEN: POXXA7 PATENT ASSIGNEE(S):

DOCUMENT TYPE: Patent Polish

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

APPLICATION NO. KIND DATE DATE PL 138692 PRIORITY APPLN. INFO.: В1 PL 1983-245531 PL 1983-245531 19831230 19831230 19861031

OTHER SOURCE(S): CASREACT 114:130228

ANSWER 8 OF 11 CAPLUS COPYRIGHT 2004 ACS on STN

The resistance of Ti alloys to corrosion in the presence of C1-containing amine in a H atmospheric (Treseder and Miller, CA 45, 6985f; McQuillan, CA 45, 6033C) was exptl. investigated. A series of tests was carried out under the conditions of catalytic reduction of nitro-chlorobenzene to chloroaniline (USSR 166,037, CA 62, 10366C) at a H pressure of 200 kg./cm2, \$200°, and C1- content in the catalyst zone of approx. 3 mol. 8. A sample of 12Kh9MA steel under these conditions was decomposed in 9 hrs., the Cr-Na-W steels were badly corroded after 90 hrs., while high-quality stainless steels lasted 253 hrs.; the resp. corrosion rates were 19.7, 2.2, and 0.15 g./m.2 hr. Four types of Ti material showed no corrosion after 100 hrs. In more prolonged tests over 1700 hrs. of the same reaction at 140° and 200 kg./cm.2 H pressure, practically no mech. changes were observed. Full details of these tests are tabulated. A study of the microstructure of a reactor wall made from VTI-1 Ti after 2 years under the conditions of the nitrochlorobenzene reduction showed the absence of Ti hydrides. At the CI- concentration increased to 202 mol. % the same material was corroded in the area of previously made incisions after 5500 hrs. of exposure. An exptl. reactor was made from a 30KhMA steel, diameter 30 mm., height 1.5 m., having a 3-mm.

thick VT1-1 sleeve, and the thermal zone, catalyst separating sieves,

thick VTI-1 sieve, and the chiant arms.

and other

details made from Ti materials. The reactor was used under the above conditions for >3 years. To investigate the effect of the catalyst regeneration conditions, 3-mm. thick samples of the VTI-1 Ti were exposed to 15-hr. periods at 325-50 and 350-450° to the action of H at 200 kg./cm.2 pressure, followed by a temperature drop to

to the action of H at ZOU KG./CM.2 Pleasure, 1210-00 W, and the Ti 200°. In both cases hydrides of Ti were formed, and the Ti material was disintegrating.

ACCESSION NUMBER: 1966:402648 CAPLUS
DOCUMENT NUMBER: 65:2648

FITTLE: The use of titanium for apparatus for chloroorganic synthesis in a hydrogen atmosphere

AUTHOR(S): Antonovskaya, E. I.; Pozdeeva, A. A.

Khim. Prom. (1966), 42(4), 304-5

DOCUMENT TYPE: Journal

Russian

L30 ANSWER 9 OF 11 CAPLUS COPYRIGHT 2004 ACS ON STN AB See U.S. 2,671,763 (c.A. 48, 6620b). ACCESSION NUMBER: 1954:50635 CAPLUS DOCUMENT NUMBER: 48:50635 CAPLUS ORIGINAL REFERENCE NO.: 48:8990d

TITLE: PATENT ASSIGNEE(S): Catalyst regeneration
Allied Chemical & Dye Corp.

DOCUMENT TYPE: Patent
LANGUAGE: Unavailable
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

APPLICATION NO. DATE PATENT NO. KIND DATE 19540106 GB GB 702018

ANSWER 11 OF 11 CAPLUS COPYRIGHT 2004 ACS on STN
Hydrogenation-promoting metals such as Ni, Cu and Co
are separated in finely divided form from aqueous solns, of their salts ne use
of Fe, 2n and Al (which may be effected in the presence of a non-metallic
carrier such as kieselguhr, silica gel or charcoal) and the catalysts

thus

thus

prepared may be used in the hydrogenation of aromatic hydrocarbons or their derivs. such as toluene or phenol, in the hydrogenation of unsatd. compds. such as cinnamic acid or fats in the production of alcs. from aldehydes and ketones, conversion of quinones to hydroquinones,
and in reduction of nitriles, nitro, azoxy, azo and hydrazo compds. to the corresponding maines (these reactions may be effected in the liquid phase and at comparatively low temps.). If the solution from which the catalyst is prepared is alkaline the oxide of the more

solution from which the catalyst is prepared is alealed the more electropositive element is deposited and serves as an activator. The catalyst may be dried in vacuo or in a current of H before use.

ACCESSION NUMBER: 1928:29948 CAPLUS
DOCUMENT NUMBER: 22:29848
CAPLUS
CATALUS

DOCUMENT TRIPS: Catalysts
PATENT ASSIGNEE(S): I. G. Farbenindustrie AG
PATENT ASSIGNEE (S): Unavailable
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION: DATE

PATENT NO. KIND DATE APPLICATION NO. DATE GB 281218 19261127 GB

AB A catalyst comprising Nis associated with a substance of the group consisting of activated Al2O3 is regenerated by treating the partially spent catalyst with a gas containing O at about 300-700° until all carbonaceous deposits on the catalyst have been removed and without any intervening sulfiding step. The oxidized catalyst is treated with a gas containing H at about 200-700° until the catalyst has been reactivated. The process is applicable to regenerating the Nis-Al2O3-type catalysts used in the catalytic hydrogenation of sromatic nitro compds., particularly the vapor-phase catalytic hydrogenation of POCUMENT NUMBER: 1954:36975 CAPLUS DOCUMENT NUMBER: 48:36975 CAPLUS DOCUMENT NUMBER: 48:36975 CAPLUS DOCUMENT NUMBER: 48:36975 CAPLUS DOCUMENT NUMBER: 48:36975 CAPLUS DOCUMENT TYPE: Catalyst regeneration Winstrom, Leon O.; Harris, Wm. B. PATENT ASSIGNEE(S): Allied Chemical & Dye Corp. Patent LANGUAGE: Unavailable FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

KIND DATE PATENT NO. APPLICATION NO. DATE US US 2671763 19540309

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